



LAKE ROE PROJECT – KOPAI CRESCENT AND BOMBARA WASTE CHARACTERISATION

Ramelius Resources Ltd
April 2025



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1. INTRODUCTION

1.1 Overview

Landloch Pty Ltd (Landloch) has been engaged by Ramelius Resources Limited (Ramelius) to conduct a waste material characterisation study related to the mining activities of the Lake Roe Gold Project (the Project). This project involves the development of high-grade gold deposits, specifically the Kopai Crescent and Bombora deposits, which are part of a continuous mineralisation within an approximately 9 km long greenfields gold district. Gold extraction at Kopai Crescent is proposed to occur through open-pit mining, while Bombora will be mined using both open-pit and underground methods. The results of this study are expected to inform the submission of the Mining Proposal (MP) and Mine Closure Plan (MCP) for the Project.

1.2 Regulatory context

This study is anticipated to inform development of the MCP in accordance with the Department of Energy, Mines, Industry Regulation and Safety's (DEMIRS) MCP guidance document (DMIRS 2023a) that supports the MCP statutory guidelines (DMIRS 2023b). The MCP guidance document states that:

"Comprehensive characterisation of materials (including soils and mine waste) is critical to effective closure planning and successful progressive rehabilitation. This process should start during the exploration phase and continue throughout the life of the mine. Characterisation of materials allows for separation and selective placement of materials considered beneficial to rehabilitation and segregation of materials that may inhibit rehabilitation or cause detrimental effects".

DEMIRS provides additional information relating to characterisation of wastes in the Draft Guidance for Materials Characterisation Baseline Data Requirements for Mining Proposals (DMP 2016). Broadly, characterisation of wastes should be undertaken to evaluate their physical stability and plant growth potential, and to identify risks associated with materials that are dispersive, asbestiform, radioactive, and that may produce acidic, metalliferous, or saline drainage.

1.3 Scope of work

The scope of this waste characterisation assessment included:

- Review of existing geological information;
- Submission of samples for assessment of geochemical and physical characteristics, presence of Normally Occurring Radioactive Materials (NORMs), absence/presence of fibrous mineral, durability of rocky wastes, and erosion potential;
- Collation of all laboratory data and interpretation of the results using industry-accepted classification schemes; and
- Definition of the suitability of waste materials for rehabilitation, including strategies for their management.

2. ENVIRONMENTAL SETTING

2.1 Project location and setting

The Lake Roe Gold Project is located approximately 100 km east of Kalgoorlie in the Eastern Goldfields region, within the Shire of Kalgoorlie in Western Australia (WA). The Project is situated on the south-western edge of Lake Roe, a large, naturally saline playa. The Project is within tenement M28/388, covering an area of approximately 3,753 hectares (ha) (Figure 1).

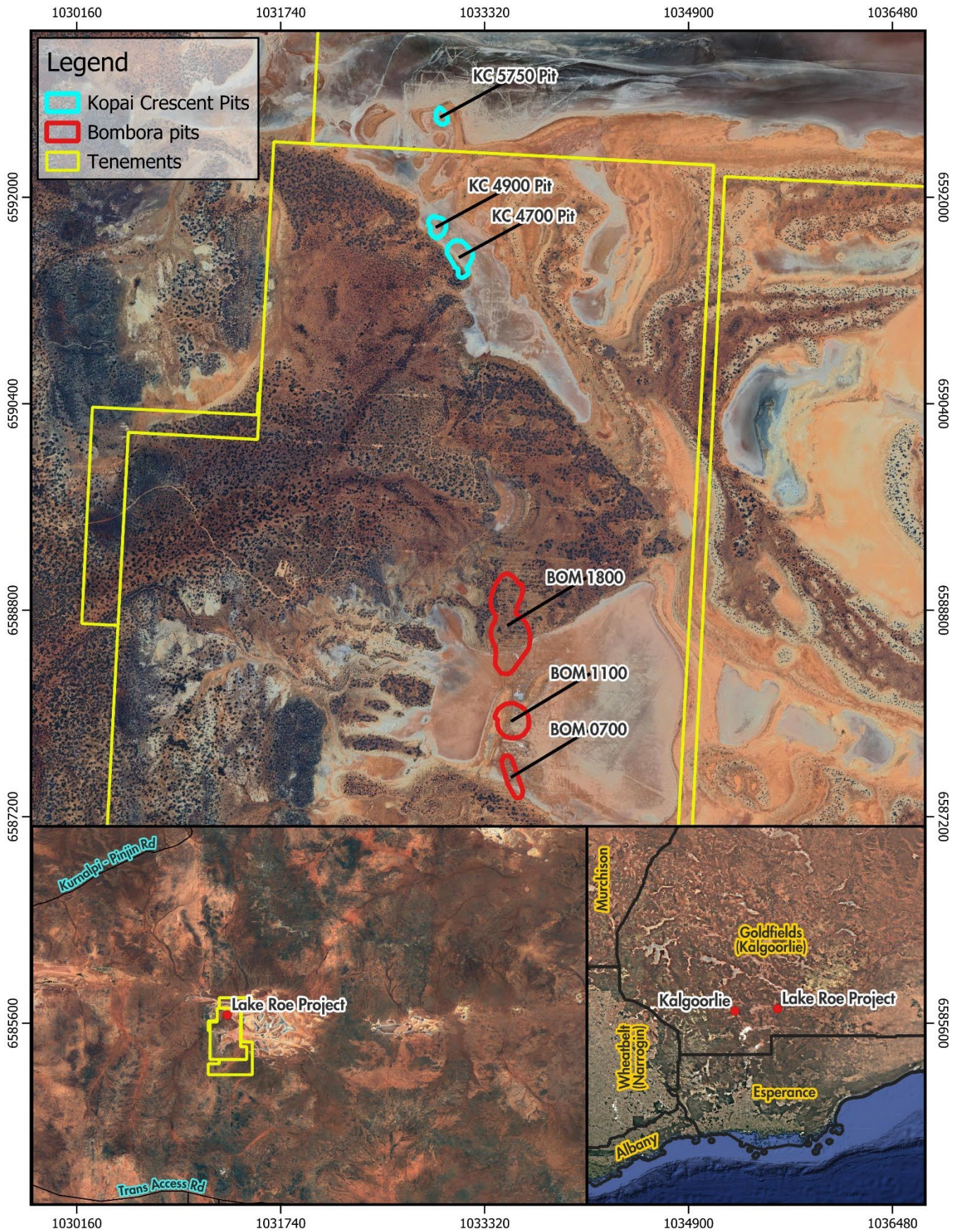
The Kopai Crescent and Bombora deposits each consist of three open pits. At Bombora, the pit areas are known as BOM 1800, BOM 1100, and BOM 0700. At Kopai Crescent, the pit areas are known as KC 5750 Pit, KC 4900 Pit, and KC 4700 Pit (Figure 1).

At Bombara, a decline and associated underground mining is anticipated to be developed at BOM 1800 (referred to as Kidney Lake (KL) underground) and BOM 0700 (referred to as Tura underground). The anticipated depth of the underground is ~555 m and ~180 m below ground level for KL and Tura, respectively. The underground mining will commence on completion of open pit mining.

2.2 Topography and surface water hydrology

The Project is situated along the south-western margin of Lake Roe, which forms part of the Roe Palaeodrainage system. This system originates west of Kalgoorlie, with drainage flowing eastward toward Lake Yindarlgooda and Lake Roe. The salt lake playas along this drainage system contain clay, silt, and sand, interbedded with evaporite minerals such as gypsum and halite. Although the playas are naturally saline, they are bordered by numerous smaller peripheral claypans that often contain non-saline water. Although typically dry, the region experiences infrequent flood events, usually triggered by intense winter rains or cyclonic activity (Campagna 2007; Kern 1996a; Stantec 2022).

Contaminants potentially mobilised within surface runoff from the Project's waste materials, if present, could be transported to nearby water systems and the broader landscape. Runoff from constructed waste landforms containing materials that generate soluble contaminants may adversely affect downstream water quality. To mitigate this risk, the presence and mobility of contaminants in leachates generated from the waste were assessed.



**Lake Roe Project
 Geochemical Waste Characterisation
 Ramelius Resources**

Figure 1: Location of the Project and deposits relevant to this study.

2.3 Climate

The climate of the Project is characterised as “Desert Hot (persistently dry)” under the modified Köppen classification system. This climate is characterised by hot summers and cool winters. Average monthly rainfall is greatest in February (31.7 mm) and lowest in September (13.4 mm) (BOM 2025a) (Table 1). January has the warmest mean maximum daily temperature (33.7° C); the mean minimum daily temperature for the same month is 18.4° C. In contrast, July is the coolest month, with a mean daily maximum temperature of 16.9° C and a mean daily minimum temperature of 5.1° C. Mean monthly pan evaporation (BoM 2006) exceeds mean monthly rainfall for all months throughout the year (Table 1).

Given that mean annual pan evaporation exceeds mean annual rainfall, water availability is likely limited and landform surfaces are likely to be dry for much of the time. As a result, infiltration of water and the associated movement of dissolved salts and/or metals (if present) is likely to be less than would occur in higher rainfall environments.

Table 1: Mean monthly rainfall and pan evaporation and mean daily temperature for Kalgoorlie-Boulder Airport.

Month	Mean Monthly Rainfall* (mm)	Mean Monthly Pan Evaporation^ (mm)	Mean Daily Max. Temperature* (°C)	Mean Daily Min. Temperature* (°C)
January	26.4	387.8	33.7	18.4
February	31.7	304.0	32.2	18.0
March	25.6	267.0	29.5	16.2
April	20.1	174.9	25.3	12.7
May	24.1	111.6	20.8	8.7
June	27.7	80.9	17.6	6.3
July	24.0	90.4	16.9	5.1
August	21.3	121.6	18.8	5.8
September	13.4	182.4	22.4	8.2
October	15.5	261.9	26.0	11.3
November	19.6	311.3	29.1	14.2
December	15.8	368.5	32.1	16.7
	265.3 (MEAN ANNUAL)	2 656.9 (MEAN ANNUAL)	25.4 (MEAN DAILY)	11.8 (MEAN DAILY)

* Rainfall and temperature data was sourced from the Kalgoorlie-Boulder Airport weather station (BOM station 12038) for the period of 1939-2024 (BoM 2025a).

^ Pan evaporation data was sourced from gridded data supplied by the Bureau of Meteorology (BoM 2006).

2.4 Project geology and associated wastes

2.4.1 Regional geology

The Project is situated within the Eastern Goldfields geological province of the Yilgarn Craton which forms a large section of the Western Shield geotectonic unit (Pilgrim 1979; Stantec 2022). The terrain at Lake Roe is characterised by gently undulating plains interrupted in the west by low hills and ridges of Archaean greenstones and in the east

by a host of Proterozoic basaltic granulite. The underlying strata are eroded and covered with Tertiary sand and gravel soils, scattered exposures of bedrock, and plains of calcareous earths (Cowan 2001; Kern 1996b; Stantec 2022).

2.4.2 Project geology

The Bombora resource lies on the eastern limb of the Bombora Antiform, within differentiated mafic volcanic and intrusive rocks (primarily basalt and dolerite), flanked by mafic sediments to the east and west (Figure 2). An unnamed syenite pluton extends beneath the deposit and beyond Lake Roe.

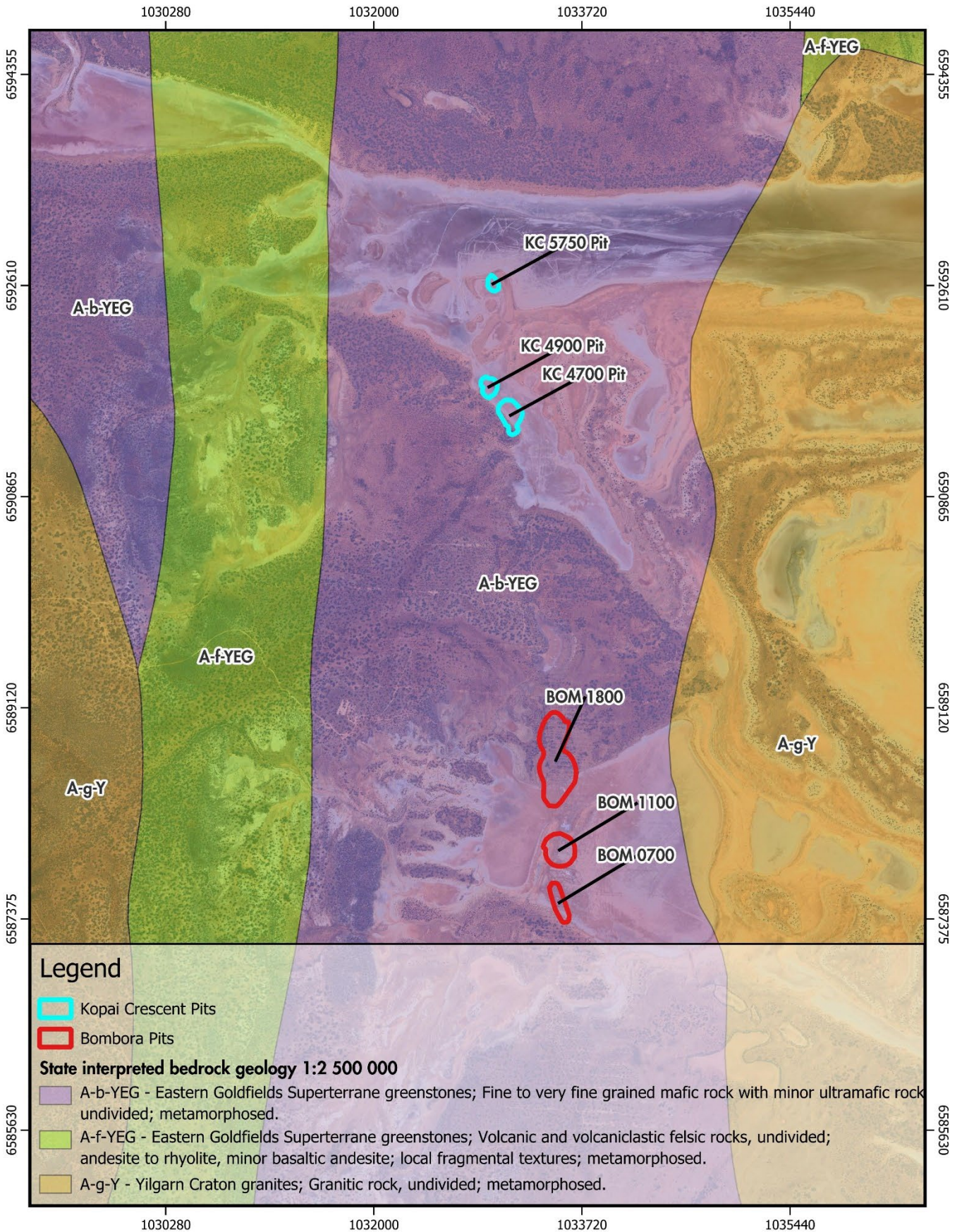
The deposit is hosted within a domed, fractionated dolerite with lamprophyre intrusions, extensive lineaments, and crosscutting de-magnetisation features, indicating potential alteration zones. Mineralisation is associated with iron sulphides, accompanied by quartz, biotite, albite, and carbonate alteration in unweathered zones.

The gold-bearing system consists of a quartz-sulphide vein stockwork at the quartz-dolerite and dolerite contact, overlain by highly weathered saprolite and less weathered saprock. Mineralisation is shallow, occurring from as little as 5 m deep, and extends across multiple lodes within a 150 m wide zone over a 3.7 km strike (Watson 2019; Ramelius 2025; Stantec 2022). Table 2 provides a detailed description of lithology at the Bombara deposit.

Mineralisation at Kopai Crescent is similar to Bombara in that it is hosted by iron-rich mafic host rocks (basalt and dolerite) and is associated with quartz veining and silica-albite-biotite-carbonate-sulphide alteration (Breaker Resources 2020).

Table 2: Detailed description of host lithologies at Bombora deposit (Source: Ramelius, pers. comm. G. Lister, September 2024)

Lithology		Description
Western Package	Footwall	Interbedded meta sediments (fine grained and finely bedded, usually silicified) and high Mg basalt (foliated to massive)
Bombora Sill	Mafic quartz dolerite	Most prospective unit of the Bombora Sill; granophyric
	Mafic dolerite	Medium grained dolerite, lacking granophyric textures
Eastern Package	Hanging wall	Foliated andesitic basalts and metasediments (incl. siltstones, shales, cherts and mafic volcanoclastics)
Lamprophyre		Late -stage intrusions characterised by coarse biotite, amphibole and carbonate
Cover		Transported cover



2.5 Expected mass of waste and ore

Approximately 24 Mt of waste material is proposed to be generated over the life of mine for the Project. A breakdown of waste materials across deposits, lithology, and weathering is provided in Table 3.

Kopai Crescent is expected to generate 2.9 Mt of waste. A large proportion of waste is expected to be dolerite (86%), with smaller proportions of basalt (14%). The majority of material is likely to be transitional (moderately weathered) (~98%) (Ramelius 2024a).

Bombora is expected to generate 21.4 Mt of waste. The dominant waste is expected to be quartz dolerite (68%) with smaller proportions of footwall (13%), dolerite (10%), and transported cover (8%). Less than 1% of lamprophyre is expected to be intersected. The majority of material brought to the surface is expected to be transitional (41%) and fresh waste (47%) (Ramelius 2024b).

Table 3: Anticipated life of mine waste mass for Kopai Crescent and Bombora (Source: Ramelius).

Deposit	Lithology	Lithology Code	Overall Proportion (%)	Weathering	Mass (t)	Proportion (%)
Kopai Crescent	Basalt	MB	14	Oxide	27,104	0.9
				Transitional	380,457	13
				Fresh	0	0
	Dolerite	MD	86	Oxide	33,750	1
				Transitional	2,476,247	85
				Fresh	4,350	<1
SUB-TOTAL					2,921,908	100
Bombora	Lamprophyre	ALP	1	Oxide	119	0.0
				Transition	23,495	0.1
				Fresh	127,013	0.6
	Dolerite	MD	10	Oxide	332,499	1.6
				Transition	1,713,380	8.0
				Fresh	102,814	0.5
	Quartz dolerite	MDQ	68	Oxide	522,674	2.4
				Transition	5,969,667	27.9
				Fresh	8,147,228	38.1
	Footwall	MR_FW	13	Oxide	59,279	0.3
				Transition	1,024,051	4.8
				Fresh	1,738,377	8.1
Transported cover	TRANSP	8	Oxide	1,574,304	7.4	
			Transition	69,738	0.3	
SUB-TOTAL					21,404,638	100
TOTAL					24,326,546	

2.6 Previous test work

Preliminary waste characterisation at Bombora was conducted by Stantec on behalf of Breaker Resources (Stantec 2022). Sixty (60) drill core samples were analysed to evaluate waste materials expected from open-pit mining. The findings indicated that the waste materials were predominantly non-acid forming (NAF) with a low risk of metalliferous drainage. Salinity levels ranged from slightly saline to extremely saline. Most wastes were inferred to be physically stable, although a small subset of samples showed partial clay dispersion.

The assessment did not evaluate the suitability of waste rock for rehabilitation purposes, such as their use as growth media or rock armour. Additionally, no testing was conducted for the presence of asbestiform/fibrous minerals or Naturally Occurring Radioactive Material (NORM). To date, no waste characterisation studies have been conducted for materials at Kopai Crescent.

3. METHODOLOGY

3.1 Sample selection

Ramelius provided Landloch with 73 diamond core samples selected from 14 diamond drill (DD) holes at Kopai Crescent and 60 pulp samples selected from 16 rotary drill (RC) holes at Bombara (Figure 3).

The samples were selected to represent the waste lithologies in terms of their volumes, spatial distribution, and weathering state across each of the open pits and underground. Table 4 provides an overview of the sample numbers split by lithology and weathering. No transported cover samples were available for analysis as this waste lithology was not intersected frequently during the drilling. Further detail on selected samples, including targeted depth intervals and associated cross sections, are provided in Appendix A.

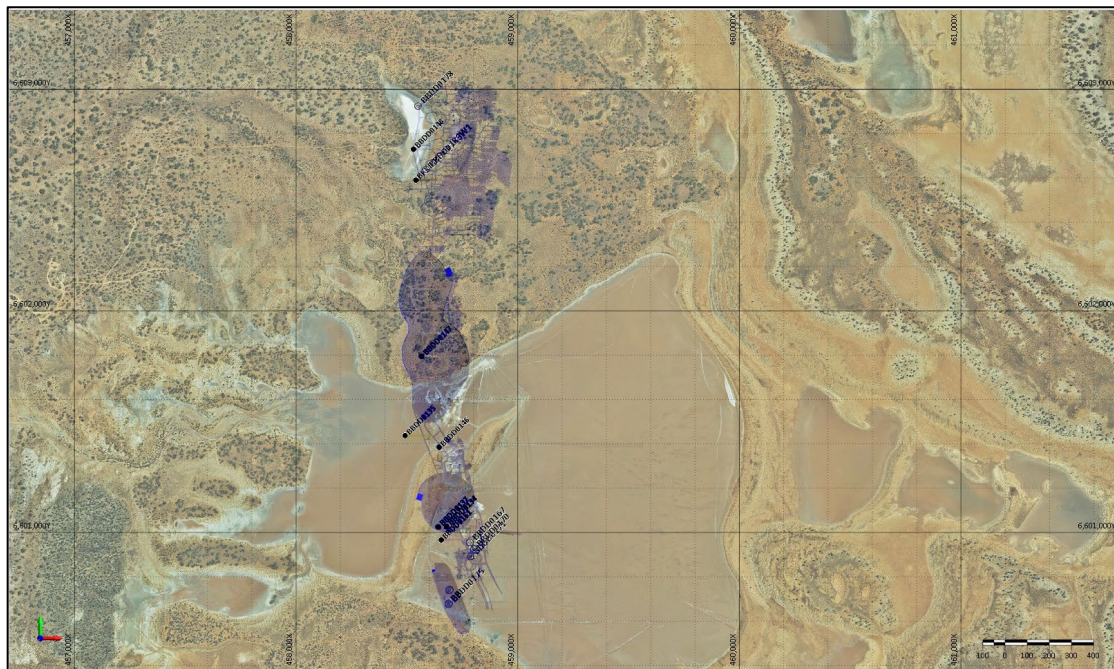
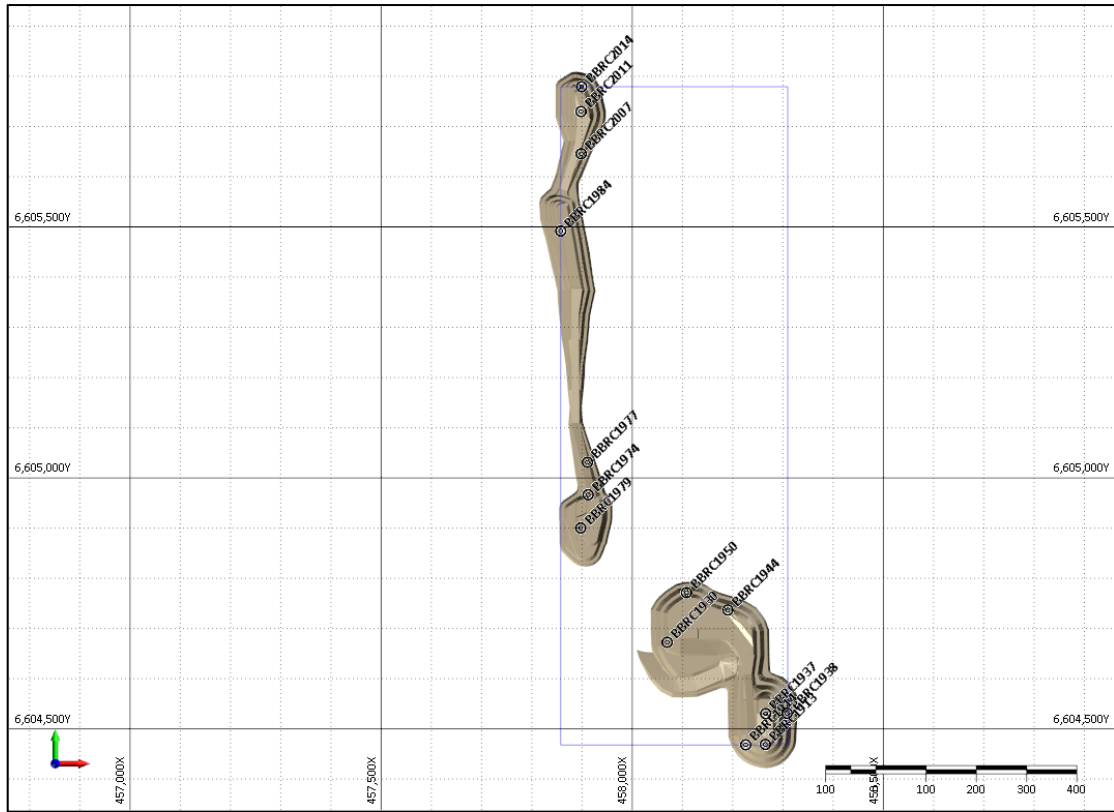


Figure 3: Collar locations of drill holes selected for characterisation at Kopai Crescent (top) and Bombara (bottom) (Source: Ramelius).

Table 4: Sample numbers selected for this study split by lithology and weathering

Deposit	Lithology Codes	Weathering	Proportion (%)	Sample Number	Proportion (%)
Kopai Crescent	MB	Oxide	0.9	0	0
		Transitional	13	15	21
		Fresh	0	3	4
	MD	Oxide	1	3	4
		Transitional	85	49	67
		Fresh	<1	3	4
SUB-TOTAL				73	100
Bombora	ALP	Oxide	0.0	0	0
		Transition	0.1	0	0
		Fresh	0.6	3	5
	MD	Oxide	1.6	0	0
		Transition	8.0	7	12
		Fresh	0.5	3	5
	MDQ	Oxide	2.4	0	0
		Transition	27.9	19	32
		Fresh	38.1	19	32
	MR_FW	Oxide	0.3	0	0
		Transition	4.8	4	7
		Fresh	8.1	5	8
	TRANSP	Oxide	7.4	0	0
		Transition	0.3	0	0
	SUB-TOTAL				60

3.2 Test work program

All 133 waste samples provided for characterisation were initially screened for:

- Soil pH_w (1:5 solid:deionised water solution);
- Electrical Conductivity (EC_{1:5});
- Total S (LECO), sulphide S, and sulphate S; and
- Naturally Occurring Radioactive Materials (NORMs).

Samples considered to be at higher risk of generating acid based on the screening total S values were further assessed for acid-base accounting (ABA) parameters and other indicators of acid generation potential, including:

- Total C;
- Net Acid Generation (NAG)¹;
- NAG_{4.5}²;

¹ Sulphur released by reaction with strong hydrogen peroxide at pH 4.5 and pH 7.0.

² Free acid (i.e. H₂SO₄) in the NAG liquor at pH 4.5.

- Final NAG pH³;
- pH (ox);
- Maximum Potential Acidity (MPA)⁴;
- Acid Producing Potential (APP);
- Acid Neutralising Capacity (ANC); and
- Net Acid Production Potential (NAPP)⁵.

A subset of 33 samples were subjected to 4-acid digest followed by ICP-MS and assessed for total metal concentrations in the solids⁶. Leachate testing was carried out for the same samples using a 1:20 solids:water solution and the concentrations of 21 elements within the leachate were assessed to consider drainage water quality⁷.

A subset of 25 samples were tested for rock durability indices including rock water absorption and rock particle density (11 DD core, and 14 RC rock chips).

A subset of 33 samples were assessed for chemical indicators of structural stability:

- Exchangeable cations and dispersion risk, including:
 - Exchangeable cations (Ca²⁺, Mg²⁺, Na⁺, K⁺, Al³⁺)
 - Effective Cation Exchange Capacity (ECEC)⁸;
 - Exchangeable Na Percentage (ESP)⁹;
 - Exchangeable Mg Percentage (EMP)¹⁰;
 - Electrochemical Stability Index (ESI)¹¹; and
 - Ca:Mg ratio¹²;

A subset of 13 samples were also assessed for presence of asbestiform materials via optical microscopy.

3.3 Description of characterisation parameters

3.3.1 pH_w

The pH of water extracts can be used as an indication of whether a sample is acidic or alkaline (pH) on short-term contact with water.

³ pH of the sample after oxidisation with hydrogen peroxide.

⁴ Calculated as total S multiplied by 30.6.

⁵ Calculated as the difference between APP and ANC.

⁶ **Elements assessed via 4-acid digest include:** Ag, Al, As, B, Ba, Be, Bi, Ca, Cd, Ce, Co, Cr, Cs, Cu, Fe, Ga, Ge, Hf, Hg, In, K, La, Li, Mg, Mn, Mo, Na, Nb, Ni, P, Pb, Rb, Re, S, Sb, Sc, Se, Sn, Sr, Ta, Te, Th, Ti, Tl, U, V, W, Y, Zn, and Zr.

⁷ **Leachate elements assessed include:** Al, As, B, Ba, Be, Ca, Cd, Cr, Co, Cu, Fe, K, Mg, Mn, Na, Ni, Pb, Se, SO₄²⁻, V, Zn, and water leachable Hg.

⁸ Calculated as the sum of exchangeable cations.

⁹ Calculated as the ratio of Ex. Na to ECEC expressed as a percentage.

¹⁰ Calculated as the ratio of Ex. Mg to ECEC expressed as a percentage.

¹¹ Calculated as the ratio of EC1:5 to ESP.

¹² Calculated as the ratio of exchangeable Ca to exchangeable Mg.

A general classification of pH values when measured in a 1:5 solids:deionised water solution (pH_w) is given in Table 5 (Hazelton and Murphy 2016). This classification allows for a range of pH values to be grouped but does not define whether a certain classification is suitable or not for vegetation growth. This is because vegetation in different areas is adapted to different pH values.

Responses in plant growth to varying pH is complex and determination of guidelines for pH suitability relevant for a wide range of materials is impractical. A rational approach when considering the potential impact of waste pH on rehabilitation success is to compare them with naturally occurring soil pH values and the vegetation that is endemic in soil having these pH values. Materials that have similar pH can be considered suitable for plant growth, at least with respect to pH.

Baseline soil assessments at Lake Roe recorded pH_w values for surface soils (0–0.15 m) ranging from 7.9 to 8.8 (median 8.4), classifying the soils as slightly to moderately alkaline (Stantec 2019). Therefore, for the purpose of comparison, pH values falling within ± 1 pH unit of 8 to 9 were considered suitable for plant growth. This range is also consistent with pH classifications for moderately alkaline soils.

Wastes with pH_w values outside this range may still support vegetation, but for the purposes of this report, they are classified as unsuitable unless additional work is completed to demonstrate their suitability or amendments are applied to adjust their pH.

Table 5: General material pH_w classification (Hazelton and Murphy 2016).

Material pH_w (pH units)	Classification
<4.0	Very strongly acidic
4.0–5.5	Strongly acidic
5.5–6.0	Moderately acidic
6.0–7.0	Slightly acidic
7.0–8.0	Slightly alkaline
8.0–9.0	Moderately alkaline
9.0–10.0	Strongly alkaline
>10.0	Very strongly alkaline

3.3.2 Salinity ($EC_{1:5}$)

Salinity is often measured using the electrical conductivity of a solution containing 1 part solids to 5 parts deionised water ($EC_{1:5}$). Most published salinity suitability ranking systems assume agricultural vegetation, and application of these ranking systems have little relevance for plant species used in rangeland rehabilitation.

The Western Australian Department of Primary Industries and Regional Development (DPIRD) provides a summary of published salinity tolerance values for a range of common species used in revegetation of disturbed rangelands in Western Australia (DAFWA 2004). These tolerance values are given as EC values measured not using the 1:5 soil:water solution method, but the alternative saturation extract method (EC_e). Conversion between $EC_{1:5}$ and EC_e is dependent on soil texture. Hazelton and Murphy (2016) suggest that for sandy and sandy loam soils EC_e should be divided by ~ 20 to estimate $EC_{1:5}$.

Using these conversion factors, salt sensitive species such as *Eucalyptus* have salt tolerance values ($EC_{1.5}$) ranging from 0.5–2.0 dS/m. Salinity trials performed near Kalgoorlie indicated that $EC_{1.5}$ in excess of 4 dS/m is likely to cause significant adverse impacts on germination and establishment of salt tolerant species (Jennings *et al.* 1993). Therefore, salinity values greater than 4 dS/m was used to indicate materials that are sufficiently saline to potentially impact vegetation growth.

These values correlate with the range of $EC_{1.5}$ values recorded for surface soils at Lake Roe (0.01–4.56 dS/m) (Stantec 2019). Stantec (2019) noted saline soils at Lake Roe to support growth of local salt tolerant species such as samphire.

3.3.3 Acid-base accounting (ABA)

Measurements of total S were used to determine the presence of potentially acid producing materials. A screening cut-off value for total S of $\geq 0.3\%$ was used to provide a preliminary characterisation of samples with the potential for acid generation (Price 1997). Samples with total S $< 0.3\%$ were classed as NAF irrespective of further ABA test work results. The classification scheme by which potentially acid forming (PAF) and NAF materials were identified is provided in Table 6. The classification system is based on the *Guidelines on Preventing Acid and Metalliferous Drainage* (DFAT 2016) and the *ARD Test Handbook* (AMIRA International 2002). The latter is advocated for by the *Global Acid Rock Drainage Guidelines* (INAP 2009).

Table 6: Acid generation risk classification scheme used for this study.

Classification*	Total S (%)	NAPP (kgH ₂ SO ₄ /tonne)	Final NAG pH	NAG _{4.5} (kgH ₂ SO ₄ /tonne)	ANC/MPA
NAF	≤ 0.3	-	-	-	-
	> 0.3	< 0	≥ 4.5	< 5	> 2
NAF-HS	> 1.0	< 0	≥ 4.5	< 5	> 2
PAF	> 0.3	≥ 0	< 4.5	> 5	< 2
PAF LC	> 0.3	≥ 0	< 4.5	< 5	< 2
Uncertain	> 0.3	≥ 0	> 4.5	-	< 2
	> 0.3	< 0	≤ 4.5	-	< 2

*NAF: non-acid forming; NAF-HS: non-acid forming, high sulphur; PAF: potentially acid forming; PAF-LC: potentially acid forming, low capacity.

3.3.4 Elemental enrichment (solids)

The screening of solids for metals and metalloids serves the purpose of detecting any elements that might raise environmental concerns in terms of revegetation and the quality of surface water and groundwater. The results obtained can serve as a basis for further investigation if necessary. To determine if any environmentally sensitive elements are significantly enriched, the concentrations of metals and metalloids in each sample were compared with the median concentration of that element within a reference material, in this case the median crustal abundance values provided by AusIMM (2011). The degree of enrichment was assessed using the Geochemical Abundance Index (GAI) equation:

$$GAI = \log_2 \left[\left(\frac{C}{1.5 * S} \right) \right]$$

in which C represents the concentration of the element in the sample and S denotes the median concentration of that element in the reference material (INAP 2009). GAI values are rounded to their integer value (from 0 to 6). A GAI of 0 indicates that the element is present at a concentration similar to or lower than the median abundance, while a GAI of 6 suggests an enrichment of approximately 100 times or more above the median abundance. As a general guideline, a GAI of 3 or higher is considered significant, indicating the need for further examination of such enrichment (INAP 2009).

The results were also compared to the background concentrations provided by the National Environmental Protection Council (NEPC) for Health Investigation Level (HIL C) and Ecological Investigation Level (open public spaces) ranges applicable to soils (NEPC 2011a, 2011b). HILs and EILs serve as indicators for the requirement of a more detailed risk assessment and do not imply the presence of significant health or ecological risks. If the guidelines do not specify criteria, Landloch cannot comment on the appropriateness of the measured values, but the data are included for completeness.

3.3.5 Elemental enrichment (leachate)

Drainage (produced by leaching materials) that contains elevated concentrations of soluble metals or metalloids can have adverse impacts on downstream surface water and/or groundwater. Metals at high concentrations can be toxic to biota. Additionally, dissolution of salts from saline materials may result in local or regional dispersion of salts through leaching, or accumulation of salt in soil pore water which can inhibit plant growth. Short-term static leach tests provide a means of screening for the potential of a material to become a mine waste contaminant source. The tests provide insight into aspects such as an element's solubility, mobility, bioavailability, and toxicity. The deionised water extraction undertaken for this study (1:20 solid:water) seeks to provide an indication of the mobile fraction under field conditions (an unbuffered water source, i.e. rainfall).

Evaluation of leachate data for each element was undertaken as follows:

- Comparison of the measured element concentration within the leachate with:
 - Freshwater and livestock (beef cattle) drinking water guideline limits (ANZECC 2000), or where freshwater and livestock limits did not exist;
 - Australian drinking water guideline limits (NHMRC and NRMCC 2011).
- Identification of element(s) with measurable but not elevated concentrations (i.e. significantly above the detection limit but below the guideline limits);
- Identification of element(s) with elevated concentrations (i.e. above one or more guideline limit); and
- Consideration of probable physical characteristics that may control solubility in certain pH ranges.

3.3.6 Structural stability

A materials' structural stability was assessed by considering the interrelated properties of exchangeable cations, salinity, and particle size distribution. These were considered

from three different perspectives. The order in which they are discussed should not be taken to indicate the relative importance of the different perspectives. Other perspectives, such as instability due to an abundance of fine sands, silts, and clays in the samples, were not considered due to the disturbed nature of the sample provided; the particle size distributions of the samples are unlikely to be representative of the particle sizes resulting from mining.

The proportion of exchangeable sodium (Na) held on the material's cation exchange complex in relation to other exchangeable cations is important. This is referred to as the Exchangeable Sodium Percent (ESP). McKenzie *et al.* (2004) considers the measurement of ESP as suitable for assessing the potential for clay dispersion when a material's Effective Cation Exchange Capacity (ECEC) is $\geq 3 \text{ meq}/100\text{g}$ and exchangeable Na $\geq 0.3 \text{ meq}/100\text{g}$. Further, the risk of clay dispersion is relevant only in materials with loam or clay textures (clay fraction $\geq 10\%$); sand-dominated materials are not prone to structural instability via clay dispersion. When these conditions are met, ESP $\geq 6\%$ is considered to indicate a material at prone to structural instability.

Dispersion potential for a material is also influenced by interactions between clay content, ESP, and EC_{1:5}. The Electrochemical Stability Index (ESI) is a way of considering these relationships for loam and clay textured materials (clay fraction $\geq 10\%$). ESI is the ratio of EC_{1:5} expressed in dS/m to ESP expressed as a percent. A tentative critical ESI value is 0.05 (McKenzie 1998), with ESI < 0.05 indicating a material that is prone to structural instability when the materials have sufficient clay.

Magnesian materials can also be prone to structural instability. Fenton and Conyers (2002) indicated that Mg by itself has little effect on spontaneous clay dispersion. However, it can enhance the effect of exchangeable Na in causing clay dispersion. This is assessed using a combination of the Exchangeable Magnesium Percent (EMP), Ca:Mg ratio, and ESP. When EMP $\geq 30\%$, dispersion is more likely to occur when the Ca:Mg ratio is < 1 and when either ESP $\geq 4\%$, or $\text{ESP} + (\text{EMP} \div 10)$ is $\geq 6\%$, assuming clay content is also $\geq 10\%$.

Relevant to all three of these perspectives, a material can have a fine fraction that is prone to structural instability, but if the proportion of fines is small, the material as a whole may be structurally stable. Typically, a binary mixture containing a $> 30\text{--}40\%$ fine fraction and $< 60\text{--}70\%$ coarse fraction could be considered a fines-dominated material that contains some coarse-grained particles; a binary mixture containing a $< 30\text{--}40\%$ fine fraction and $> 60\text{--}70\%$ coarse fraction could be considered a coarse-dominated material that contains some fine-grained particles. For this report, a fines cut-off of $\geq 40\%$ was adopted, with a material that contains a $\geq 40\%$ fine fraction being at risk of structural instability if it meets one or more of the criteria described above. Measurement of the fine/coarse fraction was not possible due to the nature of the samples provided. Therefore, for the purposes of this report, it was assumed that the fine fraction for all samples was $> 40\%$.

To capture interactions between physical and chemical properties and their effect on structural stability, the samples were assessed against these three criteria. If any of these criteria are met, that material was classified as being prone to structural instability. The assessment assumes that all materials had a clay content $\geq 10\%$ and fine fraction $\geq 40\%$. This is because the particle size distributions of the samples are unlikely to be representative of the particle sizes resulting from mining. This provides a conservative

assessment of structural stability, particularly for materials that are less weathered (i.e. fresh). The criteria are detailed in Table 7.

Table 7: Structural stability criteria, assuming clay content $\geq 10\%$ and fine fraction $\geq 40\%$.

ESP-based criteria:	EMP-based criteria:
<ul style="list-style-type: none"> • ECEC $\geq 3\text{meq}/100\text{g}$, and • Ex. Na $\geq 0.3\text{meq}/100\text{g}$, and • ESP $\geq 6\%$. 	<ul style="list-style-type: none"> • ECEC $\geq 3\text{meq}/100\text{g}$, and • EMP $\geq 30\%$, Ca:Mg < 1, & ESP $\geq 4\%$, or • EMP $\geq 30\%$, Ca:Mg < 1, & (ESP + (EMP \div 10)) $\geq 6\%$
ESI-based criteria:	
<ul style="list-style-type: none"> • ESI < 0.05. 	

3.3.7 Rock durability

Understanding the susceptibility of the rock fraction of a waste to breakdown is important when considering them as a potential armour against erosion in rehabilitation plans. Where rock particles are observed to breakdown, this can result in an increase in the proportion of fines over time, which would either increase erosion potential in the long term or sustain elevated erosion rates for longer periods (i.e. surface armouring by rainfall that can lead to reductions in erosion would be impeded). The breakdown of rock particles can also occur during landform re-shaping using heavy machinery, whereby rock particles of a certain size become smaller as machinery traverses over them, push them, and rip into them. Where rocks are shown to be durable, they could be considered for use as a rock armour to reduce erosion potential of rehabilitation landform batters if required.

In this study the durability of rock particles is classed as 'excellent', 'good', 'marginal', or 'poor' based on their lithology, weathering grade, particle density, and water absorption values (Table 8). This rating system is based on CIRIA *et al.* (2007) criteria for determining the quality and durability of armour-stone from a quarry source.

The degree to which a rocky waste can successfully mitigate risks associated with erosion is related not only to rock durability, but also to the size and abundance of rock, the geometry of the landforms being considered, and the climate in which the landform is situated. Therefore, the rock durability results provide a useful indication of potential rock armour materials, but more comprehensive particle size distribution and erodibility testing, and landform design modelling is still required to determine the limits of its use as a rock armour material.

Table 8: Rock durability rating system (adapted from CIRIA *et al.* (2007)).

Criteria	Excellent	Good	Marginal	Poor
Lithology	Unfoliated igneous and metamorphic rocks, quartzites and high silica cemented sandstones, compact crystalline limestones	Crystalline dolomites, crystalline limestone and moderately well cemented sandstones	Argillaceous limestones, poorly cemented sandstones, dolomite reef rock with void cavities	Shaly limestones, reef breccia, shale, siltstone, slate, schist, chalk, gypsiferous carbonates
Weathering grade	Fresh, unweathered	Faintly weathered (staining on major surfaces)	Slightly weathered (staining persists through a greater part of rock mass)	Moderately weathered (less than half the rock mass is decomposed)
Particle density (g/cm ³)	>2.7	2.5-2.7	2.3-2.5	<2.3
Water absorption (%)	<0.5	0.5-2.0	2.0-6.0	>6.0

3.3.8 Naturally occurring radioactive material

All mining wastes contain radionuclides of natural origin. For most activities involving minerals and raw materials, the levels of exposure to these radionuclides are not significantly greater than normal background levels and are not of concern for radiation protection. However, some materials may give off radiant energy at levels higher than background levels. Human activity around these types of materials can increase the risk of radiological exposure. Material giving rise to these enhanced exposures is known as naturally occurring radioactive material (NORM).

The WA Radiological Council (2019) and the DMP (2010) require that the dose limit and surface concentration for beta, gamma, and low toxicity alpha emitters falls below 10 μ Sv/hr and 0.40Bq/cm², respectively. Dose limits and surface concentrations in samples are assessed against these criteria.

3.3.9 Asbestiform minerals

“Asbestos” is a commercial term referring to six types of naturally occurring silicate mineral fibres that can be separated into two broad mineral categories: amphibole and serpentine. These minerals are hydrated silicates with varying metal compositions and are known as asbestiform minerals. Figure 4 shows asbestos mineral categories that are considered to be harmful to human health based on the Department of Energy, Mines, Industry Regulation and Safety’s (DEMIRS) guidance document *Management of fibrous minerals in Western Australian mining operations* (DMP 2015).

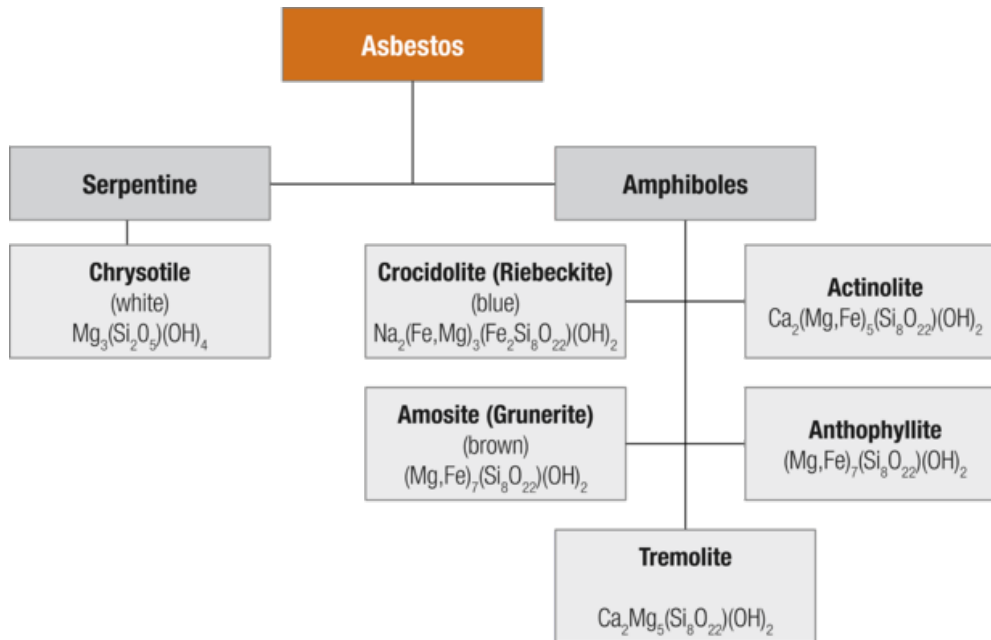


Figure 4: Asbestos mineral categories that are considered to be harmful to human health (DMP 2015).

Amphibole and serpentine minerals are major components of mafic and ultramafic rocks found in Western Australia's greenstone belts, which contain the State's major gold and base metal deposits. Amphibole minerals may also be encountered in the banded iron formations of the Hamersley Basin (DMP 2015). Exposure to asbestiform minerals can have adverse impacts on human health, when fibres are of a size that can be airborne and respirable.

Assessment of samples for presence/absence of mineral fibres is undertaken in accordance with the polarised light microscopy and dispersion staining method established in AS:4964-2004 *Method for the qualitative identification of asbestos in bulk samples*.

Samples returning mineral fibres are further analysed for fibre characteristics using scanning electron microscopy (SEM) with elemental analysis by energy dispersive spectroscopy (EDS) following AS: 4964-2004 (modified for SEM). Under AS 4964-2004 asbestos is defined as:

- Many particles with aspect ratios (i.e. length to width ratios) ranging from 20:1 to 100:1 or higher for particles >5 micrometres (µm) in length. Bundles of fibres may have lower aspect ratios;
- Sets of fibre bundles generally less than 0.5 µm, but always less than 1.0 µm in width, unless in thick bundles;
- In addition to the mandatory fibrillar crystal growth, one or more, and preferably three of the following aspects:
 - Parallel fibres occurring in bundles;
 - Fibre bundles displaying splayed ends;
 - Matted masses of individual fibres;
 - Fibres showing characteristic curvature.

- Respirable asbestos fibres are defined as:
 - Asbestos fibres less than 3 μm in width, and greater than 5 μm in length, and with a length to width ratio greater than 3 to 1.

The detection limit for asbestiform minerals using the AS 4964-2004 method is influenced by sample condition, composition, and fibre type. Typically, this method can detect asbestos at concentrations ranging from 1 in 1,000 (0.1 wt%) to 1 in 10,000 (0.01 wt%) by weight, which is equivalent to 1 to 0.1 g/kg. For the purposes of this assessment, a Limit of Reporting (LoR) of 0.01 wt% has been applied in alignment with AS 4964.

Concentrations greater than 0.01 wt% are considered potentially hazardous to human health based on limits set by the Globally Harmonised System (GHS) of classification and labelling of chemicals and soil screening levels set by the Department of Health (DoH 2009).

4. TEST WORK RESULTS

This section provides a summary of the completed test work. Detailed laboratory results can be found in Appendix B. To simplify data presentation, the results have been grouped according to deposit and lithology classification.

4.1 pH_w

The majority (65%) of the Kopai Crescent samples analysed recorded pH values within the range considered suitable for vegetation establishment and plant growth (Figure 5). The pH values ranged from 6.2–10.0, with a median of 7.9, classifying them as slightly acidic to strongly alkaline.

A number of dolerite samples from Kopia Crescent had lower pH_w than measured for surface soils at the Project. However, the median pH_w of the dolerite (pH_w 8.2) remained within the range considered suitable for vegetation establishment. The pH_w of all but 2 basalt samples are classed as suitable, and the median pH_w (8.2) was within the acceptable range. Subsequently, all waste materials from Kopai Crescent are considered suitable for use as growth media based on pH.

All of the samples from Bombara recorded pH levels within (or very close to) the range considered suitable for vegetation establishment (pH_w 7–10) (Figure 6). Waste materials from Bombara recorded pH_w values ranging from 7.4 to 10.1, with a median of 9.6. All waste material from this deposit is considered suitable for use in rehabilitation as growth media based on pH.

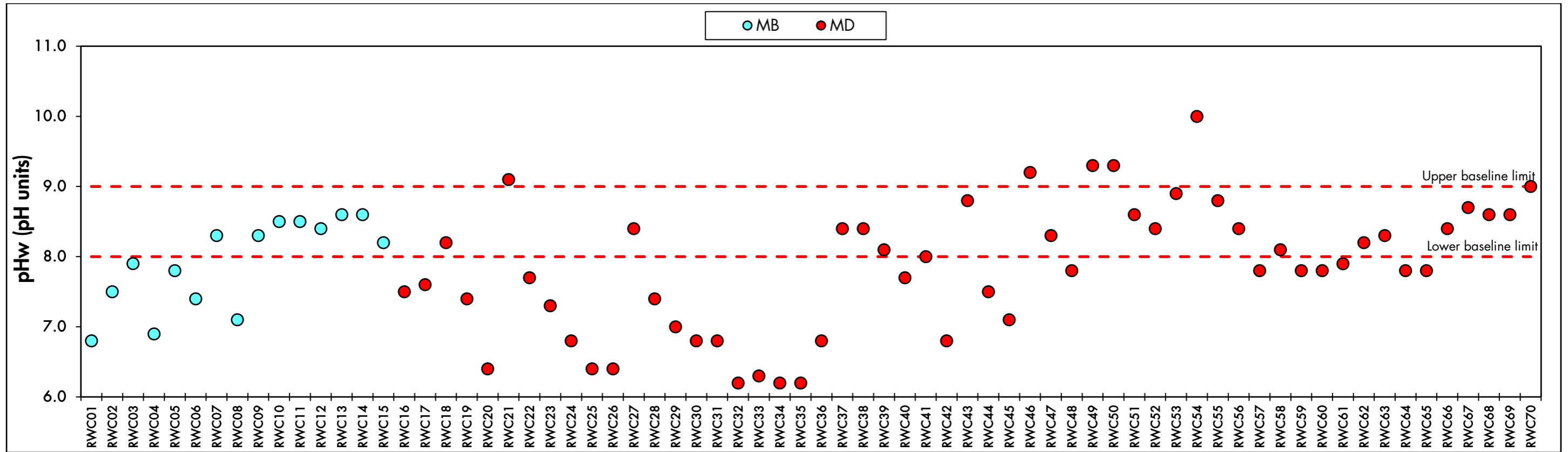


Figure 5: pH_w recorded for 70 Kopai Crescent samples. Data points are compared to limits stipulated in section 3.3.1. MB=Basalt, MD=Dolerite.

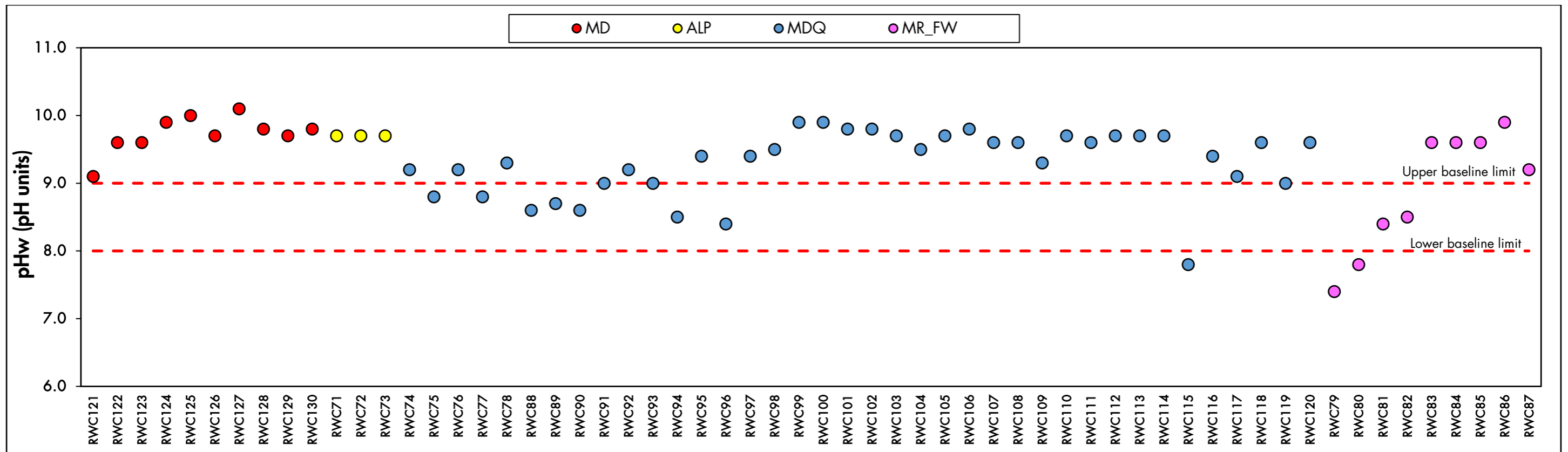


Figure 6: pH_w recorded for 60 Bombora samples. Data points are compared to limits stipulated in section 3.3.1. ALP= Lamprophyre, MD=Dolerite, MDQ= Quartz dolerite, MR_FW=Footwall.

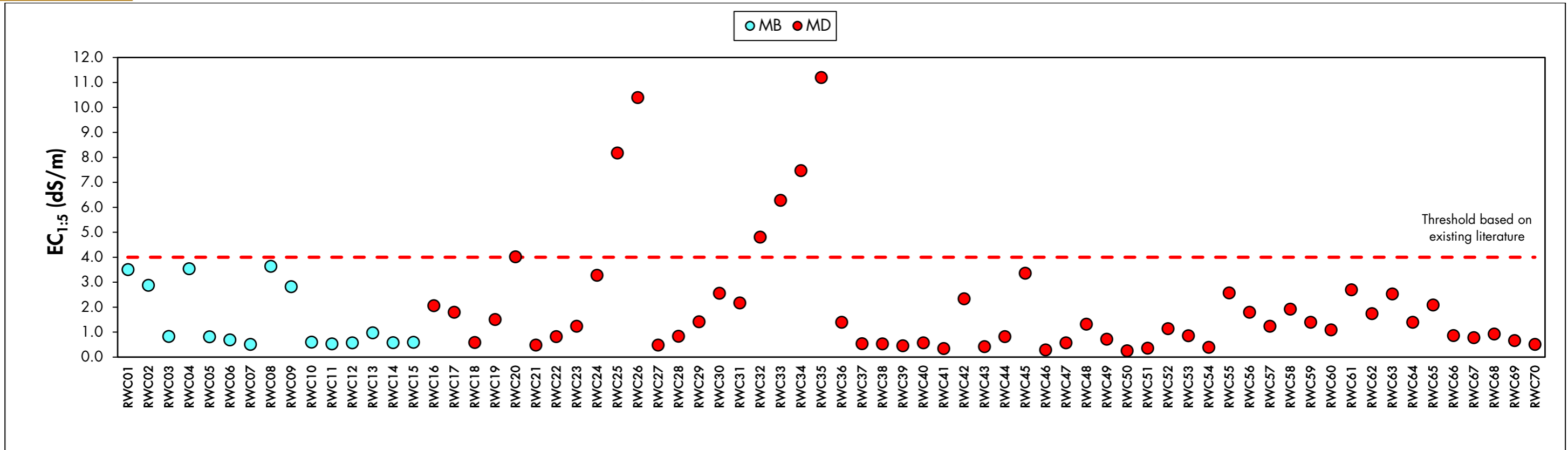


Figure 7: EC_{1:5} recorded for 70 Kopai Crescent samples. Data points are compared to limits stipulated in section 3.3.2. MB=Basalt, MD=Dolerite.

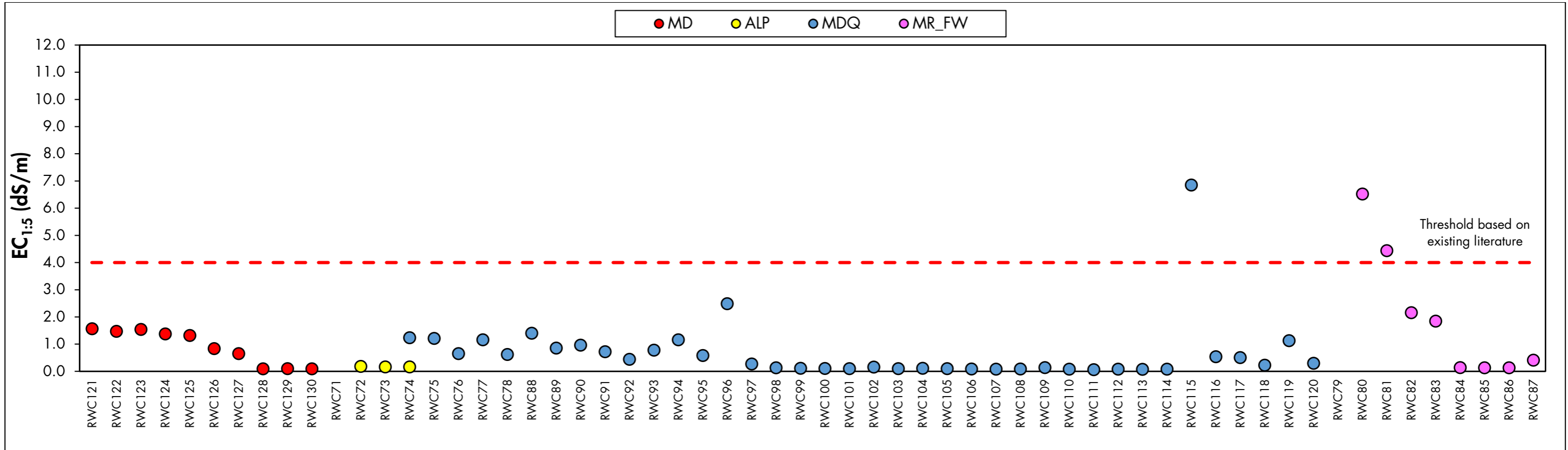


Figure 8: EC_{1:5} recorded for 60 Bombora samples. Data points are compared to limits stipulated in section 3.3.2. ALP= Lamprophyre, MD=Dolerite, MDQ= Quartz dolerite, MR_FW=Footwall.

4.2 Salinity (EC_{1:5})

The majority of samples (91%) from both Kopai Crescent and Bombara exhibited EC_{1:5} values within the acceptable limits (Figure 7). Samples from Kopai Crescent exhibited higher salinity values compared to Bombara with recorded EC_{1:5} values ranging from 0.26–11.2 dS/m (median 1.12 dS/m). Samples from Bombara had EC_{1:5} values ranging from 0.07–6.9 dS/m (median 0.36 dS/m) (Figure 8).

A small proportion (9%) of samples recorded EC_{1:5} levels >4 dS/m. These samples originated from oxidised and transitional dolerite at Kopai Crescent. Stockpiling and handling of this material with other waste materials of lower salinity is likely to dilute salinity concentrations. When considered as a whole, waste from both Kopai Crescent and Bombara deposits are considered suitable for use in rehabilitation based on their salinity.

4.3 Acid-base accounting (ABA)

Of the 133 samples tested, 125 (94%) recorded total S values <0.3%. Further assessment of sulphur speciation showed that the majority of S present in these samples was in the form of sulphate (SO₄²⁻), a non-acid generating form of sulphur. These samples were classified as NAF.

Five (5) samples returned a total S value >0.3%. Two (2) of the 5 samples originated from Kopai Crescent and were logged as fresh dolerite (MD) and three of the samples originated from Bombara and were logged as fresh quartz dolerite (MDQ). Further assessment of acid-base accounting parameters for these samples showed that 4 of the 5 samples could be classified as NAF based on:

- Final NAG pH >4;
- NAG_{4.5} < 5 kg H₂SO₄/tonne; and
- NAPP < 0 kg H₂SO₄/tonne.

One fresh quartz dolerite sample from the Bombara deposit, was classified as PAF based on the acid neutralising capacity (ANC 57.8 kg H₂SO₄/t) being insufficient to neutralise the maximum potential acidity (MPA 60.28 kg H₂SO₄/t) (Figure 9). This is further evidenced by a NAG pH of 3 and positive NAPP of 1.6 kg H₂SO₄/t (Figure 10). This sample resides at a depth interval of 182.5–184 m and represents quartz dolerite occurring at the maximum planned depth of the Tura underground. All quartz dolerite samples originating from depths <182.5 m for Tura underground and open pits returned a NAF classification.

Total S values, sulphur speciation, acid-base accounting parameters and assigned acid generation risk classification for all samples is provided in Appendix B.

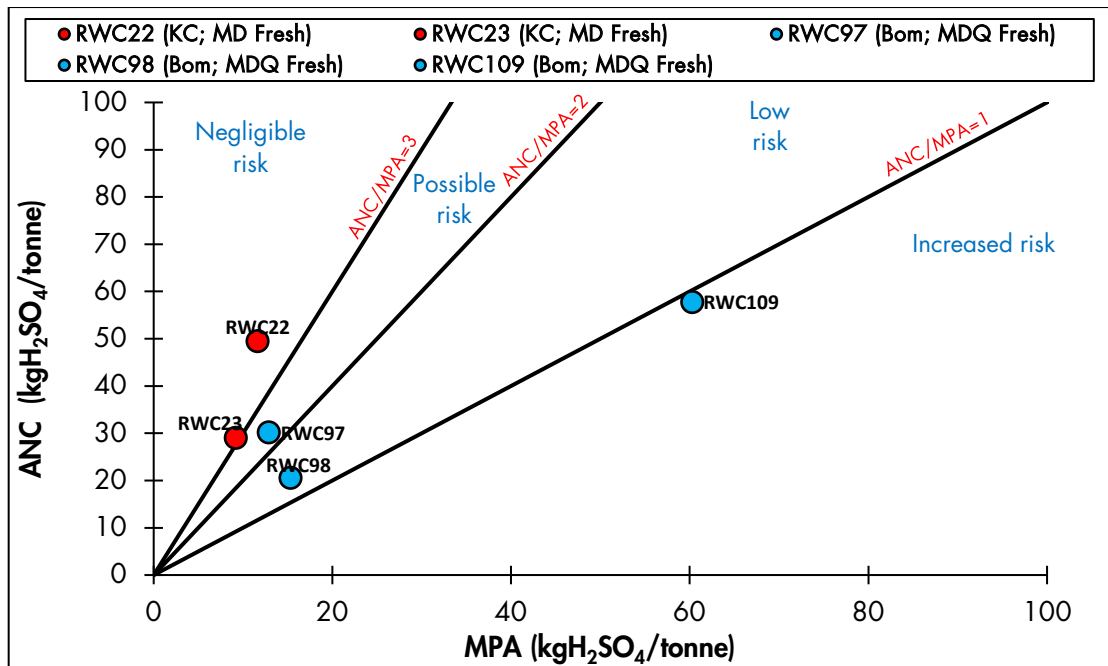


Figure 9: Acid neutralising capacity (ANC) compared to Maximum potential acidity (MPA). MD=Dolerite, MDQ= Quartz dolerite. Bom = Bomabara, KC = Kopai Crescent.

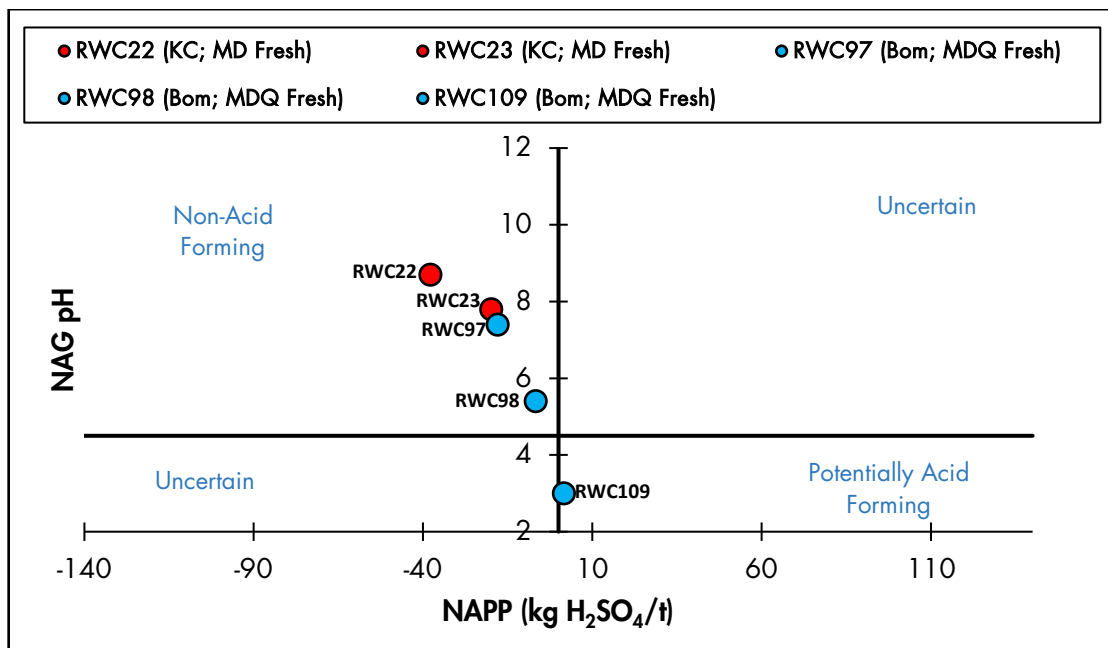


Figure 10: NAG pH compared to NAPP. MD=Dolerite, MDQ=Quartz dolerite. Bom=Bomabara, KC=Kopai Crescent.

4.4 Elemental enrichment (solids)

Elemental concentrations of 33 samples were analysed as part of the study. At least one sample from each lithology was included in the assessment. Elemental concentrations of all samples can be found in Appendix B. Elements typically shown to be enriched in the samples include:

- Iron (Fe) with a median GAI of 6;
- Selenium (Se) with a median GAI of 4; and
- Tellurium (Te) with a median GAI of 3.

Neither Fe or Te are known to be toxic to natural biota. It is noted that, although Se had a median GAI value of 4, the detection limit of the laboratory assessment (1 ppm) exceeds the median crustal abundance (0.05 ppm).

Comparison of elemental concentrations in samples to NEPC Health Investigation Level (HIL C) thresholds showed that one footwall sample (MR_FW Fresh) from Bombora exceeded the HIL threshold for chromium (Cr) with a value of 489 mg/kg.

Comparison of elemental concentrations in samples to NEPC Ecological Investigation Level (EIL) thresholds for open public spaces showed that:

- 1 sample of Bombora footwall (MR-FW) exceeded the EIL value for Cr;
- 6 samples exceeded the EIL value for copper (Cu), including;
 - 5 Kopai Crescent dolerite (MD); and
 - 1 Bombora quartz dolerite (MDQ).
- 24 samples exceeded the EIL value for nickel (Ni) notably including all lithologies across both deposits except quartz dolerite (MDQ);
- 15 samples exceeded the EIL value for zinc (Zn) including:
 - 7 Kopai Crescent samples (2 basalt (MB); and
 - 5 dolerite (MD)); and
 - 8 Bombora samples (quartz dolerite (MDQ)).

The remainder of elements with EIL or HIL values fell within the recommended threshold (i.e. As, Bi, Cd, Co, Mn, Ni, Pb).

4.5 Elemental enrichment (leachate)

Results of elemental testing of the leachate are provided in Appendix B. For all deposits, leaching of the samples with an unbuffered water source (i.e. rainfall) generated:

- Kopai Crescent:
 - Measurable but not elevated concentrations of chromium, manganese, nickel, selenium, and zinc;
 - Elevated concentrations of aluminium and chloride; and
 - Salinity levels between 0.09–3.54 dS/m (median 0.58 dS/m).
- Bombora:
 - Measurable but not elevated concentrations of chromium, nickel, selenium, and zinc; and
 - Elevated concentrations of aluminium; and
 - Salinity levels between 0.05–0.61 dS/m (median 0.09dS/m).

The mafic volcanic and intrusive rocks at Lake Roe, particularly basalt and dolerite, are natural sources of trace metals such as chromium, manganese, and nickel. These elements are commonly hosted within iron sulphides and mafic minerals, which undergo oxidation and leaching in the weathering profile.

The concentration of aluminium in the majority of samples across various lithologies exceeded the guideline trigger values. Elevated aluminium levels are likely attributed to secondary aluminium-bearing clays (e.g., kaolinite, gibbsite) formed through the weathering of feldspars and micas, particularly within the fractionated dolerite and syenite units. The enrichment and mobilisation of aluminium are driven by intense weathering and further enhanced by structural controls such as shearing, jointing, and folding, which increase rock permeability and promote fluid-driven alteration. Additionally, demagnetisation features observed in the dolerite host rock indicate zones of hydrothermal alteration, where magnetic minerals (e.g. magnetite) have been replaced by non-magnetic minerals like hematite, sulphides, and clays. These alteration processes further contribute to aluminium enrichment by facilitating mineral breakdown and secondary clay formation.

The potential for elevated levels of aluminium in leachate to pose a risk to the environment was considered further with respect to probable pH ranges and bioavailability. The higher aluminium concentrations may be attributed to the formation of soluble $\text{Al}(\text{OH})_4^-$ (negatively charged complex at neutral and slightly alkaline conditions) due to the pH of the leachate solution, which has a median pH of 9.0. The potential for adverse effects on the receiving environment due to elevated aluminium levels primarily exists when the pH is below 5. In such conditions, aluminium is more likely to be present as trivalent Al^{3+} , which is known to be highly toxic to organisms (Bojórquez-Quintal *et al.* 2017). However, in environments where the pH_w ranges from ~5.5 up to moderately alkaline conditions (pH_w 8.5), soluble aluminium concentrations are expected to remain low. In these conditions, aluminium is predominantly in the non-soluble forms of $\text{Al}(\text{OH})_3$ and $\text{Al}(\text{OH})_4^-$, and it does not pose a significant toxicity issue.

The salinity of leachate is consistent with the salinity generated on short-term contact with water (Section 3.3.2) and provides further evidence to suggest the salinity of the waste materials is likely to remain below 4 dS/m. The Project's location within a low-lying terrain, influenced by an enhanced weathering profile over syenite, has led to the accumulation of gypseous loams, saline loams, and lake clays (CSIRO and BRS 1991). These deposits, formed in a semi-arid environment with limited drainage, may contribute to the elevated salinity levels observed in the leachate. The proximity of Project to the salt lakes also plays a role in chloride enrichment, as evaporative processes concentrate dissolved salts in the soil and shallow groundwater. Due to the presence of *in situ* saline soils, it is expected that vegetation in the vicinity has some degree of salt tolerance.

The impact of saline drainage on groundwater quality will depend on existing groundwater salinity. Given the close proximity of the deposit to Lake Roe, it is likely that groundwater in the area is already saline. Additional studies are necessary to assess whether the salinity levels of drainage from waste materials are comparable to that of local groundwater.

4.6 Structural stability

Thirty-two (32) samples (97%) assessed were found to be prone to structural instability, triggering the ESP and ESI criteria. These samples exhibited elevated ESP, high exchangeable sodium (Ex. Na) concentrations, increased effective cation exchange capacity (ECEC), and/or low ESI values (Table 9). This assessment of structural stability is conservative as it assumes that all materials have a clay content $\geq 10\%$ and fine fraction $\geq 40\%$. A small proportion of waste material will be oxidised (12%). This waste type is likely to be dominated by fine grained particles and clay minerals and so will be at greatest risk of structural instability.

The remainder of materials mined from Kopai Crescent and Bombara will be transitional (moderately weathered) and fresh. If the post-mined matrix of transitional and fresh materials is predominantly coarse-grained, these materials may retain structural stability. Based on the mining sequence fresh waste rock is expected to be extracted last. The fresh rock may be used on the outer surface of the waste rock landform to encapsulate materials that are at greater risk of structural instability (i.e. oxides) pending it carries an abundance of coarse fraction ($>60\%$).

The susceptibility of the transitional and fresh materials to structural instability was further considered based on their geology. Basalt is likely to consist of more than 40% dense, silica-rich minerals (e.g., quartz), which offer increased resistance to weathering and are expected to break down into coarser sandy material. The dolerite primarily comprises feldspar (~40–70%) and pyroxene minerals. Feldspars weather into kaolinite and gibbsite, while pyroxenes degrade into 1:1 and 2:1 clays, contributing to material breakdown. The transitional dolerite may have greater dominance of fine-grained particles and clay minerals. That being said, the increased quartz content in quartz dolerite at Bombara is likely to improve resistance to weathering, with the degree of resistance directly dependent on the proportion of quartz within the material. Based on the above, transitional and fresh materials are likely to retain their structural stability, however further physical testing of extracted material during mining operations is recommended to determine the proportion of coarse versus fine fractions, assess clay content, and re-evaluate their structural stability.

4.7 Rock durability

The durability of the majority of samples is highly variable, ranging from marginal to poor. This is expected given their variability in weathering and alteration. The lamprophyre, oxidised dolerite and transitional basalt waste materials are shown to have poor to marginal durability and are considered unsuitable for use as rock armour.

The fresh quartz dolerite and transitional and fresh dolerite at both Kopai Crescent and Bombara exhibit good durability and may be considered as potential rock armour materials, if these materials also carry an abundance coarse fraction ($>60\%$).

Table 9: Assessment of structural stability of fine fraction in wastes.

Sample ID	Material	Oxidation	ECEC	Ex. Na	Ca:Mg	ESP	EMP	ESI	Assessment Criteria			Prone to Structural Instability
			meq/100g	meq/100g	-	%	%	-	ESP	EMP	ESI	
Kopai Crescent												
RWC01	Basalt (MB)	Transitional	13.4	9.5	0.6	70.5	16.8	0.063	✓	✗	✗	✓
RWC05	Basalt (MB)	Transitional	13.5	9.9	0.2	73.2	20.9	0.056	✓	✗	✗	✓
RWC09	Basalt (MB)	Transitional	18.6	14.3	0.2	77.1	18.0	0.049	✓	✗	✓	✓
RWC13	Basalt (MB)	Transitional	12.4	9.4	0.2	75.9	17.9	0.025	✓	✗	✓	✓
RWC17	Dolerite (MD)	Transitional	7.3	5.5	0.2	74.8	19.4	0.032	✓	✗	✓	✓
RWC22	Dolerite (MD)	Fresh	3.4	1.2	3.1	34.6	15.2	0.029	✓	✗	✓	✓
RWC26	Dolerite (MD)	Oxide	18.3	11.6	0.2	63.2	29.0	0.212	✓	✗	✗	✓
RWC30	Dolerite (MD)	Transitional	2.3	1.0	1.2	42.5	25.4	0.076	✗	✗	✗	✗
RWC34	Dolerite (MD)	Transitional	4.1	0.9	2.6	23.0	20.5	0.404	✓	✗	✗	✓
RWC38	Dolerite (MD)	Transitional	18.8	14.6	0.2	77.9	17.7	0.026	✓	✗	✓	✓
RWC42	Dolerite (MD)	Transitional	6.2	3.8	0.5	60.1	25.1	0.055	✓	✗	✗	✓
RWC46	Dolerite (MD)	Transitional	1.4	0.7	1.4	50.4	17.7	0.008	✗	✗	✓	✓
RWC50	Dolerite (MD)	Transitional	1.6	0.7	1.3	42.0	23.7	0.008	✗	✗	✓	✓
RWC54	Dolerite (MD)	Transitional	3.1	1.7	0.7	53.8	25.6	0.010	✓	✗	✓	✓
RWC58	Dolerite (MD)	Transitional	6.6	4.5	0.3	68.7	22.7	0.039	✓	✗	✓	✓
RWC62	Dolerite (MD)	Transitional	7.8	5.8	0.3	74.5	18.9	0.035	✓	✗	✓	✓
RWC66	Dolerite (MD)	Transitional	10.2	7.5	0.3	74.2	19.3	0.032	✓	✗	✓	✓
RWC70	Dolerite (MD)	Transitional	7.8	5.9	0.3	75.5	16.9	0.020	✓	✗	✓	✓
Bombora												
RWC73	Lamprophyre (ALP)	Fresh	1.7	0.4	3.3	21.7	11.2	0.014	✗	✗	✓	✓
RWC77	Quartz dolerite (MDQ)	Transitional	9.1	7.0	0.4	76.9	15	0.015	✓	✗	✓	✓
RWC81	Interbedded meta sediments and high Mg basalt (MR_FW)	Transitional	19.5	17.1	0.2	87.5	9.44	0.029	✓	✗	✓	✓
RWC85	MR_FW Fresh	Fresh	1.4	0.6	1.9	38.1	16.20	0.006	✗	✗	✓	✓
RWC89	Quartz dolerite (MDQ)	Transitional	3.3	2.2	0.7	65.9	18.00	0.014	✓	✗	✓	✓
RWC93	Quartz dolerite (MDQ)	Transitional	4.5	3.3	0.7	72.7	14.1	0.013	✓	✗	✓	✓

Sample ID	Material	Oxidation	ECEC	Ex. Na	Ca:Mg	ESP	EMP	ESI	Assessment Criteria			Prone to Structural Instability
			meq/100g	meq/100g	-	%	%	-	ESP	EMP	ESI	
RWC97	Quartz dolerite (MDQ)	Fresh	1.9	0.8	4.4	42.8	6.31	0.007	*	*	✓	✓
RWC101	Quartz dolerite (MDQ)	Fresh	1.1	0.5	4.6	42.0	7.59	0.004	*	*	✓	✓
RWC105	Quartz dolerite (MDQ)	Fresh	0.9	0.3	14.0	33.6	10.40	0.005	*	*	✓	✓
RWC109	Quartz dolerite (MDQ)	Fresh	0.9	0.2	14.7	27.6	9.68	0.007	*	*	✓	✓
RWC113	Quartz dolerite (MDQ)	Fresh	1.0	0.3	4.4	32.1	8.74	0.005	*	*	✓	✓
RWC117	Quartz dolerite (MDQ)	Transitional	2.5	1.6	11.0	65.8	13.90	0.007	*	*	✓	✓
RWC121	Dolerite (MD)	Transitional	4.8	3.3	0.7	67.7	18.20	0.025	✓	*	✓	✓
RWC125	Dolerite (MD)	Transitional	4.8	3.3	0.9	69.1	15.10	0.016	✓	*	✓	✓
RWC129	Dolerite (MD)	Fresh	0.9	0.3	1.2	36.4	23.00	0.004	*	*	✓	✓

Table 10: Durability classification for wastes.

Deposit	Material	Weathering	Average rock density (g/cm ³)	Average water absorption (%)	Durability rating	
Kopai Crescent	Basalt (MB)	Transitional	3.41	4.99	Marginal	
	Basalt (MB)	Transitional	3.62	6.91	Marginal	
	Dolerite (MD)	Transitional	1.77	4.33	Poor	
	Dolerite (MD)	Fresh	2.66	2.51	Marginal	
	Dolerite (MD)	Fresh	3.18	2.93	Marginal	
	Dolerite (MD)	Oxide	1.90	13.16	Poor	
	Dolerite (MD)	Oxide	Disintegrated on wetting after 24 hrs.		Poor	
	Dolerite (MD)	Transitional	3.43	1.97	Good	
	Dolerite (MD)	Transitional	2.63	17.87	Poor	
	Dolerite (MD)	Transitional	2.59	7.72	Poor	
	Dolerite (MD)	Transitional	3.40	1.32	Good	
	Bombara	Lamprophyre (ALP)	Fresh	2.48	0.10	Good
		Lamprophyre (ALP)	Fresh	2.55	0.11	Good
Footwall (MR_FW)		Transitional	2.80	3.23	Marginal	
Quartz dolerite (MDQ)		Fresh	Disintegrated on wetting after 24 hrs.		Poor	
Quartz dolerite (MDQ)		Fresh	2.02	6.51	Poor	
Quartz dolerite (MDQ)		Fresh	3.15	0.25	Good	
Quartz dolerite (MDQ)		Fresh	2.78	0.22	Good	
Quartz dolerite (MDQ)		Fresh	3.07	0.18	Good	
Quartz dolerite (MDQ)		Fresh	3.39	0.10	Good	
Quartz dolerite (MDQ)		Fresh	4.61	8.92	Poor	
Dolerite (MD)		Fresh	1.40	1.52	Poor	
Dolerite (MD)		Fresh	3.08	1.25	Good	
Dolerite (MD)		Fresh	2.90	0.14	Good	
Dolerite (MD)	Fresh	3.03	0.13	Good		

4.8 Naturally occurring radioactive material

All samples provided by Ramelius to Landloch were tested for NORMs. None of the samples had values for either alpha/beta or gamma that exceeded the specified limit. Therefore, none of the samples give rise to enhanced exposures of radioactivity (Table 11).

Table 11: Results of NORM assessment

Measurement	Surface activity (Bq/cm ²)	Dose rate (μSv/hr)
Sample average	0.0098	0.0008
Specified limit	0.40	10

4.9 Asbestiform minerals

A subset of thirteen (13) samples was assessed for the presence of asbestiform minerals using optical microscopy, in accordance with AS 4964-2004—the standard qualitative method for bulk material analysis employing polarised light microscopy (PLM) and dispersion staining. Of these 13 samples, 7 (54%) returned indications of unknown mineral fibres. These samples were logged as basalt and dolerite from the Kopai Crescent area, as well as footwall and dolerite from the Bombora deposit.

Although these 7 samples were not associated with any known asbestos seams, they may contain trace amounts of asbestiform minerals, particularly where alteration or mineral enrichment has occurred.

In line with standard industry procedure and as permitted under the modified application of AS 4964-2004, the samples were subjected to further analysis using scanning electron microscopy (SEM) coupled with energy dispersive spectroscopy (EDS) to determine the mineralogical composition and morphological characteristics of the fibres.

The samples were first assessed to determine if asbestos fibres were visible to the naked eye or under low powered microscope. All 7 samples returned a “no fibrous mineral” visually apparent in first inspection. Subsamples were then taken and analysed using SEM to 2,000x magnification.

SEM/EDS analysis confirmed that all 7 samples contained asbestos mineral fibres at concentrations below the limit of reporting (LoR) of 0.01 wt%.

At Kopai Crescent, the asbestos-type and mineral fibres were identified as:

- Hornblende (82.1%);
- Chlorite group (10.7%); and
- Actinolite (7.1%).

At Bombora, the asbestos-type and mineral fibres were identified as:

- Chlorite (47.6%);
- Hornblende (28.6%);
- Halite (14.3%);
- Albite (4.8%); and
- Iron oxide (4.8%).

Hornblende and Actinolite are the only asbestos-type minerals identified under SEM that fall under the Amphibole classification in the DEMIRS guidelines (Section 3.3.9). Although the identified fibres meet the broader morphological and compositional definitions of asbestos, none exceeded 0.01 wt%, and therefore pose no risk to human health under current occupational exposure standards.

A summary of the characteristics of the samples containing asbestos mineral fibres is provided in Table 12.

Table 12: Characteristics of samples returning asbestos mineral fibre,

Deposit	Sample ID	Lithology	Weathering	Wt % of Asbestos in Sample	Depth From (m)	Depth To (m)	No. of Asbestos Fibres Observed	Assigned Dominant Mineralogy	Mineral Type 2 (subordinate)	Mineral Type 3 (subordinate)
Kopai Crescent	RWC11	Basalt	Transitional	<0.01	10	11	<3	Hornblende		
	RWC21	Dolerite	Fresh	<0.01	30	31	<3	Hornblende		
	RWC32	Dolerite	Transitional	<0.01	15	16	<5	Hornblende	Chlorite	Actinolite
	RWC56	Dolerite	Transitional	<0.01	5	6	<5	Hornblende	Chlorite	Actinolite
Bombara	RWC79	Footwall	Transitional	<0.01	6	8	<5	Chlorite	Halite	
	RWC123	Dolerite	Transitional	<0.01	14	16	<5	Hornblende	Chlorite	Albite
	RWC128	Dolerite	Fresh	<0.01	592	594	<5	Chlorite	Chlorite	Fe oxide

5. KEY FINDINGS AND RECOMMENDATIONS

Overall, the results of test work regime suggests that waste materials from Kopai Crescent and Bombara present low risk of generating acid and metalliferous drainage. Wastes from these deposits are not expected to give rise to enhanced exposure of NORMs. These findings are consistent with the results of previous characterisation test work undertaken for Bombara.

PAF quartz dolerite waste material was encountered at the Bombara deposit between 182.5–184 m, approaching the maximum planned Tura underground depth. All quartz dolerite samples from depths shallower than this returned a NAF classification. This suggests that PAF may be encountered at Tura underground at depths beyond 182 m. If mining is planned to extend beyond 180 m, further sampling should be conducted of materials below this depth to help delineate locations of sulphide mineralisation and estimate the volumes of PAF that may be intersected (at the time of this study, the number of drill holes within Tura underground extending >200m below ground level was limited). If the anticipated volume of PAF is less than 1% of the total waste volume, co-mingling PAF with NAF may be a viable management strategy. For volumes exceeding 1% of the total waste volume, PAF may need to be managed via encapsulation within a designated PAF cell.

Seven (7) samples analysed were found to contain asbestos mineral fibres. These 7 samples originated from a number of lithologies (basalt, dolerite, and footwall) across different locations within the deposit. The samples were not part of a known asbestos seam. There is potential for waste material to contain traces of asbestos where it occurs in zones where alteration or mineral enrichment has occurred. Although some of the identified fibres meet the broader morphological and compositional definitions of asbestos, none exceeded 0.01 wt%, and therefore pose no risk to human health under current occupational exposure standards.

Waste materials from both Kopai Crescent and Bombara are considered suitable for use as growth media based on pH and salinity. Drainage from these materials is likely to generate salinity <4 dS/m. It is expected that vegetation in the vicinity of the project has some degree of salt tolerance however due to the presence of saline soils. Salt tolerant plants have also been observed to grow on these soils. It is recommended that salt tolerant species be included in the rehabilitation seed mix to account for the presence of these plant species *in situ* soils and promote vegetation establishment. The effect of saline drainage on the receiving environment will be dependent on the existing salinity of groundwater.

The post-mining matrix of the majority of waste material placed at surface is likely to be predominantly coarse-grained based on the weathering state (transitional and fresh). These materials are likely to remain stable when used in rehabilitation. Further physical testing of the waste materials is recommended once they become available in mining, to re-assess their physical characteristics and susceptibility to structural instability. There is potential for the fresh quartz dolerite and transitional and fresh dolerite at both deposits to be used in rehabilitation as rock armour, provided they retain a predominantly coarse-grained composition post-mining (>60%). The inclusion of rock armour on the outer surface of landforms may help to improve their resistance to erosion.

A small proportion of waste placed in waste rock landforms will be oxidised (<10%) and likely to be prone to structural instability. This waste type is likely to be dominated by fine grained particles and clay minerals and should not be placed at the surface of the landform (i.e. encapsulated).

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APPENDIX A: SAMPLE DETAILS AND DEPOSIT CROSS SECTIONS

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Table A-1: Sample details for Kopai Crescent.

Drill Hole ID	Sample ID	Lithology Code	Lithology	Oxidation	Depth From (m)	Depth To (m)	Location
BBRC1913	RWC01	MB	Basalt	Transitional	10	11	KC 4700
BBRC1913	RWC02	MB	Basalt	Transitional	11	12	KC 4700
BBRC1913	RWC03	MB	Basalt	Transitional	12	13	KC 4700
BBRC1913	RWC04	MB	Basalt	Transitional	13	14	KC 4700
BBRC1913	RWC05	MB	Basalt	Transitional	14	15	KC 4700
BBRC1913	RWC06	MB	Basalt	Transitional	15	16	KC 4700
BBRC1913	RWC07	MB	Basalt	Transitional	16	17	KC 4700
BBRC1913	RWC08	MB	Basalt	Transitional	17	18	KC 4700
BBRC1913	RWC09	MB	Basalt	Transitional	18	19	KC 4700
BBRC1913	RWC10	MB	Basalt	Transitional	19	20	KC 4700
BBRC1914	RWC11	MB	Basalt	Transitional	10	11	KC 4700
BBRC1914	RWC12	MB	Basalt	Transitional	11	12	KC 4700
BBRC1914	RWC13	MB	Basalt	Transitional	12	13	KC 4700
BBRC1914	RWC14	MB	Basalt	Transitional	13	14	KC 4700
BBRC1914	RWC15	MB	Basalt	Transitional	14	15	KC 4700
BBRC1930	RWC16	MD	Dolerite	Transitional	10	11	KC 4700
BBRC1930	RWC17	MD	Dolerite	Transitional	11	12	KC 4700
BBRC1930	RWC18	MD	Dolerite	Transitional	12	13	KC 4700
BBRC1930	RWC19	MD	Dolerite	Transitional	13	14	KC 4700
BBRC1930	RWC20	MD	Dolerite	Transitional	14	15	KC 4700
BBRC1937	RWC21	MD	Dolerite	Fresh	30	31	KC 4700
BBRC1937	RWC22	MD	Dolerite	Fresh	31	32	KC 4700
BBRC1937	RWC23	MD	Dolerite	Fresh	32	33	KC 4700
BBRC1938	RWC24	MD	Dolerite	Oxide	7	8	KC 4700
BBRC1938	RWC25	MD	Dolerite	Oxide	8	9	KC 4700
BBRC1938	RWC26	MD	Dolerite	Oxide	9	10	KC 4700
BBRC1944	RWC27	MD	Dolerite	Transitional	10	11	KC 4700
BBRC1944	RWC28	MD	Dolerite	Transitional	11	12	KC 4700
BBRC1944	RWC29	MD	Dolerite	Transitional	12	13	KC 4700
BBRC1944	RWC30	MD	Dolerite	Transitional	13	14	KC 4700
BBRC1944	RWC31	MD	Dolerite	Transitional	14	15	KC 4700
BBRC1950	RWC32	MD	Dolerite	Transitional	15	16	KC 4700
BBRC1950	RWC33	MD	Dolerite	Transitional	16	17	KC 4700
BBRC1950	RWC34	MD	Dolerite	Transitional	17	18	KC 4700
BBRC1950	RWC35	MD	Dolerite	Transitional	18	19	KC 4700
BBRC1950	RWC36	MD	Dolerite	Transitional	19	20	KC 4700
BBRC1974	RWC37	MD	Dolerite	Transitional	5	6	KC 4700
BBRC1974	RWC38	MD	Dolerite	Transitional	6	7	KC 4700
BBRC1974	RWC39	MD	Dolerite	Transitional	7	8	KC 4700
BBRC1974	RWC40	MD	Dolerite	Transitional	8	9	KC 4700
BBRC1974	RWC41	MD	Dolerite	Transitional	9	10	KC 4700
BBRC1977	RWC42	MD	Dolerite	Transitional	10	11	KC 4700
BBRC1977	RWC43	MD	Dolerite	Transitional	11	12	KC 4700

Drill Hole ID	Sample ID	Lithology Code	Lithology	Oxidation	Depth From (m)	Depth To (m)	Location
BBRC1977	RWC44	MD	Dolerite	Transitional	12	13	KC 4700
BBRC1977	RWC45	MD	Dolerite	Transitional	13	14	KC 4700
BBRC1977	RWC46	MD	Dolerite	Transitional	14	15	KC 4700
BBRC1979	RWC47	MD	Dolerite	Transitional	9	10	KC 4900
BBRC1979	RWC48	MD	Dolerite	Transitional	10	11	KC 4900
BBRC1979	RWC49	MD	Dolerite	Transitional	11	12	KC 4900
BBRC1979	RWC50	MD	Dolerite	Transitional	12	13	KC 4900
BBRC1984	RWC51	MD	Dolerite	Transitional	5	6	KC 4900
BBRC1984	RWC52	MD	Dolerite	Transitional	6	7	KC 4900
BBRC1984	RWC53	MD	Dolerite	Transitional	7	8	KC 4900
BBRC1984	RWC54	MD	Dolerite	Transitional	8	9	KC 4900
BBRC1984	RWC55	MD	Dolerite	Transitional	9	10	KC 4900
BBRC2007	RWC56	MD	Dolerite	Transitional	5	6	KC 5750
BBRC2007	RWC57	MD	Dolerite	Transitional	6	7	KC 5750
BBRC2007	RWC58	MD	Dolerite	Transitional	7	8	KC 5750
BBRC2007	RWC59	MD	Dolerite	Transitional	8	9	KC 5750
BBRC2007	RWC60	MD	Dolerite	Transitional	9	10	KC 5750
BBRC2011	RWC61	MD	Dolerite	Transitional	15	16	KC 5750
BBRC2011	RWC62	MD	Dolerite	Transitional	16	17	KC 5750
BBRC2011	RWC63	MD	Dolerite	Transitional	17	18	KC 5750
BBRC2011	RWC64	MD	Dolerite	Transitional	18	19	KC 5750
BBRC2011	RWC65	MD	Dolerite	Transitional	19	20	KC 5750
BBRC2014	RWC66	MD	Dolerite	Transitional	8	9	KC 5750
BBRC2014	RWC67	MD	Dolerite	Transitional	9	10	KC 5750
BBRC2014	RWC68	MD	Dolerite	Transitional	10	11	KC 5750
BBRC2014	RWC69	MD	Dolerite	Transitional	11	12	KC 5750
BBRC2014	RWC70	MD	Dolerite	Transitional	12	13	KC 5750

Table A-2: Sample details for Bombora.

Drill Hole ID	Sample ID	Lithology Code	Lithology	Oxidation	Depth From (m)	Depth To (m)	Location
BBDD0135	RWC71	ALP	Lamprophyre	Fresh	160.85	162.21	BOM1800
BBDD0146	RWC72	ALP	Lamprophyre	Fresh	105.42	107.01	Tura UG
BBDD0178	RWC73	ALP	Lamprophyre	Fresh	426	428	KL UG
BBDD0134	RWC74	MDQ	Q. dolerite	Trans	5	7.1	BOM1100
BBDD0134	RWC75	MDQ	Q. dolerite	Transitional	8	10	BOM1100
BBDD0134	RWC76	MDQ	Q. dolerite	Transitional	10	12	BOM1100
BBDD0134	RWC77	MDQ	Q. dolerite	Transitional	12	14	BOM1100
BBDD0134	RWC78	MDQ	Q. dolerite	Transitional	14	16	BOM1100
BBDD0135	RWC79	MR_FW	Footwall	Transitional	6	8	BOM1800
BBDD0135	RWC80	MR_FW	Footwall	Transitional	8	10	BOM1800
BBDD0135	RWC81	MR_FW	Footwall	Transitional	10	12	BOM1800
BBDD0135	RWC82	MR_FW	Footwall	Transitional	12	14	BOM1800
BBDD0135	RWC83	MR_FW	Footwall	Fresh	147	149	BOM1800
BBDD0135	RWC84	MR_FW	Footwall	Fresh	149	151	BOM1800
BBDD0135	RWC85	MR_FW	Footwall	Fresh	151	153	BOM1800
BBDD0135	RWC86	MR_FW	Footwall	Fresh	153	155	BOM1800
BBDD0135	RWC87	MR_FW	Footwall	Fresh	155	157	BOM1800
BBDD0137	RWC88	MDQ	Q. dolerite	Transitional	5	7	BOM1100
BBDD0137	RWC89	MDQ	Q. dolerite	Transitional	7	9	BOM1100
BBDD0137	RWC90	MDQ	Q. dolerite	Transitional	9	11	BOM1100
BBDD0137	RWC91	MDQ	Q. dolerite	Transitional	11	13	BOM1100
BBDD0137	RWC92	MDQ	Q. dolerite	Transitional	13	15	BOM1100
BBDD0137	RWC93	MDQ	Q. dolerite	Transitional	15	17	BOM1100
BBDD0147	RWC94	MDQ	Q. dolerite	Transitional	6.9	9	BOM1800
BBDD0147	RWC95	MDQ	Q. dolerite	Transitional	11	13	BOM1800
BBDD0147	RWC96	MDQ	Q. dolerite	Fresh	78	80	BOM1800
BBDD0147	RWC97	MDQ	Q. dolerite	Fresh	89.37	90.71	BOM1800
BBDD0147	RWC98	MDQ	Q. dolerite	Fresh	90.71	92	BOM1800
BBDD0147	RWC99	MDQ	Q. dolerite	Fresh	102	104	BOM1800
BBDD0147	RWC100	MDQ	Q. dolerite	Fresh	104	106	BOM1800
BBDD0147	RWC101	MDQ	Q. dolerite	Fresh	106	108.09	BOM1800
BBDD0147	RWC102	MDQ	Q. dolerite	Fresh	108.72	110	BOM1800
BBDD0147	RWC103	MDQ	Q. dolerite	Fresh	110	112	BOM1800
BBDD0147	RWC104	MDQ	Q. dolerite	Fresh	112	114	BOM1800
BBDD0161	RWC105	MDQ	Q. dolerite	Fresh	175	177	Tura UG
BBDD0161	RWC106	MDQ	Q. dolerite	Fresh	177	179	Tura UG
BBDD0161	RWC107	MDQ	Q. dolerite	Fresh	179	181	Tura UG
BBDD0161	RWC108	MDQ	Q. dolerite	Fresh	181	182.5	Tura UG
BBDD0161	RWC109	MDQ	Q. dolerite	Fresh	182.5	184	Tura UG
BBDD0154W1	RWC110	MDQ	Q. dolerite	Fresh	544.04	545.72	KL UG
BBDD0154W1	RWC111	MDQ	Q. dolerite	Fresh	545.72	547	KL UG
BBDD0183W1	RWC112	MDQ	Q. dolerite	Fresh	550	552	KL UG
BBDD0183W1	RWC113	MDQ	Q. dolerite	Fresh	552	554	KL UG
BBDD0183W1	RWC114	MDQ	Q. dolerite	Fresh	554	556	KL UG

Drill Hole ID	Sample ID	Lithology Code	Lithology	Oxidation	Depth From (m)	Depth To (m)	Location
BBDD0175	RWC115	MDQ	Q. dolerite	Transitional	5	7	BOM 0700, above Tura UG
BBDD0175	RWC116	MDQ	Q. dolerite	Transitional	7	9	BOM 0700, above Tura UG
BBDD0175	RWC117	MDQ	Q. dolerite	Transitional	9	11	BOM 0700, above Tura UG
BBDD0173	RWC118	MDQ	Q. dolerite	Transitional	5	7	BOM 0700, above Tura UG
BBDD0173	RWC119	MDQ	Q. dolerite	Transitional	7	9	BOM 0700, above Tura UG
BBDD0173	RWC120	MDQ	Q. dolerite	Transitional	9	11	BOM 0700, above Tura UG
BBDD0171	RWC121	MD	Dolerite	Transitional	10	12	Tura UG
BBDD0171	RWC122	MD	Dolerite	Transitional	12	14	Tura UG
BBDD0171	RWC123	MD	Dolerite	Transitional	14	16	Tura UG
BBDD0170	RWC124	MD	Dolerite	Transitional	18	20	Tura UG
BBDD0170	RWC125	MD	Dolerite	Transitional	20	22	Tura UG
BBDD0167	RWC126	MD	Dolerite	Transitional	16	18	Tura UG
BBDD0167	RWC127	MD	Dolerite	Transitional	18	19.85	Tura UG
BBDD0148	RWC128	MD	Dolerite	Fresh	592	594	KL UG
BBDD0148	RWC129	MD	Dolerite	Fresh	594	596	KL UG
BBDD0148	RWC130	MD	Dolerite	Fresh	596	598	KL UG

Kopai Crescent

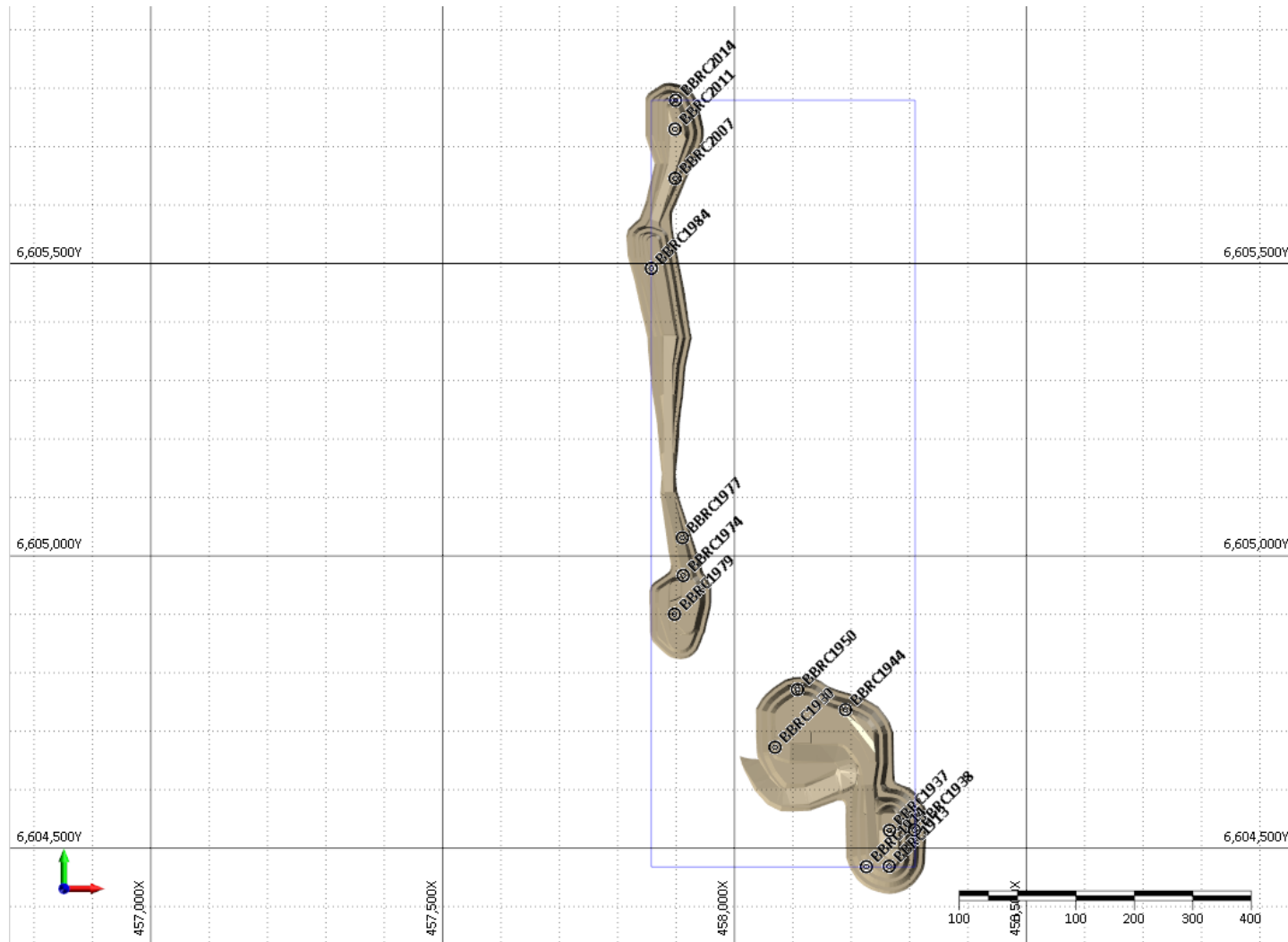


Figure A-1: Locality plan view showing selected drill holes and pit shell used for this study (Source: Ramelius).

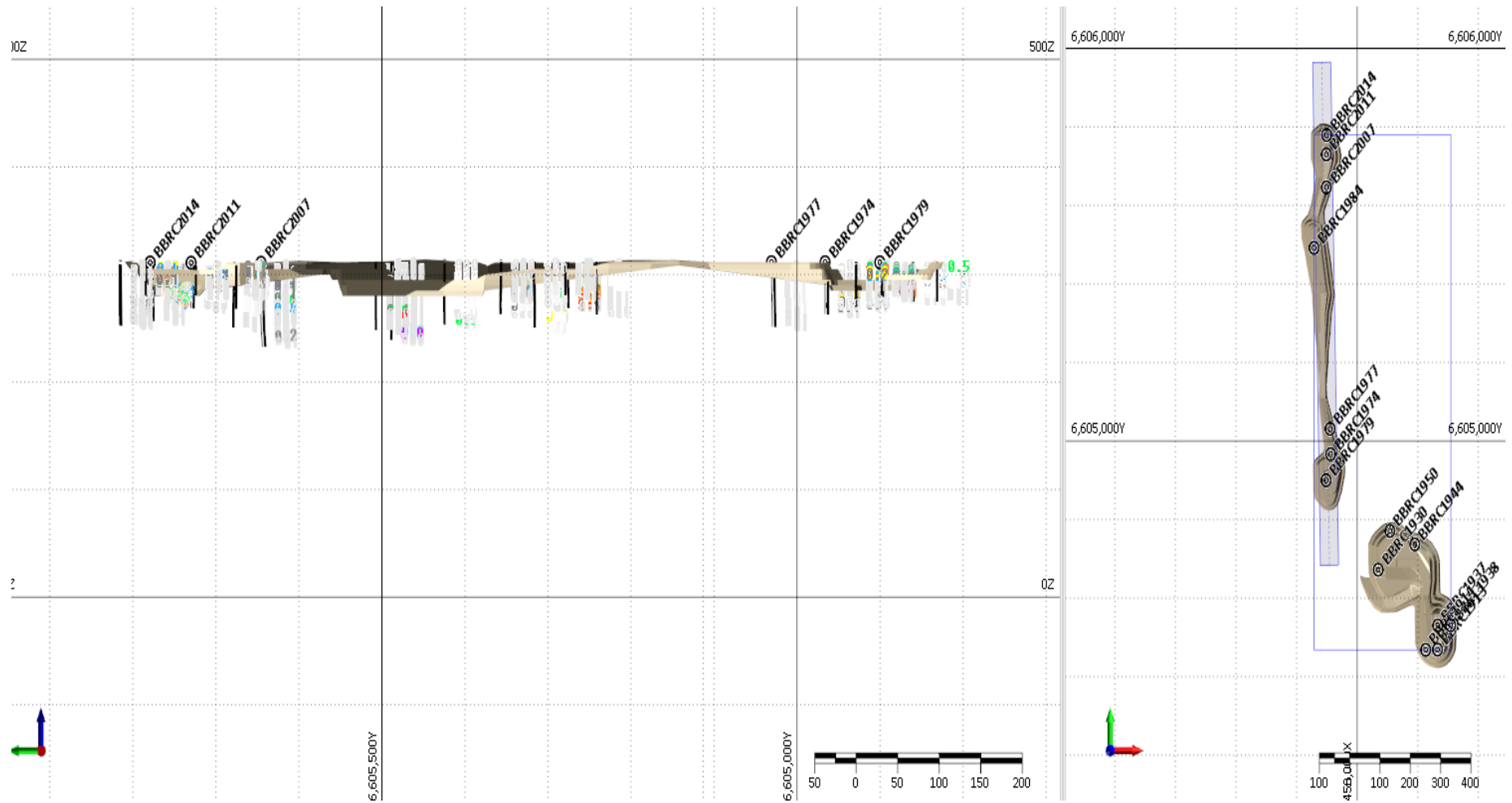


Figure A-2: Long section of the middle and northern pits (Source: Ramelius).

The following cross-sections depict the drill string coloured by lithology code, with light green representing basalt and dark green indicating dolerite. The information displayed to the right of the drill string indicates the assay grade of gold (Au) in ppm, while the depth in meters is shown on the left side.

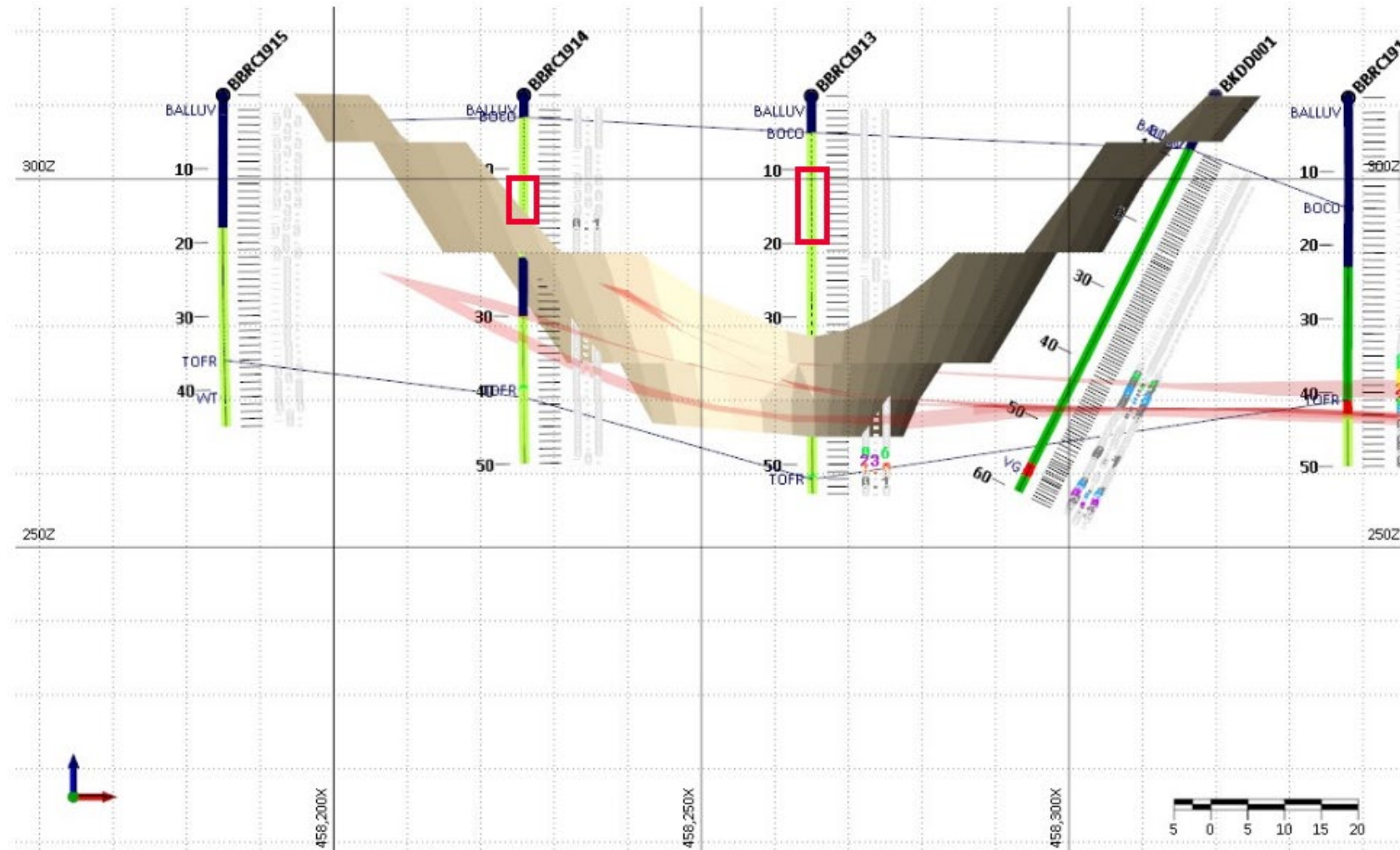


Figure A-4: 6604470N BBRC1913 samples 10-20m and BBRC1914 10-15m in transitional basalt (Source: Ramelius).

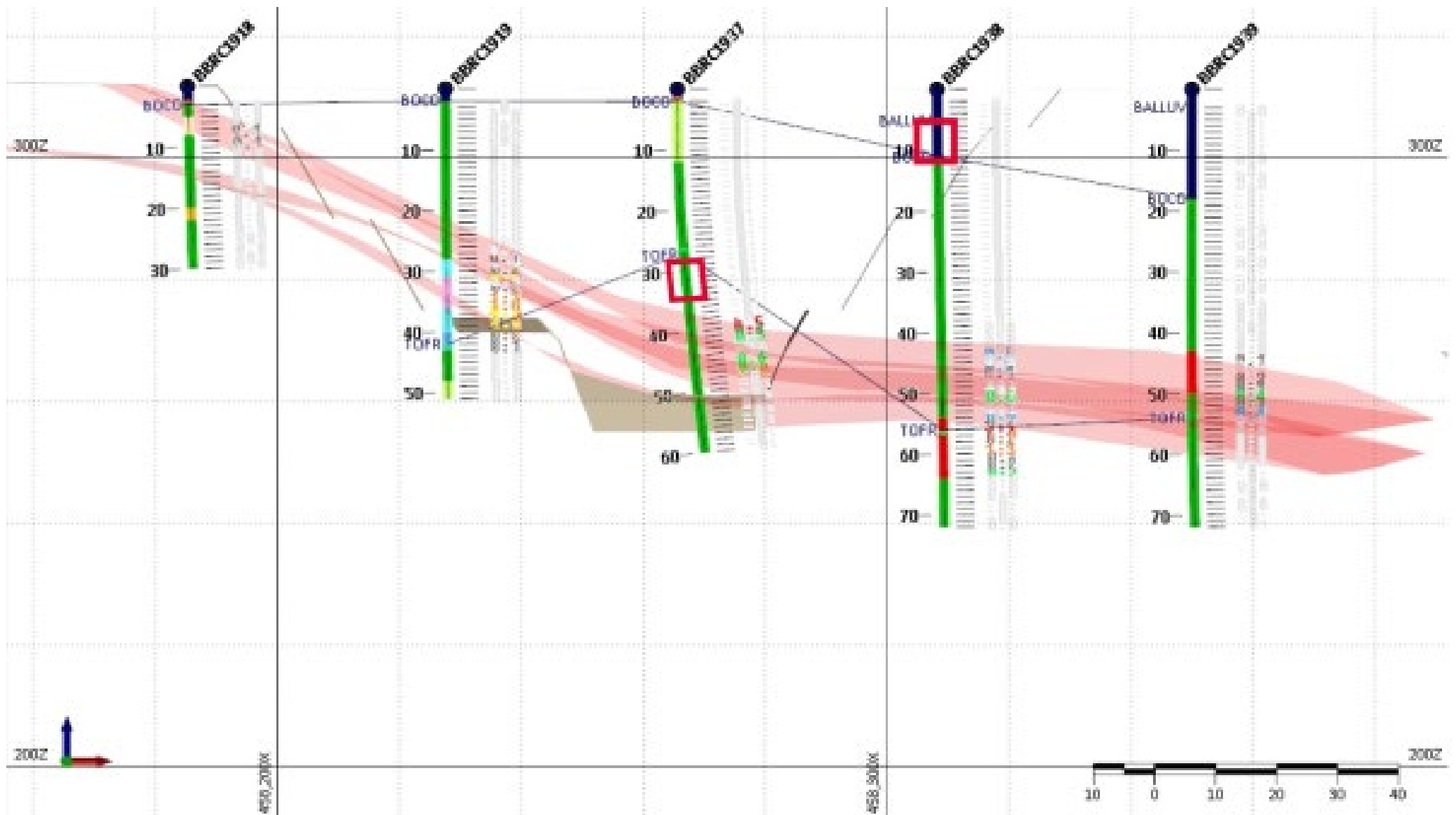


Figure A-5: 6604530N BBRC1937 28-33m fresh dolerite and BBRC1938 5-10m oxide dolerite (Source: Ramelius).

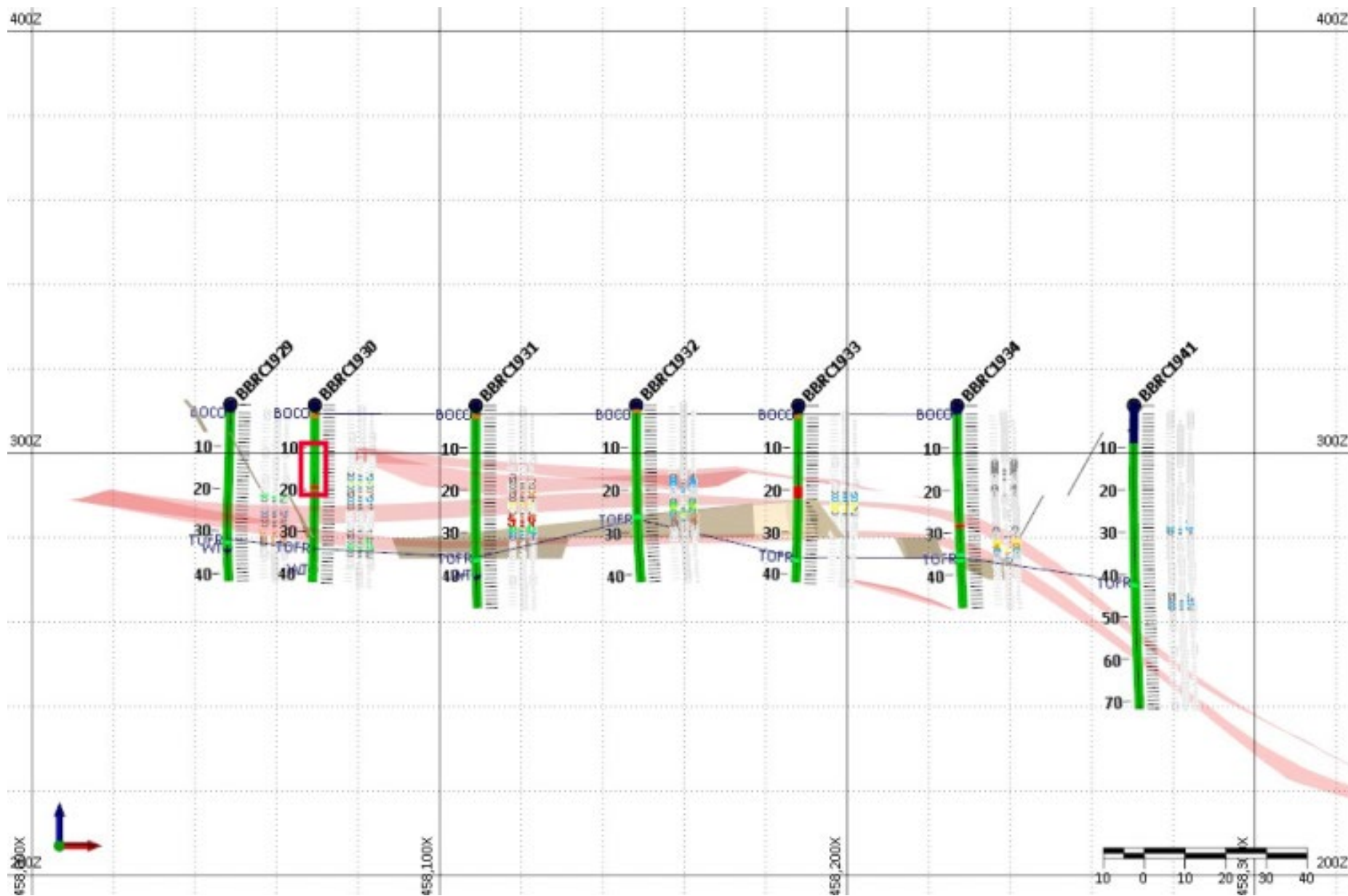


Figure A-6: 6604670N BBRC1930 10-15m transitional dolerite (Source: Ramelius).

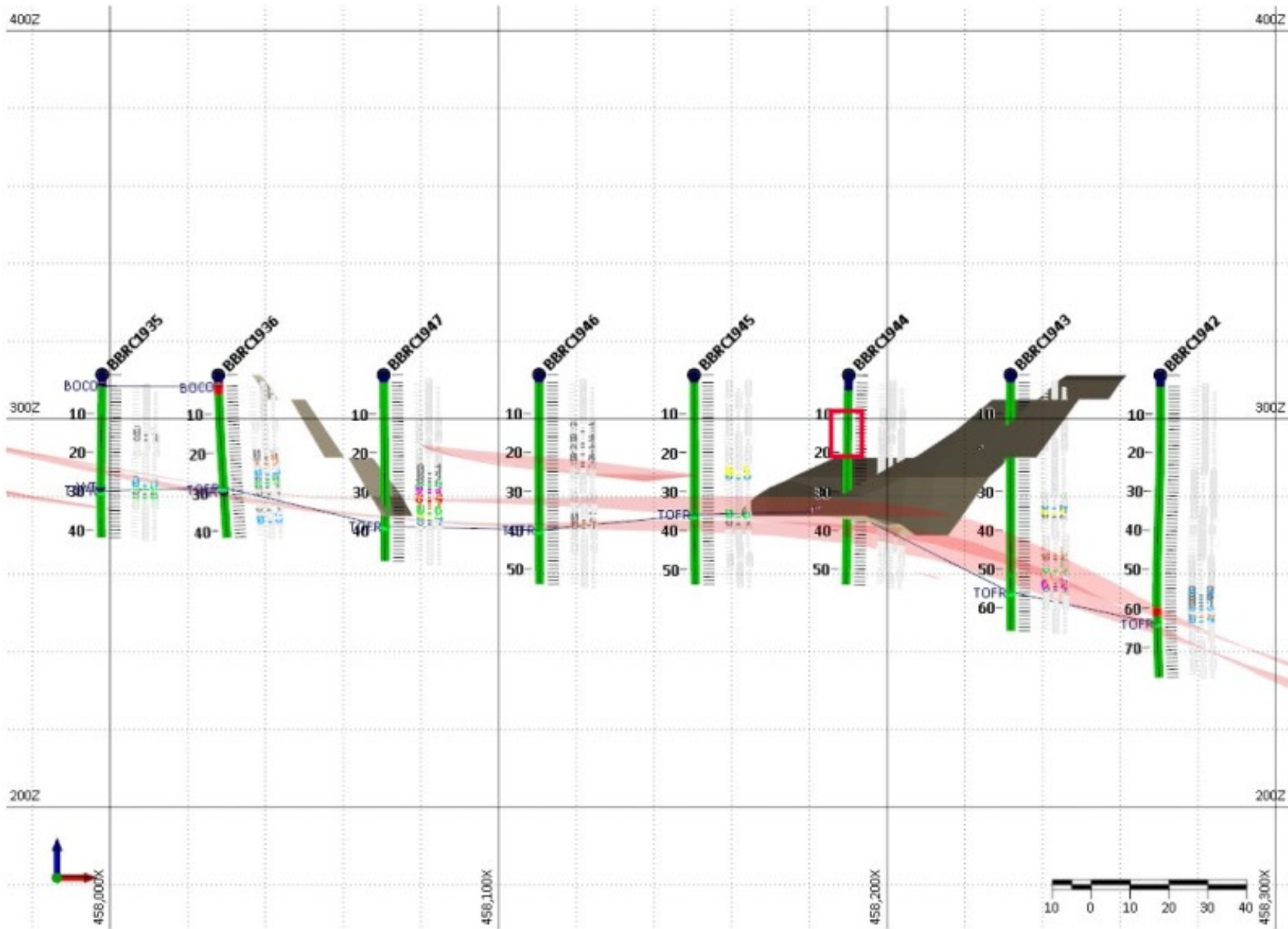


Figure A-7: 6604740N BBRC1944 10 to 15m transitional dolerite (Source: Ramelius).

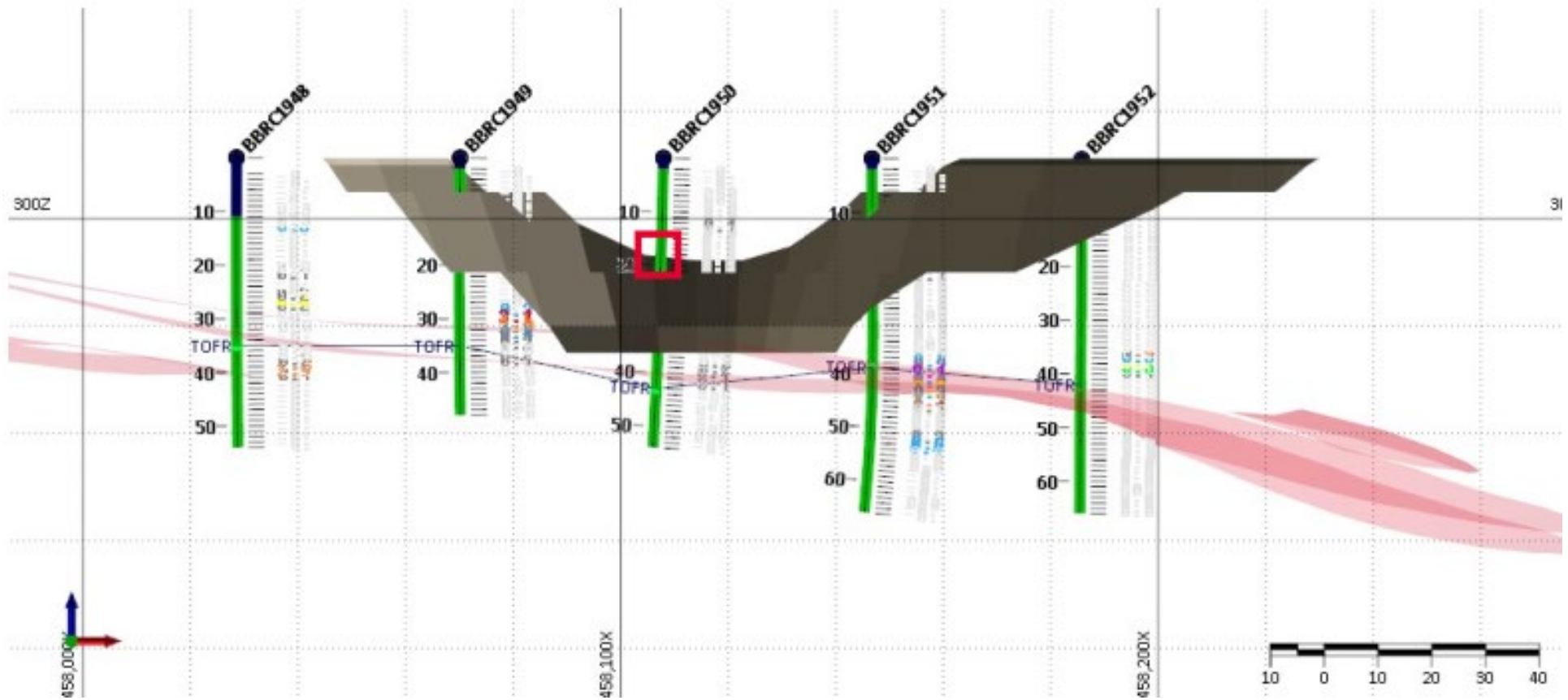


Figure A-8: 6604770N BBRC1950 15-20m transitional dolerite (Source: Ramelius).

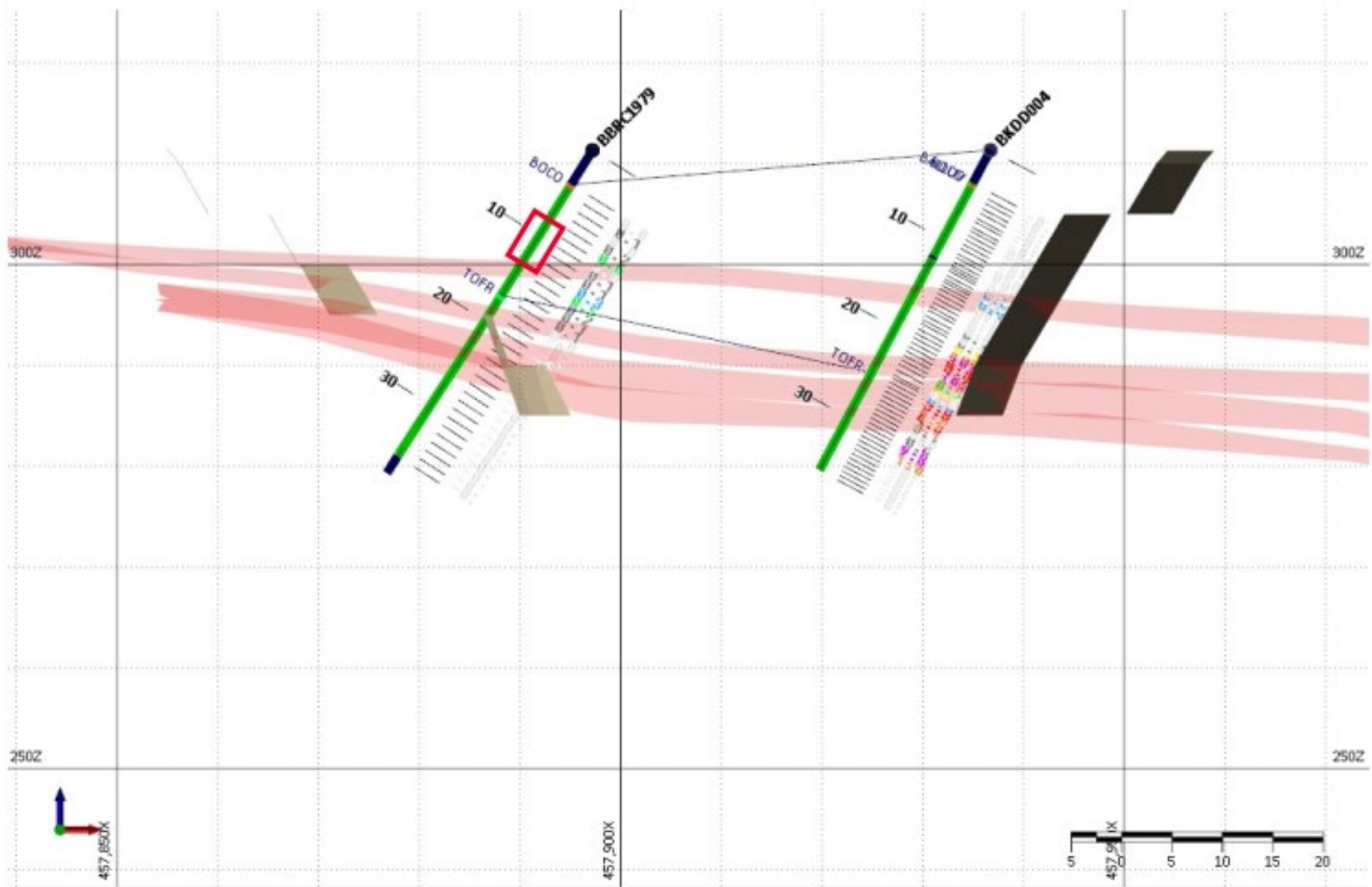


Figure A-9: 6604900N BBRC1979 9-13m transitional dolerite (Source: Ramelius).

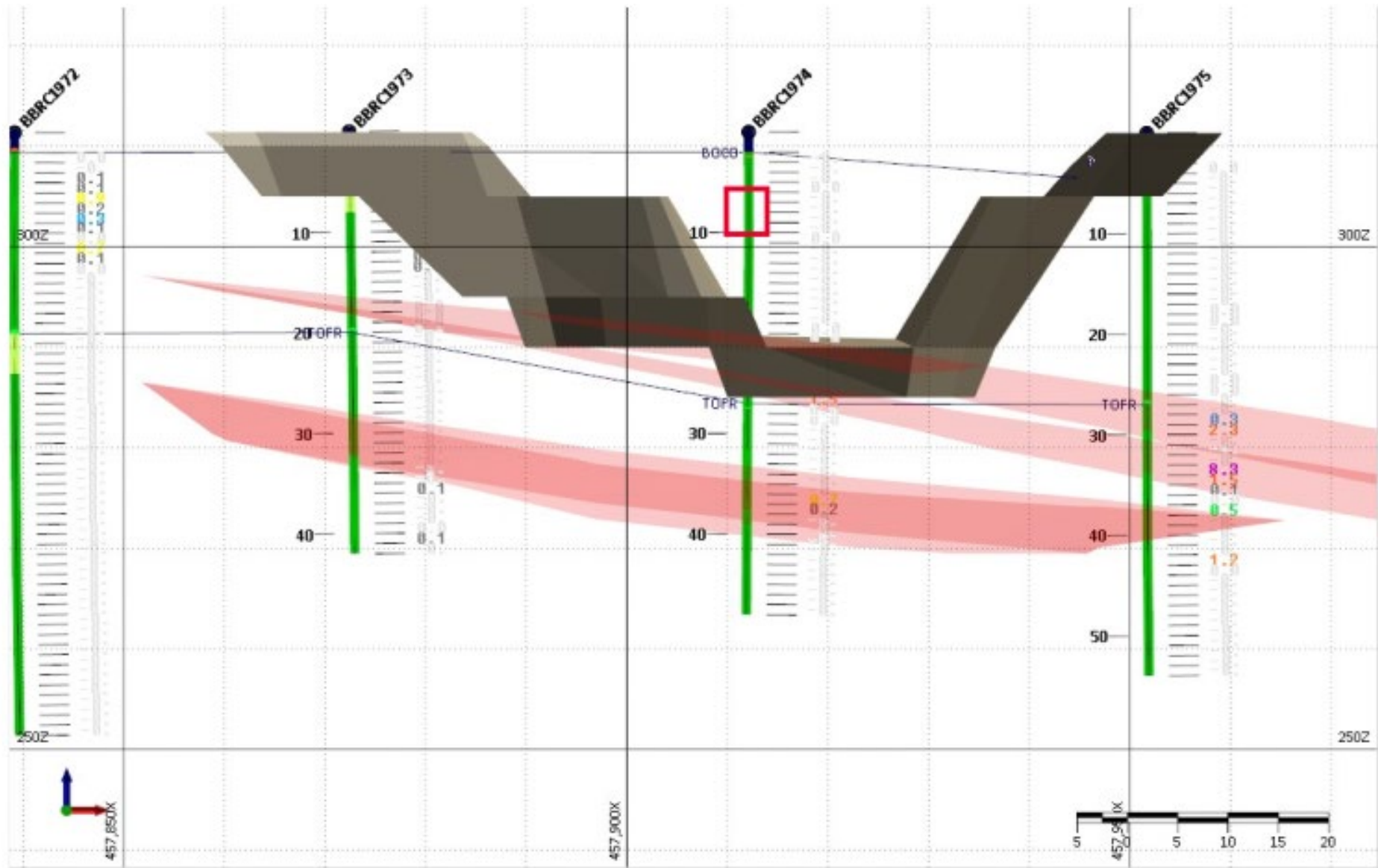


Figure A-10: 6604970 BBRC1974 5 to 10m transitional dolerite (Source: Ramelius).

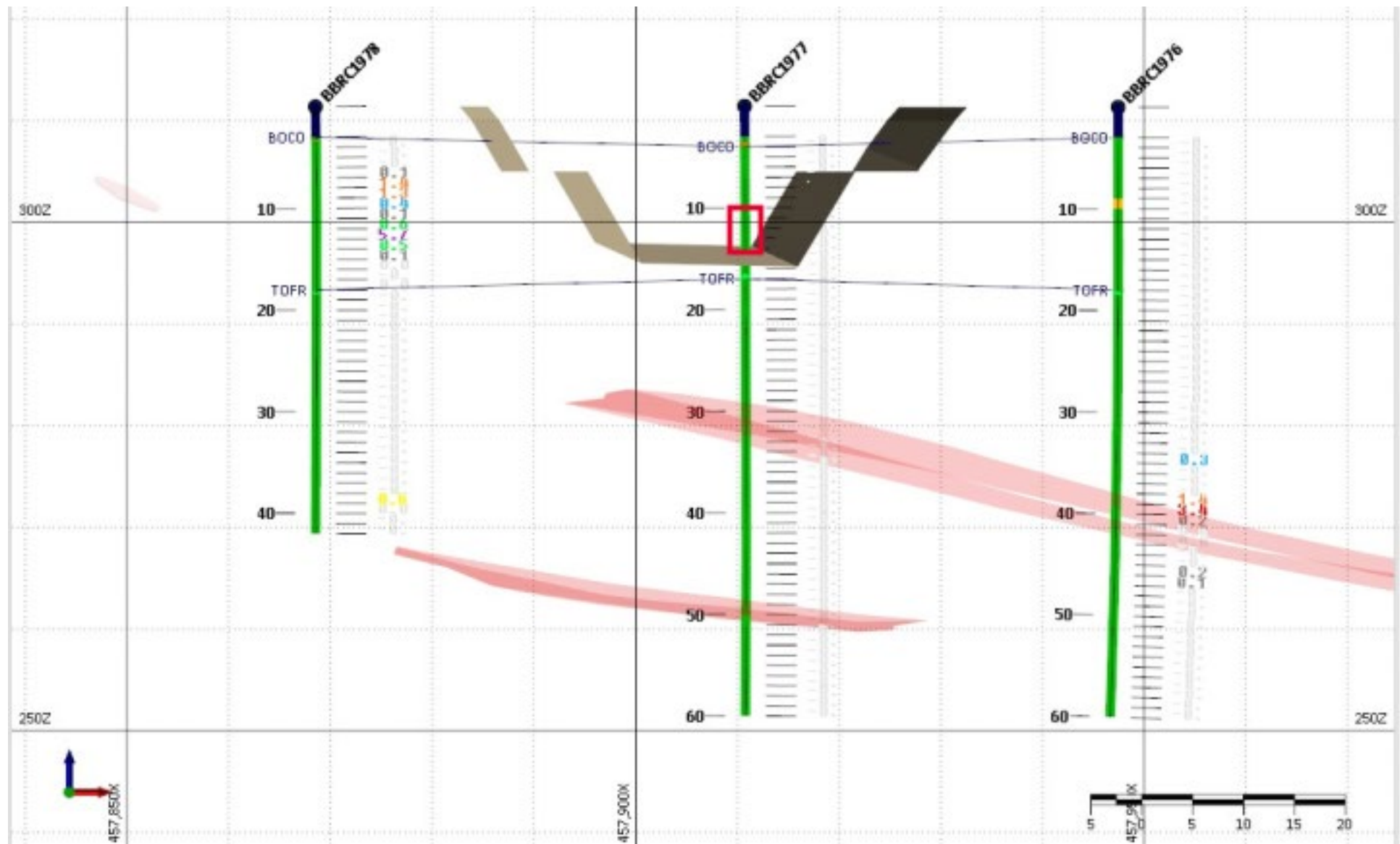


Figure A-11: 6605030N BBRC1977 10 to 15m transitional dolerite (Source: Ramelius).

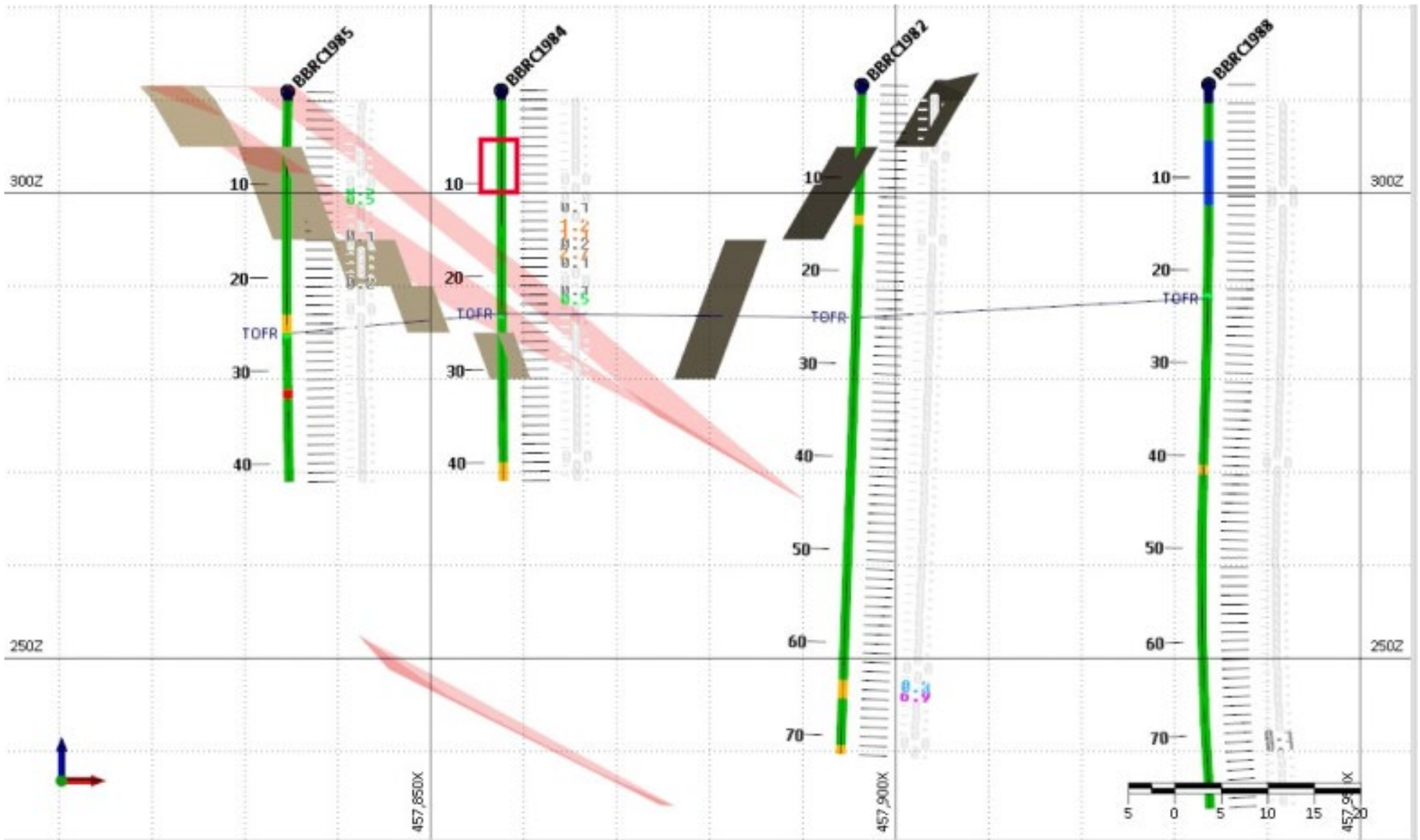


Figure A-12: 6605490N BBRC1984 5-10m transitional dolerite (Source: Ramelius).

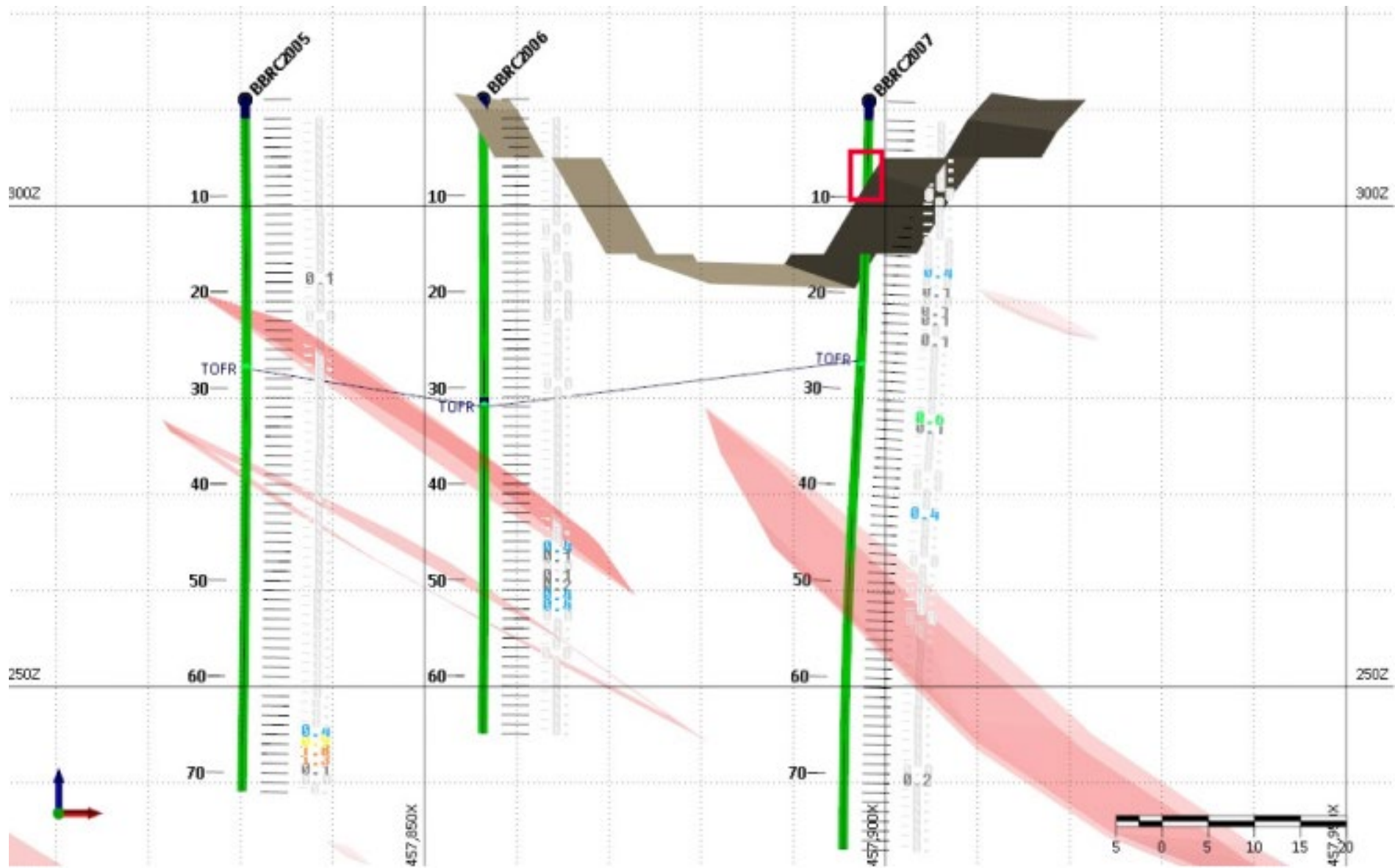


Figure A-13: 6605640N BBRC2007 5-10m transitional dolerite (Source: Ramelius).

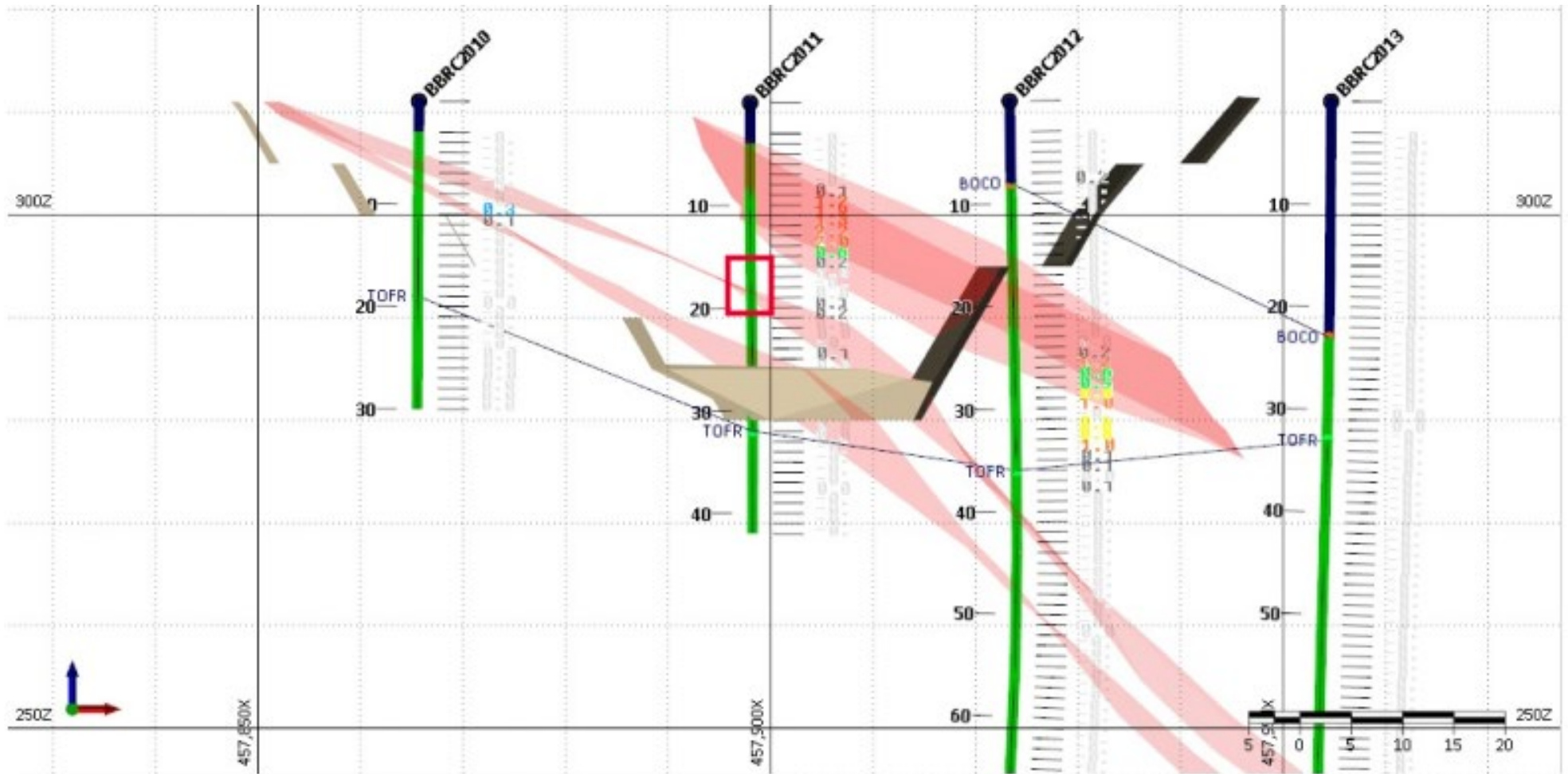


Figure A-14: 6605730N BBRC2011 15-20m transitional dolerite (Source: Ramelius).

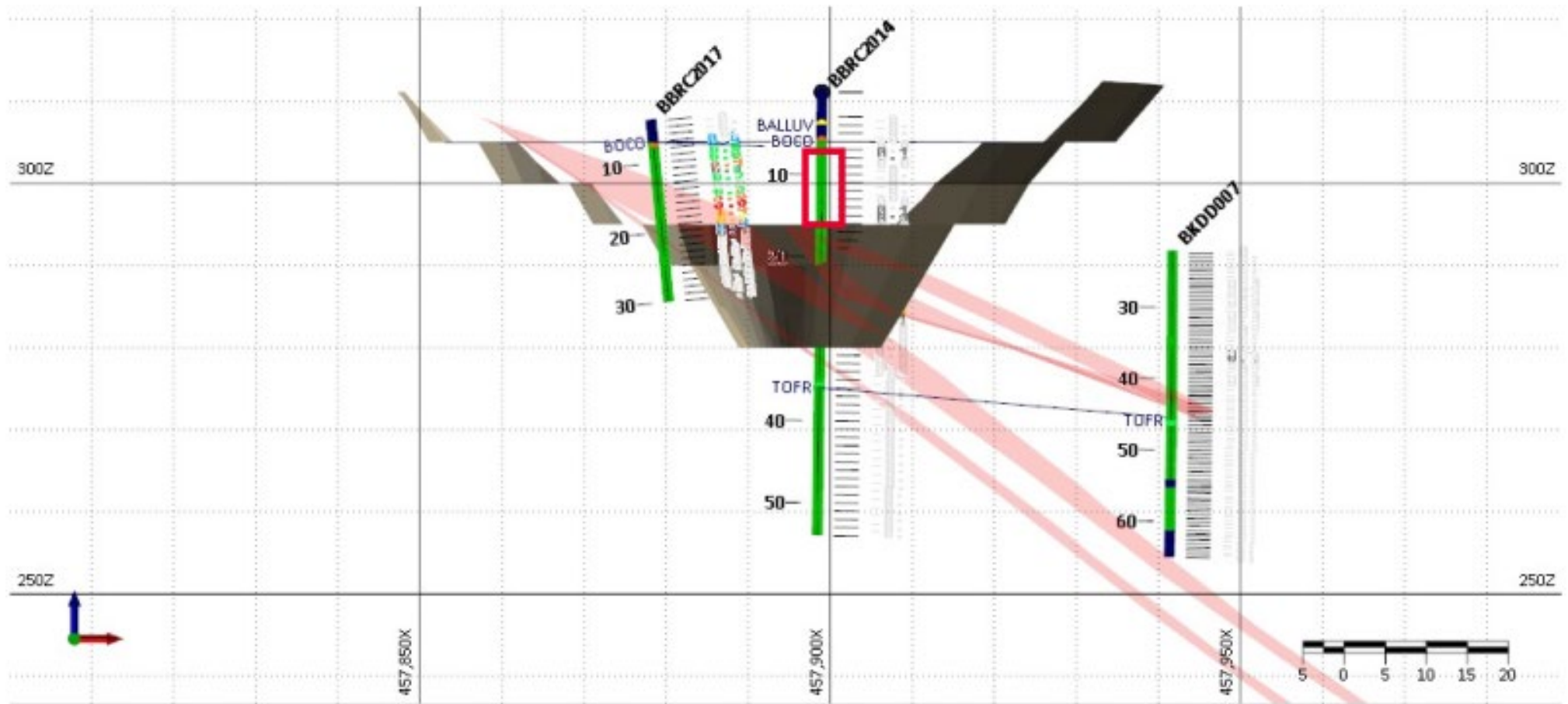


Figure A-15: 6605780 BBRC2014 8-13m transitional dolerite (Source: Ramelius).

Bombora



Figure A-16: Map showing all available drill holes <3 years old, for waste characterisation (Source: Ramelius).



Figure A-17: Showing location of selected drill holes for the waste characterisation study. Hollow points indicate 2021-2022 drilling campaign (Breaker Resources) and solid points indicate 2023-2024 Ramelius resources drilling campaign) (Source: Ramelius).

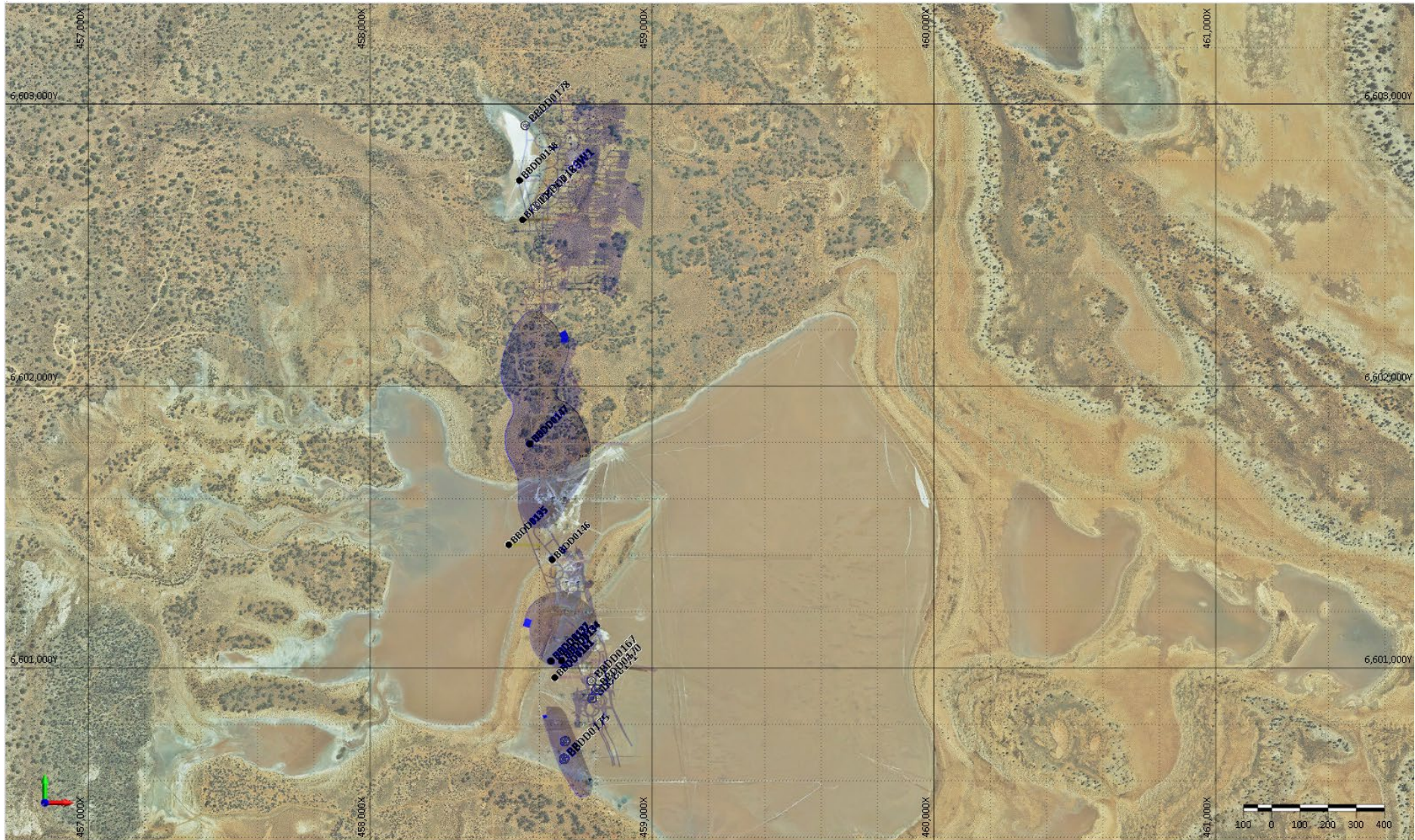


Figure A-18: Showing collar locations with planned open pit and underground development (Source: Ramelius)

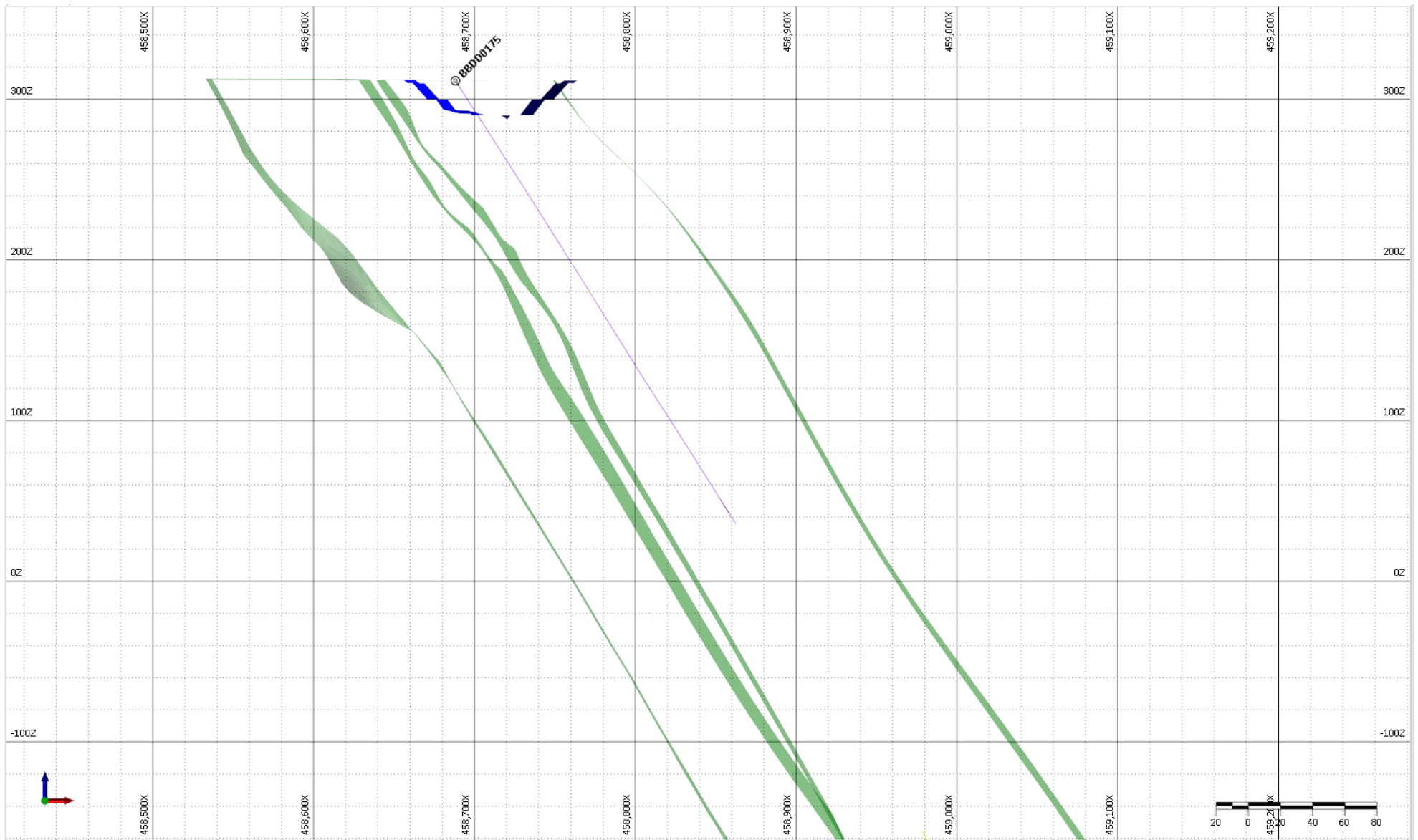


Figure A-19: BBDD0175 6600666 Northing, sampled from 5-11 in transitional weathered MDQ, within BOM0700 (Source: Ramelius).

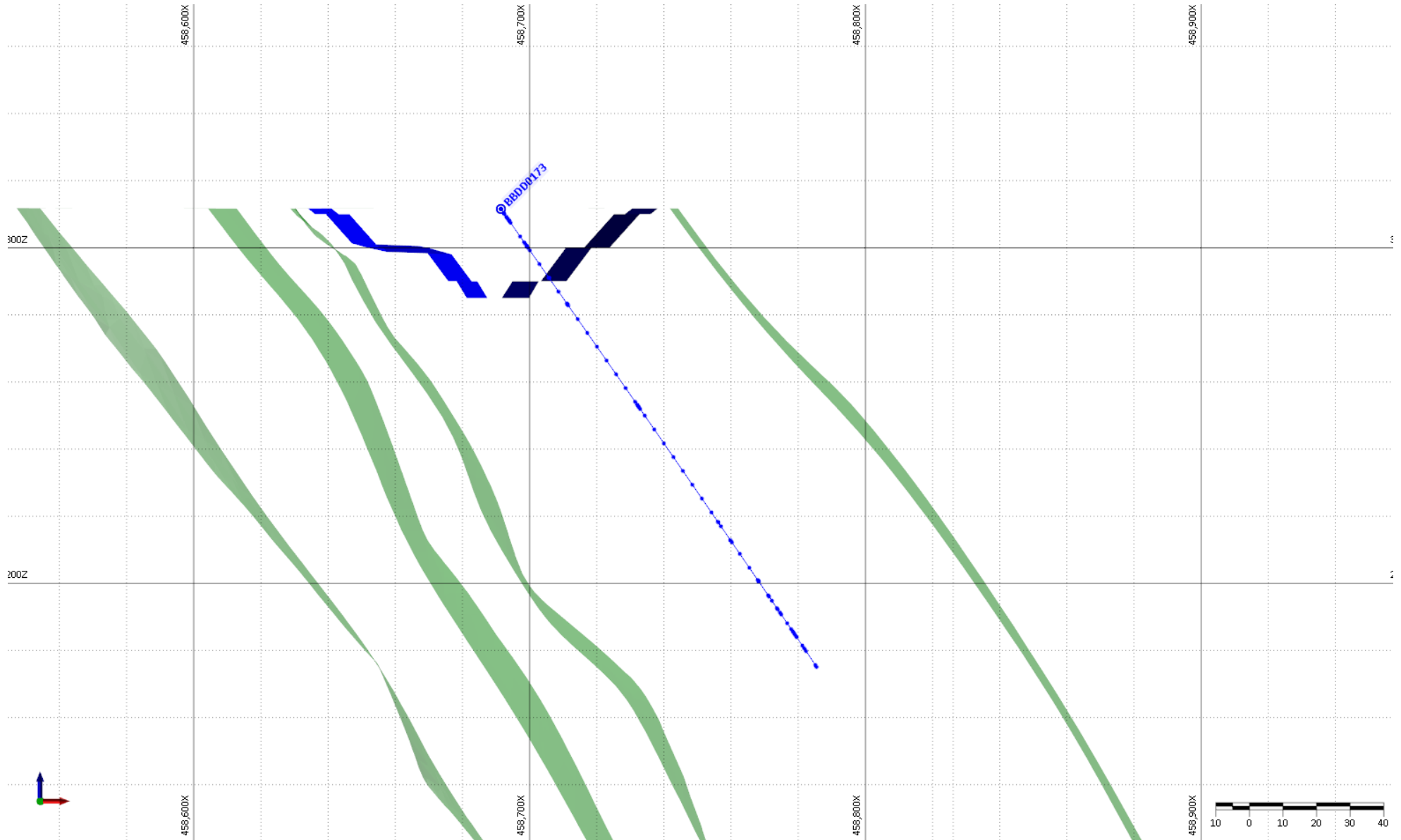


Figure A-20: BBDD0173 6600740 Northing sampled from 5-11 m in transitional MDQ within BOM 0700 (Source: Ramelius).



Figure A-21: BBDD0171 6600892 Northing sampled from 10-16m in transitional MD (Source: Ramelius).

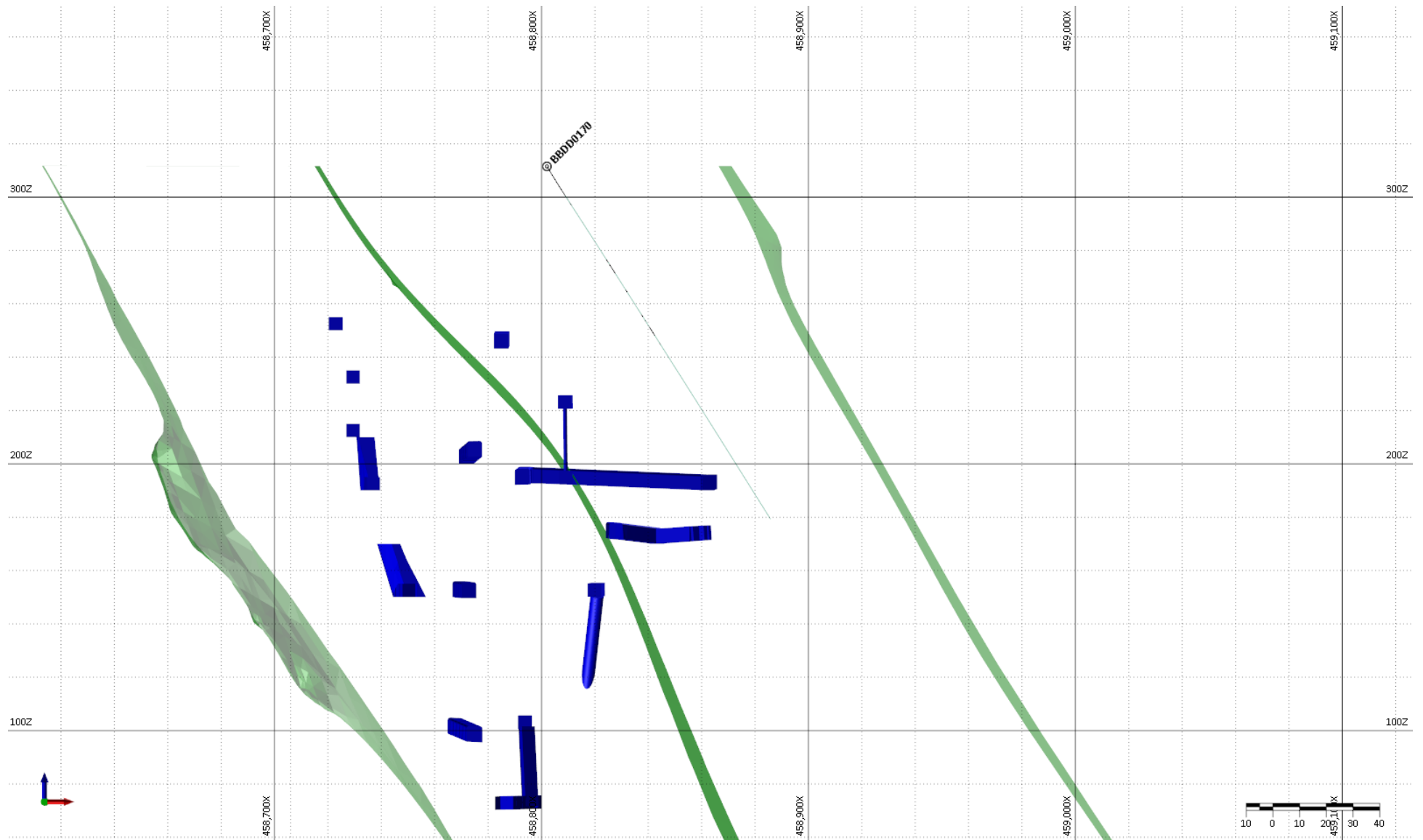


Figure A-22: BBDD0170 6600925 northing sampled from 18-22m in transitional MD close to Tura underground (Source: Ramelius).

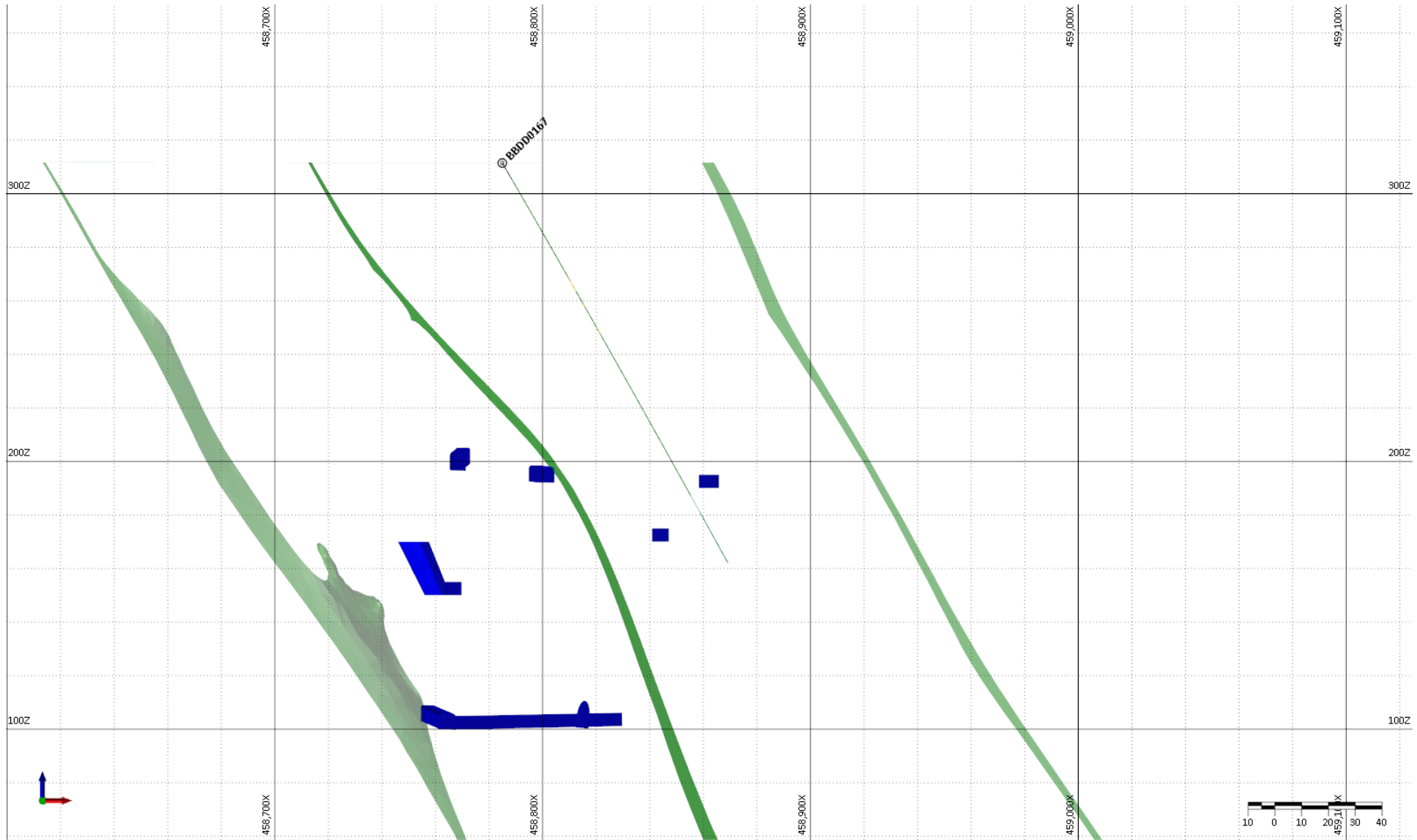


Figure A-22: BBDD0167 6600954 Northing sampled from 16-19.85 in transitional MD located above Tura underground (Source: Ramelius).

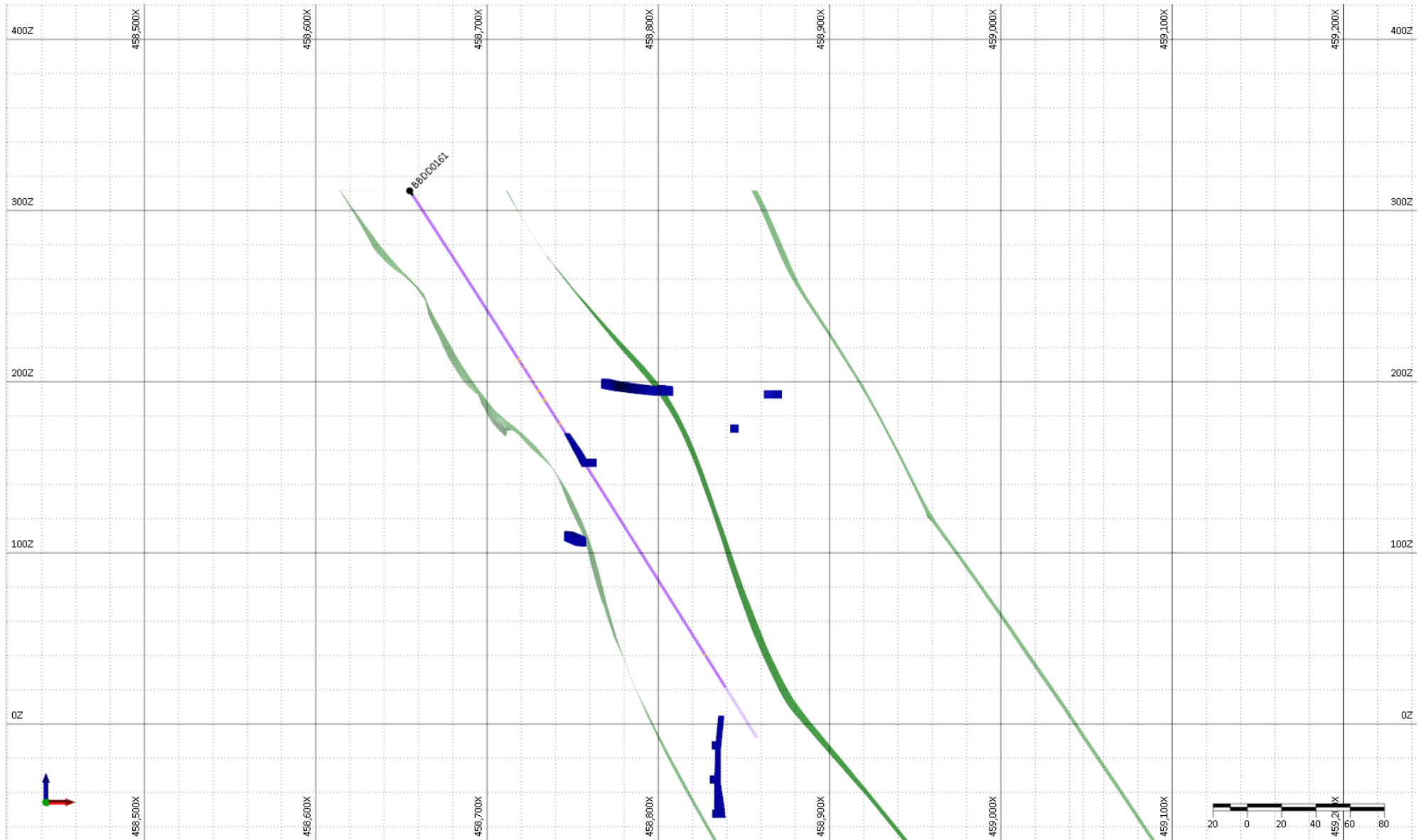


Figure A-23: BBDD0161 6600966 northing sampled from 174-184m in Fresh MDQ, within Tura underground (Source: Ramelius).

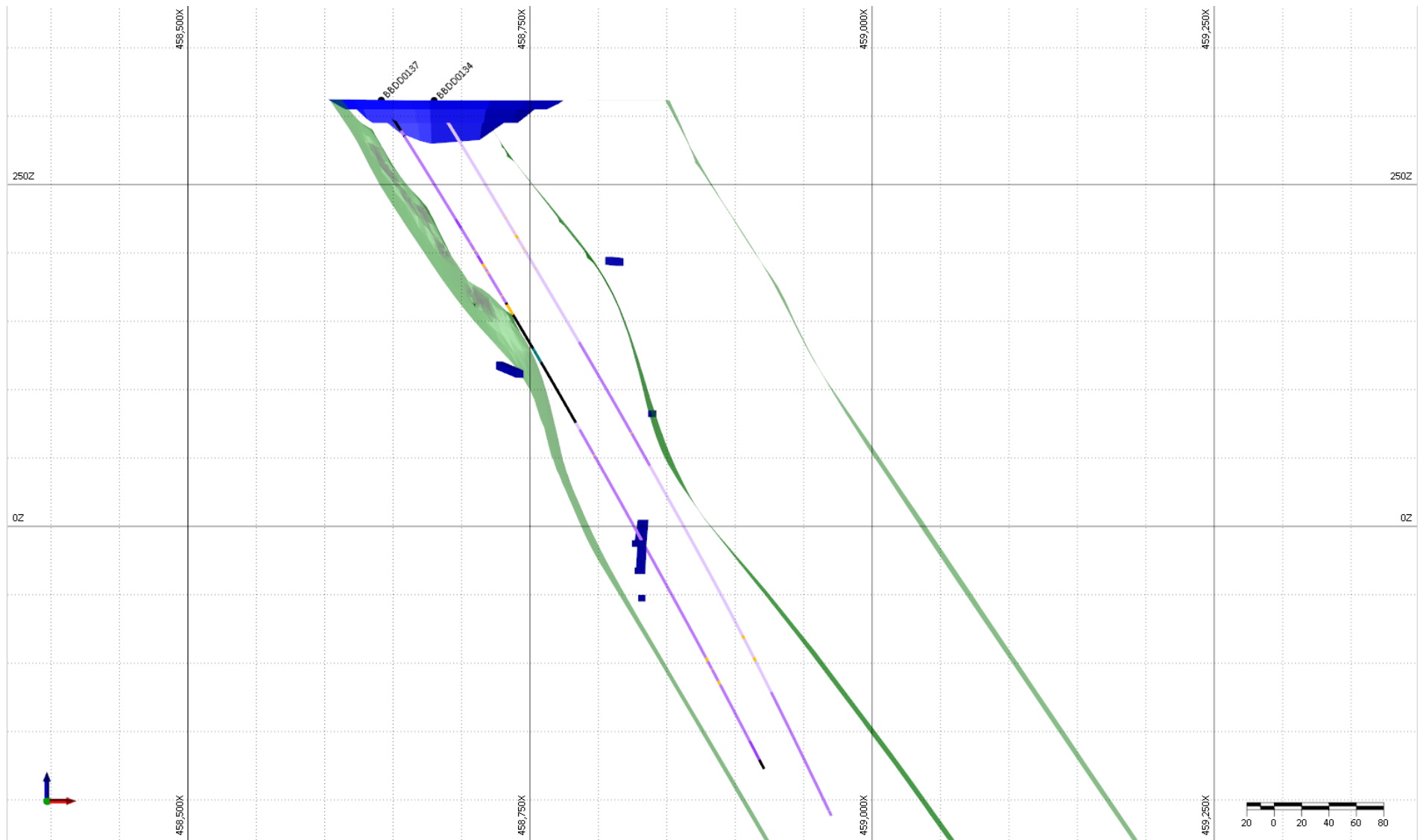


Figure A-24: BBDD0137 and BBDD0134 6601026 Northing BBDD0137 sampled between 5-17m transitional MDQ and BBDD0134 sampled between 5-16m transitional MDQ, within BOM1100 (Source: Ramelius).

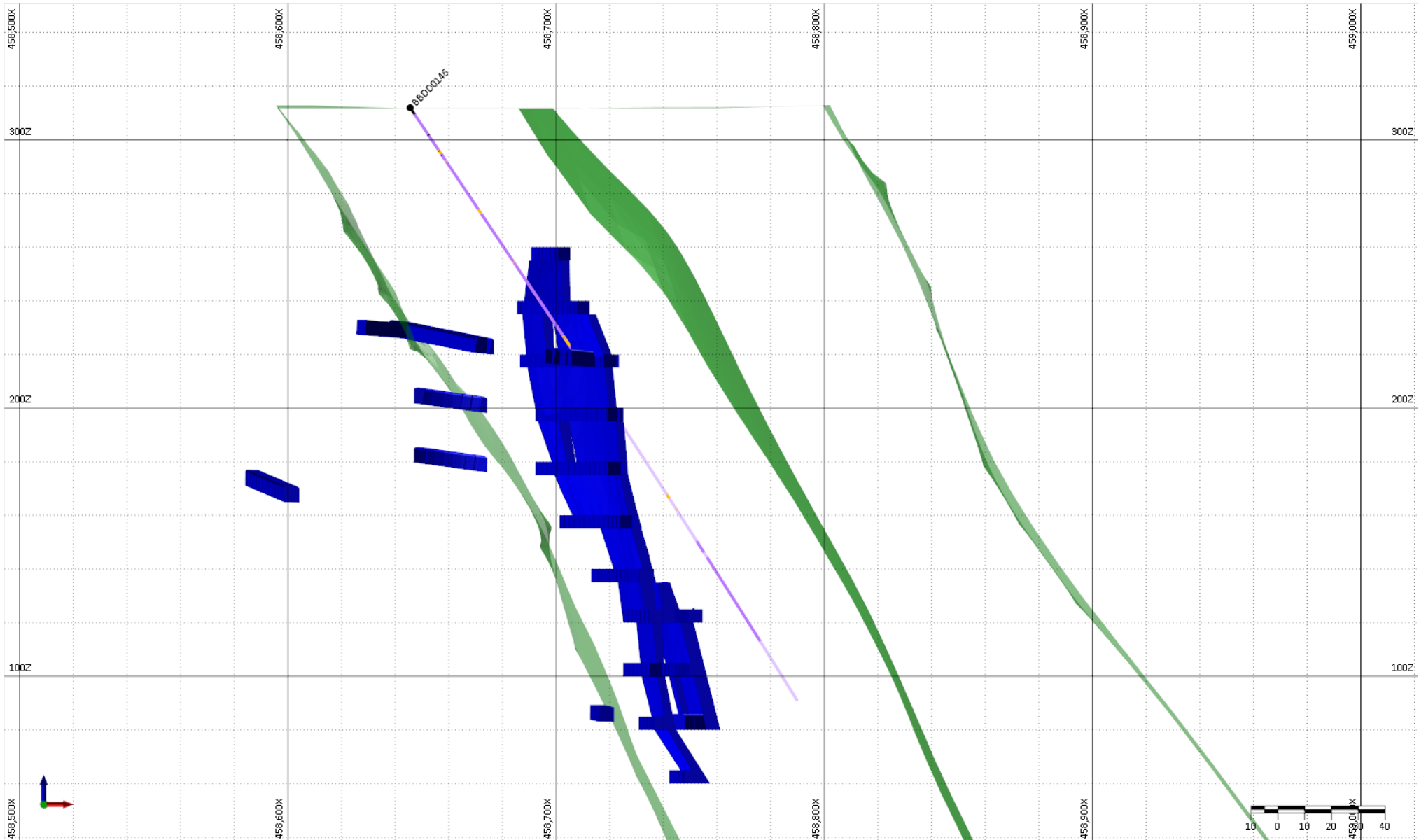


Figure A-25: BBDD0146 6601383 Northing sampled from 105.42-107.01m Fresh Lamprophyre Tura underground (Source: Ramelius).

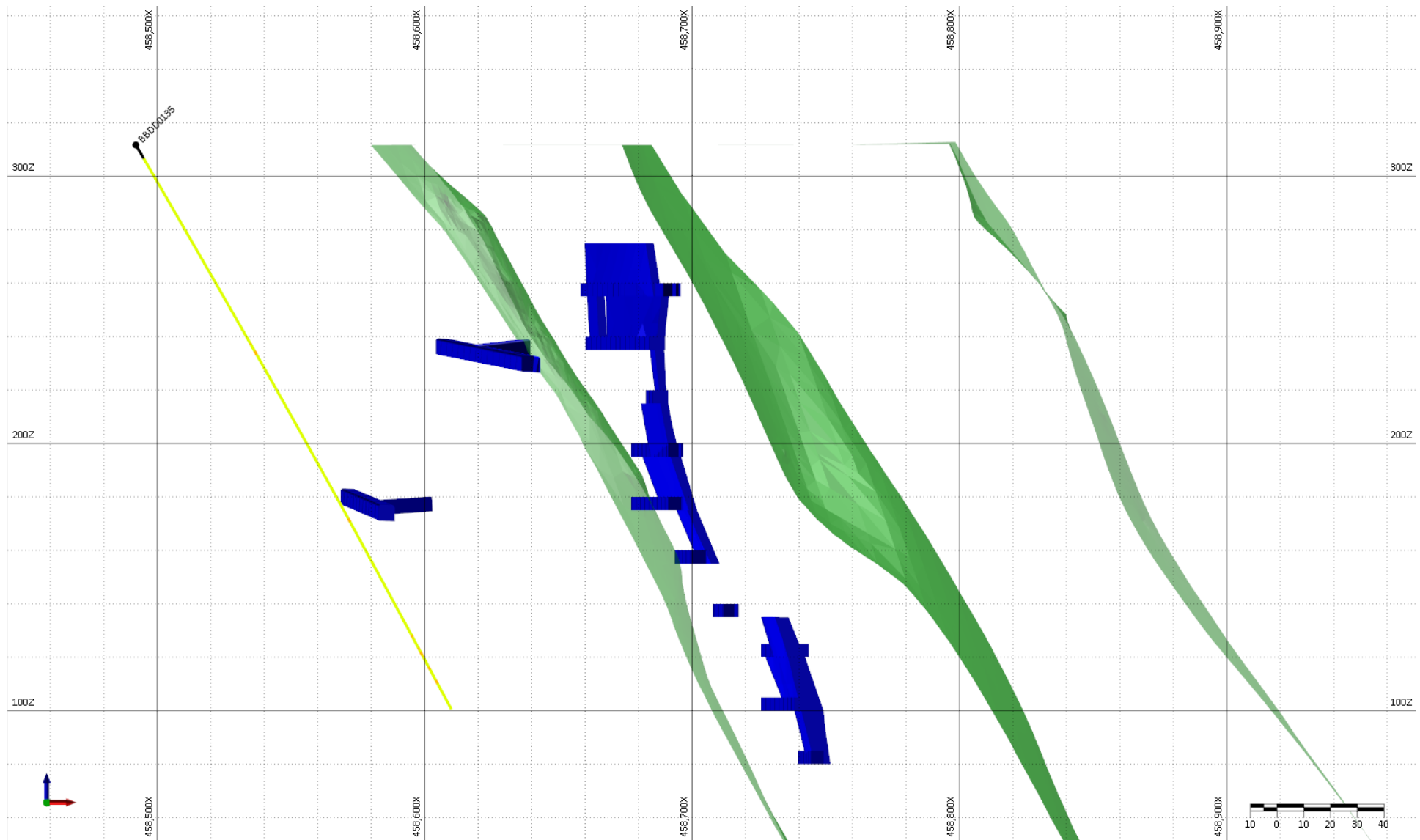


Figure A-26: BBDD0135 6601437 Northing sampled from 6-14m transitional MR_FW and 147-157m fresh MR_FW. This is the only hole that intercepted MR_FW. Close to Tura underground (Source: Ramelius).

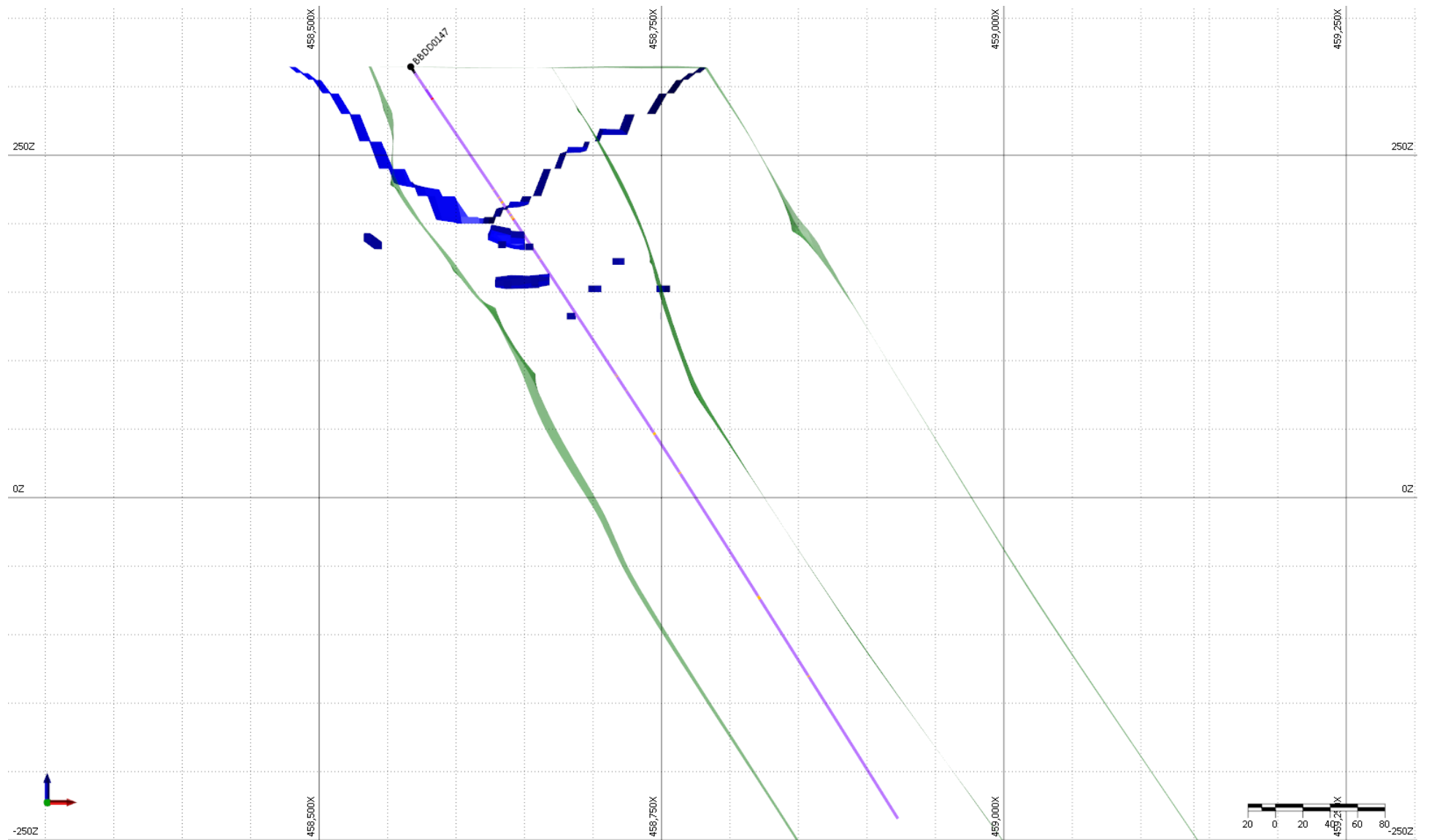


Figure A-27: BBDD0147 6601794 Northing MDQ sampled from 6.9-13m (Transitional), 78-92m (Fresh), 102-114m (Fresh) within BOM1800 pit shell (Source: Ramelius).

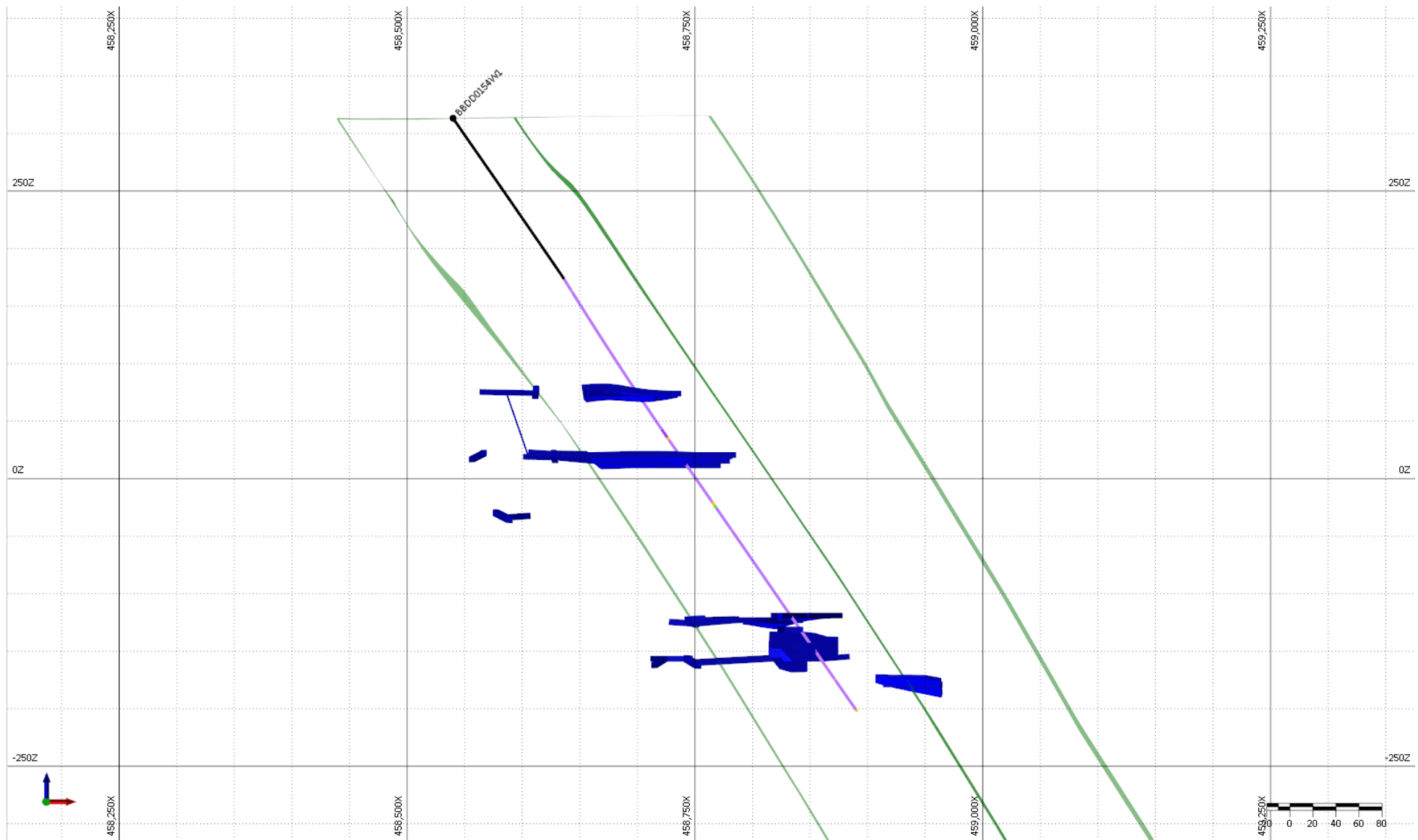


Figure A-28: BBDD0154W1 6602589 northing sampled from 544.04 – 547m fresh MDQ, within or proximal to KL underground (Source: Ramelius).

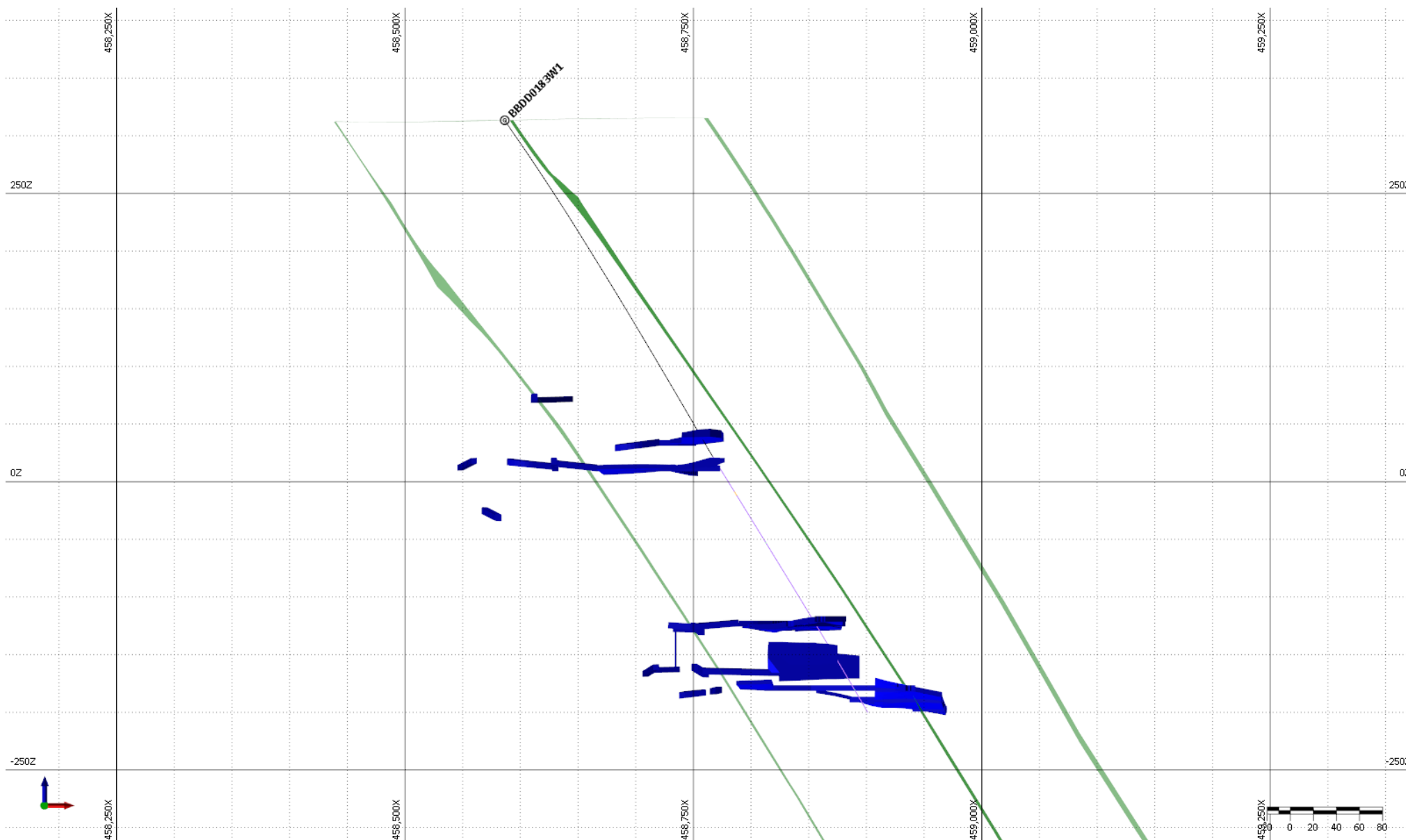


Figure A-29: BBDD0183W1 6602637 Northing, sampled from 550-556m fresh MDQ KL underground (Source: Ramelius).

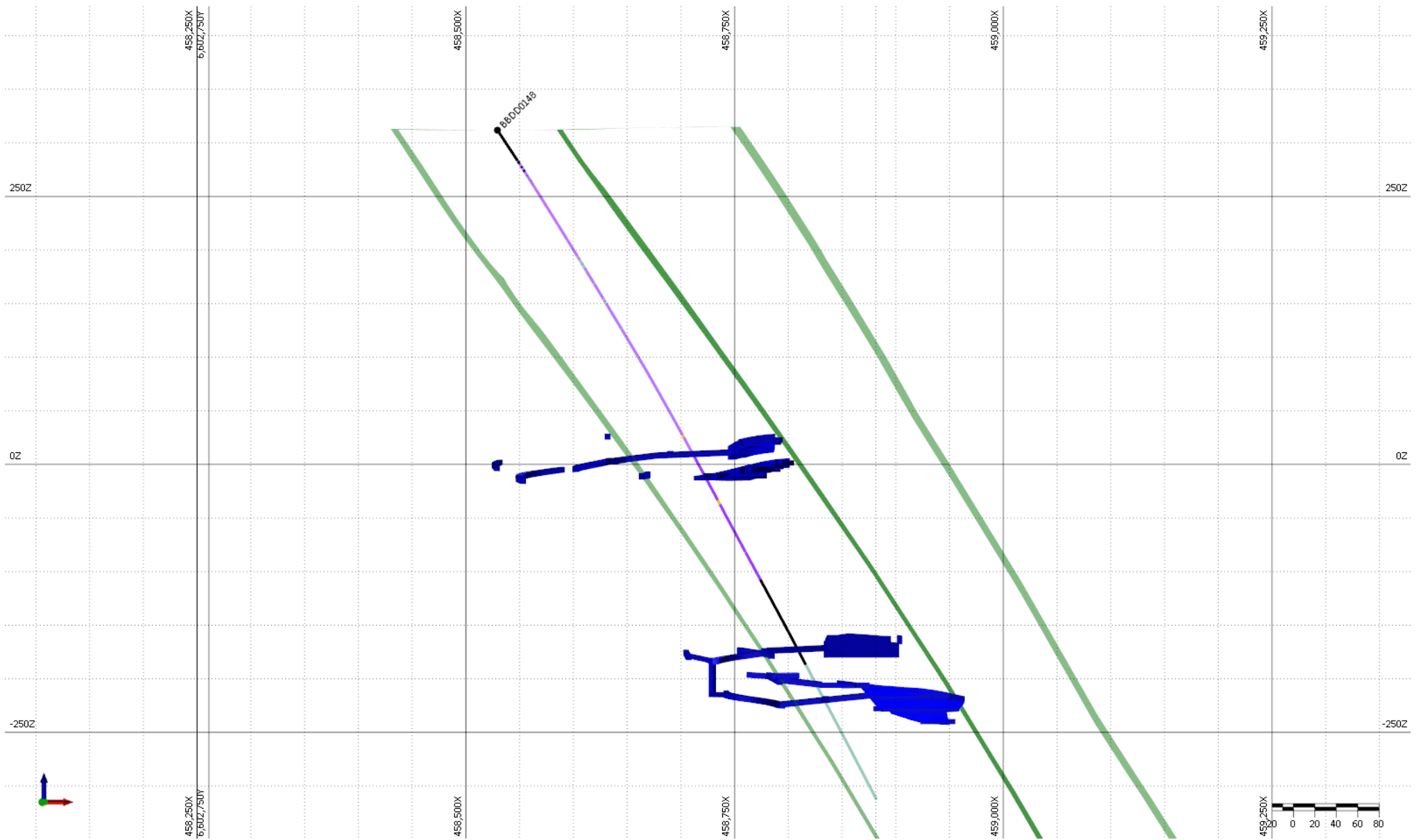


Figure A-30: BBDD0148 6602729 Northing, sampled from 592-598m fresh MD, KL underground (Source: Ramelius).

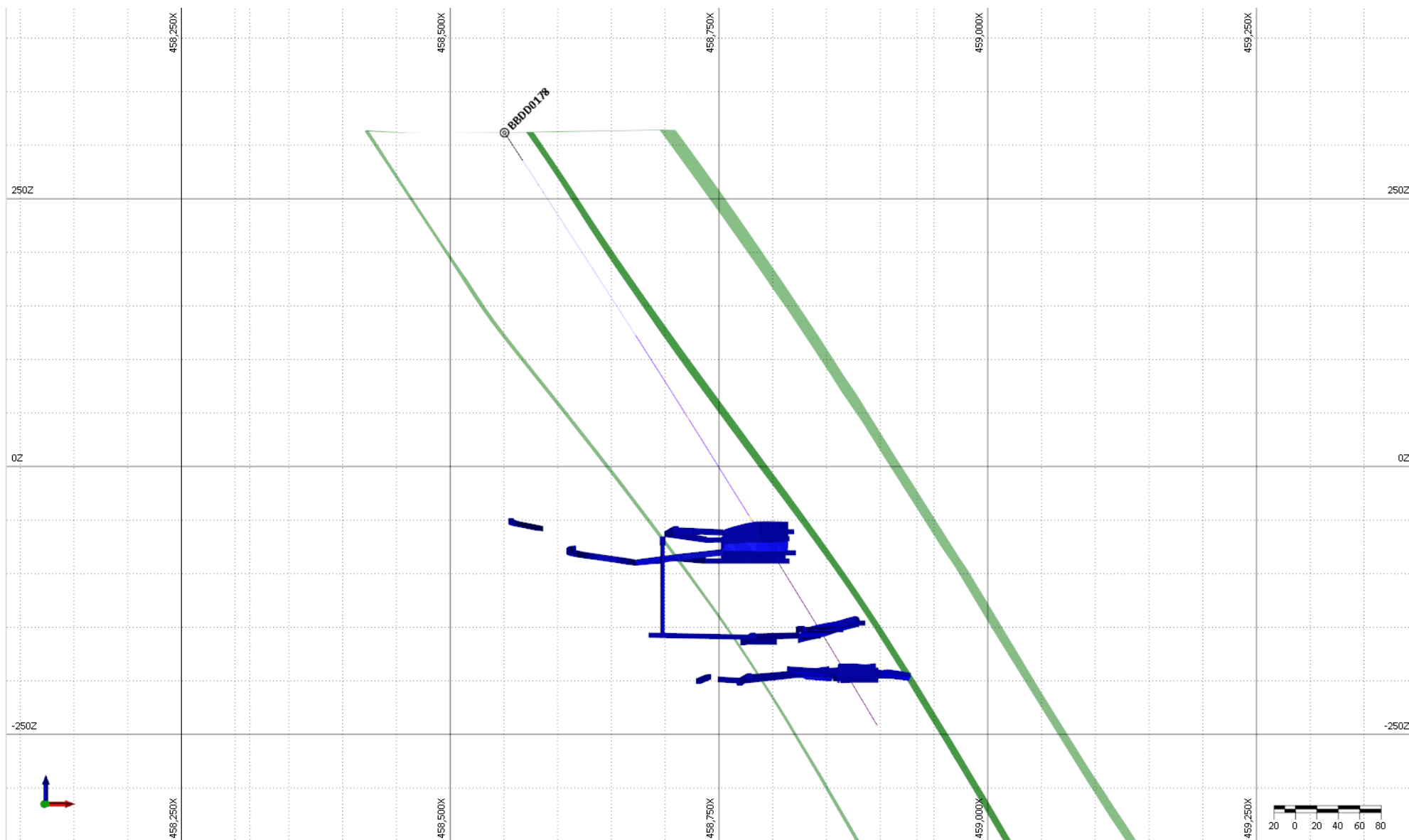


Figure A-31: BBDD0178 6602924 Northing, sampled from 426-428m fresh ALP Lamprophyre, KL underground (Source: Ramelius).

APPENDIX B: LABORATORY RESULTS

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Table B-1: Baseline laboratory results for Kopai Crescent samples

Analysis	Unit	RWC01	RWC05	RWC09	RWC13	RWC17	RWC22	RWC26	RWC30	RWC34	RWC38	RWC42	RWC46
		RRK000188	RRK000193	RRK000197	RRK000248	RRK000831	RRK001178	RRK001218	RRK001693	RRK002029	RRK003208	RRK003383	RRK003387
		10m-11m	14m-15m	18m-19m	12m-13m	11m-12m	31m-32m	9m-10m	13m-14m	17m-18m	6m-7m	10m-11m	14m-15m
		MB Trans	MB Trans	MB Trans	MB Trans	MD Trans	MD Fresh	MD Oxide	MD Trans	MD Trans	MD Trans	MD Trans	MD Trans
		241968-1	241968-2	241968-3	241968-4	241968-5	241968-6	241968-7	241968-8	241968-9	241968-10	241968-11	241968-12
pH _w	pH units	6.61	7.10	7.59	8.52	7.90	8.67	6.54	8.45	6.61	8.61	6.61	7.10
Electrical Conductivity (EC _{1:5})	dS/m	4.41	4.13	3.76	1.88	2.43	1.01	13.40	3.23	9.30	2.00	4.41	4.13
Chloride	mg/kg	6306	5608	4992	2275	3326	643	26539	3824	16586	2484	6306	5608
Cation Extraction Method	Rayment & Lyons	15A2	15A2	15C1	15C1	15C1	15C1	15A2	15C1	15A2	15C1	15A2	15A2
Effective Cation Exchange Capacity	meq/100g	13.40	13.50	18.60	12.40	7.30	3.40	18.30	2.30	4.10	18.80	13.40	13.50
Ex Sodium Percent	%	70.52	73.16	77.09	75.92	74.80	34.62	63.23	42.53	23.03	77.91	70.52	73.16
Ex Magnesium Percent	%	16.79	20.90	17.98	17.88	19.37	15.16	28.96	25.40	20.53	17.67	16.79	20.90
Exchangeable Calcium	meq/100g	1.44	0.48	0.63	0.54	0.30	1.63	1.19	0.69	2.13	0.57	1.44	0.48
Exchangeable Magnesium	meq/100g	2.26	2.83	3.34	2.22	1.42	0.52	5.29	0.58	0.83	3.32	2.26	2.83
Exchangeable Potassium	meq/100g	0.25	0.29	0.26	0.21	0.11	0.06	0.22	0.03	0.14	0.24	0.25	0.29
Exchangeable Sodium	meq/100g	9.48	9.89	14.33	9.41	5.47	1.18	11.55	0.97	0.93	14.62	9.48	9.89
Exchangeable Aluminium	meq/100g	0.02	0.02	0.02	0.02	0.02	0.02	0.02	0.02	0.02	0.02	0.02	0.02
Calcium/Magnesium Ratio	-	0.64	0.17	0.19	0.24	0.21	3.13	0.22	1.19	2.57	0.17	0.64	0.17

Analysis	Unit	RWC42	RWC46	RWC50	RWC54	RWC58	RWC62	RWC66	RWC70
		RRK003383	RRK003387	RRK003491	RRK003742	RRK005131	RRK005411	RRK005625	RRK005629
		10m-11m	14m-15m	12m-13m	8m-9m	7m-8m	16m-17m	8m-9m	12m-13m
		MD Trans	MD Trans	MD Trans	MD Trans	MD Trans	MD Trans	MD Trans	MD Trans
		241968-11	241968-12	241968-13	241968-14	241968-15	241968-16	241968-17	241968-18
pH _w	pH units	8.06	9.06	9.01	9.67	8.29	8.44	8.47	8.78
Electrical Conductivity (EC _{1:5})	dS/m	3.32	0.38	0.33	0.54	2.71	2.58	2.35	1.51
Chloride	mg/kg	3824	325	198	414	3001	2667	3592	1658
Cation Extraction Method	Rayment & Lyons	15C1	15C1	15C1	15C1	15C1	15C1	15C1	15C1
Effective Cation Exchange Capacity	meq/100g	6.20	1.40	1.60	3.10	6.60	7.80	10.20	7.80
Ex Sodium Percent	%	60.05	50.44	41.99	53.77	68.67	74.53	74.15	75.52
Ex Magnesium Percent	%	25.07	17.68	23.69	25.62	22.71	18.89	19.34	16.93
Exchangeable Calcium	meq/100g	0.81	0.36	0.51	0.56	0.50	0.40	0.55	0.45
Exchangeable Magnesium	meq/100g	1.57	0.25	0.39	0.79	1.50	1.47	1.97	1.33
Exchangeable Potassium	meq/100g	0.10	0.07	0.03	0.05	0.05	0.09	0.09	0.12
Exchangeable Sodium	meq/100g	3.75	0.71	0.68	1.65	4.53	5.79	7.54	5.91
Exchangeable Aluminium	meq/100g	0.02	0.02	0.02	0.02	0.02	0.02	0.02	0.02
Calcium/Magnesium Ratio	-	0.52	1.44	1.31	0.71	0.33	0.27	0.28	0.34

Table B-2: Baseline laboratory results for Bombara samples

Analysis	Unit	RWC73	RWC77	RWC81	RWC85	RWC89	RWC93	RWC97	RWC101	RWC105	RWC109	RWC113	RWC117
		BBDD0134	BBDD0134	BBDD0135	BBDD0135	BBDD0137	BBDD0137	BBDD0147	BBDD0147	BBDD0161	BBDD0161	BBDD0183W1	BBDD0175
		426m-428m	12m-14m	10m-12m	151m-153m	7m-9m	15m-17m	89.37m-90.71m	106m-108.09m	175m-177m	182.5m-184m	552m-554m	9m-11m
		ALP Fresh	MDQ Trans	MR_FW Trans	MR_FW Fresh	MDQ Trans	MDQ Trans	MDQ Fresh	MDQ Fresh	MDQ Fresh	MDQ Fresh	MDQ Fresh	MDQ Trans
		242109-1	242109-2	242109-3	242109-4	242109-5	242109-6	242109-7	242109-8	242109-9	242109-10	242109-11	242109-12
pH _w	pH units	9.41	8.51	8.52	9.10	8.76	8.61	8.95	9.27	9.18	8.75	9.01	9.00
Electrical Conductivity (EC _{1:5})	dS/m	0.31	1.19	2.54	0.24	0.90	0.91	0.31	0.18	0.16	0.18	0.17	0.48
Chloride	mg/kg	154	1370	3300	114	1080	1035	109	102	63	137	85	480
Cation Extraction Method	Rayment & Lyons	15C1	15C1	15C1	15C1	15C1	15C1	15C1	15C1	15C1	15C1	15C1	15C1
Effective Cation Exchange Capacity	meq/100g	1.66	9.08	19.50	1.44	3.26	4.51	1.85	1.10	0.89	0.86	0.95	2.50
Ex Sodium Percent	%	21.70	76.90	87.50	38.10	65.90	72.70	42.80	42.00	33.60	27.60	32.10	65.80
Ex Magnesium Percent	%	11.20	15.00	9.44	16.20	18.00	14.10	6.31	7.59	10.40	9.68	8.74	13.90
Exchangeable Calcium	meq/100g	0.63	0.52	0.32	0.44	0.41	0.47	0.53	0.37	0.42	0.44	0.40	0.44
Exchangeable Magnesium	meq/100g	0.19	1.37	1.84	0.23	0.59	0.64	0.12	0.08	0.09	0.08	0.08	0.35
Exchangeable Potassium	meq/100g	0.46	0.19	0.25	0.18	0.09	0.10	0.37	0.16	0.03	0.03	0.09	0.04
Exchangeable Sodium	meq/100g	0.36	6.99	17.10	0.55	2.15	3.28	0.79	0.46	0.30	0.24	0.31	1.64
Exchangeable Aluminium	meq/100g	0.02	0.02	0.02	0.03	0.02	0.02	0.04	0.03	0.05	0.07	0.07	0.02
Calcium/Magnesium Ratio	-	3.32	0.38	0.17	1.91	0.69	0.73	4.42	4.63	14.00	14.67	4.44	11.00

Analysis	Unit	RWC121	RWC125	RWC129
		BBDD0171	BBDD0170	BBDD0148
		10m-12m	20m-22m	594m-596m
		MD Trans	MD Trans	MD Fresh
		242109-13	242109-14	242109-15
pH _w	pH units	9.00	9.54	9.03
Electrical Conductivity (EC _{1:5})	dS/m	1.66	1.14	0.14
Chloride	mg/kg	1830	1140	53
Cation Extraction Method	Rayment & Lyons	15C1	15C1	15C1
Effective Cation Exchange Capacity	meq/100g	4.84	4.82	0.91
Ex Sodium Percent	%	67.70	69.10	36.40
Ex Magnesium Percent	%	18.20	15.10	23.00
Exchangeable Calcium	meq/100g	0.59	0.66	0.25
Exchangeable Magnesium	meq/100g	0.88	0.73	0.21
Exchangeable Potassium	meq/100g	0.07	0.09	0.09
Exchangeable Sodium	meq/100g	3.28	3.33	0.33
Exchangeable Aluminium	meq/100g	0.02	0.02	0.03
Calcium/Magnesium Ratio	-	0.67	0.90	1.19

Table B-3: Total metals concentrations in solid component of a subset 33 samples, with EIL exceedances in yellow and HIL exceedances in red (both EIL and HIL values exceeded text is highlighted in red).

Element	Unit	Detection Limit	RWC01	RWC05	RWC09	RWC13	RWC17	RWC22	RWC26	RWC30	RWC34	RWC38	RWC42	RWC46	RWC50	RWC54	EIL	HIL
			Kopai Crescent															
			MB Trans 10m-11m	MB Trans 14m-15m	MB Trans 18m-19m	MB Trans 12m-13m	MD Trans 11m-12m	MD Fresh 31m-32m	MD Oxide 9m-10m	MD Trans 13m-14m	MD Trans 17m-18m	MD Trans 6m-7m	MD Trans 10m-11m	MD Trans 14m-15m	MD Trans 12m-13m	MD Trans 8m-9m		
Ag	ppm	0.01	0.03	0.03	0.02	0.02	0.02	0.07	0.03	0.02	0.04	0.03	0.08	0.08	0.06	<0.01	NA	NA
Al	ppm	100	78500	76810	77900	77000	87900	75600	115500	91600	85000	73900	74900	72100	73400	71700	NA	NA
As	ppm	0.2	1.7	1.6	2.8	2.9	1.7	2.1	1.9	3.1	1.7	3.1	2.6	2.3	0.9	1.2	40	300
Ba	ppm	10	260	170	180	130	160	100	20	60	280	50	50	40	10	50	NA	NA
Be	ppm	0.05	0.53	0.62	0.53	0.31	0.6	0.42	0.92	0.5	0.45	0.38	0.33	0.32	0.35	0.37	NA	NA
Bi	ppm	0.01	0.07	0.13	0.06	0.05	0.04	0.02	0.05	0.02	0.02	0.02	0.05	0.02	0.03	0.04	NA	100
B	ppm	50	<50	<50	<50	<50	<50	<50	<50	<50	<50	<50	<50	<50	<50	<50	NA	NA
Ca	ppm	100	25900	32900	35800	53100	37000	65100	700	58200	45100	45800	36700	67300	62200	47900	NA	NA
Cd	ppm	0.02	0.06	0.06	0.09	0.08	0.05	0.23	0.08	0.03	0.06	0.07	0.05	0.11	0.09	0.1	NA	90
Ce	ppm	0.01	28.40	29.80	19.55	8.04	26.40	12.70	21.40	24.00	16.90	10.20	11.15	9.50	9.37	9.20	NA	NA
Co	ppm	0.1	30.0	30.7	48.3	50.4	30.2	53.5	38.9	42.4	41.2	45.3	52.9	50.6	45.3	49.4	NA	300
Cr	ppm	1	76	83	105	151	130	116	149	141	124	110	107	98	103	97	400	240
Cs	ppm	0.05	1.28	1.48	1.40	1.30	0.79	0.61	2.15	0.38	2.13	0.59	0.86	0.19	0.19	0.32	NA	NA
Cu	ppm	0.2	51.6	68.0	119.0	113.5	40.5	133.0	65.8	28.5	37.0	142.0	186.0	101.5	24.2	91.0	130	20,000
Fe	ppm	100	56300	68000	95600	96700	60700	64200	51100	64700	71900	106000	106000	99700	99900	101000	NA	NA
Ga	ppm	0.05	16.7	17.6	20.0	17.4	18.8	21.1	21.7	18.8	16.1	20.1	19.5	18.7	19.4	18.4	NA	NA
Ge	ppm	0.05	0.16	0.18	0.20	0.21	0.18	0.16	0.21	0.15	0.20	0.23	0.31	0.25	0.23	0.30	NA	NA
Hf	ppm	0.1	2.90	2.90	2.00	1.20	2.80	1.70	3.40	2.40	2.40	1.60	1.60	1.60	1.60	1.70	NA	NA
Hg	ppm	0.01	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	0.006	0.006	<0.005	<0.005	<0.005	<0.005	NA	NA
In	ppm	0.005	0.049	0.057	0.085	0.073	0.065	0.093	0.08	0.066	0.062	0.088	0.093	0.087	0.088	0.093	NA	NA
K	ppm	100	7000	6300	7000	6100	4700	3300	1200	1100	6800	3900	1400	2100	1600	1400	NA	NA
La	ppm	0.5	14.2	14.9	8.6	2.6	9.2	4.4	5.8	9.5	5.9	3.5	3.3	3.2	3.4	3.4	NA	NA
Li	ppm	0.2	11.6	9.5	11.7	13.3	9.1	12.5	14.3	11.0	17.6	7.7	14.6	7.8	11.5	10.5	NA	NA
Mg	ppm	0.01	17200	14200	22400	38100	15000	22400	4200	21200	40700	21100	36900	35200	37000	37400	NA	NA
Mn	ppm	5	529	510	1015	1385	927	1500	522	1370	1175	1105	1340	1635	1610	1365	NA	9,000
Mo	ppm	0.05	1.41	1.90	2.29	0.53	2.20	1.38	1.14	3.24	0.92	1.66	0.80	1.32	0.45	0.26	NA	NA
Na	ppm	100	31200	21800	20900	17800	31900	20700	15100	23500	34800	16500	15900	15400	10200	23900	NA	NA
Nb	ppm	0.1	4.7	4.9	4.0	2.5	5.5	3.6	6.6	4.9	4.1	3.2	3.2	3.1	3.3	3.1	NA	NA
Ni	ppm	0.2	125.5	88	100	84.1	97.3	89.7	163.5	124	107.5	69.1	70.8	56.5	48.2	57.7	30	800
P	ppm	10	540	520	510	330	830	470	500	670	580	510	470	430	450	450	NA	NA
Pb	ppm	0.5	5.7	7.2	5.1	1.9	2.8	1.6	3.6	1.5	1.5	1.5	1.6	0.8	1.2	1.4	1100	600
Rb	ppm	0.1	28.8	33.2	33.9	29.4	10.3	13.2	8.3	3.1	17.7	10.2	7.2	6.4	4.6	3.8	NA	NA
Re	ppm	0.002	0.002	<0.002	0.002	<0.002	<0.002	0.003	<0.002	<0.002	<0.002	0.002	0.002	0.002	0.002	0.002	NA	NA
S	ppm	100	300	300	300	100	200	4500	1100	300	500	200	200	700	300	<100	NA	NA
Sb	ppm	0.05	0.13	0.21	0.25	0.23	0.28	0.18	0.14	0.16	2.83	0.35	0.23	0.26	0.28	0.15	NA	NA
Sc	ppm	0.1	19.7	19.1	39.2	48.2	31.8	43.8	41.8	35.0	28.5	52.2	50.6	48.3	52.3	50.4	NA	NA
Se	ppm	1	1	<1	1	1	1	2	<1	1	1	1	1	1	1	<1	NA	NA
Sn	ppm	0.2	1	1.3	1	0.7	2.1	0.9	1	1.3	0.8	0.7	0.8	0.8	0.8	0.8	NA	NA
Sr	ppm	0.2	151.5	110.5	167.5	150.5	155.0	153.0	10.8	130.5	318.0	129.0	107.0	135.5	94.8	141.5	NA	NA
Ta	ppm	0.05	0.33	0.35	0.26	0.15	0.36	0.22	0.42	0.32	0.27	0.19	0.20	0.19	0.19	0.19	NA	NA
Te	ppm	0.05	0.07	0.12	0.1	0.08	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	NA	NA
Th	ppm	0.01	2.95	3.12	1.19	0.2	1.65	0.26	2.23	1.47	1.23	0.24	0.25	0.23	0.21	0.2	NA	NA
Ti	ppm	0.005	4160	4280	7010	6120	7430	8360	7500	7430	6440	7960	8130	7690	8340	8060	NA	NA
Tl	ppm	0.02	0.42	0.44	0.35	0.25	0.10	0.21	0.05	0.04	0.27	0.07	0.04	0.13	0.03	0.03	NA	NA
U	ppm	0.1	0.90	0.80	0.50	0.10	0.90	0.10	1.60	0.40	0.30	0.30	0.40	0.10	0.10	0.30	NA	NA
V	ppm	1	133	139	298	294	211	348	243	211	195	371	354	345	373	358	NA	NA
W	ppm	0.1	0.8	1.4	0.8	0.8	0.8	3.4	0.4	3.7	0.9	1.1	1.7	1.5	0.7	0.3	NA	NA
Y	ppm	0.1	15.8	19.4	30.8	27.2	23.0	31.8	12.2	24.0	21.4	33.4	28.1	30.7	33.2	32.1	NA	NA
Zn	ppm	2	114	154	200	108	75	122	137	58	69	121	122	110	113	112	130	30,000
Zr	ppm	0.5	110.5	112.0	69.4	38.2	105.5	76.0	123.0	86.1	90.7	44.2	55.0	56.0	48.7	78.8	NA	NA

Table B-3 (cont'd): Total metals concentrations in solid component of a subset 33 samples, with EIL exceedances in yellow and HIL exceedances in red (both EIL and HIL values exceeded text is highlighted in red).

Element	Unit	Detection Limit	RWC58	RWC62	RWC66	RWC70	RWC73	RWC77	RWC81	RWC85	RWC89	RWC93	RWC97	RWC101	RWC105	RWC109	EIL	HIL	
			Kopai Crescent						Bombora										
			MD Trans 7m-8m	MD Trans 16m-17m	MD Trans 8m-9m	MD Trans 12m-13m	ALP Fresh 426m-428m	MDQ Trans 12m-14m	MR_FW Trans 10m-12m	MR_FW Fresh 151m-153m	MDQ Trans 7m-9m	MDQ Fresh 15m-17m	MDQ Fresh 89.37m-90.71m	MDQ Fresh 106m-108.09m	MDQ Fresh 175m-177m	MDQ Fresh 182.5m-184m			
Ag	ppm	0.01	0.02	0.18	0.03	<0.01	0.08	0.05	0.1	0.01	0.05	0.03	0.06	0.03	0.02	0.13	NA	NA	
Al	ppm	100	75400	76700	75400	73800	76400	62400	78600	72400	59800	61100	56300	58400	58400	55200	NA	NA	
As	ppm	0.2	1.6	1.2	9.2	3.5	0.7	1.1	0.5	0.4	1.1	1.2	1.1	2.7	0.6	1.4	40	300	
Ba	ppm	10	60	70	30	40	740	20	20	210	40	40	30	40	40	40	NA	NA	
Be	ppm	0.05	0.35	0.28	0.63	0.55	1.25	0.33	0.81	0.36	0.67	0.71	0.82	0.91	0.68	0.85	NA	NA	
Bi	ppm	0.01	0.02	0.04	0.04	0.06	0.03	0.01	0.01	0.01	0.02	0.04	0.04	0.01	0.02	0.08	NA	100	
B	ppm	50	<50	<50	<50	<50	<50	<50	<50	<50	<50	<50	<50	<50	<50	<50	NA	NA	
Ca	ppm	100	27900	5300	61300	58500	40700	47800	4000	63000	42100	42000	42500	46100	51500	51800	NA	NA	
Cd	ppm	0.02	0.08	0.02	0.09	0.12	0.1	0.1	0.03	0.02	0.12	0.11	0.35	0.12	0.12	0.06	NA	90	
Ce	ppm	0.01	13.05	7.62	21.50	10.65	91.40	8.19	16.45	16.40	13.85	14.60	24.70	27.90	19.55	18.10	NA	NA	
Co	ppm	0.1	70.6	104.0	67.5	55.8	23.3	68.8	58.8	58.0	45.6	38.0	37.0	24.1	38.1	49.5	NA	300	
Cr	ppm	1	115	103	108	100	106	5	378	489	3	6	8	4	4	3	400	240	
Cs	ppm	0.05	0.74	4.08	0.25	0.29	3.98	0.30	0.17	1.25	0.19	0.19	2.13	0.70	0.22	0.74	NA	NA	
Cu	ppm	0.2	147.0	245.0	61.4	115.5	37.0	78.1	2.9	3.7	30.5	32.1	70.6	6.7	24.8	587	130	20,000	
Fe	ppm	100	104500	104500	84400	99000	36000	136000	76900	74400	139000	137000	153000	123000	130500	137500	NA	NA	
Ga	ppm	0.05	19.8	18.7	19.8	17.9	19.6	20.6	12.9	13.8	24.3	24.3	25.1	24.1	23.1	21.8	NA	NA	
Ge	ppm	0.05	0.28	0.26	0.19	0.27	0.20	0.27	0.19	0.26	0.26	0.28	0.31	0.25	0.27	0.26	NA	NA	
Hf	ppm	0.1	1.70	1.30	1.80	1.80	3.40	0.90	2.20	1.90	2.1	2.4	2.4	3.1	2.1	2.1	NA	NA	
Hg	ppm	0.01	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	0.009	NA	NA	
In	ppm	0.005	0.1	0.083	0.111	0.089	0.033	0.098	0.058	0.053	0.149	0.141	0.253	0.18	0.141	0.142	NA	NA	
K	ppm	100	1100	2500	1600	1900	18300	3000	600	5600	3700	3100	2800	2500	2100	2000	NA	NA	
La	ppm	0.5	5.4	2.8	7.5	3.6	42.3	3	10.1	7.4	4.3	4.3	8.5	9.7	6.6	5.8	NA	NA	
Li	ppm	0.2	14.7	22.7	6.0	7.3	31.0	10.0	26.5	16.4	7.8	8.1	13.4	7.1	8.4	14.7	NA	NA	
Mg	ppm	0.01	36300	42400	33300	36000	24600	26800	39000	56300	12000	11700	11600	7000	12200	13800	NA	NA	
Mn	ppm	5	1520	1185	1535	1750	553	1715	786	1470	2280	1965	2460	2060	2130	1855	NA	9,000	
Mo	ppm	0.05	1.30	0.43	0.78	0.37	0.21	0.34	0.26	0.41	0.68	0.42	1.54	0.82	1.06	0.71	NA	NA	
Na	ppm	100	18000	23200	23100	23200	35200	16900	28300	16400	20500	20600	11800	19700	18400	16300	NA	NA	
Nb	ppm	0.1	3.3	2.8	3.5	3.1	5.0	3.0	3.4	3.1	6.4	6.6	7.2	9.7	6.6	6.0	NA	NA	
Ni	ppm	0.2	87.2	109.5	85	73.8	106	16.6	357	393	3.3	9.7	3.6	1	0.9	0.4	30	800	
P	ppm	10	490	440	520	460	1550	380	620	470	920	930	1410	1100	850	890	NA	NA	
Pb	ppm	0.5	2.6	1.5	1.5	1.3	10.4	0.9	0.7	0.7	0.9	0.9	0.9	1.5	0.7	1.3	1100	600	
Rb	ppm	0.1	4.9	14.3	5.0	4.5	60.9	5.5	0.9	19.2	8	5	5.8	4.3	3.6	4.9	NA	NA	
Re	ppm	0.002	0.002	<0.002	<0.002	<0.002	0.002	0.003	<0.002	0.002	0.002	<0.002	0.002	0.002	0.002	0.002	NA	NA	
S	ppm	100	300	300	100	100	1900	100	100	100	100	100	4300	200	1600	20800	NA	NA	
Sb	ppm	0.05	0.36	0.14	0.41	0.31	0.13	0.12	0.06	0.08	0.1	0.08	0.1	0.1	0.1	0.3	NA	NA	
Sc	ppm	0.1	51.8	48.4	56.5	53.4	10.4	59.0	29.2	26.7	47.5	47.6	34.6	39.9	44.4	40.4	NA	NA	
Se	ppm	1	1	<1	1	1	<1	1	<1	<1	1	1	<1	1	1	4	NA	NA	
Sn	ppm	0.2	1.1	0.7	0.8	0.9	0.9	0.8	1.5	2	1.1	1.2	2.4	1.5	1	1.8	NA	NA	
Sr	ppm	0.2	109.0	55.2	239.0	230.0	748.0	75.9	61.8	161.0	91.0	97.5	28.0	119.0	97.8	85.5	NA	NA	
Ta	ppm	0.05	0.20	0.17	0.21	0.19	0.27	0.19	0.23	0.20	0.37	0.39	0.43	0.56	0.39	0.35	NA	NA	
Te	ppm	0.05	0.06	0.12	<0.05	0.05	0.05	<0.05	<0.05	<0.05	<0.05	<0.05	0.05	<0.05	<0.05	0.35	NA	NA	
Th	ppm	0.01	0.22	0.18	0.29	0.2	5.64	0.18	1.04	0.95	0.41	0.44	0.45	0.57	0.4	0.34	NA	NA	
Ti	ppm	0.005	8380	7770	8810	8310	3790	11950	6070	5320	12350	11950	6230	7970	11150	10150	NA	NA	
Tl	ppm	0.02	0.02	0.06	0.05	0.02	0.29	0.03	0.03	0.13	0.03	0.03	0.03	0.02	0.03	0.10	NA	NA	
U	ppm	0.1	1.30	0.40	0.30	0.20	1.40	0.20	0.40	0.30	0.10	0.10	0.10	0.20	0.10	0.10	NA	NA	
V	ppm	1	370	350	392	371	88	601	208	173	144	128	4	5	92	86	NA	NA	
W	ppm	0.1	0.5	1.4	0.4	0.5	3.1	0.6	0.3	0.5	0.5	0.6	0.4	0.4	0.4	11.4	NA	NA	
Y	ppm	0.1	35.5	23.7	59.3	41.5	12.1	31.1	36.8	16.6	55.7	57.5	69.1	79.7	56.6	52.6	NA	NA	
Zn	ppm	2	224	144	145	141	73	137	93	54	164	150	381	154	159	111	130	30,000	
Zr	ppm	0.5	44.3	46.6	87.7	59.8	138.5	25.2	80.3	68.8	68.5	79.6	83.8	113.5	75.3	78.9	NA	NA	

Table B-3 (cont'd): Total metals concentrations in solid component of a subset 33 samples, with EIL exceedances in yellow and HIL exceedances in red (both EIL and HIL values exceeded text is highlighted in red).

Element	Unit	Detection Limit	RWC113	RWC118	RWC121	RWC126	RWC129	EIL	HIL
			Bombora						
			MDQ Fresh	MDQ Trans	MD Trans	MD Trans	MD Fresh		
			552m-554m	5m-7m	10m-12m	16m-18m	594m-596m		
	EP2417121_029	EP2417121_030	EP2417121_031	EP2417121_032	EP2417121_033				
Ag	ppm	0.01	0.11	0.04	0.18	0.07	0.04	NA	NA
Al	ppm	100	59400	62700	83500	78400	80000	NA	NA
As	ppm	0.2	0.4	1.2	1.5	1.1	<0.2	40	300
Ba	ppm	10	50	30	30	20	120	NA	NA
Be	ppm	0.05	0.91	0.52	0.27	0.25	0.6	NA	NA
Bi	ppm	0.01	0.04	0.01	0.01	0.02	<0.01	NA	100
B	ppm	50	<50	<50	<50	<50	<50	NA	NA
Ca	ppm	100	45400	52500	65600	66500	42400	NA	NA
Cd	ppm	0.02	0.14	0.1	0.07	0.07	0.08	NA	90
Ce	ppm	0.01	28.70	8.61	6.57	6.54	28.40	NA	NA
Co	ppm	0.1	23.2	60.6	50.4	51.5	42.8	NA	300
Cr	ppm	1	3	2	74	139	170	400	240
Cs	ppm	0.05	0.54	0.23	0.64	0.65	1.89	NA	NA
Cu	ppm	0.2	26.5	39.3	49.1	96.6	56.8	130	20,000
Fe	ppm	100	119000	136500	84600	91000	74100	NA	NA
Ga	ppm	0.05	24.8	21.7	18.9	18.0	18.6	NA	NA
Ge	ppm	0.05	0.22	0.22	0.18	0.21	0.20	NA	NA
Hf	ppm	0.1	2.6	1.4	0.9	0.8	2.6	NA	NA
Hg	ppm	0.01	<0.005	<0.005	<0.005	<0.005	<0.005	NA	NA
In	ppm	0.005	0.181	0.112	0.073	0.071	0.062	NA	NA
K	ppm	100	2300	2200	1700	1700	5100	NA	NA
La	ppm	0.5	10.6	2.8	3.1	2.6	13.2	NA	NA
Li	ppm	0.2	10.5	6.0	6.6	5.8	20.5	NA	NA
Mg	ppm	0.01	7700	23500	31600	36800	45700	NA	NA
Mn	ppm	5	1935	1950	1215	1440	1185	NA	9,000
Mo	ppm	0.05	1.53	0.26	0.27	0.26	0.62	NA	NA
Na	ppm	100	18500	24900	20200	16800	28700	NA	NA
Nb	ppm	0.1	8.5	3.8	2.3	2.2	5.6	NA	NA
Ni	ppm	0.2	0.2	7.9	56.2	69.2	137.5	30	800
P	ppm	10	1110	500	300	280	800	NA	NA
Pb	ppm	0.5	1.3	0.6	0.8	0.7	2.7	1100	600
Rb	ppm	0.1	5	6.3	3.3	5.6	23.2	NA	NA
Re	ppm	0.002	0.003	0.002	<0.002	0.002	<0.002	NA	NA
S	ppm	100	1700	<100	200	100	100	NA	NA
Sb	ppm	0.05	0.2	0.1	0.1	0.1	0.1	NA	NA
Sc	ppm	0.1	39.8	58.8	46.2	51	24.3	NA	NA
Se	ppm	1	1	1	1	1	<1	NA	NA
Sn	ppm	0.2	1.7	0.8	0.6	0.5	1.1	NA	NA
Sr	ppm	0.2	112.5	91.8	116.0	106.0	144.5	NA	NA
Ta	ppm	0.05	0.50	0.22	0.13	0.13	0.35	NA	NA
Te	ppm	0.05	0.17	<0.05	<0.05	<0.05	<0.05	NA	NA
Th	ppm	0.01	0.55	0.23	0.17	0.13	1.34	NA	NA
Ti	ppm	0.005	8520	12600	6220	5870	5700	NA	NA
Tl	ppm	0.02	0.03	0.04	0.04	0.04	0.12	NA	NA
U	ppm	0.1	0.10	0.30	0.20	0.20	0.40	NA	NA
V	ppm	1	4	403	292	287	162	NA	NA
W	ppm	0.1	1.1	0.4	0.9	0.5	0.2	NA	NA
Y	ppm	0.1	76.6	38.4	29.2	25.7	21.9	NA	NA
Zn	ppm	2	157	151	81	84	67	130	30,000
Zr	ppm	0.5	97.3	41.8	28.0	22.0	97.6	NA	NA

Table B-4: GAI values for subset of 33 samples with GAI values >3 in yellow.

Element	Unit	Median Crustal Abundance	RWC01	RWC05	RWC09	RWC13	RWC17	RWC22	RWC26	RWC30	RWC34	RWC38	RWC42	RWC46	RWC50	RWC54	RWC58	RWC62
			Kopai Crescent															
			MB Trans 10m-11m	MB Trans 14m-15m	MB Trans 18m-19m	MB Trans 12m-13m	MD Trans 11m-12m	MD Fresh 31m-32m	MD Oxide 9m-10m	MD Trans 13m-14m	MD Trans 17m-18m	MD Trans 6m-7m	MD Trans 10m-11m	MD Trans 14m-15m	MD Trans 12m-13m	MD Trans 8m-9m	MD Trans 7m-8m	MD Trans 16m-17m
Ag	ppm	0.07	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	1
Al	ppm	82,000	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
As	ppm	1.5	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Ba	ppm	500	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Be	ppm	2.6	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Bi	ppm	0.048	0	1	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Cd	ppm	0.11	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Ce	ppm	68	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Co	ppm	20	0	0	1	1	0	1	0	0	0	1	1	1	1	1	1	2
Cr	ppm	100	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Cs	ppm	3	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Cu	ppm	50	0	0	1	1	0	1	0	0	0	1	1	0	0	0	1	2
Fe	ppm	950	5	6	6	6	5	5	5	6	6	6	6	6	6	6	6	6
Ga	ppm	18	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Ge	ppm	1.8	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Hf	ppm	5.3	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Hg	ppm	0.05	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
In	ppm	0.049	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
La	ppm	32	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Li	ppm	20	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Mn	ppm	950	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Mo	ppm	1.5	0	0	0	0	0	0	0	1	0	0	0	0	0	0	0	0
Nb	ppm	20	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Ni	ppm	80	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Pb	ppm	14	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Rb	ppm	90	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Re	ppm	0.0004	2	2	2	2	2	2	2	2	2	2	2	2	2	2	2	2
Sb	ppm	0.2	0	0	0	0	0	0	0	0	3	0	0	0	0	0	0	0
Sc	ppm	16	0	0	1	1	0	1	1	1	0	1	1	1	1	1	1	1
Se	ppm	0.05	4	4	4	4	4	5	4	4	4	4	4	4	4	4	4	4
Sn	ppm	2.2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Sr	ppm	370	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Ta	ppm	2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Te	ppm	0.005	3	4	4	3	3	3	3	3	3	3	3	3	3	3	3	4
Th	ppm	12	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Ti	ppm	5600	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Tl	ppm	0.6	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
U	ppm	2.4	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
V	ppm	160	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
W	ppm	1	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Y	ppm	30	0	0	0	0	0	1	0	0	0	1	1	1	1	1	1	1
Zn	ppm	75	0	0	0	0	0	1	0	1	0	0	0	0	0	0	0	0
Zr	ppm	190	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0

Table B-4 (cont'd): GAI values for subset of 33 samples with GAI values >3 in yellow.

Element	Unit	Median Crustal Abundance	RWC66	RWC70	RWC73	RWC77	RWC81	RWC85	RWC89	RWC93	RWC97	RWC101	RWC105	RWC109	RWC113	RWC118	RWC121	RWC126	RWC129		
			Kopai Crescent					Bombora													
			MD Trans	MD Trans	ALP Fresh	MDQ Trans	MR_FW Trans	MR_FW Fresh	MDQ Trans	MDQ Trans	MDQ Fresh	MDQ Fresh	MDQ Fresh	MDQ Fresh	MDQ Fresh	MDQ Fresh	MDQ Trans	MD Trans	MD Trans	MD Fresh	
			8m-9m	12m-13m	426m-428m	12m-14m	10m-12m	151m-153m	7m-9m	15m-17m	89.37m-90.71m	106m-108.09m	175m-177m	182.5m-184m	552m-554m	5m-7m	10m-12m	16m-18m	594m-596m		
Ag	ppm	0.07	0	0	0	0	0	0	0	0	0	0	0	0	0	0	1	0	0		
Al	ppm	82,000	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
As	ppm	1.5	2	1	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Ba	ppm	500	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Be	ppm	2.6	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Bi	ppm	0.048	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Cd	ppm	0.11	0	0	0	0	0	0	0	0	1	0	0	0	0	0	0	0	2		
Ce	ppm	68	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Co	ppm	20	1	1	0	1	1	1	1	0	0	0	0	1	0	1	1	1	1		
Cr	ppm	100	0	0	0	0	1	2	0	0	0	0	0	0	0	0	0	0	0		
Cs	ppm	3	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Cu	ppm	50	0	1	0	0	0	0	0	0	0	0	0	3	0	0	0	0	0		
Fe	ppm	950	6	6	5	7	6	6	7	7	7	6	7	7	6	7	6	6	6		
Ga	ppm	18	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Ge	ppm	1.8	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Hf	ppm	5.3	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Hg	ppm	0.05	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
In	ppm	0.049	1	0	0	0	0	0	1	1	2	1	1	1	1	1	0	0	0		
La	ppm	32	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Li	ppm	20	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Mn	ppm	950	0	0	0	0	0	0	1	0	1	1	1	0	0	0	0	0	0		
Mo	ppm	1.5	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Nb	ppm	20	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Ni	ppm	80	0	0	0	0	2	2	0	0	0	0	0	0	0	0	0	0	0		
Pb	ppm	14	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Rb	ppm	90	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Re	ppm	0.0004	2	2	2	2	2	2	2	2	2	2	2	2	2	2	2	2	2		
Sb	ppm	0.2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Sc	ppm	16	1	1	0	1	0	0	1	1	1	1	1	1	1	1	1	1	0		
Se	ppm	0.05	4	4	4	4	4	4	4	4	4	4	4	6	4	4	4	4	4		
Sn	ppm	2.2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Sr	ppm	370	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Ta	ppm	2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Te	ppm	0.005	3	3	3	3	3	3	3	3	3	3	3	6	5	3	3	3	3		
Th	ppm	12	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
Ti	ppm	5600	0	0	0	1	0	0	1	1	0	0	0	0	0	1	0	0	0		
Tl	ppm	0.6	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
U	ppm	2.4	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		
V	ppm	160	1	1	0	1	0	0	0	0	0	0	0	0	0	1	0	0	0		
W	ppm	1	0	0	1	0	0	0	0	0	0	0	0	3	0	0	0	0	0		
Y	ppm	30	0	0	0	0	0	0	0	0	1	1	0	0	1	0	0	0	0		
Zn	ppm	75	0	0	0	0	0	0	1	0	2	0	0	0	0	0	0	0	0		
Zr	ppm	190	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		

Table B-5: Elemental concentrations in leachate for subset of 24 samples, with measurable but not elevated concentrations in yellow and elevated concentrations in red.

Element	Unit	Detection limit	RWC01	RWC05	RWC09	RWC13	RWC17	RWC22	RWC26	RWC30	RWC34	RWC38	RWC42	RWC46	RWC50	RWC54	RWC58	RWC62	Guideline				
			Kopai Crescent																		Freshwater* (ANZECC 2000)	Livestock Drinking (ANZECC 2000)	Human Drinking (NHMRC 2011)
			MB Trans 10m-11m	MB Trans 14m-15m	MB Trans 18m-19m	MB Trans 12m-13m	MD Trans 11m-12m	MD Fresh 31m-32m	MD Oxide 9m-10m	MD Trans 13m-14m	MD Trans 17m-18m	MD Trans 6m-7m	MD Trans 10m-11m	MD Trans 14m-15m	MD Trans 12m-13m	MD Trans 8m-9m	MD Trans 7m-8m	MD Trans 16m-17m					
Aluminium	mg/L	0.1	46.3	51.5	65.7	55.7	37.6	4.01	0.53	2.11	1.23	83.9	15.9	8.09	8.2	10.8	24.9	24.5	0.05	5	0.2		
Arsenic	mg/L	0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	0.013	0.5	0.01		
Barium	mg/L	0.1	0.1	0.1	<0.1	<0.1	0.2	<0.1	<0.1	<0.1	0.2	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	NA	NA	2		
Beryllium	mg/L	0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	NA	NA	0.06		
Boron	mg/L	0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	0.37	5	4		
Cadmium	mg/L	0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	0.0002	0.01	0.002		
Chloride	mg/L	1	336	307	263	100	128	32	1080	235	800	103	208	17	11	22	164	164	NA	NA	250		
Calcium	mg/L	1	6	4	4	4	4	17	5	11	38	6	4	9	7	4	3	2	NA	NA	NA		
Chromium	mg/L	0.01	0.15	0.18	0.18	0.15	0.18	0.01	<0.01	<0.01	<0.01	0.21	0.05	0.02	0.02	0.04	0.07	0.05	0.001	1	0.05		
Cobalt	mg/L	0.01	0.04	0.05	0.07	0.05	0.03	0.01	<0.01	<0.01	<0.01	0.04	0.01	0.01	0.01	0.02	0.04	0.05	NA	1	NA		
Copper	mg/L	0.01	0.1	0.16	0.23	0.32	0.06	0.04	<0.01	<0.01	<0.01	0.29	0.14	0.05	0.04	0.13	0.16	0.18	0.0014	0.5	2		
Iron	mg/L	0.05	95.9	117	153	96.2	64.2	5.6	0.81	1.84	1.8	146	30.4	14.1	17	27.8	52.5	46.4	NA	NA	NA		
Lead	mg/L	0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	0.0034	0.1	0.01		
Magnesium	mg/L	1	20	25	27	23	20	4	39	10	40	25	10	5	6	8	14	16	NA	NA	NA		
Manganese	mg/L	0.01	0.23	0.31	0.43	0.32	0.68	0.08	0.01	0.03	0.02	0.45	0.23	0.2	0.22	0.3	0.57	0.51	1.9	NA	0.5		
Mercury	mg/L	0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	0.0006	0.002	0.001		
Nickel	mg/L	0.01	0.19	0.17	0.2	0.1	0.16	0.02	0.01	0.01	0.01	0.07	0.03	0.02	0.01	0.03	0.05	0.06	0.011	1	0.02		
Potassium	mg/L	1	7	7	6	5	6	2	5	1	14	6	3	2	1	2	2	3	NA	NA	NA		
Selenium	mg/L	0.01	0.02	0.02	0.02	0.02	0.02	0.01	0.01	<0.01	<0.01	0.02	0.01	0.01	<0.01	0.02	0.01	0.01	0.011	0.02	0.01		
Sodium	mg/L	1	207	198	182	109	109	34	620	116	344	128	124	17	17	35	115	115	NA	NA	NA		
Sulphate	mg/L	1	39	33	31	17	20	44	118	23	45	21	27	4	9	6	33	37	NA	1000	NA		
Vanadium	mg/L	0.01	0.17	0.2	0.3	0.2	0.18	0.03	<0.01	<0.01	<0.01	0.32	0.08	0.04	0.04	0.08	0.13	0.17	NA	NA	NA		
Zinc	mg/L	0.01	0.14	0.27	0.39	0.12	0.09	0.02	<0.01	<0.01	<0.01	0.12	0.03	0.02	0.02	0.03	0.09	0.04	0.008	20	NA		
Total Alkalinity	mg/L	1	6	8	10	15	13	17	4	13	3	16	10	25	22	44	8	6	NA	NA	NA		
Final pH	Unit		7.9	8.2	8.5	9.1	8.4	9.1	6.8	8.6	6.8	8.9	8.6	9.4	9.3	9.8	8	8.9	NA	NA	NA		
Final EC	dS/m	0.001	1.050	0.955	0.846	0.434	0.516	0.252	3.540	0.733	2.430	0.453	0.661	0.110	0.094	0.159	0.580	0.579	NA	NA	NA		

*Level of protection (%) species = 95%.

Table B-5 (cont'd): Elemental concentrations in leachate for subset of 24 samples, with measurable but not elevated concentrations in yellow and elevated concentrations in red.

Element	Unit	Detection limit	RWC66 RWC70 RWC73 RWC77 RWC81 RWC85 RWC89 RWC93 RWC97 RWC101 RWC105 RWC109 RWC113 RWC118 RWC121 RWC126 RWC129																	Guideline		
			Kopai Crescent					Bombora												Freshwater* (ANZECC 2000)	Livestock Drinking (ANZECC 2000)	Human Drinking (NHMRC 2011)
			MD Trans 8m-9m	MD Trans 12m-13m	ALP Fresh 426m-428m	MDQ Trans 12m-14m	MR_FW Trans 10m-12m	MR_FW Fresh 151m-153m	MDQ Trans 7m-9m	MDQ Trans 15m-17m	MDQ Fresh 89.37m-90.71m	MDQ Fresh 106m-108.09m	MDQ Fresh 175m-177m	MDQ Fresh 182.5m-184m	MDQ Fresh 552m-554m	MDQ Trans 5m-7m	MD Trans 10m-12m	MD Trans 16m-18m	MD Fresh 594m-596m			
Aluminium	mg/L	0.1	63.6	47.5	2.32	31.4	44	3.94	14.6	19.6	3.15	5.02	2.42	2.18	3.11	8.27	23.4	25.6	6.86	0.05	5	0.2
Arsenic	mg/L	0.01	0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	0.013	0.5	0.01
Barium	mg/L	0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	NA	NA	2
Beryllium	mg/L	0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	NA	NA	0.06
Boron	mg/L	0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	0.37	5	4
Cadmium	mg/L	0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	0.0002	0.01	0.002
Chloride	mg/L	1	188	80	7	85	188	4	63	47	8	4	2	18	3	15	135	60	5	NA	NA	250
Calcium	mg/L	1	6	5	7	2	<1	8	3	2	11	9	9	12	9	3	3	4	6	NA	NA	NA
Chromium	mg/L	0.01	0.17	0.12	<0.01	<0.01	0.28	0.05	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	0.04	0.1	0.03	0.001	1	0.05
Cobalt	mg/L	0.01	0.28	0.14	<0.01	0.02	0.02	<0.01	0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	0.03	0.02	<0.01	NA	1	NA
Copper	mg/L	0.01	0.39	0.67	<0.01	0.1	0.1	<0.01	0.02	0.03	0.01	<0.01	<0.01	0.04	<0.01	0.05	0.08	0.22	0.01	0.0014	0.5	2
Iron	mg/L	0.05	123	104	1.57	76.2	66	4.12	41.7	47.6	7.18	10.4	4.18	4.12	4.8	22	40.9	47.7	8.86	NA	NA	NA
Lead	mg/L	0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	0.0034	0.1	0.01
Magnesium	mg/L	1	20	17	2	7	17	4	4	4	1	<1	<1	<1	<1	3	7	9	7	NA	NA	NA
Manganese	mg/L	0.01	0.32	0.54	0.03	0.2	0.45	0.07	0.11	0.1	0.09	0.14	0.06	0.04	0.08	0.08	0.09	0.11	0.12	1.9	NA	0.5
Mercury	mg/L	0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	<0.0001	0.0006	0.002	0.001
Nickel	mg/L	0.01	0.27	0.2	<0.01	<0.01	0.18	0.02	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	0.02	0.02	0.01	0.011	1	0.02
Potassium	mg/L	1	5	5	10	2	2	6	2	<1	<1	<1	<1	<1	2	<1	1	1	5	NA	NA	NA
Selenium	mg/L	0.01	0.02	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	0.01	<0.01	<0.01	<0.01	0.011	0.02	0.01
Sodium	mg/L	1	140	85	8	78	126	7	58	55	14	7	5	8	4	19	97	68	8	NA	NA	NA
Sulphate	mg/L	1	28	18	1	12	15	<1	8	7	20	<1	2	2	<1	3	14	8	<1	NA	1000	NA
Vanadium	mg/L	0.01	0.27	0.22	<0.01	0.23	0.12	<0.01	0.03	0.05	<0.01	<0.01	<0.01	<0.01	<0.01	0.08	0.12	0.1	0.02	NA	NA	NA
Zinc	mg/L	0.01	0.2	0.18	<0.01	0.04	0.04	<0.01	0.03	0.03	0.03	0.02	<0.01	<0.01	0.01	<0.01	0.02	0.02	0.01	0.008	20	NA
Total Alkalinity	mg/L	1	12	16	27	13	14	28	10	9	22	22	24	30	25	13	13	29	21	NA	NA	NA
Final pH	Unit		8.8	9	9.5	9.2	9	9.6	8.9	9	9.1	9.4	9.5	8.9	9.5	9.3	9.2	9.6	9.4	NA	NA	NA
Final EC	dS/m	0.001	0.640	0.334	0.077	0.327	0.605	0.064	0.237	0.196	0.103	0.052	0.050	0.087	0.050	0.069	0.481	0.262	0.052	NA	NA	NA

*Level of protection (%) species = 95%.

Table B-6: ARD classifications for samples tested as part of this study.

Sample ID	Deposit	Depth From (m)	Depth To (m)	Lithology	Total S (LECO) (%)	Sulfide as S ²⁻ (%)	Sulfate as S-SO ₄ ²⁻ (mg/kg)	Sulfate as S-SO ₄ ²⁻ %	Initial pH	Total Carbon	MPA (kg H ₂ SO ₄ /t)	ARD Classification
					0.01	0.01				%		
RWC01	Kopai Crescent	10	11	MB Trans	0.04	0.01	1000	0.03	6.8	0.08	1.2	NAF
RWC02	Kopai Crescent	11	12	MB Trans	0.04	0.01	1010	0.03	7.5		1.2	NAF
RWC03	Kopai Crescent	12	13	MB Trans	0.03	0.01	900	0.02	7.9		0.9	NAF
RWC04	Kopai Crescent	13	14	MB Trans	0.05	0.01	1300	0.04	6.9		1.5	NAF
RWC05	Kopai Crescent	14	15	MB Trans	0.04	0.01	880	0.03	7.8	0.07	1.2	NAF
RWC06	Kopai Crescent	15	16	MB Trans	0.05	0.01	1200	0.04	7.4		1.5	NAF
RWC07	Kopai Crescent	16	17	MB Trans	0.03	0.01	660	0.02	8.3		0.9	NAF
RWC08	Kopai Crescent	17	18	MB Trans	0.04	0.01	1220	0.03	7.1		1.2	NAF
RWC09	Kopai Crescent	18	19	MB Trans	0.03	0.01	840	0.02	8.3	0.11	0.9	NAF
RWC10	Kopai Crescent	19	20	MB Trans	0.03	0.01	620	0.02	8.5		0.9	NAF
RWC11	Kopai Crescent	10	11	MB Trans	0.03	0.01	860	0.02	8.5		0.9	NAF
RWC12	Kopai Crescent	11	12	MB Trans	0.03	0.01	780	0.02	8.4		0.9	NAF
RWC13	Kopai Crescent	12	13	MB Trans	0.02	0.01	520	0.01	8.6	0.05	0.6	NAF
RWC14	Kopai Crescent	13	14	MB Trans	0.02	0.01	610	0.01	8.6		0.6	NAF
RWC15	Kopai Crescent	14	15	MB Trans	0.02	0.01	600	0.01	8.2		0.6	NAF

Sample ID	Deposit	Depth From (m)	Depth To (m)	Lithology	Total S (LECO) (%)	Sulfide as S ²⁻ (%)	Sulfate as S-SO ₄ ²⁻ (mg/kg)	Sulfate as S-SO ₄ ²⁻ %	Initial pH	Total Carbon %	MPA (kg H ₂ SO ₄ /t)	ARD Classification
					0.01	0.01						
RWC16	Kopai Crescent	10	11	MD Trans	0.02	0.01	710	0.01	7.5		0.6	NAF
RWC17	Kopai Crescent	11	12	MD Trans	0.03	0.01	680	0.02	7.6	0.18	0.9	NAF
RWC18	Kopai Crescent	12	13	MD Trans	0.03	0.01	620	0.02	8.2		0.9	NAF
RWC19	Kopai Crescent	13	14	MD Trans	0.02	0.01	830	0.01	7.4		0.6	NAF
RWC20	Kopai Crescent	14	15	MD Trans	0.04	0.01	1100	0.03	6.4		1.2	NAF
RWC21	Kopai Crescent	30	31	MD Fresh	0.22	0.2	720	0.02	9.1		6.7	NAF
RWC22	Kopai Crescent	31	32	MD Fresh	0.42	0.37	1600	0.05	7.7	0.46	12.9	NAF
RWC23	Kopai Crescent	32	33	MD Fresh	0.32	0.25	2090	0.07	7.3		9.8	NAF
RWC24	Kopai Crescent	7	8	MD Oxide	0.1	0.01	2900	0.09	6.8		3.1	NAF
RWC25	Kopai Crescent	8	9	MD Oxide	0.12	0.01	3350	0.11	6.4		3.7	NAF
RWC26	Kopai Crescent	9	10	MD Oxide	0.1	0.01	3230	0.09	6.4	0.06	3.1	NAF
RWC27	Kopai Crescent	10	11	MD Trans	0.03	0.02	350	0.01	8.4		0.9	NAF
RWC28	Kopai Crescent	11	12	MD Trans	0.06	0.05	400	0.01	7.4		1.8	NAF
RWC29	Kopai Crescent	12	13	MD Trans	0.06	0.04	510	0.02	7		1.8	NAF
RWC30	Kopai Crescent	13	14	MD Trans	0.04	0.02	580	0.02	6.8	0.5	1.2	NAF
RWC31	Kopai Crescent	14	15	MD Trans	0.04	0.02	570	0.02	6.8		1.2	NAF

Sample ID	Deposit	Depth From (m)	Depth To (m)	Lithology	Total S (LECO) (%)	Sulfide as S ²⁻ (%)	Sulfate as S-SO ₄ ²⁻ (mg/kg)	Sulfate as S-SO ₄ ²⁻ %	Initial pH	Total Carbon	MPA (kg H ₂ SO ₄ /t)	ARD Classification
					0.01	0.01		%		%		
RWC32	Kopai Crescent	15	16	MD Trans	0.08	0.04	1340	0.04	6.2		2.4	NAF
RWC33	Kopai Crescent	16	17	MD Trans	0.04	0.01	1010	0.03	6.3		1.2	NAF
RWC34	Kopai Crescent	17	18	MD Trans	0.05	0.01	1180	0.04	6.2	0.16	1.5	NAF
RWC35	Kopai Crescent	18	19	MD Trans	0.11	0.04	2190	0.07	6.2		3.4	NAF
RWC36	Kopai Crescent	19	20	MD Trans	0.03	0.01	480	0.02	6.8		0.9	NAF
RWC37	Kopai Crescent	5	6	MD Trans	0.03	0.01	680	0.02	8.4		0.9	NAF
RWC38	Kopai Crescent	6	7	MD Trans	0.02	0.01	540	0.01	8.4	0.26	0.6	NAF
RWC39	Kopai Crescent	7	8	MD Trans	0.03	0.01	700	0.02	8.1		0.9	NAF
RWC40	Kopai Crescent	8	9	MD Trans	0.04	0.01	980	0.03	7.7		1.2	NAF
RWC41	Kopai Crescent	9	10	MD Trans	0.05	0.01	1100	0.04	8		1.5	NAF
RWC42	Kopai Crescent	10	11	MD Trans	0.03	0.01	680	0.02	6.8	0.1	0.9	NAF
RWC43	Kopai Crescent	11	12	MD Trans	0.06	0.05	280	0.01	8.8		1.8	NAF
RWC44	Kopai Crescent	12	13	MD Trans	0.06	0.05	330	0.01	7.5		1.8	NAF
RWC45	Kopai Crescent	13	14	MD Trans	0.14	0.1	1060	0.04	7.1		4.3	NAF
RWC46	Kopai Crescent	14	15	MD Trans	0.1	0.09	280	0.01	9.2	0.21	3.1	NAF
RWC47	Kopai Crescent	9	10	MD Trans	0.04	0.02	630	0.02	8.3		1.2	NAF

Sample ID	Deposit	Depth From (m)	Depth To (m)	Lithology	Total S (LECO) (%)	Sulfide as S ²⁻ (%)	Sulfate as S-SO ₄ ²⁻ (mg/kg)	Sulfate as S-SO ₄ ²⁻ %	Initial pH	Total Carbon	MPA (kg H ₂ SO ₄ /t)	ARD Classification
					0.01	0.01		%		%		
RWC48	Kopai Crescent	10	11	MD Trans	0.04	0.03	350	0.01	7.8		1.2	NAF
RWC49	Kopai Crescent	11	12	MD Trans	0.04	0.03	310	0.01	9.3		1.2	NAF
RWC50	Kopai Crescent	12	13	MD Trans	0.04	0.03	290	0.01	9.3	0.13	1.2	NAF
RWC51	Kopai Crescent	5	6	MD Trans	0.03	0.01	1000	0.02	8.6		0.9	NAF
RWC52	Kopai Crescent	6	7	MD Trans	0.02	0.01	310	0.01	8.4		0.6	NAF
RWC53	Kopai Crescent	7	8	MD Trans	0.02	0.01	360	0.01	8.9		0.6	NAF
RWC54	Kopai Crescent	8	9	MD Trans	0.01	0.01	140	0.00	10	0.29	0.3	NAF
RWC55	Kopai Crescent	9	10	MD Trans	0.04	0.01	940	0.03	8.8		1.2	NAF
RWC56	Kopai Crescent	5	6	MD Trans	0.03	0.01	570	0.02	8.4		0.9	NAF
RWC57	Kopai Crescent	6	7	MD Trans	0.03	0.01	590	0.02	7.8		0.9	NAF
RWC58	Kopai Crescent	7	8	MD Trans	0.03	0.01	700	0.02	8.1	0.25	0.9	NAF
RWC59	Kopai Crescent	8	9	MD Trans	0.04	0.02	520	0.02	7.8		1.2	NAF
RWC60	Kopai Crescent	9	10	MD Trans	0.03	0.02	390	0.01	7.8		0.9	NAF
RWC61	Kopai Crescent	15	16	MD Trans	0.03	0.01	830	0.02	7.9		0.9	NAF
RWC62	Kopai Crescent	16	17	MD Trans	0.02	0.01	840	0.01	8.2	0.04	0.6	NAF
RWC63	Kopai Crescent	17	18	MD Trans	0.03	0.01	890	0.02	8.3		0.9	NAF

Sample ID	Deposit	Depth From (m)	Depth To (m)	Lithology	Total S (LECO) (%)	Sulfide as S ²⁻ (%)	Sulfate as S-SO ₄ ²⁻ (mg/kg)	Sulfate as S-SO ₄ ²⁻ %	Initial pH	Total Carbon	MPA (kg H ₂ SO ₄ /t)	ARD Classification
					0.01	0.01		%		%		
RWC64	Kopai Crescent	18	19	MD Trans	0.02	0.01	740	0.01	7.8		0.6	NAF
RWC65	Kopai Crescent	19	20	MD Trans	0.05	0.02	1020	0.03	7.8		1.5	NAF
RWC66	Kopai Crescent	8	9	MD Trans	0.02	0.01	860	0.01	8.4	0.08	0.6	NAF
RWC67	Kopai Crescent	9	10	MD Trans	0.03	0.01	680	0.02	8.7		0.9	NAF
RWC68	Kopai Crescent	10	11	MD Trans	0.02	0.01	620	0.01	8.6		0.6	NAF
RWC69	Kopai Crescent	11	12	MD Trans	0.02	0.01	560	0.01	8.6		0.6	NAF
RWC70	Kopai Crescent	12	13	MD Trans	0.01	0.01	470	0.00	9	0.03	0.3	NAF
RWC71	Bombora	160.85	162.21	ALP Fresh	0.18	0.16	720	0.02	9.7		5.5	NAF
RWC72	Bombora	105.42	107.01	ALP Fresh	0.16	0.15	330	0.01	9.7		4.9	NAF
RWC73	Bombora	426	428	ALP Fresh	0.12	0.12	130	0.00	9.7	0.8	3.7	NAF
RWC74	Bombora	5	7.1	MDQ Trans	0.02	0.01	320	0.01	9.2		0.6	NAF
RWC75	Bombora	8	10	MDQ Trans	0.01	0.01	340	0.00	8.8		0.3	NAF
RWC76	Bombora	10	12	MDQ Trans	0.01	0.01	180	0.00	9.2		0.3	NAF
RWC77	Bombora	12	14	MDQ Trans	0.01	0.01	300	0.00	8.8	0.02	0.3	NAF
RWC78	Bombora	14	16	MDQ Trans	0.04	0.03	210	0.01	9.3		1.2	NAF
RWC79	Bombora	6	8	MR_FW Trans	0.04	0.01	1340	0.03	7.4		1.2	NAF
RWC80	Bombora	8	10	MR_FW Trans	0.03	0.01	830	0.02	7.8		0.9	NAF
RWC81	Bombora	10	12	MR_FW Trans	0.01	0.01	340	0.00	8.4	0.02	0.3	NAF
RWC82	Bombora	12	14	MR_FW Trans	0.02	0.01	350	0.01	8.5		0.6	NAF
RWC83	Bombora	147	149	MR_FW Fresh	0.01	0.01	100	0.00	9.6		0.3	NAF
RWC84	Bombora	149	151	MR_FW Fresh	0.01	0.01	100	0.00	9.6		0.3	NAF
RWC85	Bombora	151	153	MR_FW Fresh	0.01	0.01	100	0.00	9.6	0.39	0.3	NAF
RWC86	Bombora	153	155	MR_FW Fresh	0.01	0.01	120	0.00	9.9		0.3	NAF
RWC87	Bombora	155	157	MR_FW Fresh	0.01	0.01	340	0.00	9.2		0.3	NAF

Sample ID	Deposit	Depth From (m)	Depth To (m)	Lithology	Total S (LECO) (%)	Sulfide as S ²⁻ (%)	Sulfate as S-SO ₄ ²⁻ (mg/kg)	Sulfate as S-SO ₄ ²⁻ %	Initial pH	Total Carbon %	MPA (kg H ₂ SO ₄ /t)	ARD Classification
					0.01	0.01						
RWC88	Bombora	5	7	MDQ Trans	0.02	0.01	390	0.01	8.6		0.6	NAF
RWC89	Bombora	7	9	MDQ Trans	0.01	0.01	280	0.00	8.7	0.02	0.3	NAF
RWC90	Bombora	9	11	MDQ Trans	0.01	0.01	320	0.00	8.6		0.3	NAF
RWC91	Bombora	11	13	MDQ Trans	0.02	0.01	280	0.01	9		0.6	NAF
RWC92	Bombora	13	15	MDQ Trans	0.01	0.01	240	0.00	9.2		0.3	NAF
RWC93	Bombora	15	17	MDQ Trans	0.01	0.01	280	0.00	9	0.02	0.3	NAF
RWC94	Bombora	6.9	9	MDQ Trans	0.02	0.01	330	0.01	8.5		0.6	NAF
RWC95	Bombora	11	13	MDQ Trans	0.01	0.01	230	0.00	9.4		0.3	NAF
RWC96	Bombora	78	80	MDQ Fresh	0.13	0.07	1830	0.06	8.4		4.0	NAF
RWC97	Bombora	89.37	90.71	MDQ Fresh	0.42	0.41	390	0.01	9.4	0.17	12.9	NAF
RWC98	Bombora	90.71	92	MDQ Fresh	0.5	0.42	2310	0.08	9.5		15.3	NAF
RWC99	Bombora	102	104	MDQ Fresh	0.14	0.13	170	0.01	9.9		4.3	NAF
RWC100	Bombora	104	106	MDQ Fresh	0.03	0.02	160	0.01	9.9		0.9	NAF
RWC101	Bombora	106	108.09	MDQ Fresh	0.03	0.02	140	0.01	9.8	0.11	0.9	NAF
RWC102	Bombora	108.72	110	MDQ Fresh	0.25	0.24	250	0.01	9.8		7.7	NAF
RWC103	Bombora	110	112	MDQ Fresh	0.29	0.23	1740	0.06	9.7		8.9	NAF
RWC104	Bombora	112	114	MDQ Fresh	0.05	0.04	370	0.01	9.5		1.5	NAF
RWC105	Bombora	175	177	MDQ Fresh	0.16	0.13	1020	0.03	9.7	0.26	4.9	NAF
RWC106	Bombora	177	179	MDQ Fresh	0.2	0.16	1160	0.04	9.8		6.1	NAF
RWC107	Bombora	179	181	MDQ Fresh	0.28	0.24	1140	0.04	9.6		8.6	NAF
RWC108	Bombora	181	182.5	MDQ Fresh	0.24	0.22	720	0.02	9.6		7.3	NAF
RWC109	Bombora	182.5	184	MDQ Fresh	1.97	1.76	6160	0.21	9.3	0.51	60.3	PAF
RWC110	Bombora	544.04	545.72	MDQ Fresh	0.05	0.05	120	0.00	9.7		1.5	NAF
RWC111	Bombora	545.72	547	MDQ Fresh	0.02	0.02	100	0.00	9.6		0.6	NAF
RWC112	Bombora	550	552	MDQ Fresh	0.13	0.12	260	0.01	9.7		4.0	NAF
RWC113	Bombora	552	554	MDQ Fresh	0.2	0.19	370	0.01	9.7	0.28	6.1	NAF
RWC114	Bombora	554	556	MDQ Fresh	0.05	0.05	100	0.00	9.7		1.5	NAF
RWC115	Bombora	5	7	MDQ Trans	0.06	0.01	1750	0.05	7.8		1.8	NAF
RWC116	Bombora	7	9	MDQ Trans	0.01	0.01	130	0.00	9.4		0.3	NAF
RWC117	Bombora	9	11	MDQ Trans	0.02	0.02	130	0.00	9.1		0.6	NAF

Sample ID	Deposit	Depth From (m)	Depth To (m)	Lithology	Total S (LECO) (%)	Sulfide as S ²⁻ (%)	Sulfate as S-SO ₄ ²⁻ (mg/kg)	Sulfate as S-SO ₄ ²⁻ %	Initial pH	Total Carbon %	MPA (kg H ₂ SO ₄ /t)	ARD Classification
					0.01	0.01			0.1	%		
RWC118	Bombora	5	7	MDQ Trans	0.01	0.01	100	0.00	9.6	0.02	0.3	NAF
RWC119	Bombora	7	9	MDQ Trans	0.02	0.01	220	0.01	9		0.6	NAF
RWC120	Bombora	9	11	MDQ Trans	0.01	0.01	100	0.00	9.6		0.3	NAF
RWC121	Bombora	10	12	MD Trans	0.01	0.01	300	0.00	9.1	0.02	0.3	NAF
RWC122	Bombora	12	14	MD Trans	0.01	0.01	250	0.00	9.6		0.3	NAF
RWC123	Bombora	14	16	MD Trans	0.01	0.01	300	0.00	9.6		0.3	NAF
RWC124	Bombora	18	20	MD Trans	0.01	0.01	300	0.00	9.9		0.3	NAF
RWC125	Bombora	20	22	MD Trans	0.01	0.01	340	0.00	10		0.3	NAF
RWC126	Bombora	16	18	MD Trans	0.02	0.01	280	0.01	9.7	0.04	0.6	NAF
RWC127	Bombora	18	19.85	MD Trans	0.02	0.01	240	0.01	10.1		0.6	NAF
RWC128	Bombora	592	594	MD Fresh	0.01	0.01	120	0.00	9.8		0.3	NAF
RWC129	Bombora	594	596	MD Fresh	0.01	0.01	110	0.00	9.7	0.03	0.3	NAF
RWC130	Bombora	596	598	MD Fresh	0.09	0.05	1210	0.04	9.8		2.8	NAF

Table B-7: Acid-base accounting values and ARD classifications for samples with recorded total S >0.3%.

Sample ID	Deposit	Depth From	Depth To	Lithology	Sulfur - Total as S (LECO)	Sulfide as S ₂ ⁻	NAPP	NAG pH (ox)	NAG (pH 4.5)	NAG (pH 7.0)	ANC as H ₂ SO ₄	ANC as CaCO ₃	Total Carbon	MPA	ANC:MPA ratio	ARD Classification
					%	%	kg H ₂ SO ₄ /t	pH Unit	kg H ₂ SO ₄ /t	kg H ₂ SO ₄ /t	kg H ₂ SO ₄ equiv./t	% CaCO ₃	%	kg H ₂ SO ₄ /t		
					0.01	0.01	0.5	0.1	0.1	0.1	0.5	0.1	0			
RWC22	Kopai Crescent	31.00	32.00	MD Fresh	0.38	0.37	-37.90	8.70	<0.1	<0.1	49.50	5.00	0.46	11.63	4.26	NAF
RWC23	Kopai Crescent	32.00	33.00	MD Fresh	0.30	0.25	-19.90	7.80	<0.1	<0.1	29.10	3.00	0.00	9.18	3.17	NAF
RWC97	Bombora	89.37	90.71	MDQ Fresh	0.42	0.41	-18.00	7.40	<0.1	<0.1	30.20	3.10	0.17	12.85	2.35	NAF
RWC98	Bombora	90.71	92.00	MDQ Fresh	0.50	0.42	-6.80	5.40	<0.1	1.30	20.60	2.10	0.00	15.30	1.35	NAF
RWC109	Bombora	182.50	184.00	MDQ Fresh	1.97	1.76	1.60	3.00	12.80	16.70	57.80	5.90	0.51	60.28	0.96	PAF

APPENDIX C: ANALYSIS OF FIBROUS CONTENT

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Certificate of Analysis PFK1267

Client Details

Client	Meta Maya Group Pty Ltd
Contact	Jason Linford
Address	Unit 5, 78 Catalano Circuit, CANNING VALE, WA, 6155

Sample Details

Your Reference	Asbestos Bulk ID
Number of Samples	13 Aggregate
Date Samples Received	19/11/2024
Date Instructions Received	19/11/2024

Analysis Details

Please refer to the following pages for results, methodology summary and quality control data.
Samples were analysed as received from the client. Results relate specifically to the samples as received.
Results are reported on a dry weight basis for soils and on an as received basis for other matrices.

Report Details

Date Results Requested by	21/11/2024
Date of Issue	21/11/2024

NATA Accreditation Number 2901. This document shall not be reproduced except in full.

Accredited for compliance with ISO/IEC 17025. Tests not covered by NATA are denoted with *.

Authorisation Details

Asbestos Approved By	Analysed by Asbestos Approved Analyst: Lalanee Rupasinghe Authorised by Asbestos Approved Signatory: Lalanee Rupasinghe
Results Approved By	Lalanee Rupasinghe, Asbestos Analyst
Laboratory Manager	Michael Kubiak

Certificate of Analysis PFK1267

Samples in this Report

Envirolab ID	Sample ID	Matrix	Date Sampled	Date Received
PFK1267-01	RWC02 - waste	Aggregate	18/11/2024	19/11/2024
PFK1267-02	RWC11 - waste	Aggregate	18/11/2024	19/11/2024
PFK1267-03	RWC21 - waste	Aggregate	18/11/2024	19/11/2024
PFK1267-04	RWC25 - waste	Aggregate	18/11/2024	19/11/2024
PFK1267-05	RWC32 - waste	Aggregate	18/11/2024	19/11/2024
PFK1267-06	RWC56 - waste	Aggregate	18/11/2024	19/11/2024
PFK1267-07	RWC71 - waste	Aggregate	18/11/2024	19/11/2024
PFK1267-08	RWC75 - waste	Aggregate	18/11/2024	19/11/2024
PFK1267-09	RWC79 - waste	Aggregate	18/11/2024	19/11/2024
PFK1267-10	RWC84 - waste	Aggregate	18/11/2024	19/11/2024
PFK1267-11	RWC99 - waste	Aggregate	18/11/2024	19/11/2024
PFK1267-12	RWC123 - waste	Aggregate	18/11/2024	19/11/2024
PFK1267-13	RWC128 - waste	Aggregate	18/11/2024	19/11/2024

Certificate of Analysis PFK1267

Asbestos ID in Aggregate

Client ID	Envirolab ID	Description	Result
RWC02 - waste	PFK1267-01	150x150x40mm Rock Dust	No Asbestos Detected No trace fibres detected
RWC11 - waste	PFK1267-02	180x150x40mm Rock Dust	Unknown Mineral Fibre detected# Trace Analysis Not Applicable # May or may not be asbestos [see methodology]
RWC21 - waste	PFK1267-03	150x150x40mm Rock Dust	Unknown Mineral Fibre detected# Trace Analysis Not Applicable # May or may not be asbestos [see methodology]
RWC25 - waste	PFK1267-04	150x150x40mm Dust	No Asbestos Detected No trace fibres detected
RWC32 - waste	PFK1267-05	150x150x40mm Rock Dust	Unknown Mineral Fibre detected# Trace Analysis Not Applicable # May or may not be asbestos [see methodology]
RWC56 - waste	PFK1267-06	150x150x40mm Rock Dust	Unknown Mineral Fibre detected# Trace Analysis Not Applicable # May or may not be asbestos [see methodology]
RWC71 - waste	PFK1267-07	150x150x40mm Rock	No Asbestos Detected No trace fibres detected
RWC75 - waste	PFK1267-08	150x150x40mm Rock	No Asbestos Detected No trace fibres detected
RWC79 - waste	PFK1267-09	150x150x40mm Rock Dust	Unknown Mineral Fibre detected# Trace Analysis Not Applicable # May or may not be asbestos [see methodology]
RWC84 - waste	PFK1267-10	150x150x40mm Rock	No Asbestos Detected No trace fibres detected
RWC99 - waste	PFK1267-11	140x130x50mm Rock	No Asbestos Detected No trace fibres detected
RWC123 - waste	PFK1267-12	220x150x60mm Rock	Unknown Mineral Fibre detected# Trace Analysis Not Applicable # May or may not be asbestos [see methodology]
RWC128 - waste	PFK1267-13	250x150x60mm Rock	Unknown Mineral Fibre detected# Trace Analysis Not Applicable # May or may not be asbestos [see methodology]

Certificate of Analysis PFK1267

Method Summary

Method ID	Methodology Summary
ASB-001_AS4964	Asbestos ID - Qualitative identification of asbestos in bulk samples using Polarised Light Microscopy and Dispersion Staining Techniques including Synthetic Mineral Fibre and Organic Fibre as per Australian Standard 4964-2004. When mineral fibres of unknown type are detected by polarized light microscopy including dispersion staining, the fibres detected may or may not be asbestos fibres. To confirm the identities, another independent analytical technique may be required. Visible asbestos is defined as asbestos detected below the reporting limit of 0.1g/kg as per AS4964-2004

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Result Definitions

Identifier	Description
NR	Not reported
NEPM	National Environment Protection Measure
NS	Not specified
LCS	Laboratory Control Sample
RPD	Relative Percent Difference
>	Greater than
<	Less than
PQL	Practical Quantitation Limit
INS	Insufficient sample for this test
NA	Test not required
NT	Not tested
DOL	Samples rejected due to particulate overload (air filters only)
RFD	Samples rejected due to filter damage (air filters only)
RUD	Samples rejected due to uneven deposition (air filters only)
##	Indicates a laboratory acceptance criteria outlier, for further details, see Result Comments and/or QC Comments

Quality Control Definitions

Blank

This is the component of the analytical signal which is not derived from the sample but from reagents, glassware etc, and is determined by processing solvents and reagents in exactly the same manner as for samples.

Surrogate Spike

Surrogates are known additions to each sample, blank, matrix spike and LCS in a batch, of compounds which are similar to the analyte of interest, however are not expected to be found in real samples.

LCS (Laboratory Control Sample)

This comprises either a standard reference material or a control matrix (such as a blank sand or water) fortified with analytes representative of the analyte class. It is simply a check sample.

Matrix Spike

A portion of the sample is spiked with a known concentration of target analyte. The purpose of the matrix spike is to monitor the performance of the analytical method used and to determine whether matrix interferences exist.

Duplicate

This is the complete duplicate analysis of a sample from the process batch. The sample selected should be one where the analyte concentration is easily measurable.

Certificate of Analysis PFK1267

Laboratory Acceptance Criteria

Duplicate sample and matrix spike recoveries may not be reported on smaller jobs, however, were analysed at a frequency to meet or exceed NEPM requirements. All samples are tested in batches of 20. The duplicate sample RPD and matrix spike recoveries for the batch were within the laboratory acceptance criteria. Filters, swabs, wipes, tubes and badges will not have duplicate data as the whole sample is generally extracted during sample extraction. Spikes for Physical and Aggregate Tests are not applicable. For VOCs in water samples, three vials are required for duplicate or spike analysis.

General Acceptance Criteria (GAC) - Analyte specific criteria applies for some analytes and is reflected in QC recovery tables.

Duplicates: >10xPQL - RPD acceptance criteria will vary depending on the analytes and the analytical techniques but is typically in the range 20%-50% - see ELN-P05 QAQC tables for details (available on request); <10xPQL - RPD are higher as the results approach PQL and the estimated measurement uncertainty will statistically increase. Matrix Spikes, LCS and Surrogate recoveries: Generally 70-130% for inorganics/metals; 60-140% for organics (+/-50% surrogates) and 10-140% for labile SVOCs (including labile surrogates), ultra trace organics and speciated phenols is acceptable.

In circumstances where no duplicate and/or sample spike has been reported at 1 in 10 and/or 1 in 20 samples respectively, the sample volume submitted was typically insufficient in order to satisfy laboratory QA/QC protocols.

Miscellaneous Information

When samples are received where certain analytes are outside of recommended technical holding times (THTs), the analysis has proceeded. Where analytes are on the verge of breaching THTs, every effort will be made to analyse within the THT or as soon as practicable.

Where sampling dates are not provided, Envirolab are not in a position to comment on the validity of the analysis where recommended technical holding times may have been breached. We have taken the sampling date as being the date received at the laboratory.

Two significant figures are reported for the majority of tests and with a high degree of confidence, for results <10*PQL, the second significant figure may be in doubt i.e. has a relatively high degree of uncertainty and is provided for information only.

Measurement Uncertainty estimates are available for most tests upon request.

Analysis of aqueous samples typically involves the extraction/digestion and/or analysis of the liquid phase only (i.e. NOT any settled sediment phase but inclusive of suspended particles if present), unless stipulated on the Envirolab COC or by correspondence. Notable exceptions include certain Physical Tests (pH/EC/BOD/COD/Apparent Colour etc.), Solids testing, Total Recoverable metals and PFAS where sediment/solids are included by default.

Urine Analysis - The BEI values listed are taken from the 2022 edition of *TLVs and BEIs Threshold Limits by ACGIH*.

Air volume measurements are not covered by Envirolab's NATA accreditation.

Data Quality Assessment Summary PFK1267

Client Details

Client	Meta Maya Group Pty Ltd
Your Reference	Asbestos Bulk ID
Date Issued	21/11/2024

Recommended Holding Time Compliance

No recommended holding time exceedances

Quality Control and QC Frequency

QC Type	Compliant	Details
Blank	Yes	No Outliers
LCS	Yes	No Outliers
Duplicates	Yes	No Outliers
Matrix Spike	Yes	No Outliers
Surrogates / Extracted Internal Standards	Yes	No Outliers
QC Frequency	Yes	No Outliers

Surrogates/Extracted Internal Standards, Duplicates and/or Matrix Spikes are not always relevant/applicable to certain analyses and matrices. Therefore, said QC measures are deemed compliant in these situations by default. See Laboratory Acceptance Criteria for more information

Data Quality Assessment Summary PFK1267

Recommended Holding Time Compliance

Analysis	Sample Number(s)	Date Sampled	Date Extracted	Date Analysed	Compliant
Asbestos-Bulk Materials Material	1-13	18/11/2024	21/11/2024	21/11/2024	Yes

Client: Landloch
Job number: 25_0206
Lab ID: 25_0206_001
Client ID: RWC11
Analysis: Fibre characterisation by scanning electron microscopy (SEM) with elemental analysis by energy dispersive spectroscopy (EDS) following AS 4964-2004 (modified for SEM)
Revision number: 0
Comments: None

Date received: 30/01/2025
Date analysed: 24/02/2025
Date reported: 28/02/2025

Executive summary

The sample was determined to contain < **0.01 wt % asbestos** mineral fibre.

Sample preparation

The sample was supplied to Microanalysis Australia as a bulk sample. The sample was screened at 2 mm, and each size fraction was inspected visually (both with the naked eye and with a low-powered microscope) for fibrous material.

As no fibrous material was visually apparent for immediate pre-selection, a representative sub-sample of the <2 mm fraction was disaggregated and was placed on top of at least one double-sided carbon tab before being carbon coated. Non-conducting samples require coating prior to SEM analysis to prevent charging whilst being analysed by the electron beam.

Analysis

The sample was analysed using a Carl Zeiss EVO50 scanning electron microscope (SEM) fitted with an Oxford INCA energy dispersive spectrometer (EDS). The sample was scanned at low magnification to identify any possible fibre clusters before the magnification was increased to 2000x magnification for closer examination.

EDS is a semi-quantitative technique (at best) on well prepared, optically flat samples. Factors such as sample unevenness may adversely bias elemental concentration interpretation. EDS has a spatial resolution of ~5 µm meaning spectra from particles less than this size may contain elemental concentrations biased by their surroundings.

All images were acquired using backscatter electrons. Image brightness is proportional to average atomic number – the brighter the pixel, the higher the atomic number of the element.

Summary

Following AS 4964, asbestos is defined as:

- Many particles with aspect ratios (i.e. length to width ratios) ranging from 20:1 to 100:1 or higher for particles >5 micrometers in length. Bundles of fibres may have lower aspect ratios;
- Sets of fibre bundles generally less than 0.5, but always less than 1.0 micrometres in width, unless in thick bundles;
- In addition to the mandatory fibrillar crystal growth, one or more, and preferably three of the following aspects:
 - Parallel fibres occurring in bundles;
 - Fibre bundles displaying splayed ends;
 - Matted masses of individual fibres;
 - Fibres showing characteristic curvature.
- Respirable asbestos fibres are defined as:
 - Asbestos fibres less than 3 µm in width, and greater than 5 µm in length, and with a length to width ratio greater than 3 to 1.

Less than three observed fibres had an elemental composition and morphology indicative of asbestos mineral fibre, according to the definition in AS 4964.

Depending upon sample condition, composition and fibre type, the detection limit of AS 4964 has been found to lie generally in the range of 1 in 1,000 (0.1 wt %) to 1 in 10,000 (0.01 wt %) parts by weight, equivalent to 1 to 0.1 g/kg.

For this report, a limit of reporting (LoR) of 0.01 wt % has been assumed, as stated in AS 4964.

A selection of images/fields and associated elemental spectra are reported below. The fields are not representative of the area analysed. The images/fields were reported due to their higher fibre count.

Fibre #	Image/Field #	Diameter (µm)	Length (µm)	Aspect ratio	Major Elements	Minor Elements	Morphology	Assigned mineralogy
1	1/1	0.5	7.2	14 :1	O, Si	Mg, Al, Fe, Ca, Na	Irregular ends	Hornblende
2	2/3	1.2	14.2	12 :1	O, Si	Mg, Al, Fe, Ca, Na	Non-parallel sides	Hornblende
3	4/7	0.7	20.9	30 :1	O, Si	Mg, Al, Fe, Ca, Na	Non-parallel sides	Hornblende
4	6/10	0.7	7.2	10 :1	O, Si	Mg, Al, Fe, Ca, Na	Non-parallel sides	Hornblende
5	8/13	1.3	18.6	14 :1	O, Si	Mg, Al, Fe, Ca, Na	Non-parallel sides	Hornblende
6	9/15	1.0	11.4	11 :1	O, Si	Mg, Al, Fe, Ca, Na	Non-parallel sides	Hornblende
7	11/17	0.4	19.7	49 :1	O, Si	Mg, Al, Fe, Ca, Na	Non-parallel sides	Hornblende

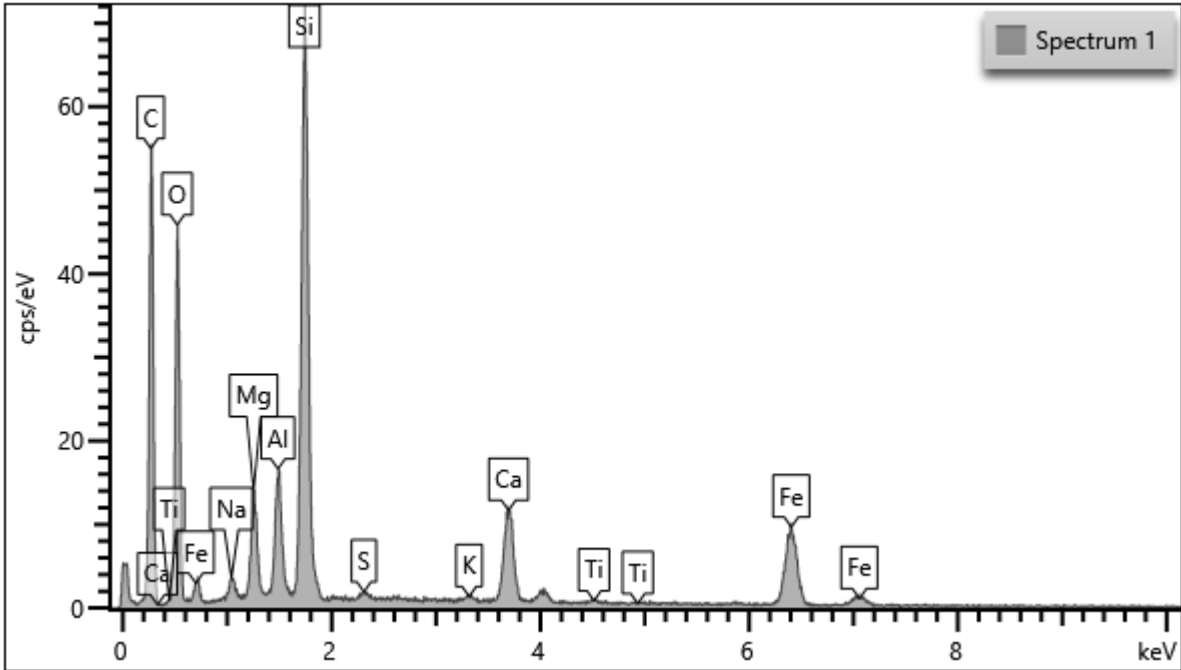
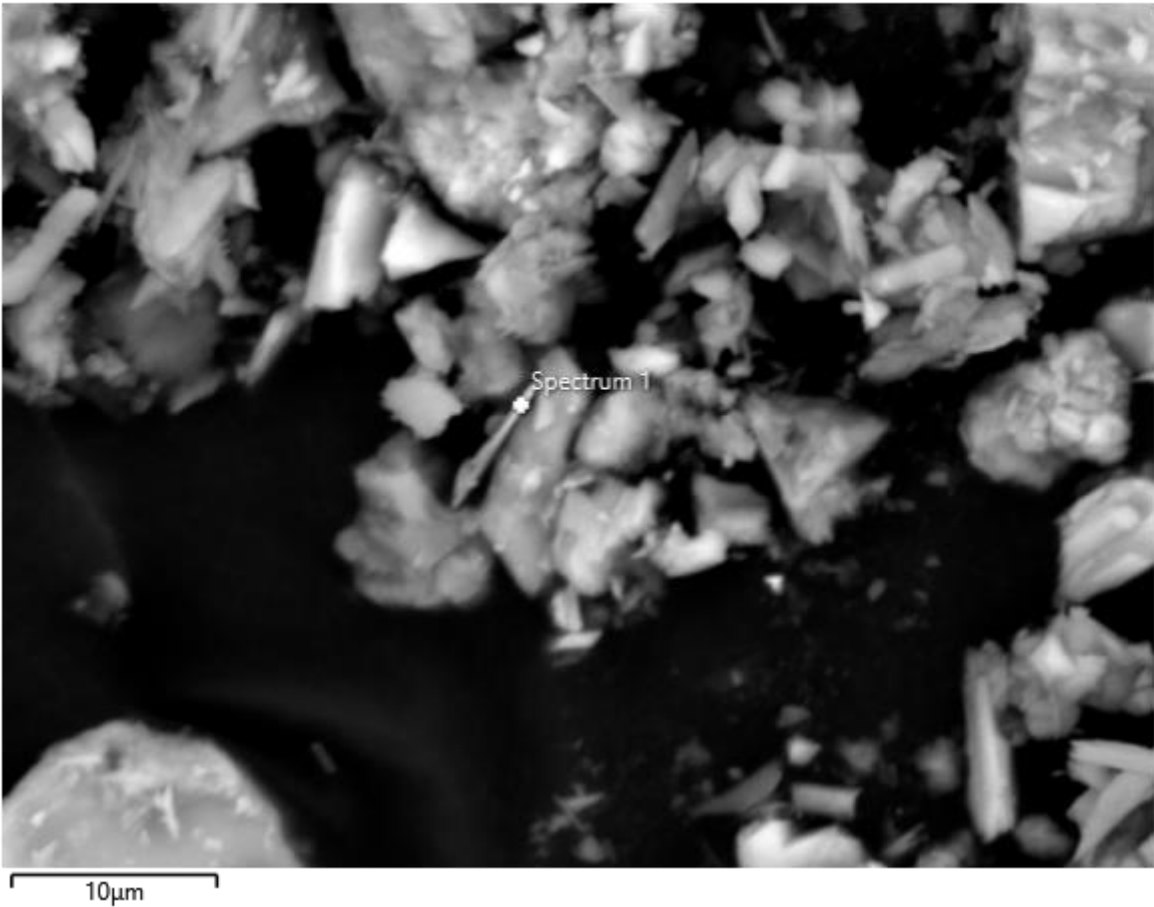
It should be noted that the higher resolution of the SEM may increase the number of fibres observed when compared with optical microscopy. Positive identification of asbestos fibres by SEM EDS uses elemental information that is not available by optical methods.

Analyst: Damon Blakey, *B.Sc. (Forensic Biology and Toxicology)*

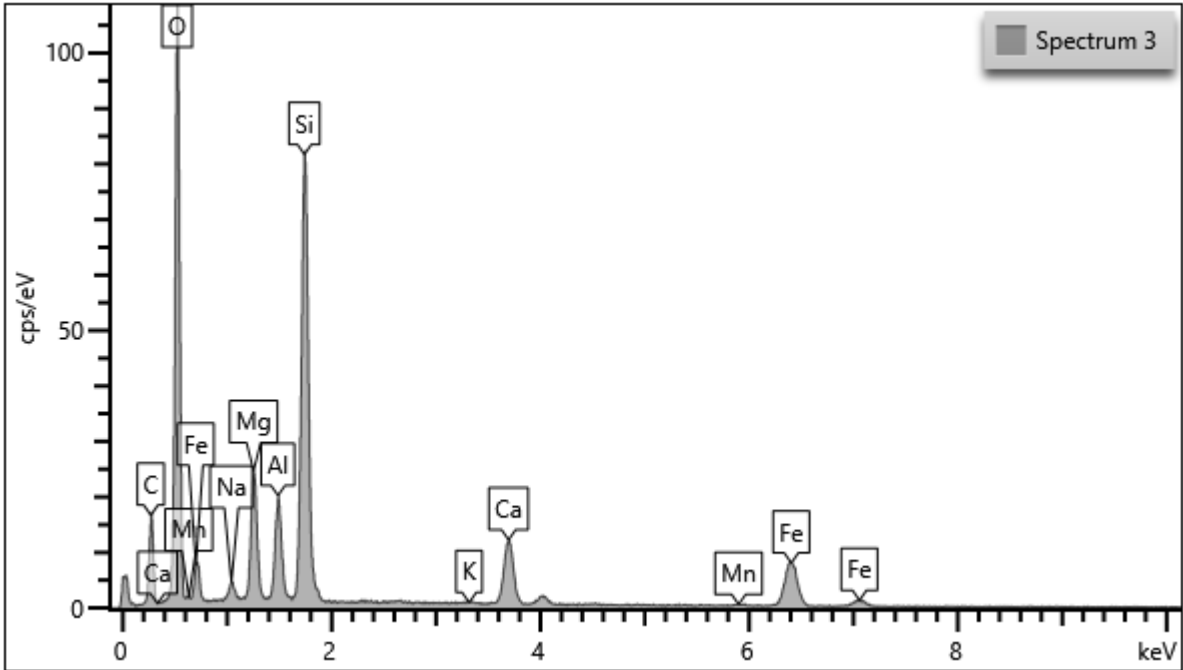
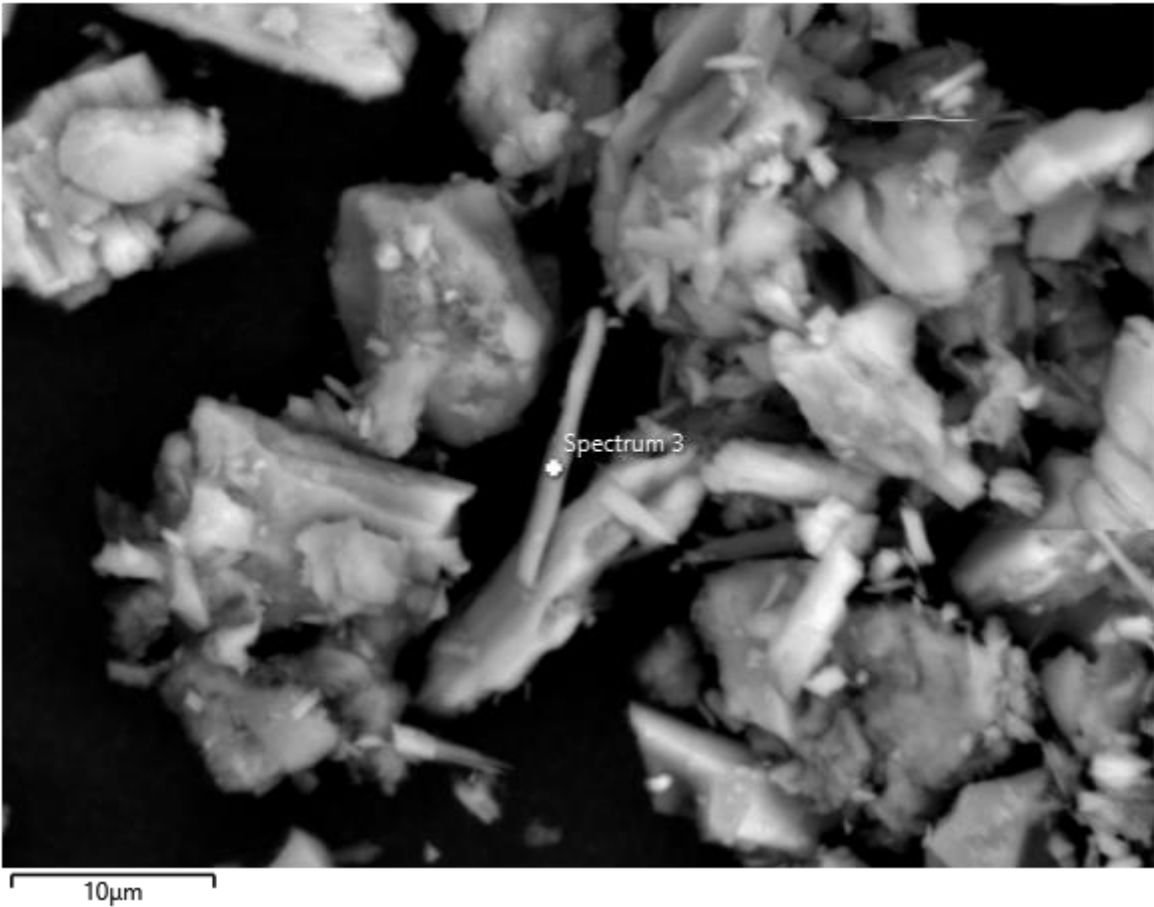
Reported by: Damon Blakey, *B.Sc. (Forensic Biology and Toxicology)*

Approved: Nimue Pendragon, *B.Sc.(Nanotechnology)*

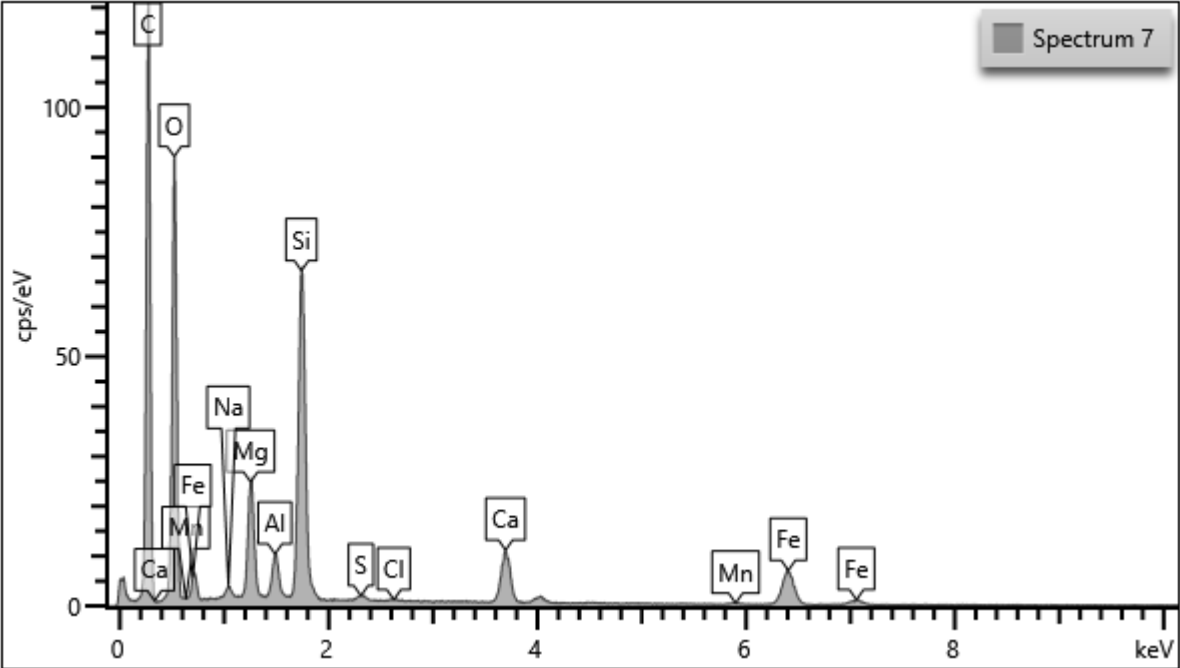
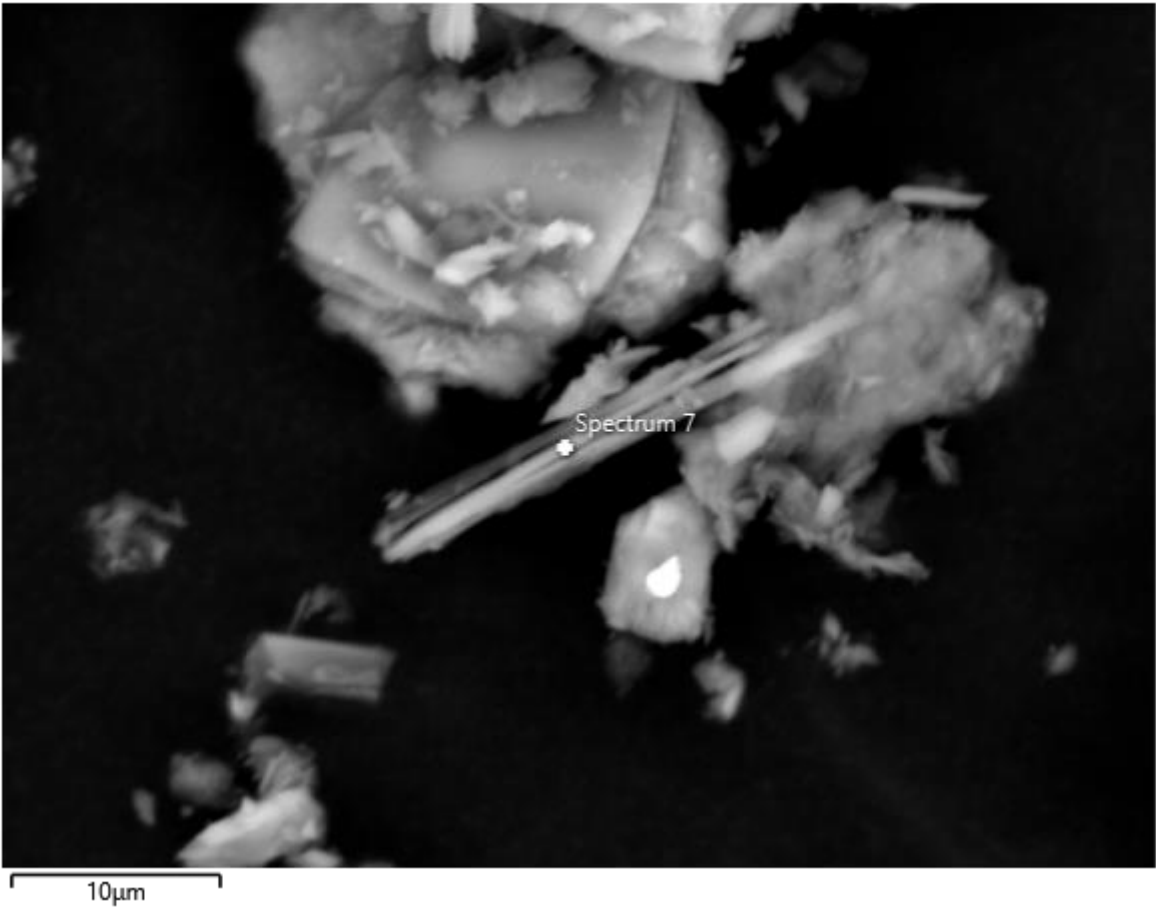
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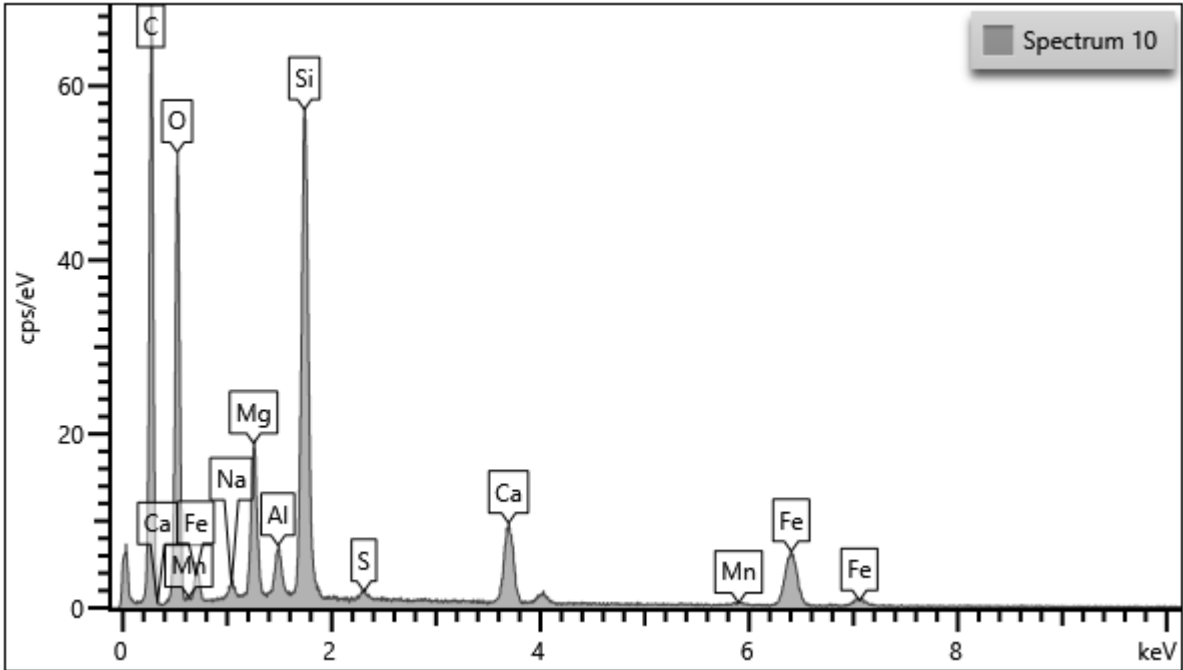
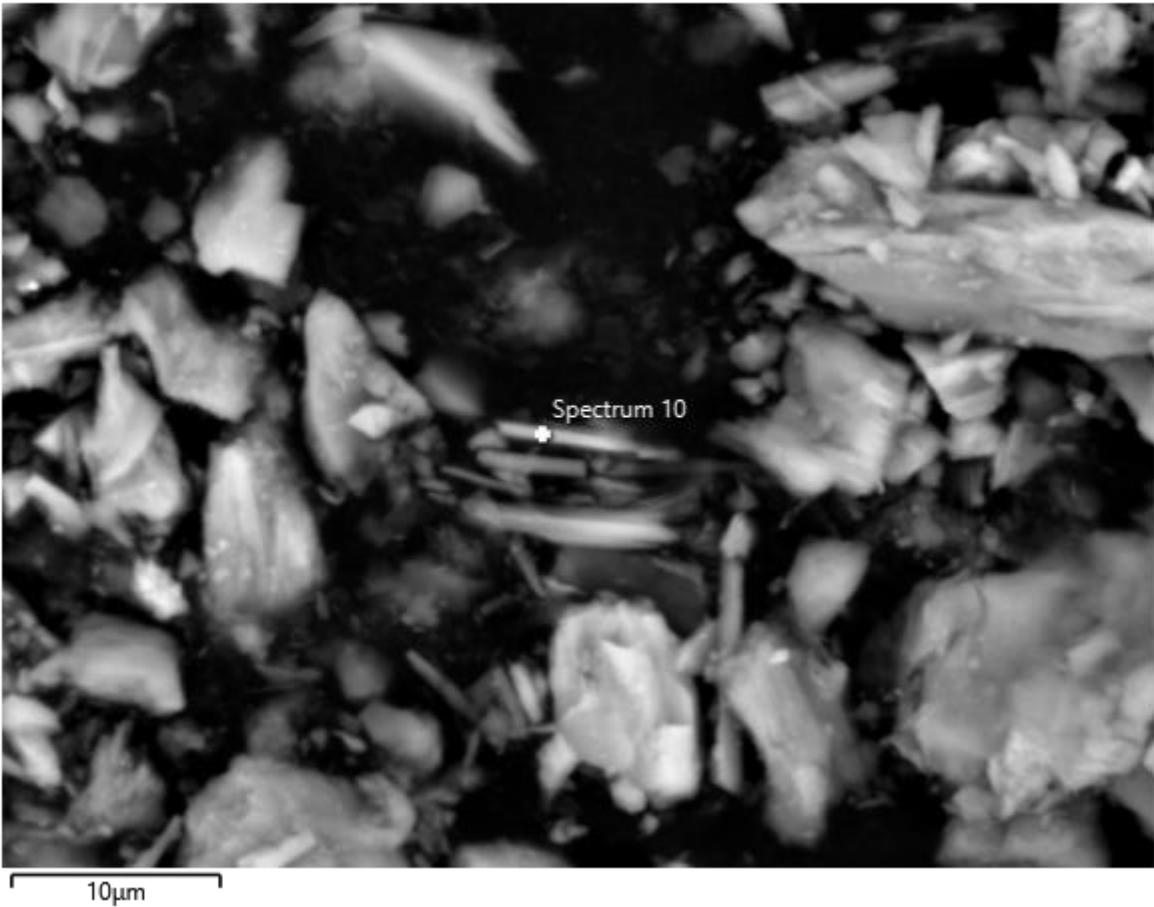
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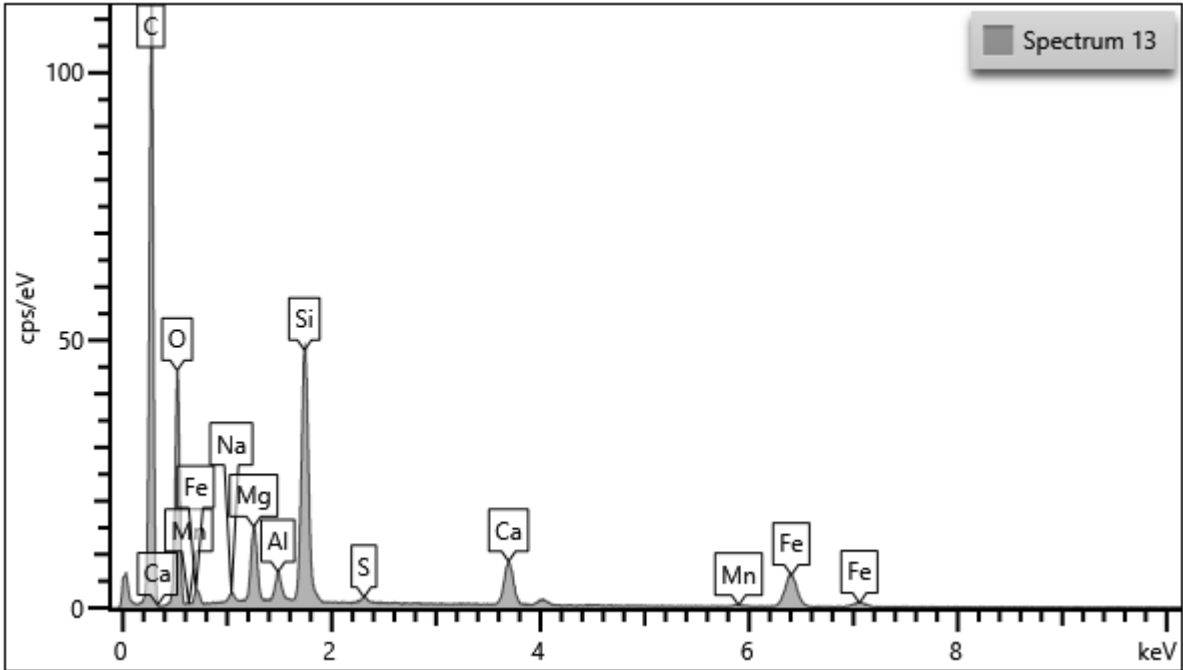
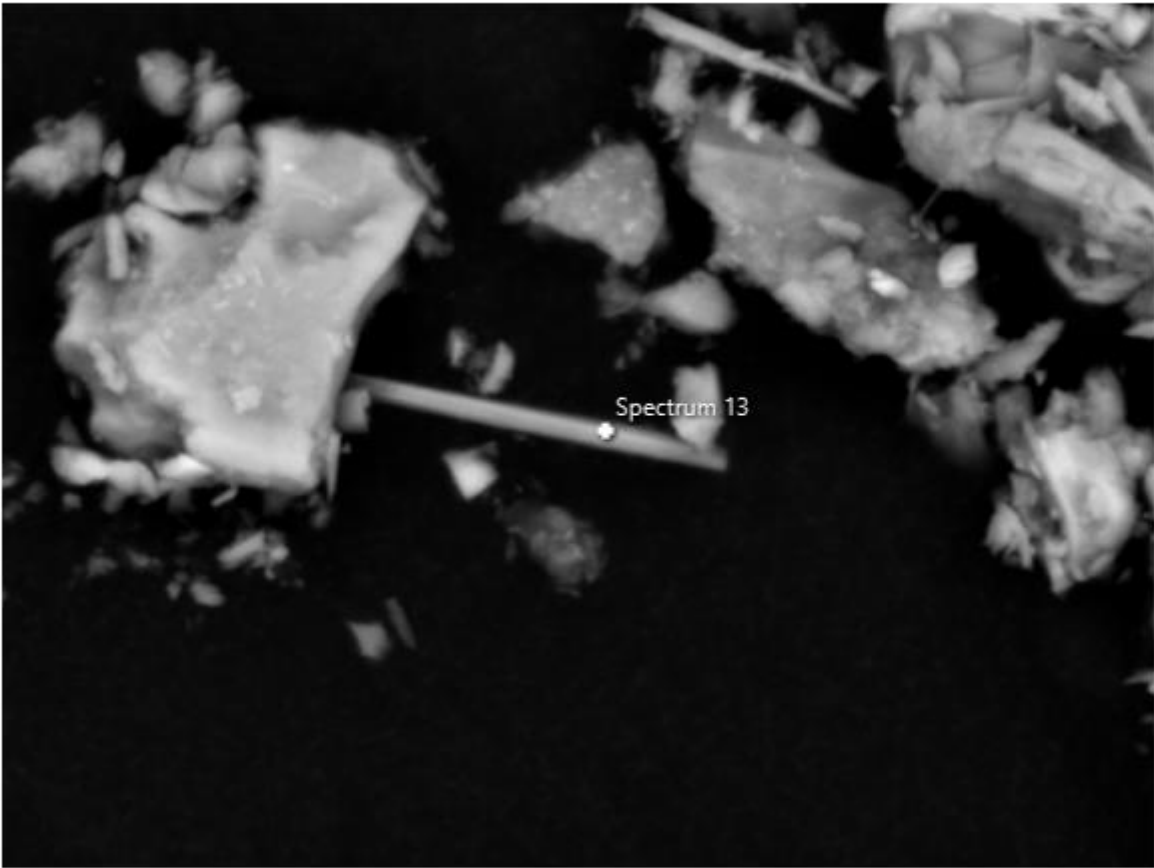
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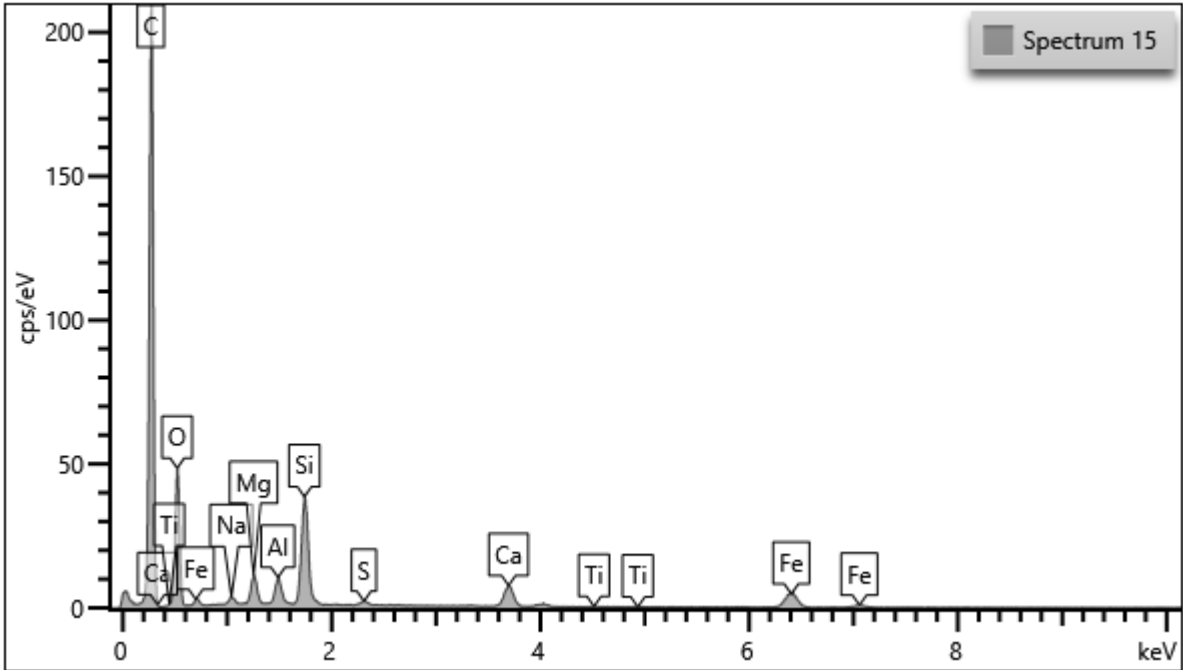
Electron Image 6 (Input1)



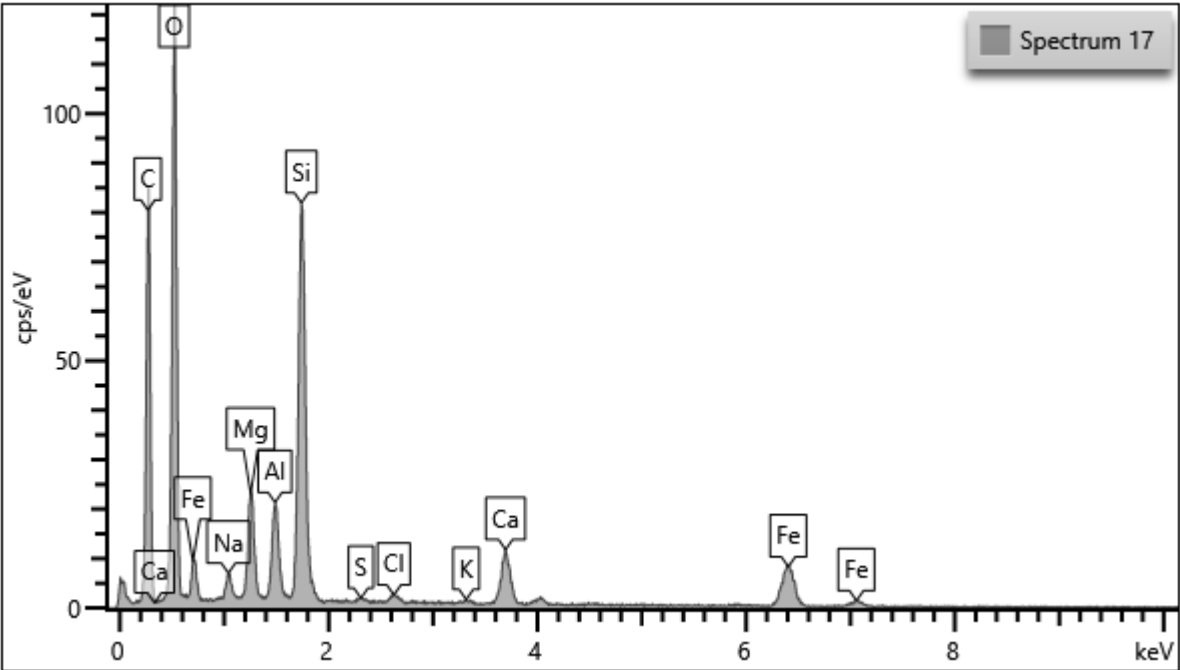
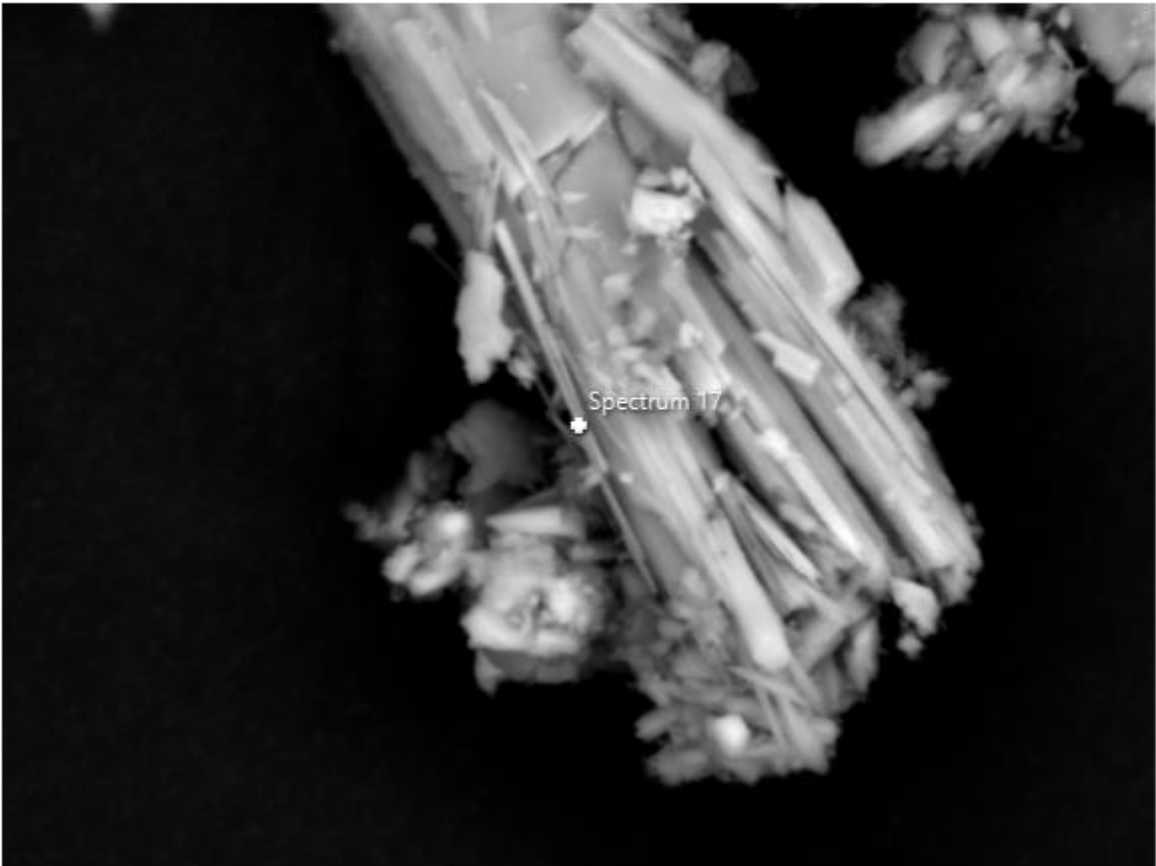
Electron Image 8 (Input1)



Electron Image 9 (Input1)



Electron Image 11 (Input1)



Client: Landloch
Job number: 25_0206
Lab ID: 25_0206_002
Client ID: RWC21
Analysis: Fibre characterisation by scanning electron microscopy (SEM) with elemental analysis by energy dispersive spectroscopy (EDS) following AS 4964-2004 (modified for SEM)
Revision number: 0
Comments: None

Date received: 30/01/2025
Date analysed: 24/02/2025
Date reported: 28/02/2025

Executive summary

The sample was determined to contain < **0.01 wt % asbestos** mineral fibre.

Sample preparation

The sample was supplied to Microanalysis Australia as a bulk sample. The sample was screened at 2 mm, and each size fraction was inspected visually (both with the naked eye and with a low-powered microscope) for fibrous material.

As no fibrous material was visually apparent for immediate pre-selection, a representative sub-sample of the <2 mm fraction was disaggregated and was placed on top of at least one double-sided carbon tab before being carbon coated. Non-conducting samples require coating prior to SEM analysis to prevent charging whilst being analysed by the electron beam.

Analysis

The sample was analysed using a Carl Zeiss EVO50 scanning electron microscope (SEM) fitted with an Oxford INCA energy dispersive spectrometer (EDS). The sample was scanned at low magnification to identify any possible fibre clusters before the magnification was increased to 2000x magnification for closer examination.

EDS is a semi-quantitative technique (at best) on well prepared, optically flat samples. Factors such as sample unevenness may adversely bias elemental concentration interpretation. EDS has a spatial resolution of ~5 µm meaning spectra from particles less than this size may contain elemental concentrations biased by their surroundings.

All images were acquired using backscatter electrons. Image brightness is proportional to average atomic number – the brighter the pixel, the higher the atomic number of the element.

Summary

Following AS 4964, asbestos is defined as:

- Many particles with aspect ratios (i.e. length to width ratios) ranging from 20:1 to 100:1 or higher for particles >5 micrometers in length. Bundles of fibres may have lower aspect ratios;
- Sets of fibre bundles generally less than 0.5, but always less than 1.0 micrometres in width, unless in thick bundles;
- In addition to the mandatory fibrillar crystal growth, one or more, and preferably three of the following aspects:
 - Parallel fibres occurring in bundles;
 - Fibre bundles displaying splayed ends;
 - Matted masses of individual fibres;
 - Fibres showing characteristic curvature.
- Respirable asbestos fibres are defined as:
 - Asbestos fibres less than 3 µm in width, and greater than 5 µm in length, and with a length to width ratio greater than 3 to 1.

Less than three observed fibres had an elemental composition and morphology indicative of asbestos mineral fibre, according to the definition in AS 4964.

Depending upon sample condition, composition and fibre type, the detection limit of AS 4964 has been found to lie generally in the range of 1 in 1,000 (0.1 wt %) to 1 in 10,000 (0.01 wt %) parts by weight, equivalent to 1 to 0.1 g/kg.

For this report, a limit of reporting (LoR) of 0.01 wt % has been assumed, as stated in AS 4964.

A selection of images/fields and associated elemental spectra are reported below. The fields are not representative of the area analysed. The images/fields were reported due to their higher fibre count.

Fibre #	Image/Field #	Diameter (µm)	Length (µm)	Aspect ratio	Major Elements	Minor Elements	Morphology	Assigned mineralogy
1	15/23	1.5	11.9	8 :1	O, Si	Mg, Al, Fe, Ca, Na	Non-parallel sides	Hornblende
2	16/24	0.5	5.4	11 :1	O, Si	Mg, Al, Fe, Ca, Na	Irregular ends	Hornblende
3	17/26	0.4	9.9	25 :1	O, Si	Mg, Al, Fe, Ca, Na	Irregular ends	Hornblende
4	18/27	0.6	10.3	17 :1	O, Si	Mg, Al, Fe, Ca, Na	Irregular ends	Hornblende
5	19/28	0.4	5.5	14 :1	O, Si	Mg, Al, Fe, Ca, Na	Irregular ends	Hornblende
6	21/33	1.1	9.6	9 :1	O, Si	Mg, Al, Fe, Ca, Na	Non-parallel sides	Hornblende
7	11/37	0.9	8.1	9 :1	O, Si	Mg, Al, Fe, Ca, Na	Irregular ends	Hornblende

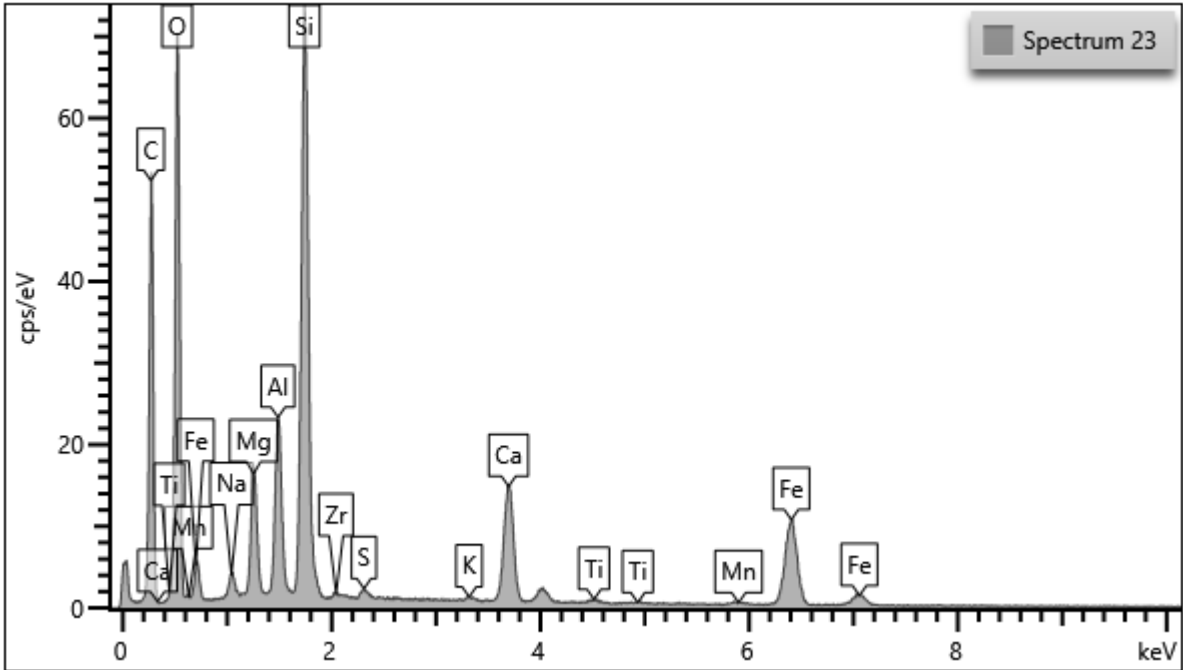
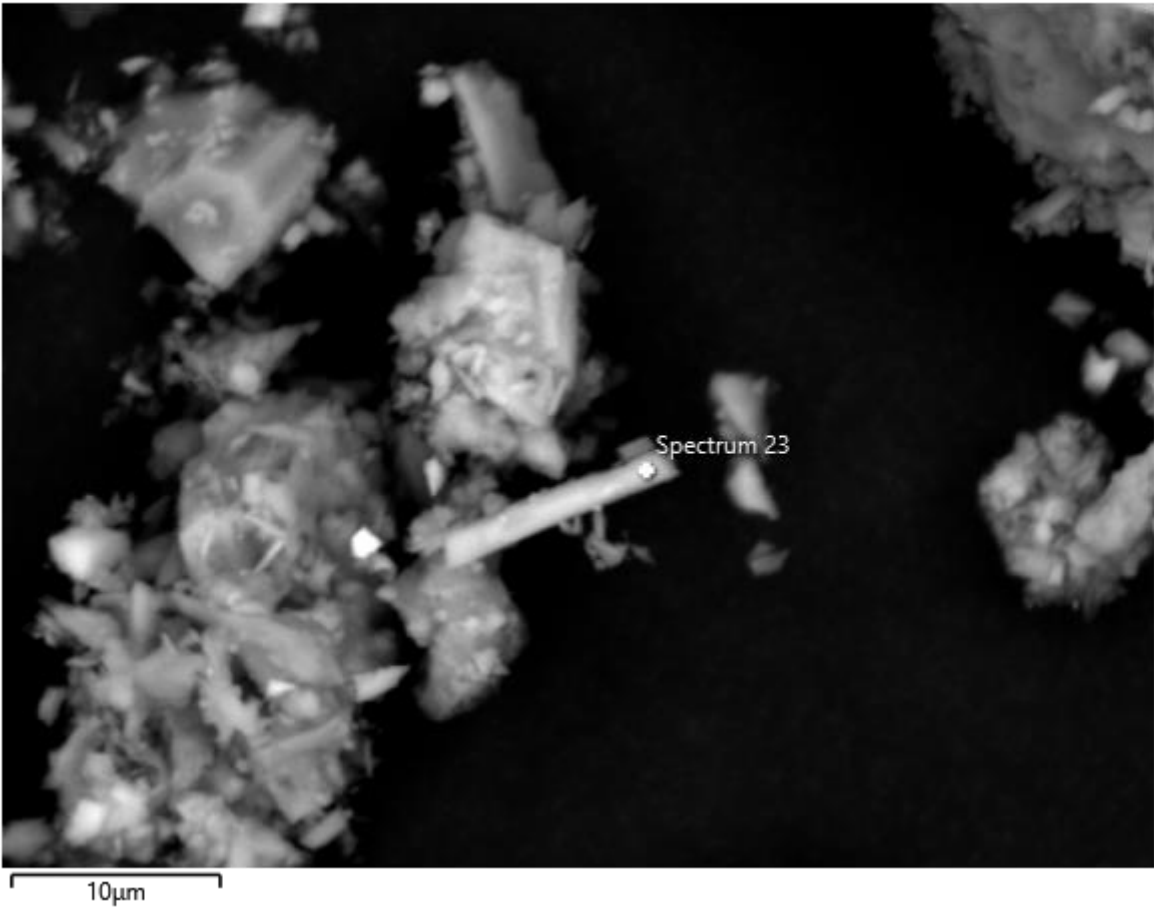
It should be noted that the higher resolution of the SEM may increase the number of fibres observed when compared with optical microscopy. Positive identification of asbestos fibres by SEM EDS uses elemental information that is not available by optical methods.

Analyst: Damon Blakey, *B.Sc. (Forensic Biology and Toxicology)*

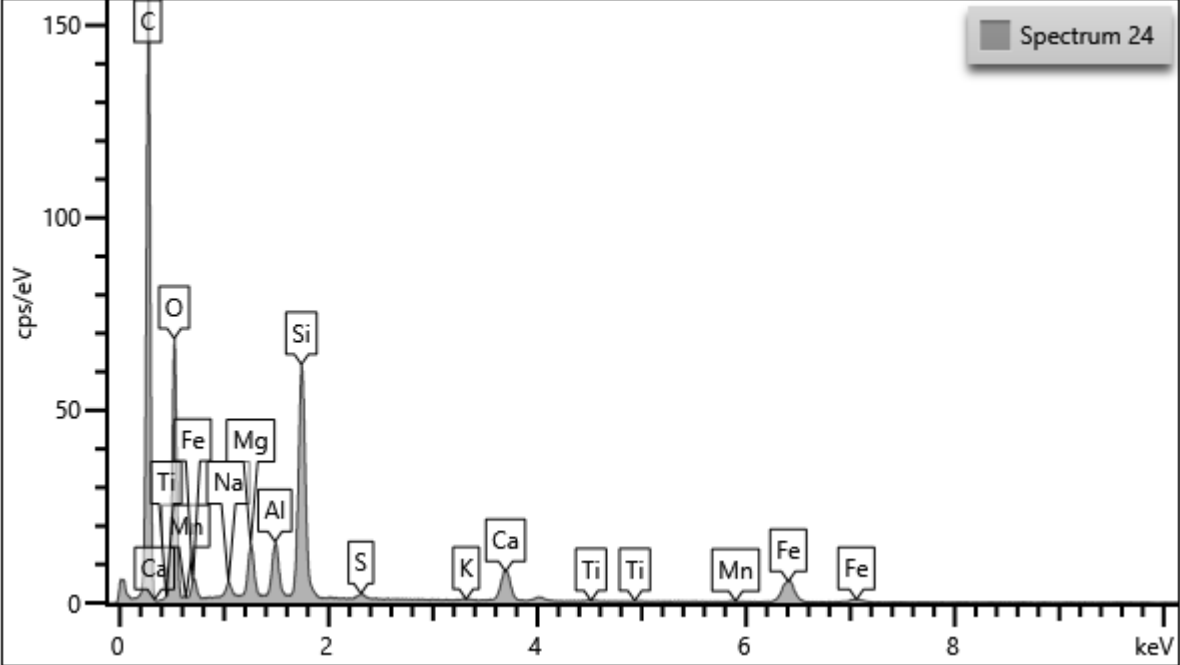
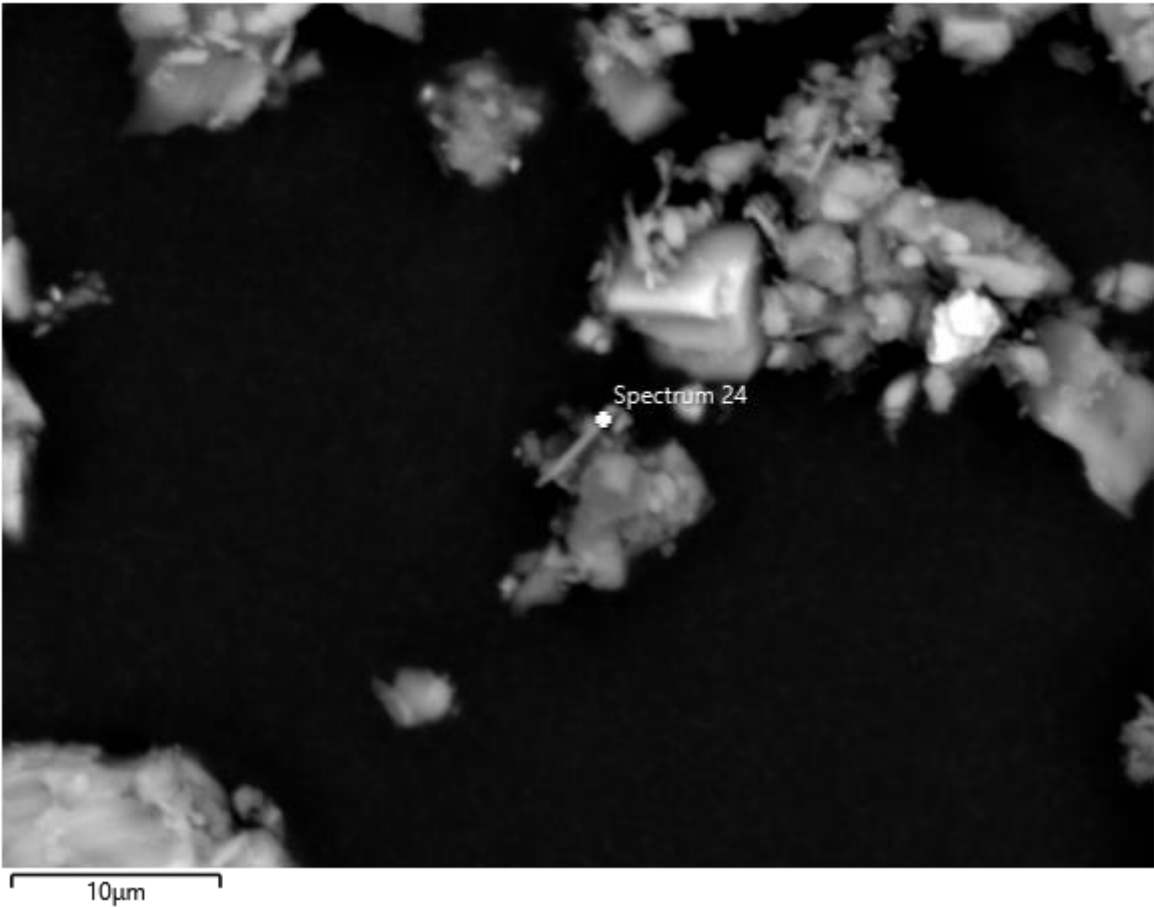
Reported by: Damon Blakey, *B.Sc. (Forensic Biology and Toxicology)*

Approved: Nimue Pendragon, *B.Sc.(Nanotechnology)*

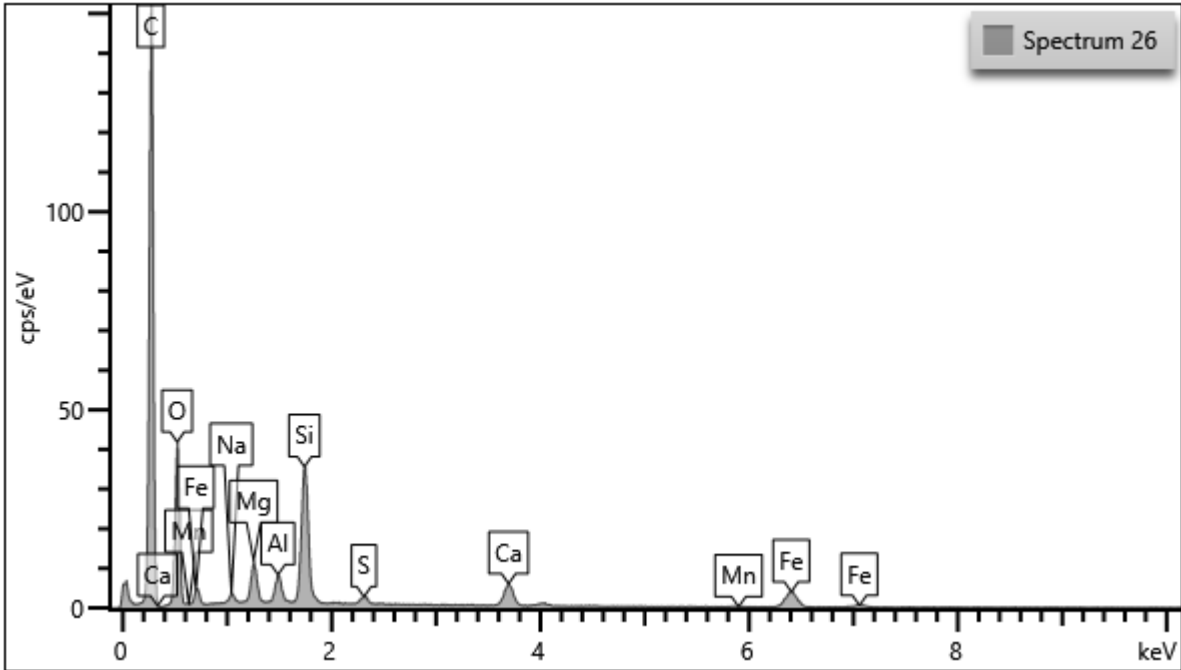
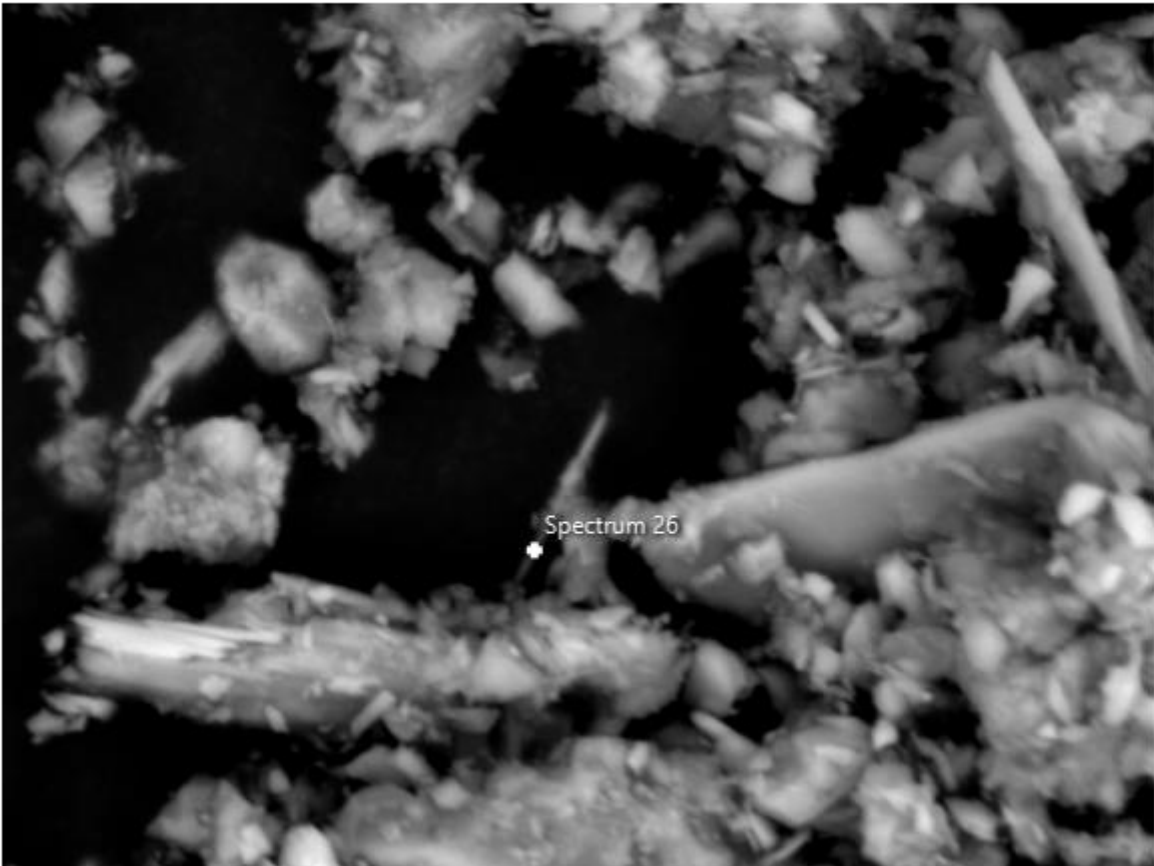
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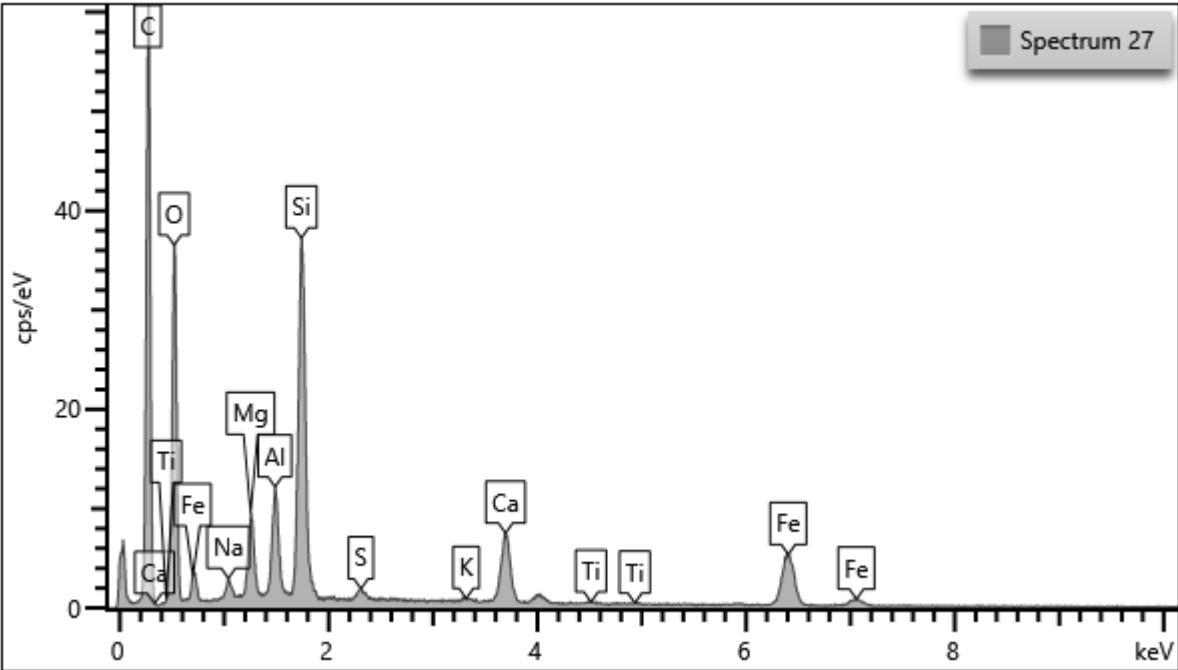
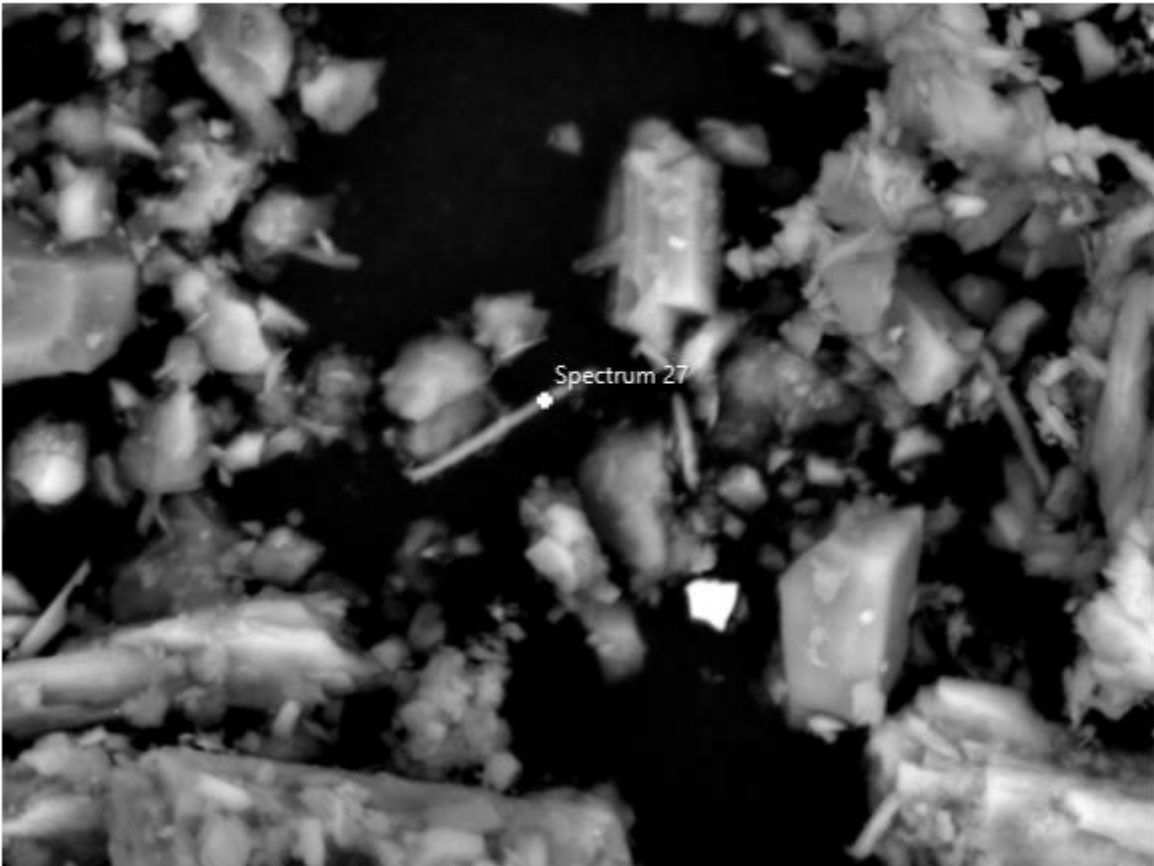
Electron Image 16 (Input1)



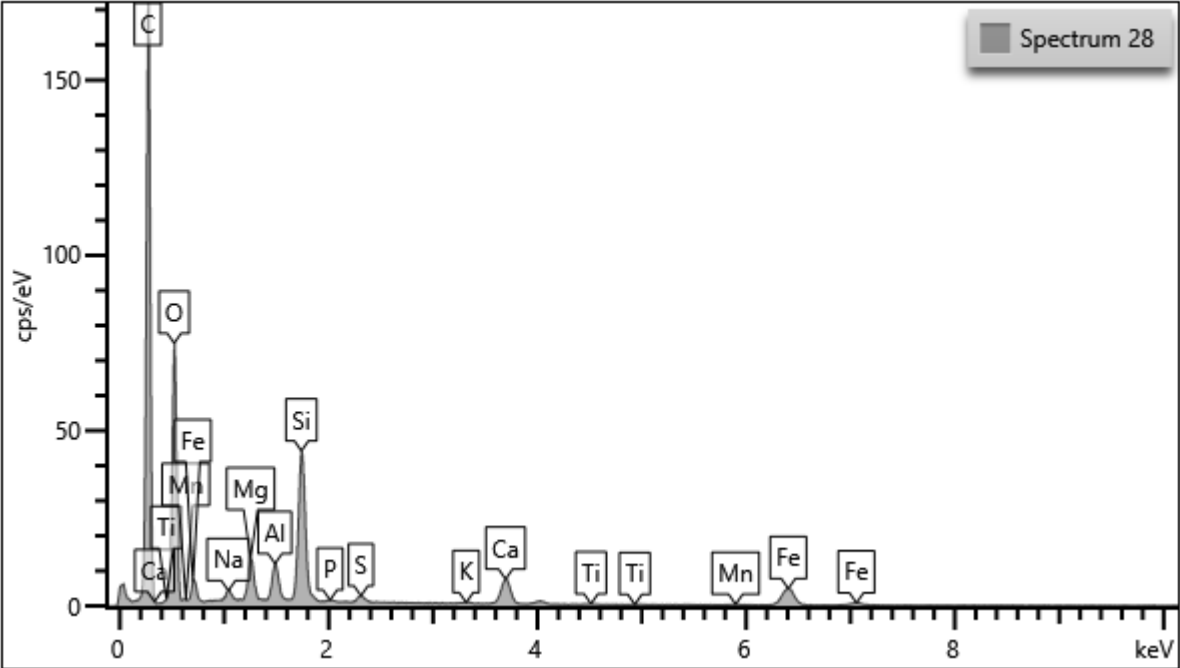
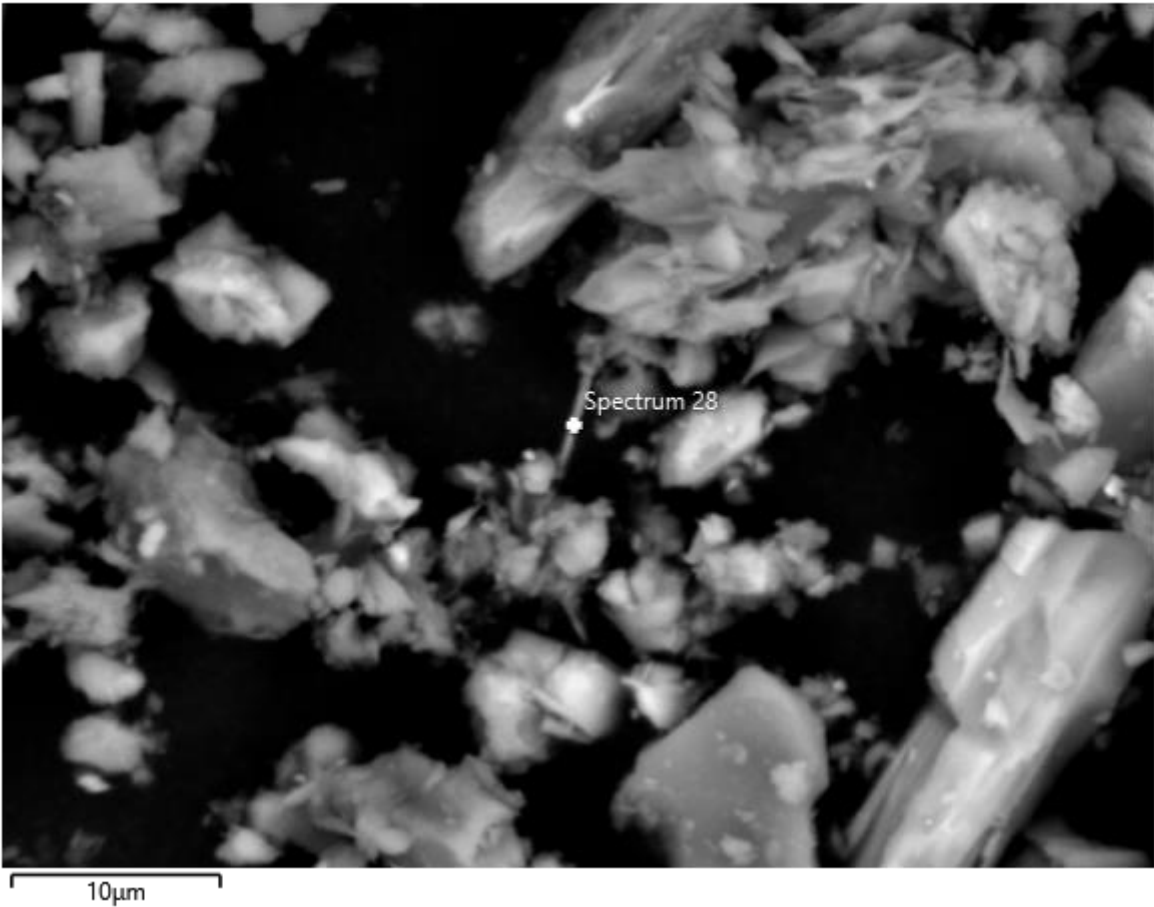
Electron Image 17 (Input1)



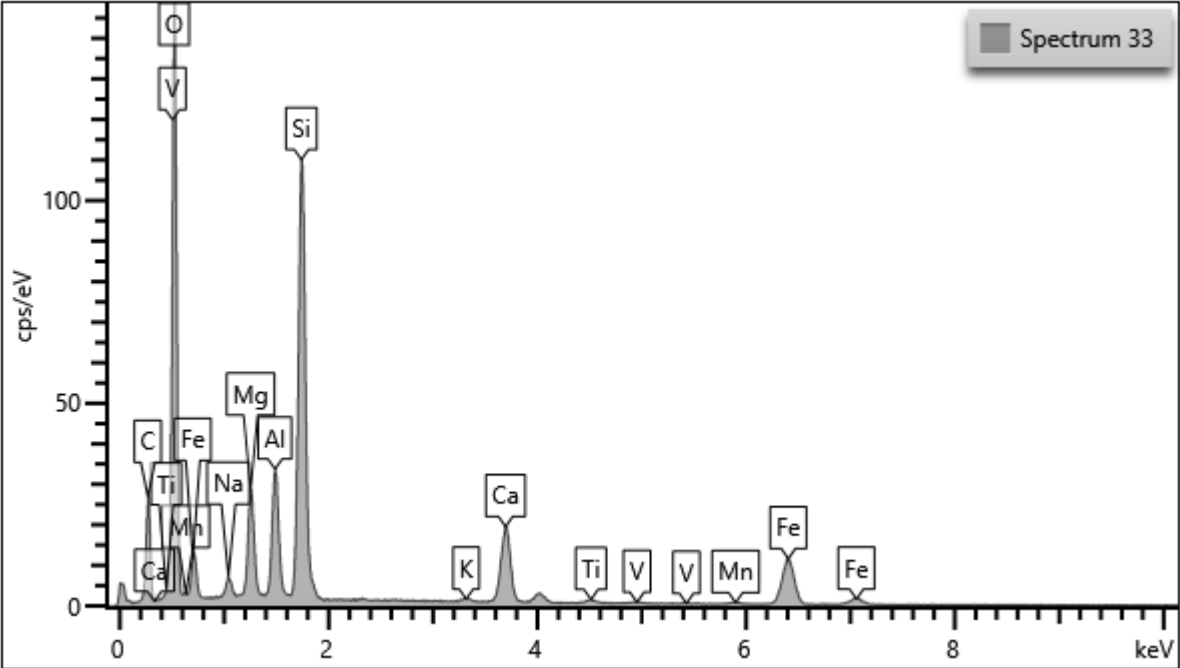
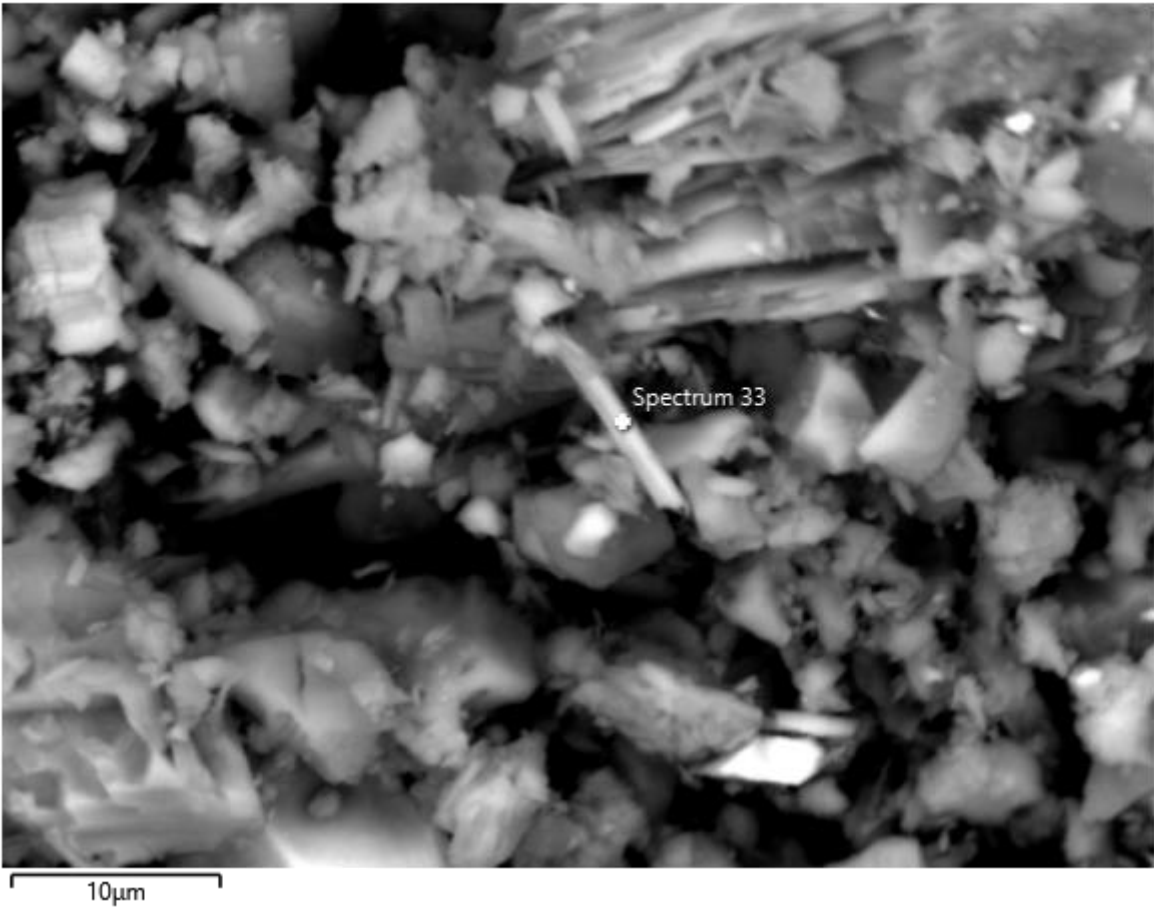
Electron Image 18 (Input1)



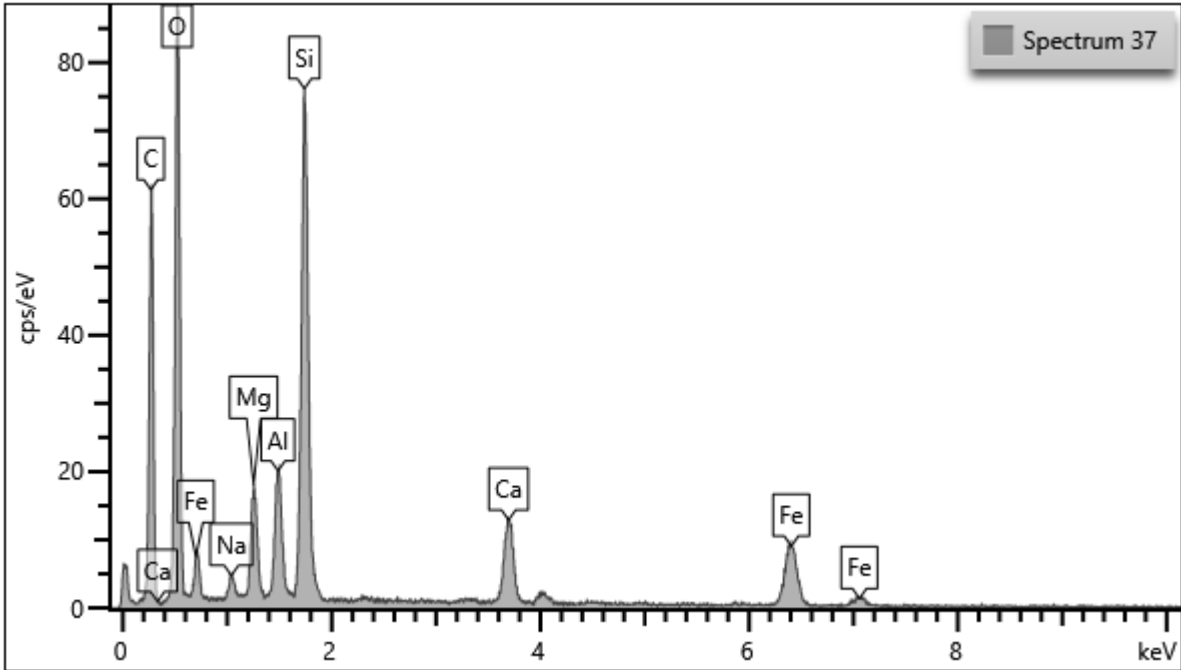
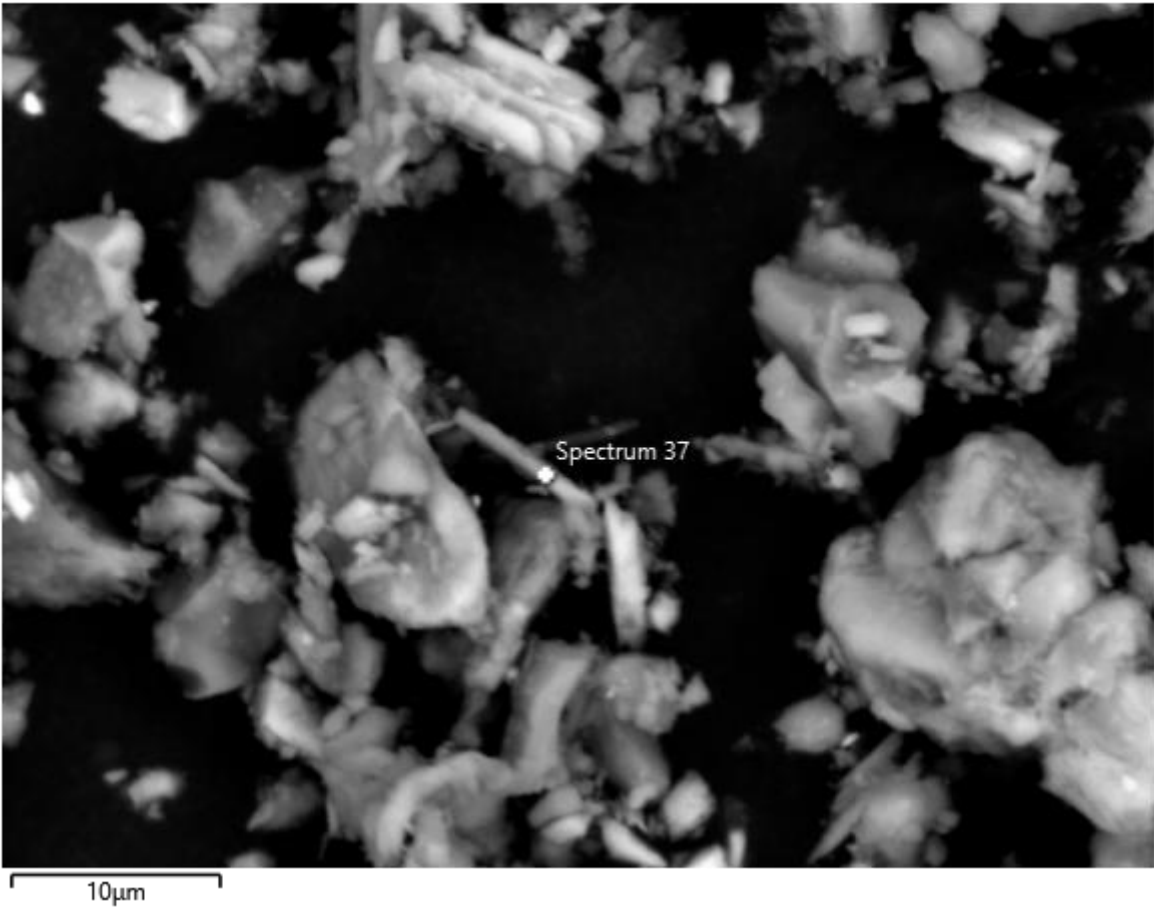
Electron Image 19 (Input1)



Electron Image 21 (Input1)



Electron Image 25 (Input1)



Client: Landloch
Job number: 25_0206
Lab ID: 25_0206_003
Client ID: RWC32
Analysis: Fibre characterisation by scanning electron microscopy (SEM) with elemental analysis by energy dispersive spectroscopy (EDS) following AS 4964-2004 (modified for SEM)
Revision number: 0
Comments: None

Date received: 30/01/2025
Date analysed: 24/02/2025
Date reported: 28/02/2025

Executive summary

The sample was determined to contain < **0.01 wt % asbestos** mineral fibre.

Sample preparation

The sample was supplied to Microanalysis Australia as a bulk sample. The sample was screened at 2 mm, and each size fraction was inspected visually (both with the naked eye and with a low-powered microscope) for fibrous material.

As no fibrous material was visually apparent for immediate pre-selection, a representative sub-sample of the <2 mm fraction was disaggregated and was placed on top of at least one double-sided carbon tab before being carbon coated. Non-conducting samples require coating prior to SEM analysis to prevent charging whilst being analysed by the electron beam.

Analysis

The sample was analysed using a Carl Zeiss EVO50 scanning electron microscope (SEM) fitted with an Oxford INCA energy dispersive spectrometer (EDS). The sample was scanned at low magnification to identify any possible fibre clusters before the magnification was increased to 2000x magnification for closer examination.

EDS is a semi-quantitative technique (at best) on well prepared, optically flat samples. Factors such as sample unevenness may adversely bias elemental concentration interpretation. EDS has a spatial resolution of ~5 µm meaning spectra from particles less than this size may contain elemental concentrations biased by their surroundings.

All images were acquired using backscatter electrons. Image brightness is proportional to average atomic number – the brighter the pixel, the higher the atomic number of the element.

Summary

Following AS 4964, asbestos is defined as:

- Many particles with aspect ratios (i.e. length to width ratios) ranging from 20:1 to 100:1 or higher for particles >5 micrometers in length. Bundles of fibres may have lower aspect ratios;
- Sets of fibre bundles generally less than 0.5, but always less than 1.0 micrometres in width, unless in thick bundles;
- In addition to the mandatory fibrillar crystal growth, one or more, and preferably three of the following aspects:
 - Parallel fibres occurring in bundles;
 - Fibre bundles displaying splayed ends;
 - Matted masses of individual fibres;
 - Fibres showing characteristic curvature.
- Respirable asbestos fibres are defined as:
 - Asbestos fibres less than 3 µm in width, and greater than 5 µm in length, and with a length to width ratio greater than 3 to 1.

Less than 5 observed fibres had an elemental composition and morphology indicative of asbestos mineral fibre, according to the definition in AS 4964.

Depending upon sample condition, composition and fibre type, the detection limit of AS 4964 has been found to lie generally in the range of 1 in 1,000 (0.1 wt %) to 1 in 10,000 (0.01 wt %) parts by weight, equivalent to 1 to 0.1 g/kg.

For this report, a limit of reporting (LoR) of 0.01 wt % has been assumed, as stated in AS 4964.

A selection of images/fields and associated elemental spectra are reported below. The fields are not representative of the area analysed. The images/fields were reported due to their higher fibre count.

Fibre #	Image/Field #	Diameter (µm)	Length (µm)	Aspect ratio	Major Elements	Minor Elements	Morphology	Assigned mineralogy
1	5/8	1.1	14.6	13 :1	O, Si	Mg, Ca, Fe, Al, Na	Non-parallel sides	Hornblende
2	7/11	1.0	14.2	14 :1	O, Si	Al, Mg, Fe	Irregular ends	Chlorite group
3	8/12	0.6	6.6	11 :1	O, Si	Mg, Fe, Ca	Parallel sides	Actinolite
4	10/14	1.7	9.8	6 :1	O, Si	Al, Mg, Fe	Non-parallel sides	Chlorite group
5	11/15	0.7	9.3	13 :1	O, Si	Mg, Ca, Fe, Al, Na	Non-parallel sides	Hornblende
6	12/16	0.6	10.6	18 :1	O, Si	Mg, Ca, Fe, Al, Na	Non-parallel sides	Hornblende
7	14/19	1.5	11.6	8 :1	O, Si	Mg, Ca, Fe, Al, Na	Irregular ends	Hornblende

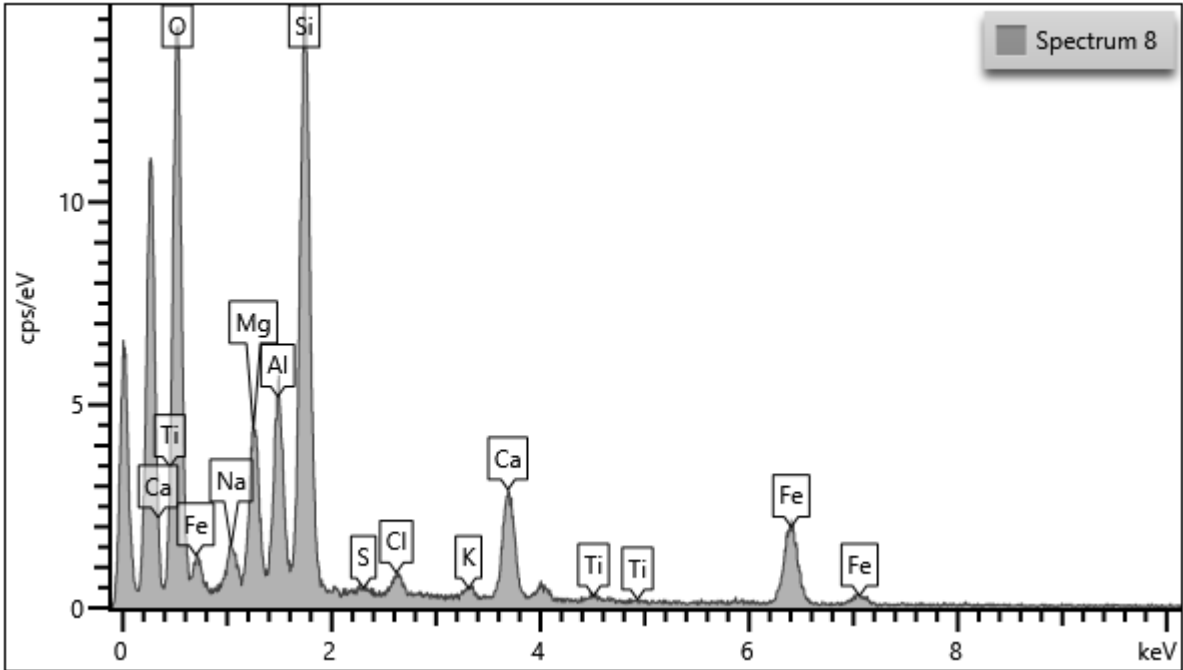
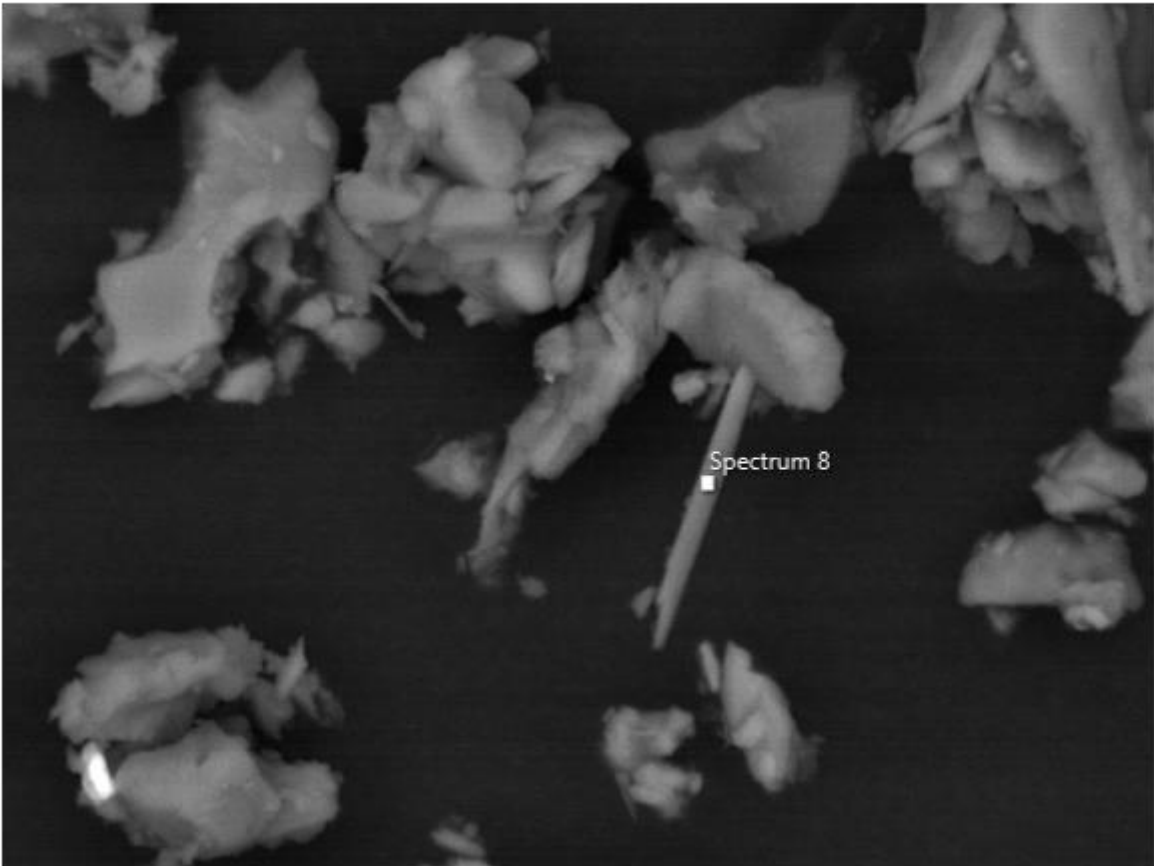
It should be noted that the higher resolution of the SEM may increase the number of fibres observed when compared with optical microscopy. Positive identification of asbestos fibres by SEM EDS uses elemental information that is not available by optical methods.

Analyst: Damon Blakey, *B.Sc. (Forensic Biology and Toxicology)*

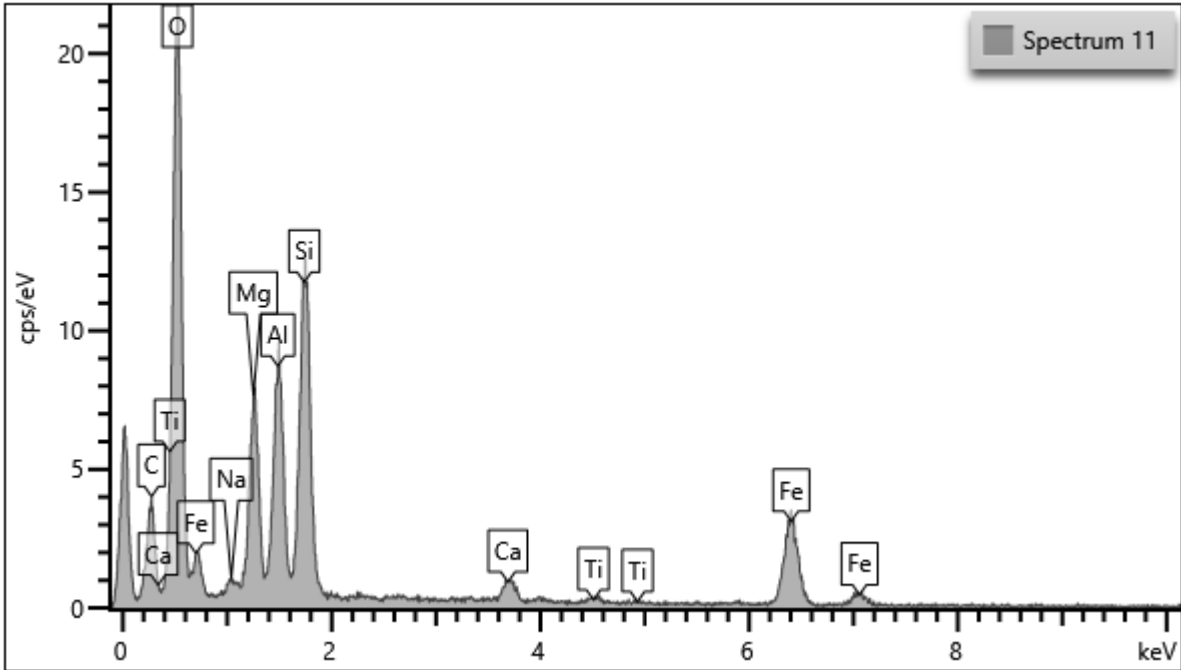
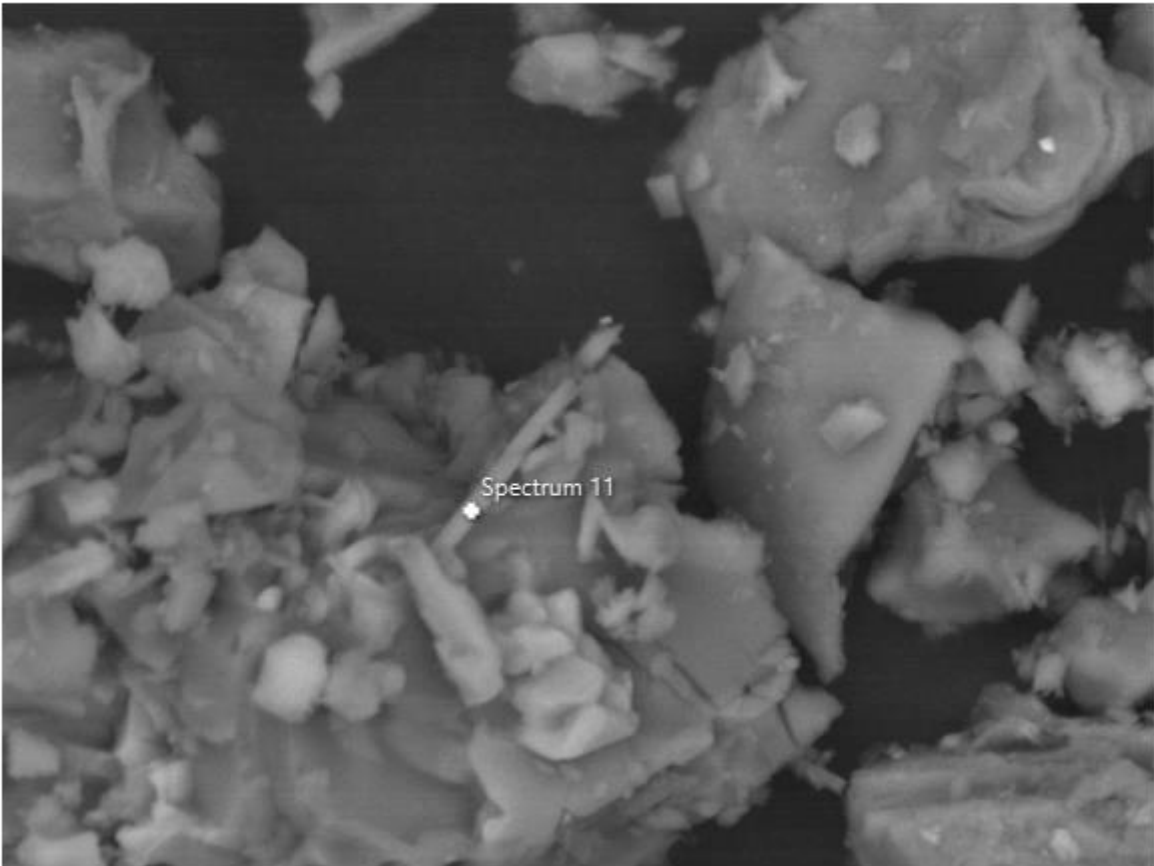
Reported by: Damon Blakey, *B.Sc. (Forensic Biology and Toxicology)*

Approved: Nimue Pendragon, *B.Sc.(Nanotechnology)*

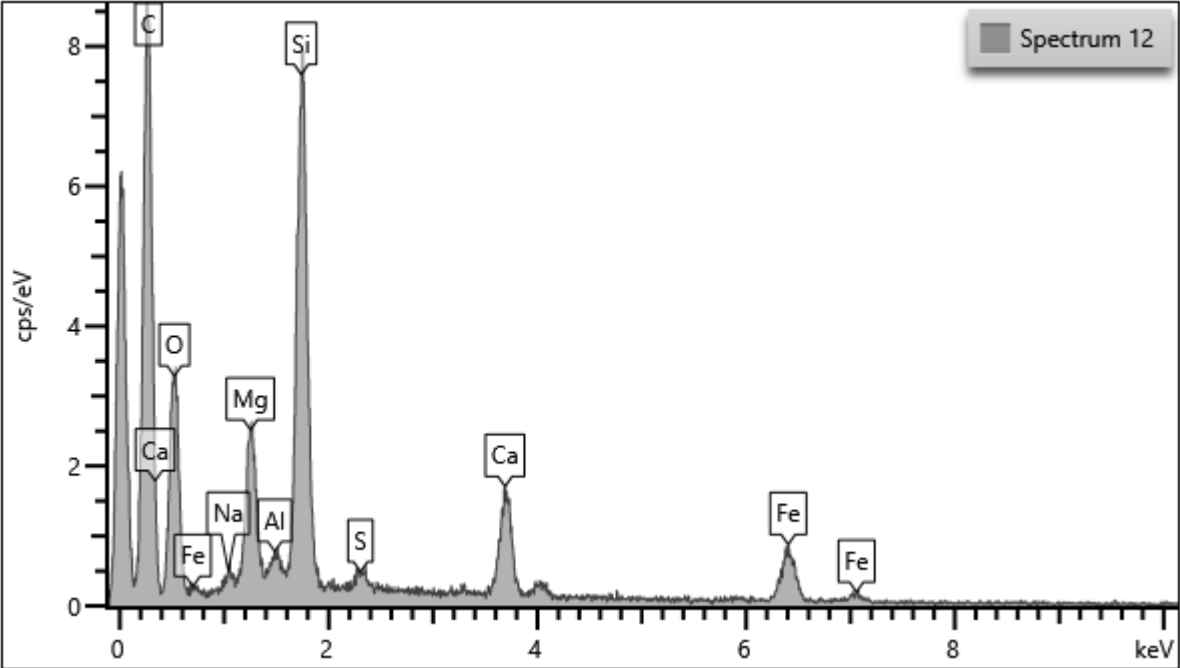
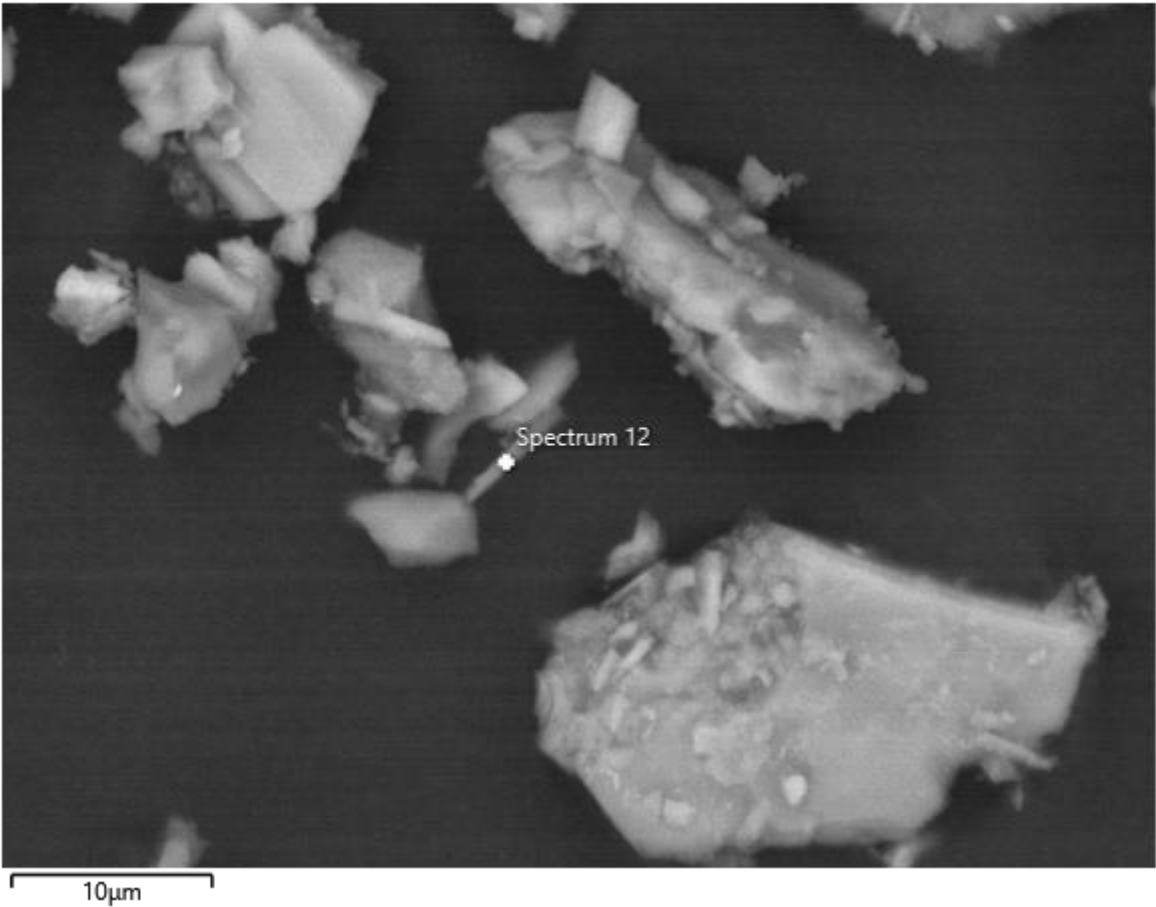
Electron Image 5 (Input1)



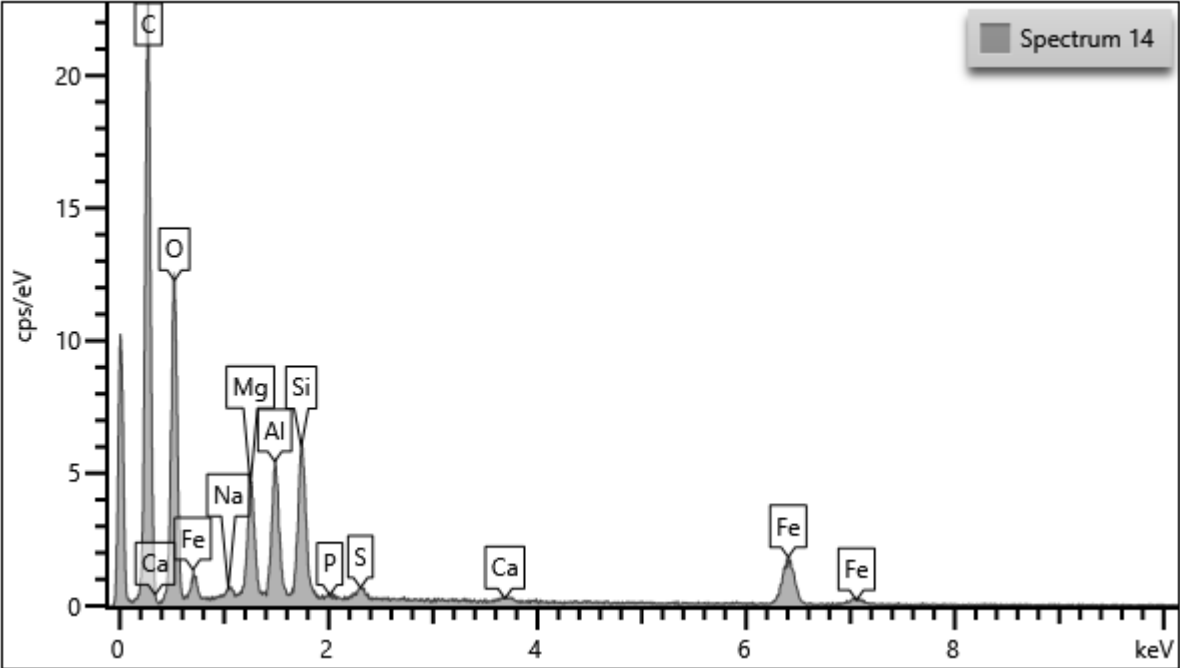
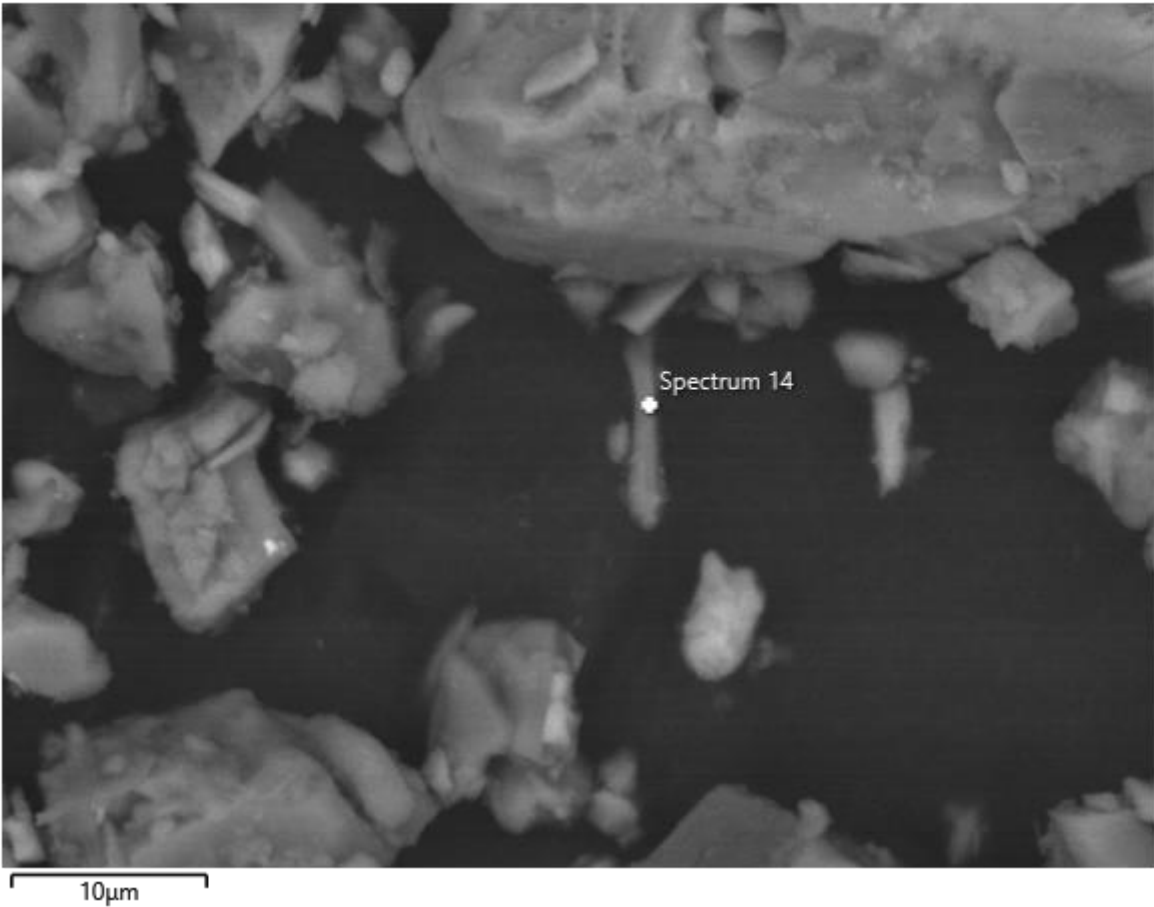
Electron Image 7 (Input1)



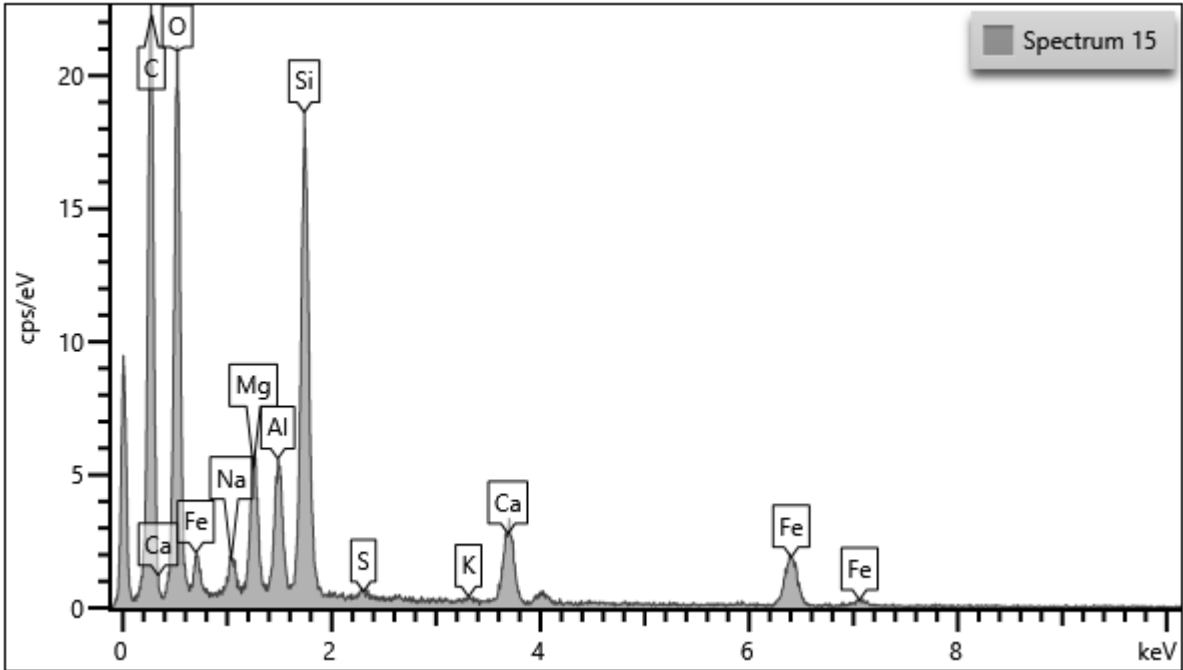
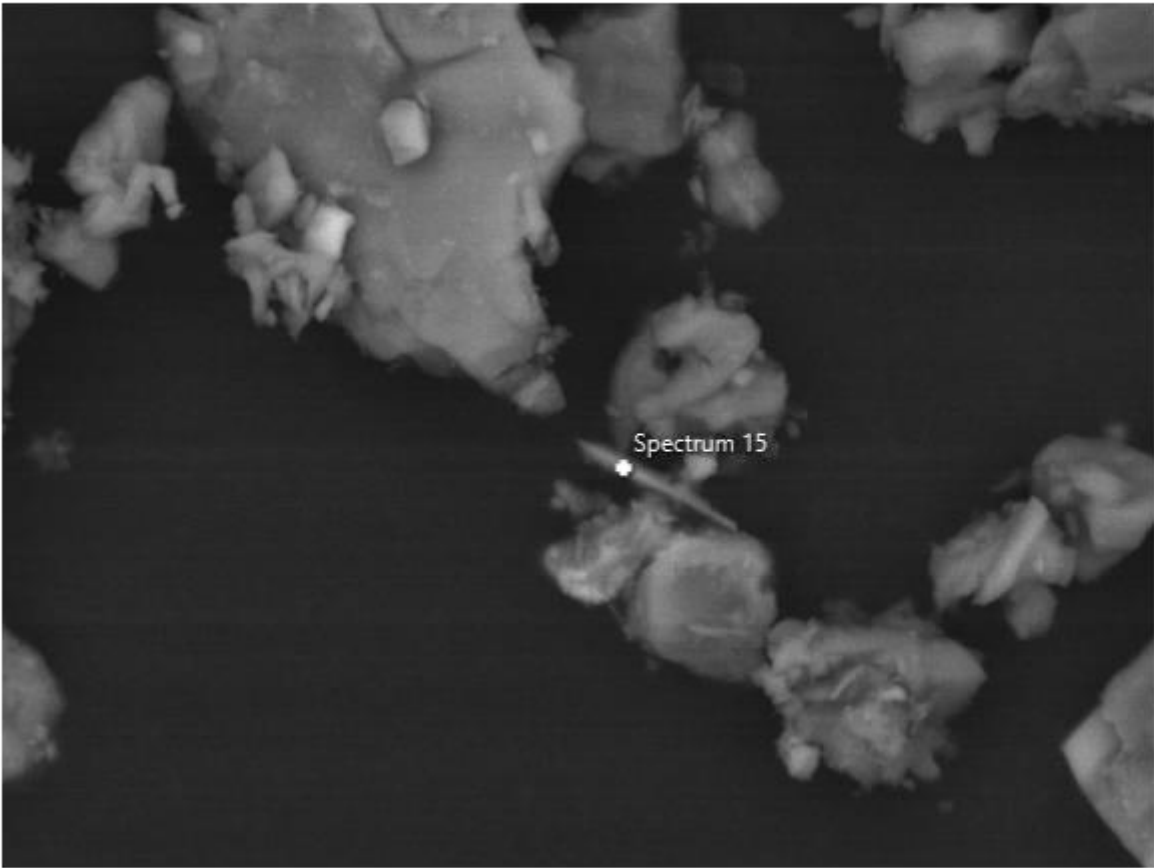
Electron Image 8 (Input1)



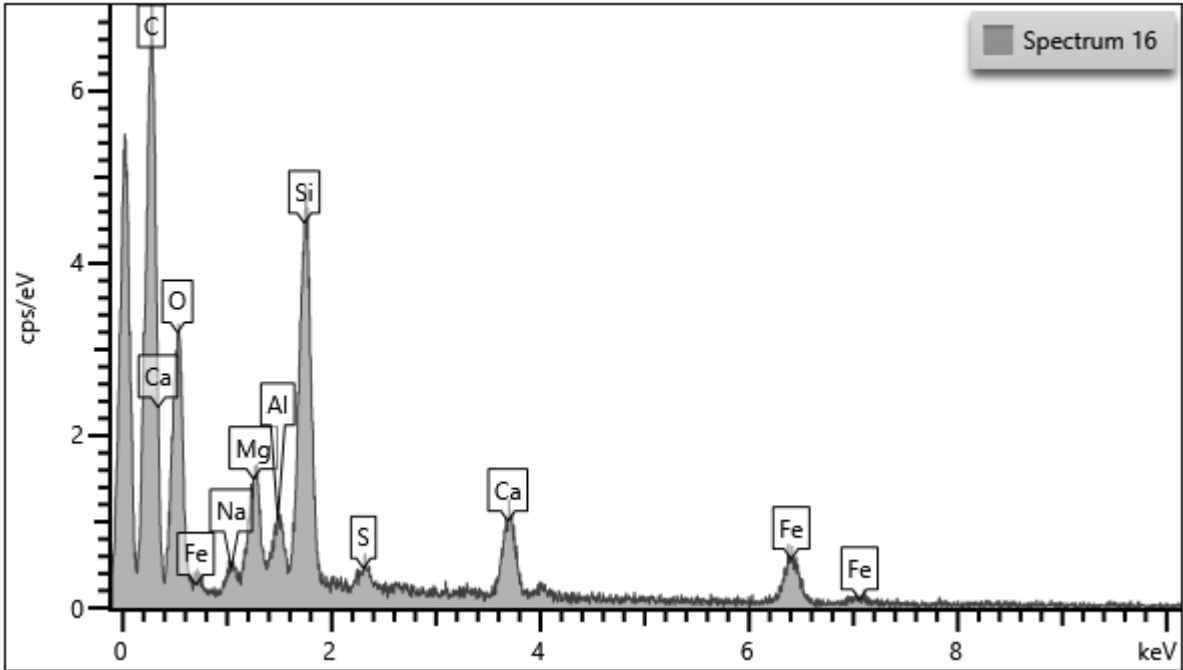
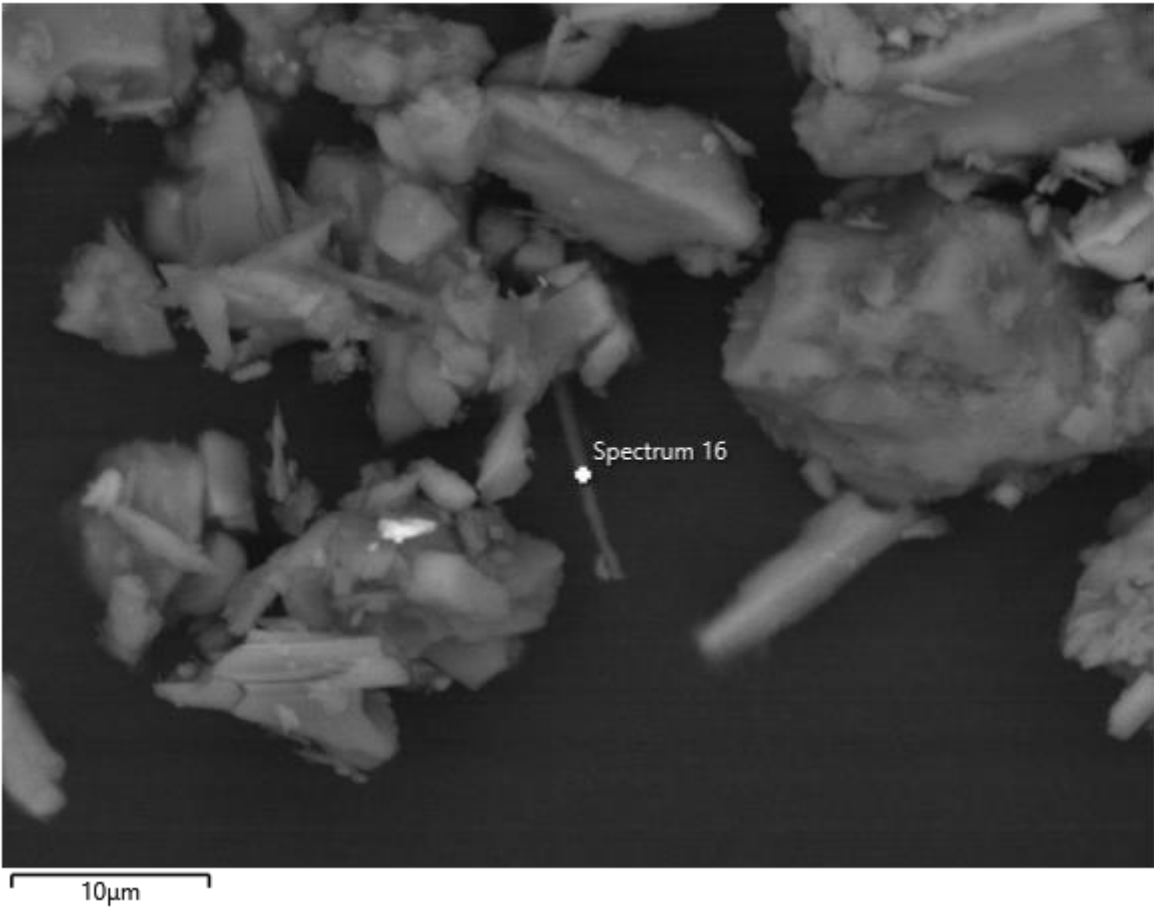
Electron Image 10 (Input1)



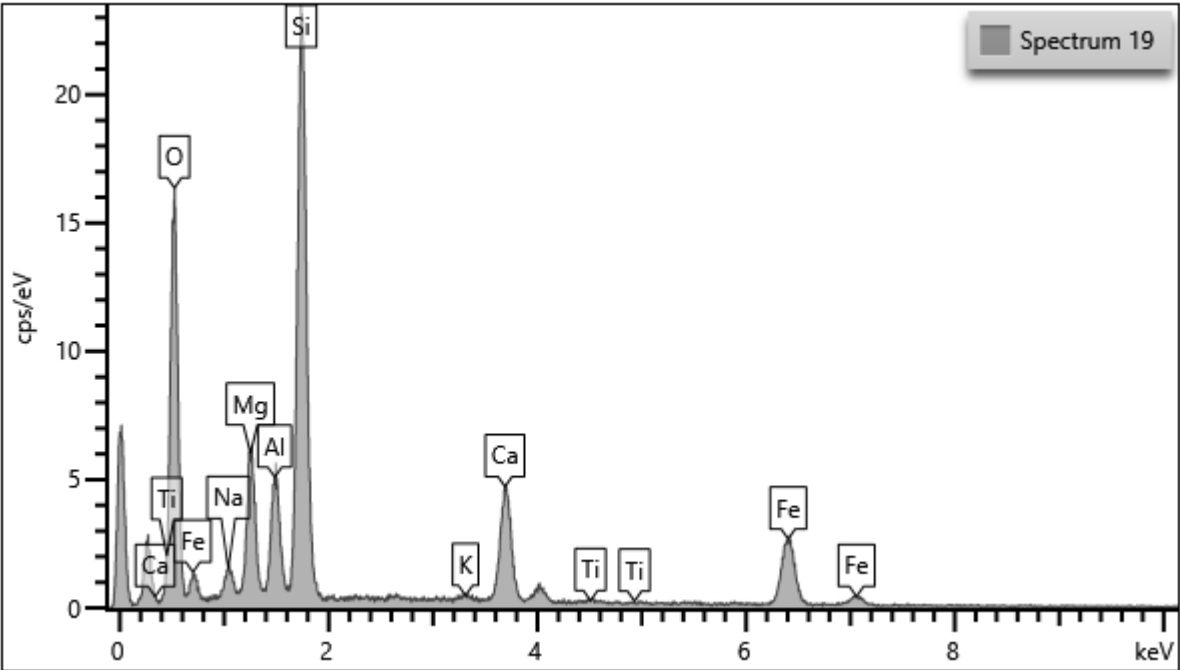
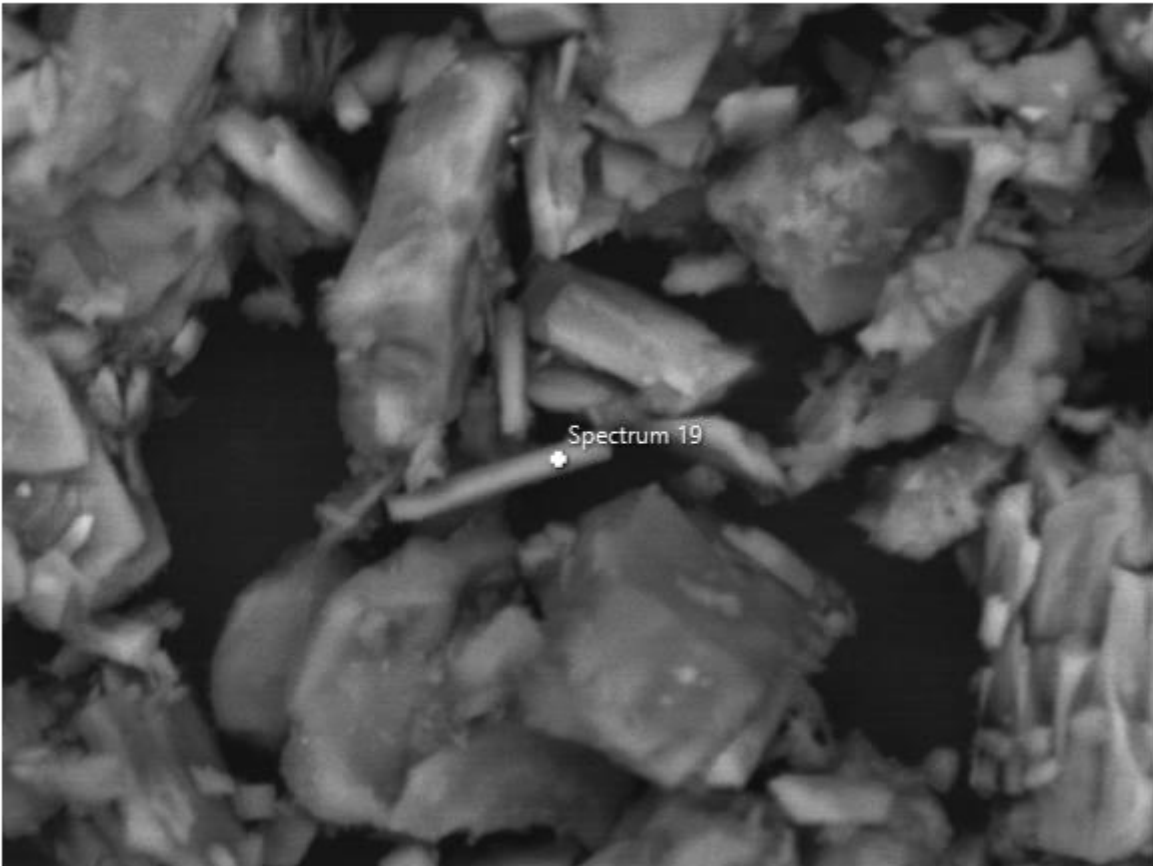
Electron Image 11 (Input1)



Electron Image 12 (Input1)



Electron Image 14 (Input1)



Client: Landloch
Job number: 25_0206
Lab ID: 25_0206_004
Client ID: RWC56
Analysis: Fibre characterisation by scanning electron microscopy (SEM) with elemental analysis by energy dispersive spectroscopy (EDS) following AS 4964-2004 (modified for SEM)
Revision number: 0
Comments: None

Date received: 30/01/2025
Date analysed: 28/02/2025
Date reported: 28/02/2025

Executive summary

The sample was determined to contain < **0.01 wt % asbestos** mineral fibre.

Sample preparation

The sample was supplied to Microanalysis Australia as a bulk sample. The sample was screened at 2 mm, and each size fraction was inspected visually (both with the naked eye and with a low-powered microscope) for fibrous material.

As no fibrous material was visually apparent for immediate pre-selection, a representative sub-sample of the <2 mm fraction was disaggregated and was placed on top of at least one double-sided carbon tab before being carbon coated. Non-conducting samples require coating prior to SEM analysis to prevent charging whilst being analysed by the electron beam.

Analysis

The sample was analysed using a Carl Zeiss EVO50 scanning electron microscope (SEM) fitted with an Oxford INCA energy dispersive spectrometer (EDS). The sample was scanned at low magnification to identify any possible fibre clusters before the magnification was increased to 2000x magnification for closer examination.

EDS is a semi-quantitative technique (at best) on well prepared, optically flat samples. Factors such as sample unevenness may adversely bias elemental concentration interpretation. EDS has a spatial resolution of ~5 µm meaning spectra from particles less than this size may contain elemental concentrations biased by their surroundings.

All images were acquired using backscatter electrons. Image brightness is proportional to average atomic number – the brighter the pixel, the higher the atomic number of the element.

Summary

Following AS 4964, asbestos is defined as:

- Many particles with aspect ratios (i.e. length to width ratios) ranging from 20:1 to 100:1 or higher for particles >5 micrometers in length. Bundles of fibres may have lower aspect ratios;
- Sets of fibre bundles generally less than 0.5, but always less than 1.0 micrometres in width, unless in thick bundles;
- In addition to the mandatory fibrillar crystal growth, one or more, and preferably three of the following aspects:
 - Parallel fibres occurring in bundles;
 - Fibre bundles displaying splayed ends;
 - Matted masses of individual fibres;
 - Fibres showing characteristic curvature.
- Respirable asbestos fibres are defined as:
 - Asbestos fibres less than 3 µm in width, and greater than 5 µm in length, and with a length to width ratio greater than 3 to 1.

Less than 5 observed fibres had an elemental composition and morphology indicative of asbestos mineral fibre, according to the definition in AS 4964.

Depending upon sample condition, composition and fibre type, the detection limit of AS 4964 has been found to lie generally in the range of 1 in 1,000 (0.1 wt %) to 1 in 10,000 (0.01 wt %) parts by weight, equivalent to 1 to 0.1 g/kg.

For this report, a limit of reporting (LoR) of 0.01 wt % has been assumed, as stated in AS 4964.

A selection of images/fields and associated elemental spectra are reported below. The fields are not representative of the area analysed. The images/fields were reported due to their higher fibre count.

Fibre #	Image/Field #	Diameter (µm)	Length (µm)	Aspect ratio	Major Elements	Minor Elements	Morphology	Assigned mineralogy
1	2/3	1.3	14.1	11 :1	O, Si	Mg, Ca, Fe, Al, Na	Non-parallel sides	Hornblende
2	3/4	1.3	9.8	8 :1	O, Si	Mg, Ca, Fe	Irregular ends	Actinolite
3	4/6	0.8	13.2	17 :1	O, Si	Mg, Ca, Fe, Al, Na	Irregular ends	Hornblende
4	6/9	1.2	9.1	8 :1	O, Si	Al, Mg, Fe	Non-parallel sides	Chlorite group
5	8/11	1.0	9.2	9 :1	O, Si	Mg, Ca, Fe, Al, Na	Irregular ends	Hornblende
6	9/12	1.0	15.5	16 :1	O, Si	Mg, Ca, Fe, Al, Na	Irregular ends	Hornblende
7	11/14	1.3	20.8	16 :1	O, Si	Mg, Ca, Fe, Al, Na	Non-parallel sides	Hornblende

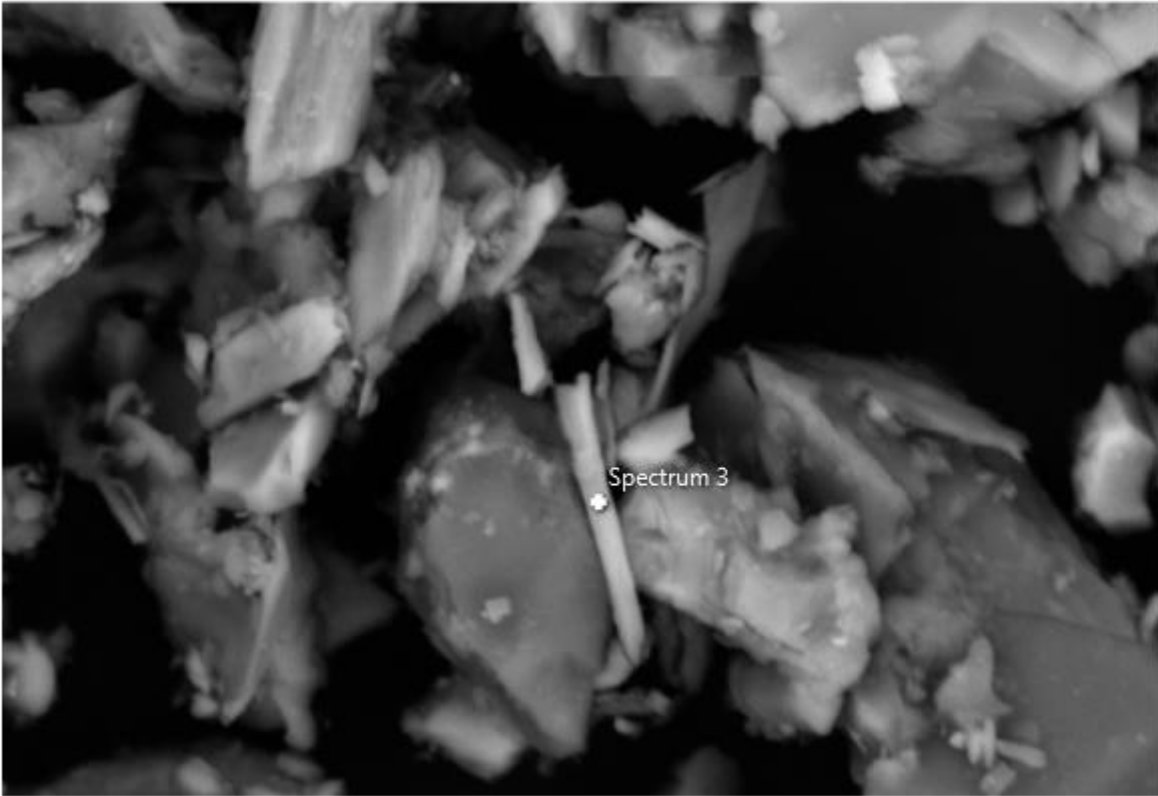
It should be noted that the higher resolution of the SEM may increase the number of fibres observed when compared with optical microscopy. Positive identification of asbestos fibres by SEM EDS uses elemental information that is not available by optical methods.

Analyst: Damon Blakey, *B.Sc. (Forensic Biology and Toxicology)*

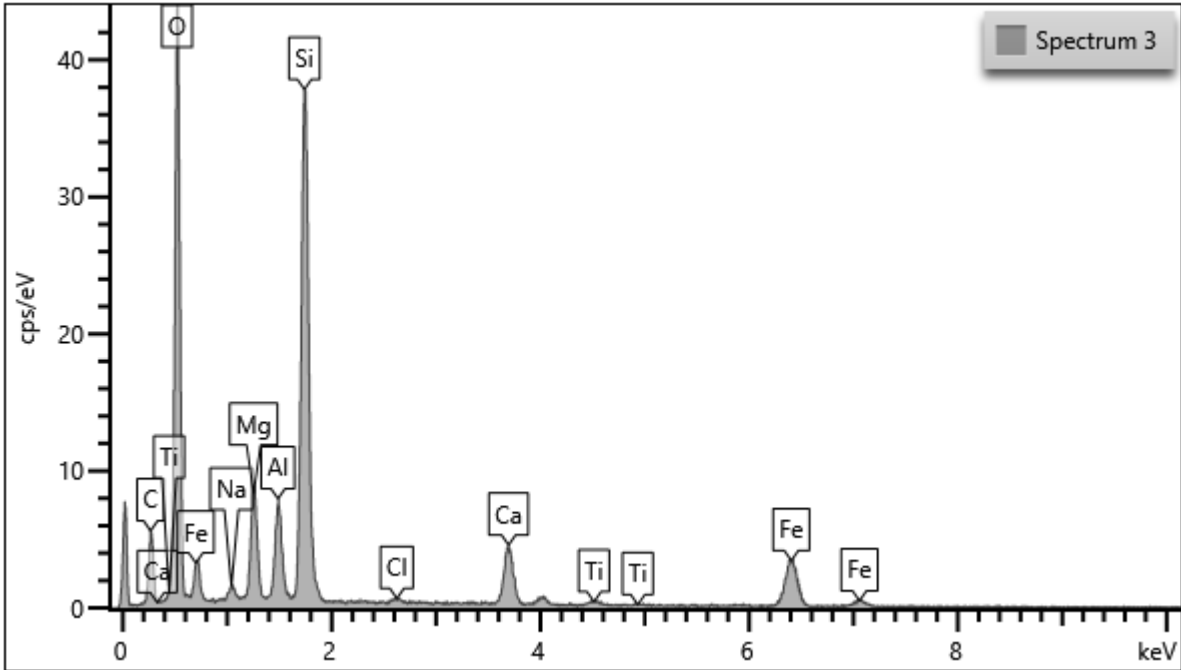
Reported by: Damon Blakey, *B.Sc. (Forensic Biology and Toxicology)*

Approved: Nimue Pendragon, *B.Sc.(Nanotechnology)*

Electron Image 2 (SE)



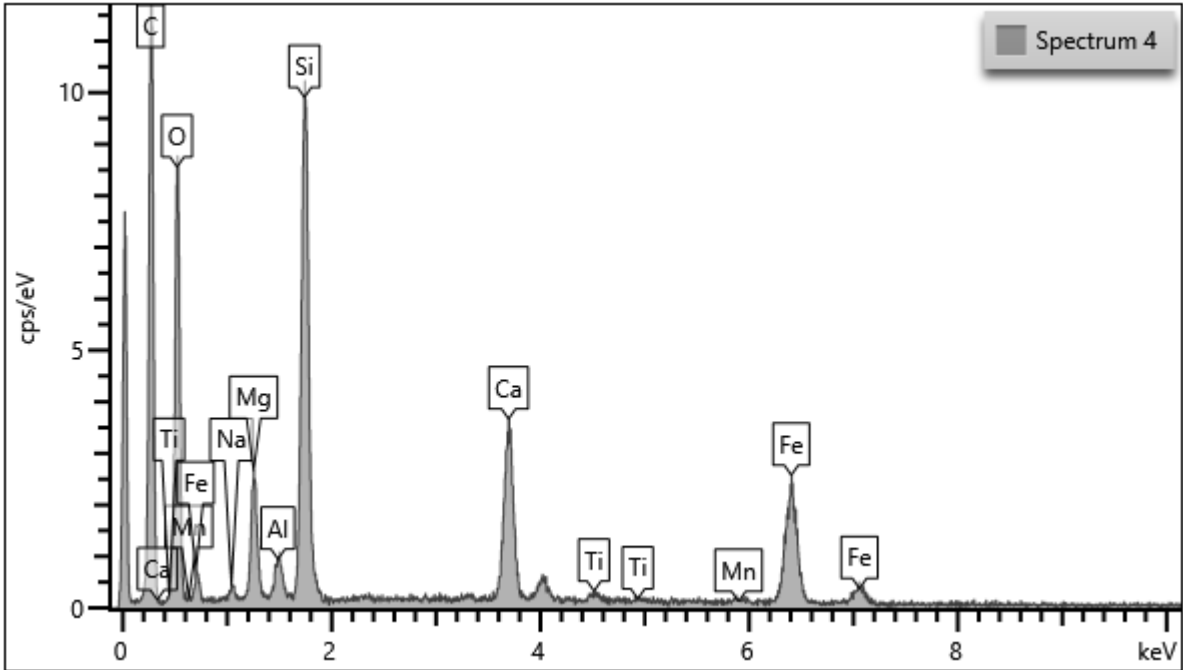
10µm



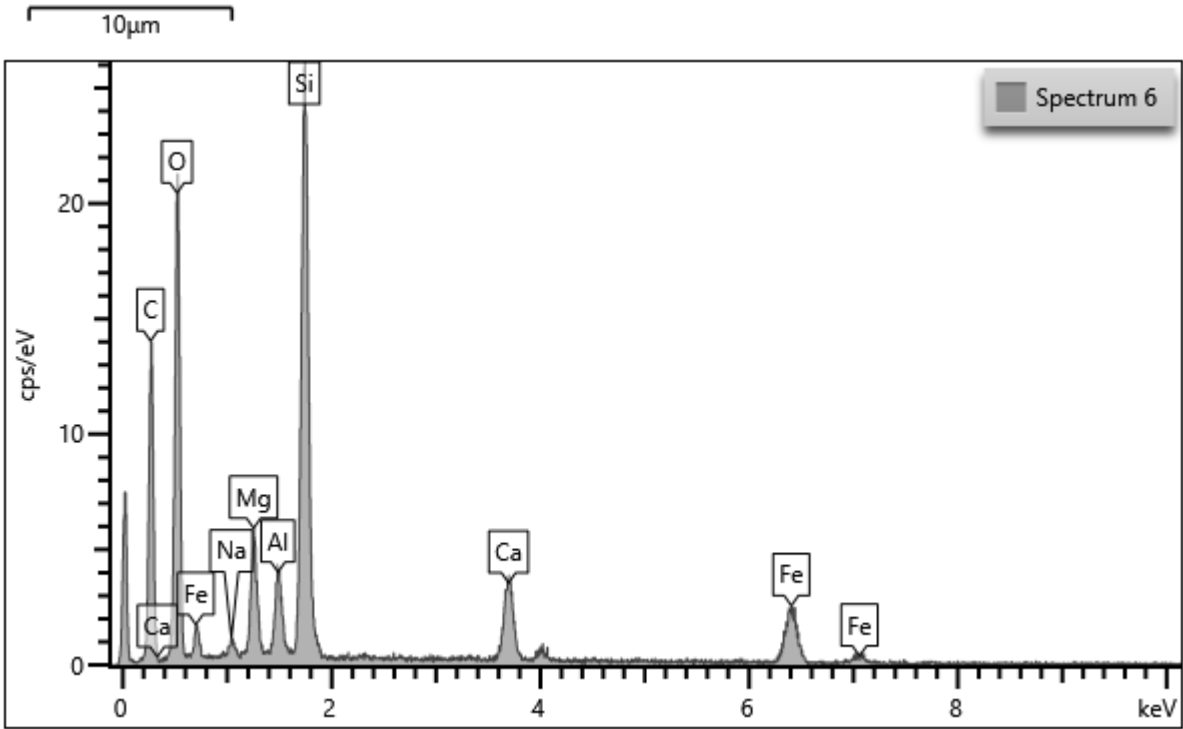
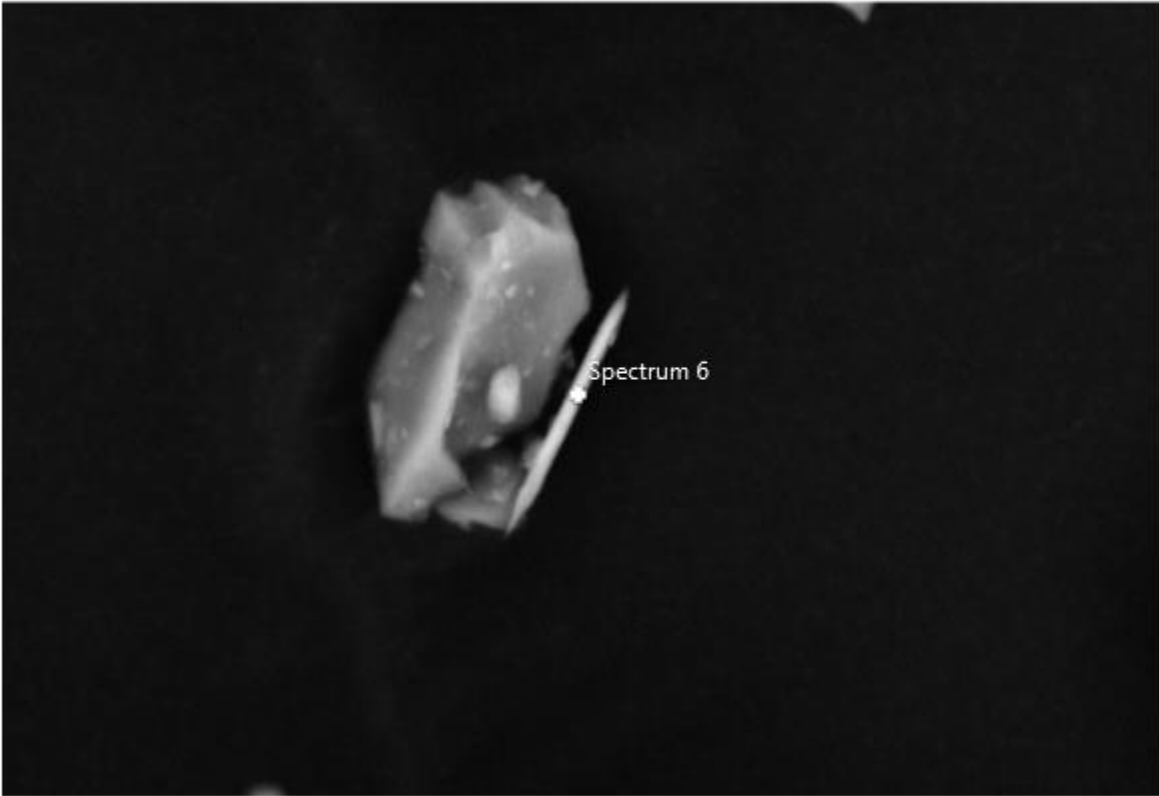
Electron Image 3 (SE)



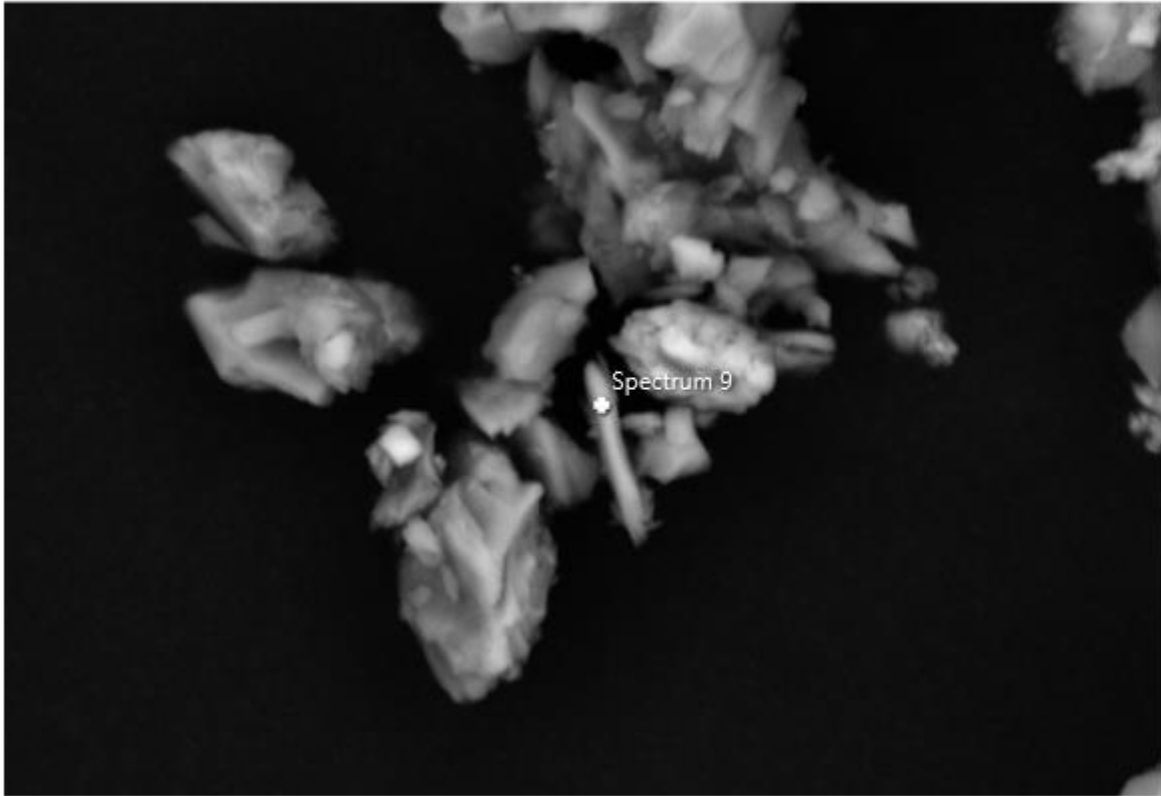
10µm



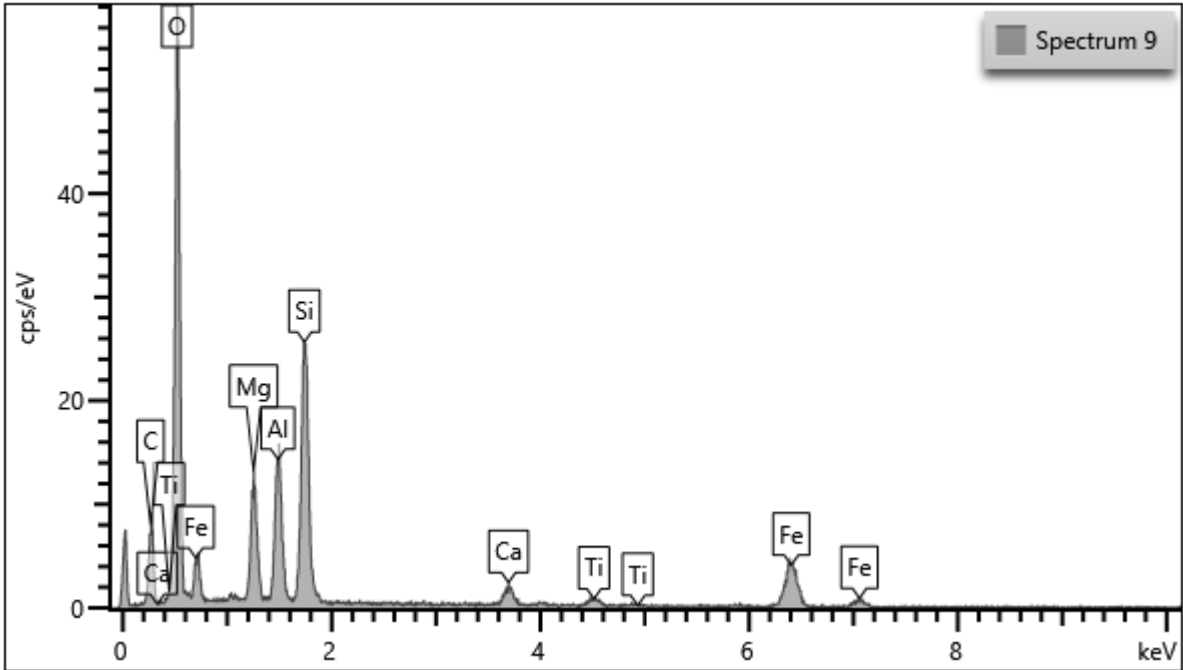
Electron Image 4 (SE)



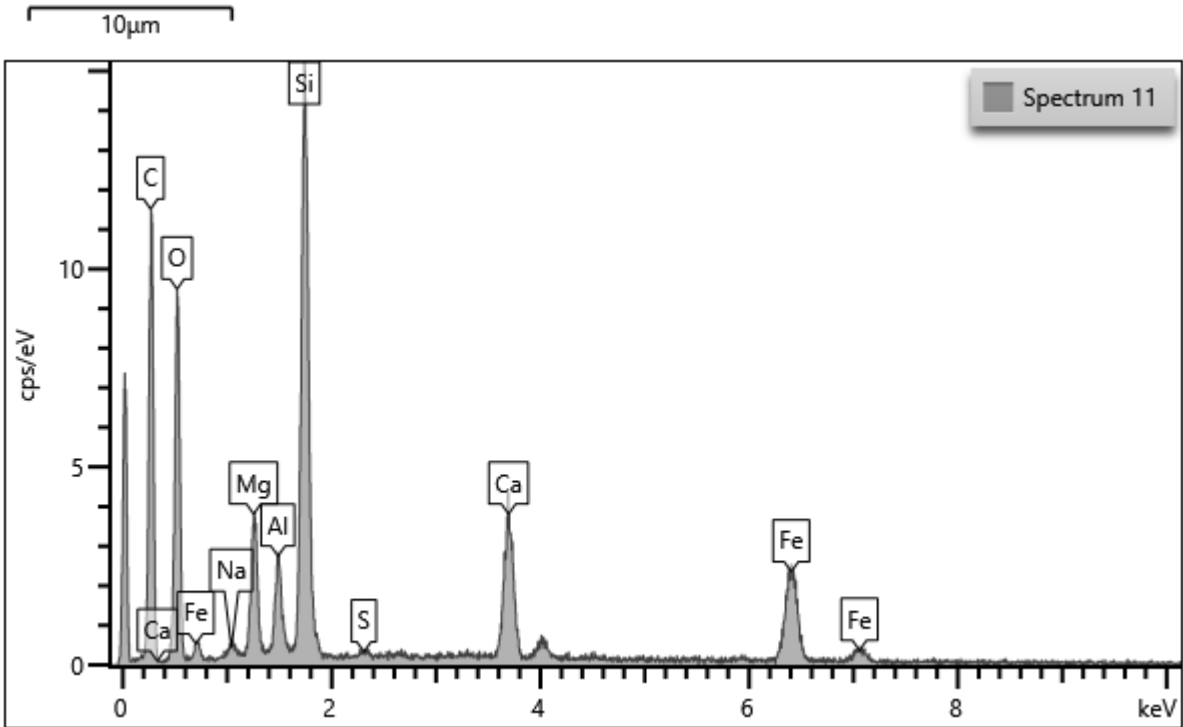
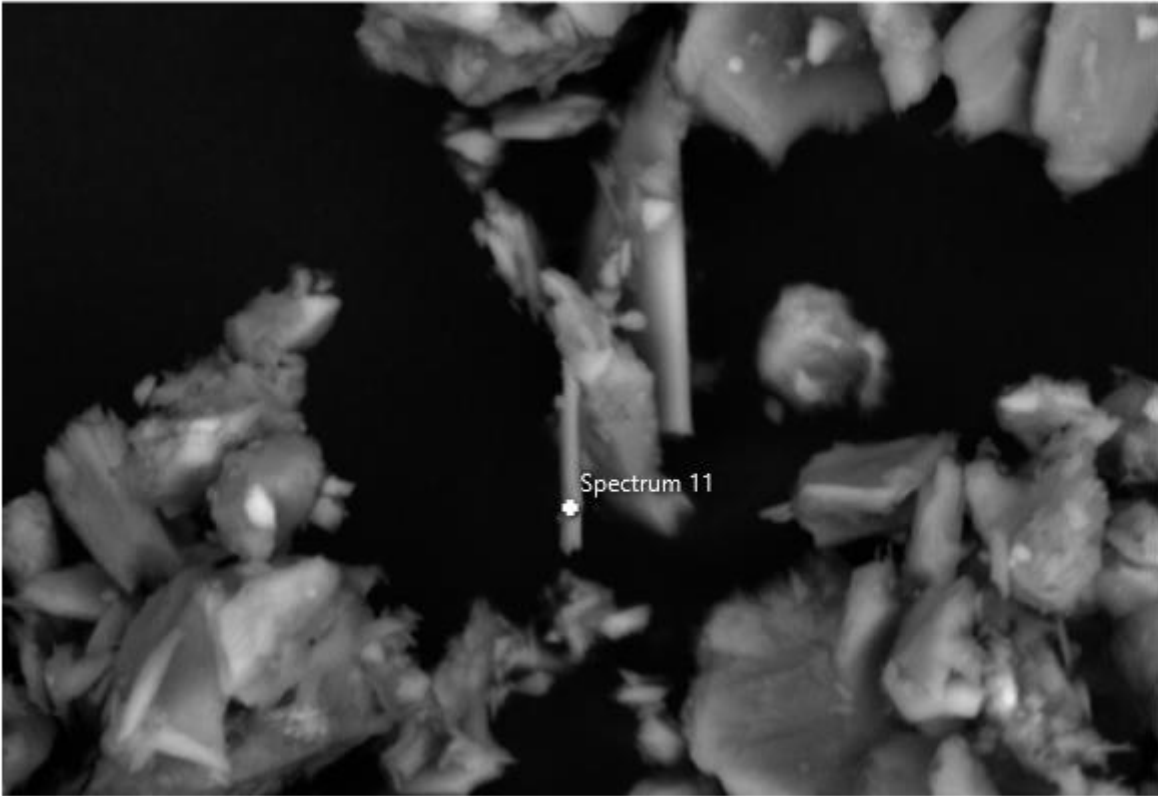
Electron Image 6 (SE)



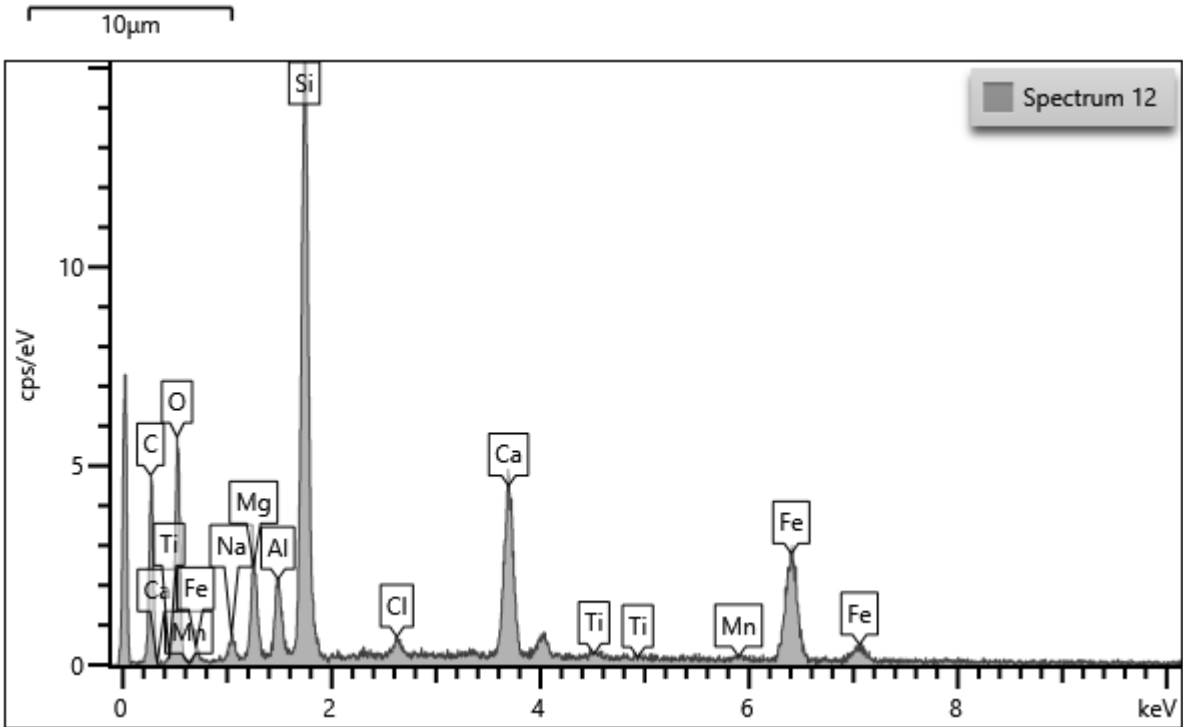
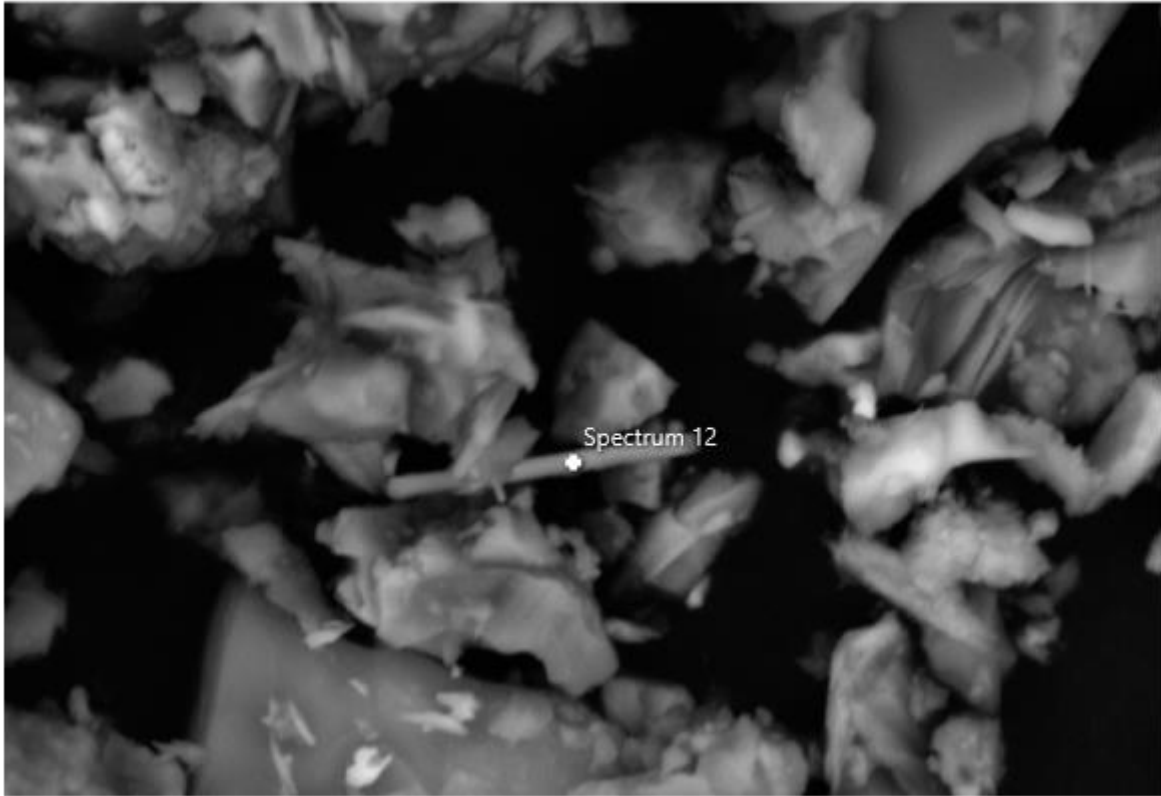
10µm



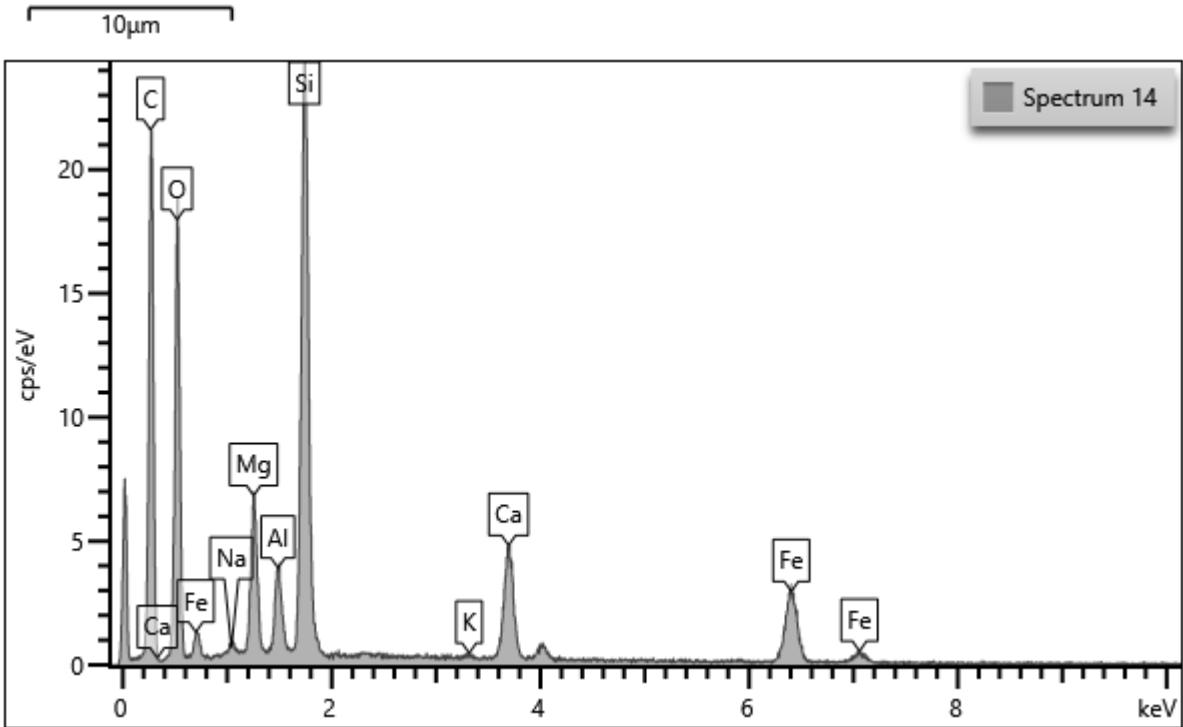
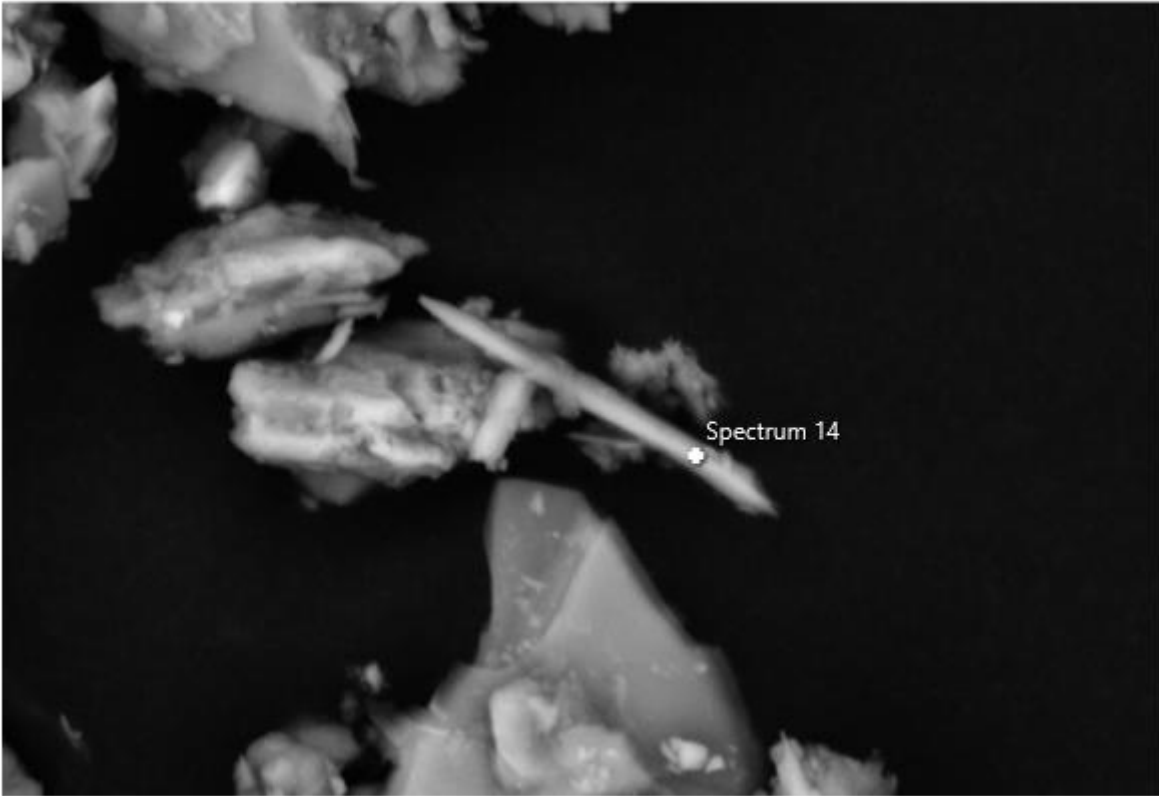
Electron Image 8 (SE)



Electron Image 9 (SE)



Electron Image 11 (SE)



Client: Landloch
Job number: 25_0206
Lab ID: 25_0206_005
Client ID: RWC79
Analysis: Fibre characterisation by scanning electron microscopy (SEM) with elemental analysis by energy dispersive spectroscopy (EDS) following AS 4964-2004 (modified for SEM)
Revision number: 0
Comments: None

Date received: 30/01/2025
Date analysed: 28/02/2025
Date reported: 28/02/2025

Executive summary

The sample was determined to contain < **0.01 wt % asbestos** mineral fibre.

Sample preparation

The sample was supplied to Microanalysis Australia as a bulk sample. The sample was screened at 2 mm, and each size fraction was inspected visually (both with the naked eye and with a low-powered microscope) for fibrous material.

As no fibrous material was visually apparent for immediate pre-selection, a representative sub-sample of the <2 mm fraction was disaggregated and was placed on top of at least one double-sided carbon tab before being carbon coated. Non-conducting samples require coating prior to SEM analysis to prevent charging whilst being analysed by the electron beam.

Analysis

The sample was analysed using a Carl Zeiss EVO50 scanning electron microscope (SEM) fitted with an Oxford INCA energy dispersive spectrometer (EDS). The sample was scanned at low magnification to identify any possible fibre clusters before the magnification was increased to 2000x magnification for closer examination.

EDS is a semi-quantitative technique (at best) on well prepared, optically flat samples. Factors such as sample unevenness may adversely bias elemental concentration interpretation. EDS has a spatial resolution of ~5 µm meaning spectra from particles less than this size may contain elemental concentrations biased by their surroundings.

All images were acquired using backscatter electrons. Image brightness is proportional to average atomic number – the brighter the pixel, the higher the atomic number of the element.

Summary

Following AS 4964, asbestos is defined as:

- Many particles with aspect ratios (i.e. length to width ratios) ranging from 20:1 to 100:1 or higher for particles >5 micrometers in length. Bundles of fibres may have lower aspect ratios;
- Sets of fibre bundles generally less than 0.5, but always less than 1.0 micrometres in width, unless in thick bundles;
- In addition to the mandatory fibrillar crystal growth, one or more, and preferably three of the following aspects:
 - Parallel fibres occurring in bundles;
 - Fibre bundles displaying splayed ends;
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 - Fibres showing characteristic curvature.
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For this report, a limit of reporting (LoR) of 0.01 wt % has been assumed, as stated in AS 4964.

A selection of images/fields and associated elemental spectra are reported below. The fields are not representative of the area analysed. The images/fields were reported due to their higher fibre count.

Fibre #	Image/Field #	Diameter (µm)	Length (µm)	Aspect ratio	Major Elements	Minor Elements	Morphology	Assigned mineralogy
1	13/16	0.8	8.7	11 :1	O, Si	Al, Mg, Fe	Non-parallel sides	Chlorite group
2	14/17	1.4	10.2	7 :1	O, Si	Al, Mg, Fe	Non-parallel sides	Chlorite group
3	15/18	1.5	13.1	9 :1	Na, Cl	-	Non-parallel sides	Halite
4	16/20	1.2	27.2	23 :1	Na, Cl	Si, Al	Non-parallel sides	Halite
5	17/21	0.9	12.1	13 :1	O, Si	Al, Mg, Fe	Non-parallel sides	Chlorite group
6	19/24	1.5	17.1	11 :1	O, Si	Al, Mg, Fe	Non-parallel sides	Chlorite group
7	21/26	2.0	19.2	10 :1	O, Si	Al, Mg, Fe	Non-parallel sides	Chlorite group

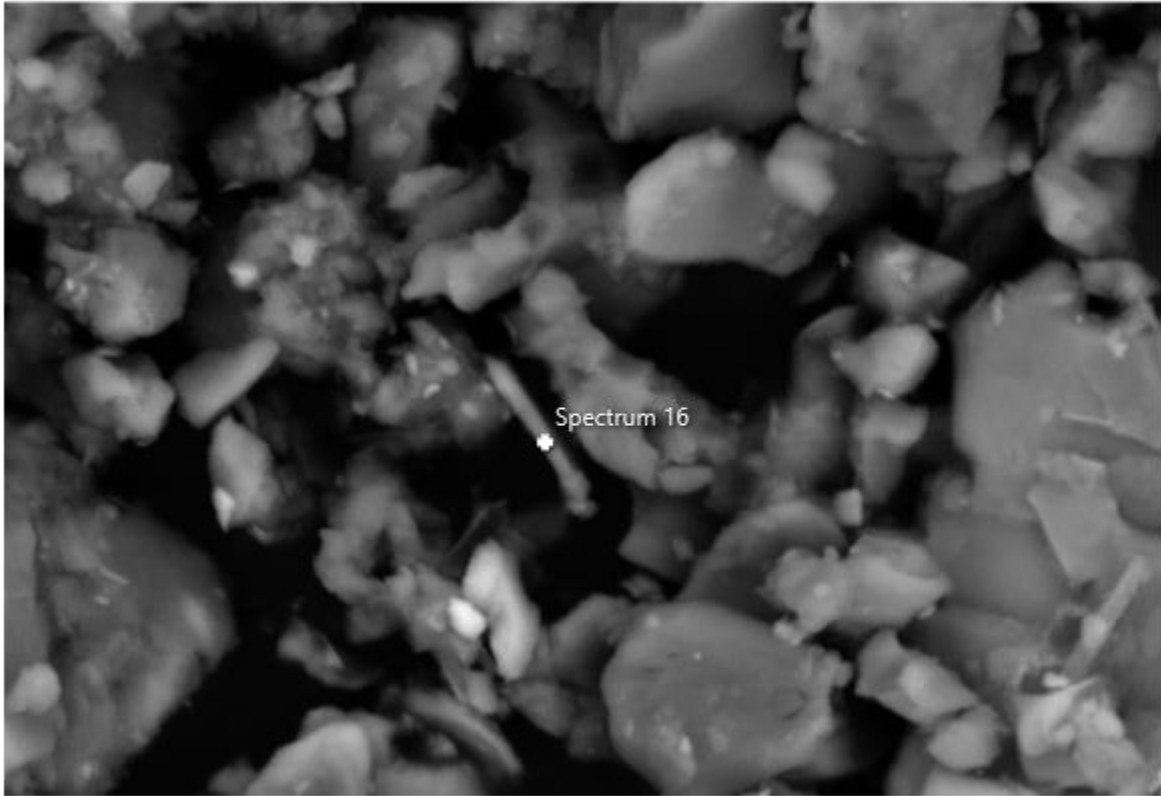
It should be noted that the higher resolution of the SEM may increase the number of fibres observed when compared with optical microscopy. Positive identification of asbestos fibres by SEM EDS uses elemental information that is not available by optical methods.

Analyst: Damon Blakey, *B.Sc. (Forensic Biology and Toxicology)*

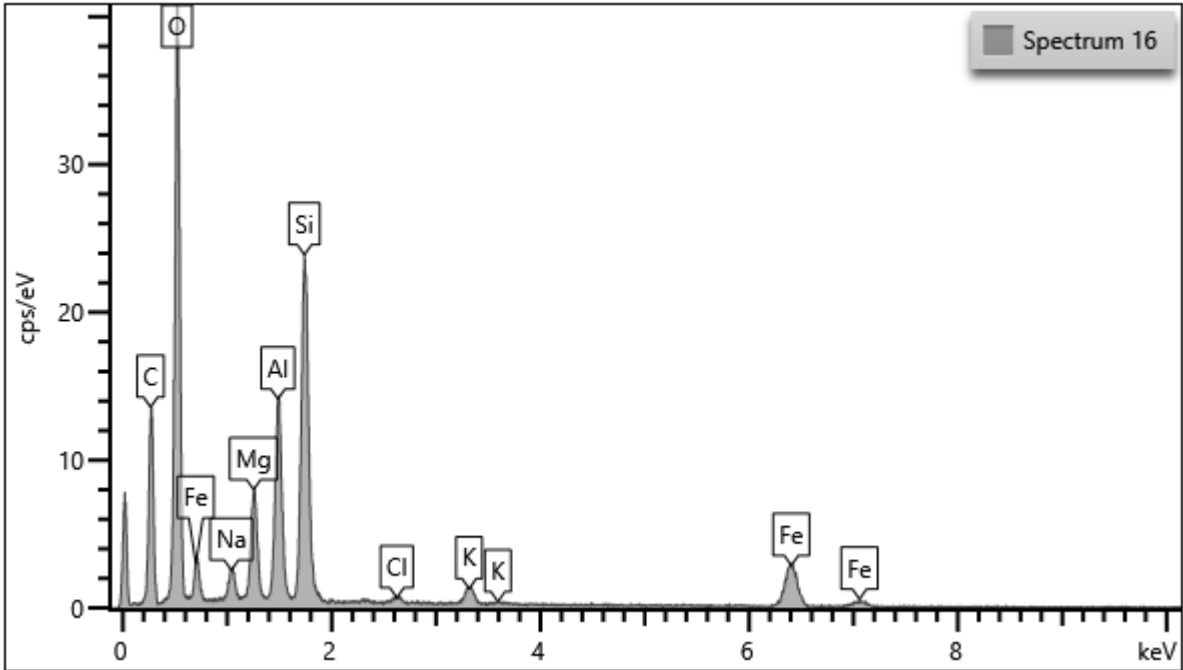
Reported by: Damon Blakey, *B.Sc. (Forensic Biology and Toxicology)*

Approved: Nimue Pendragon, *B.Sc.(Nanotechnology)*

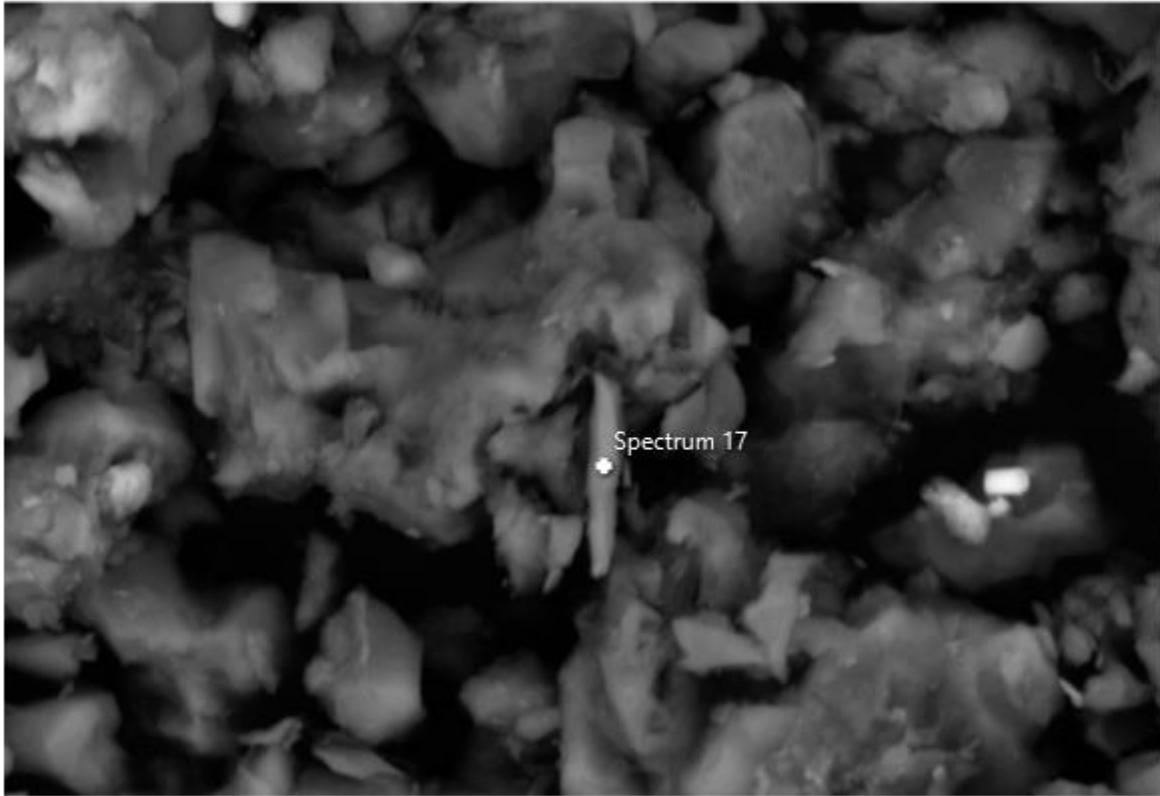
Electron Image 13 (SE)



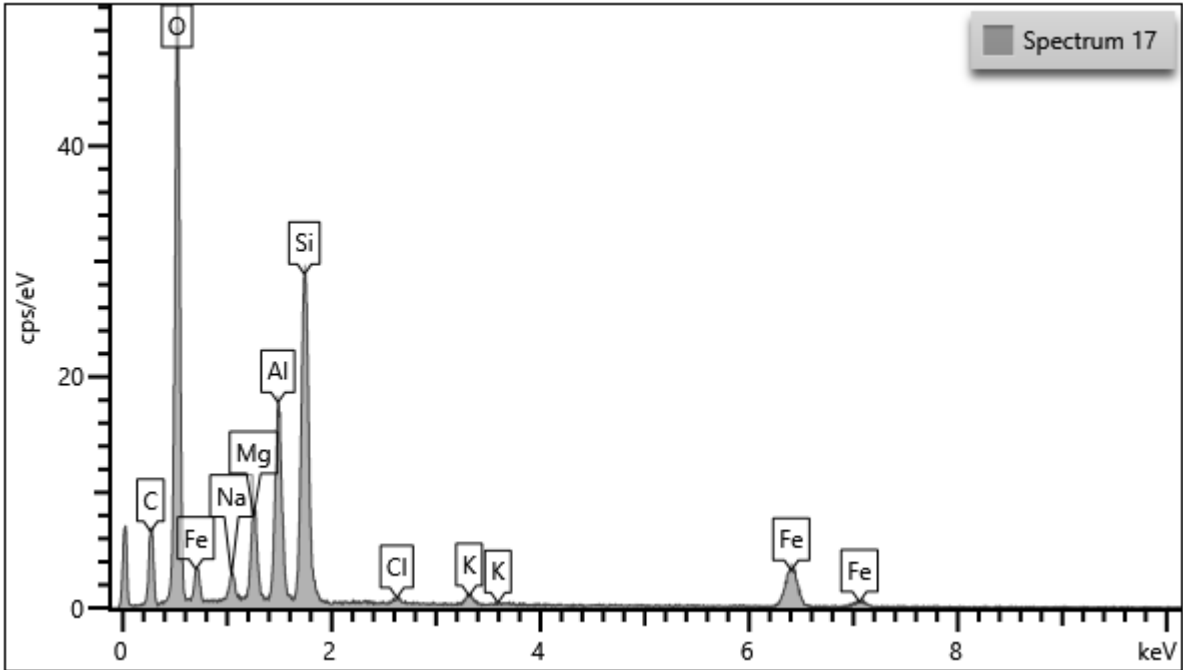
10µm



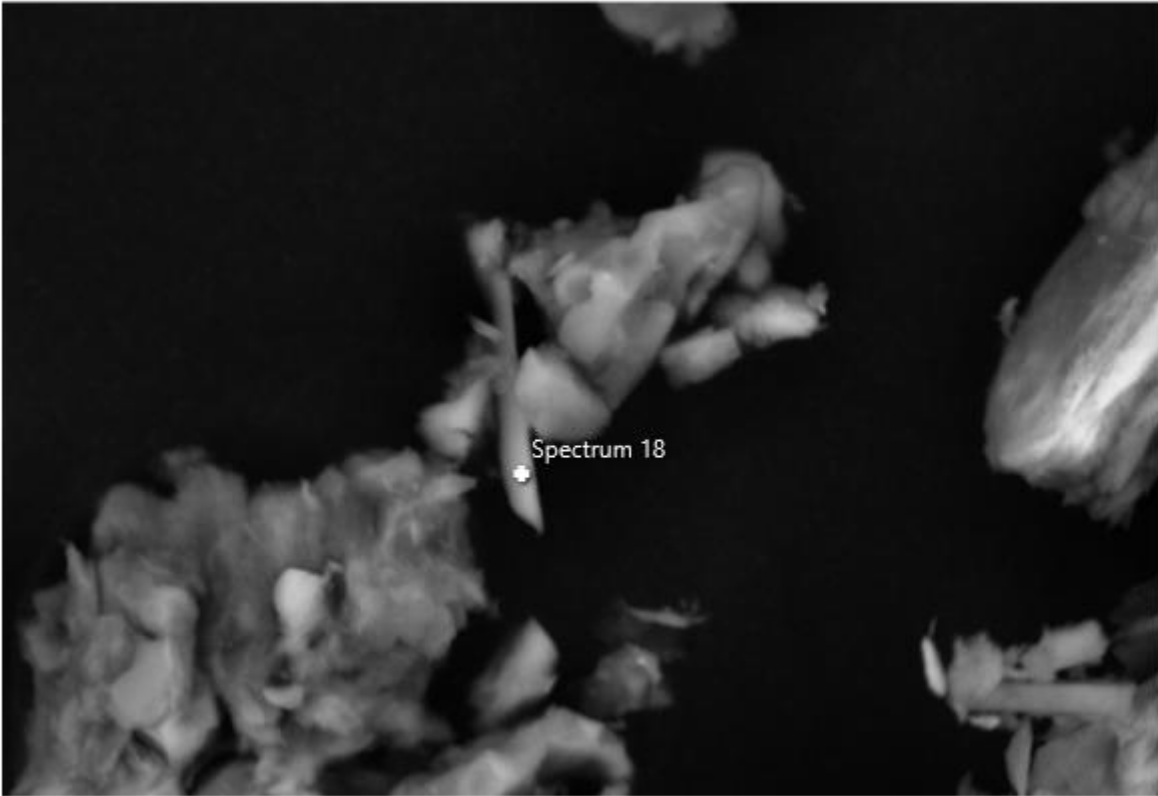
Electron Image 14 (SE)



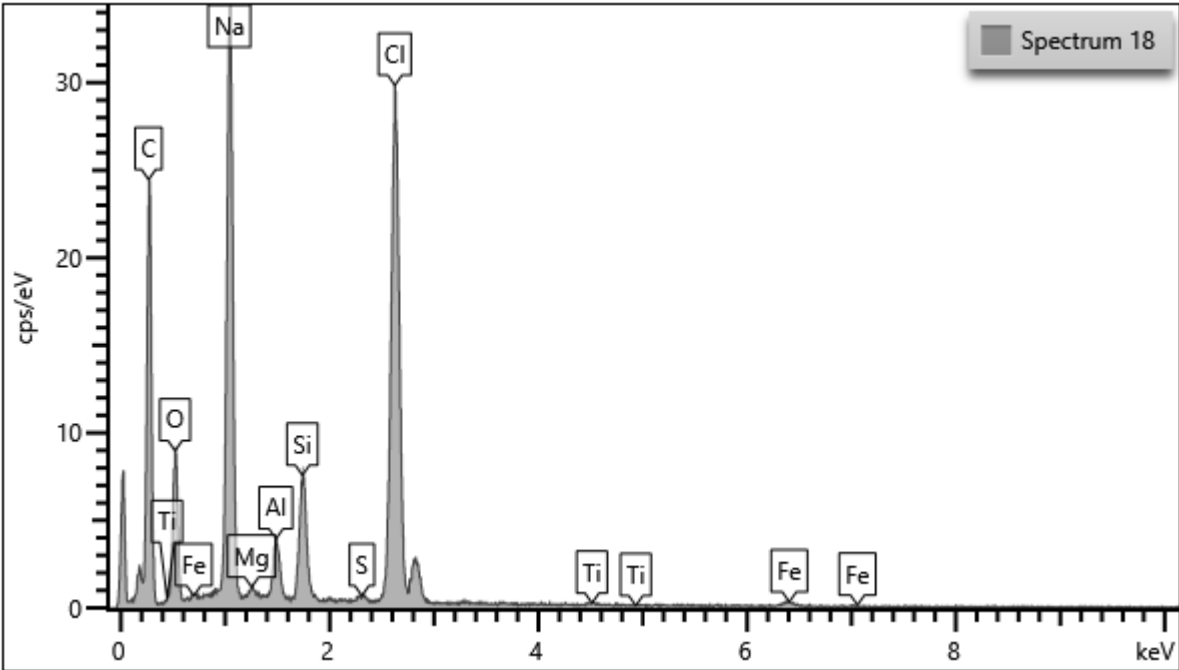
10µm



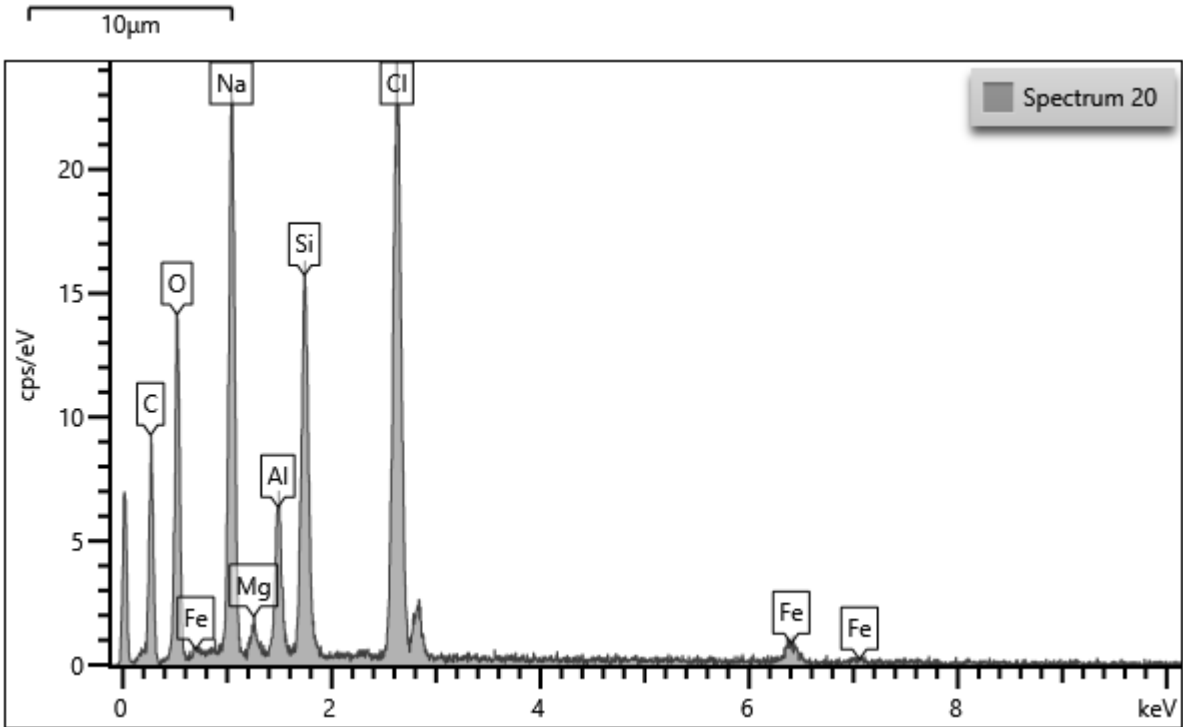
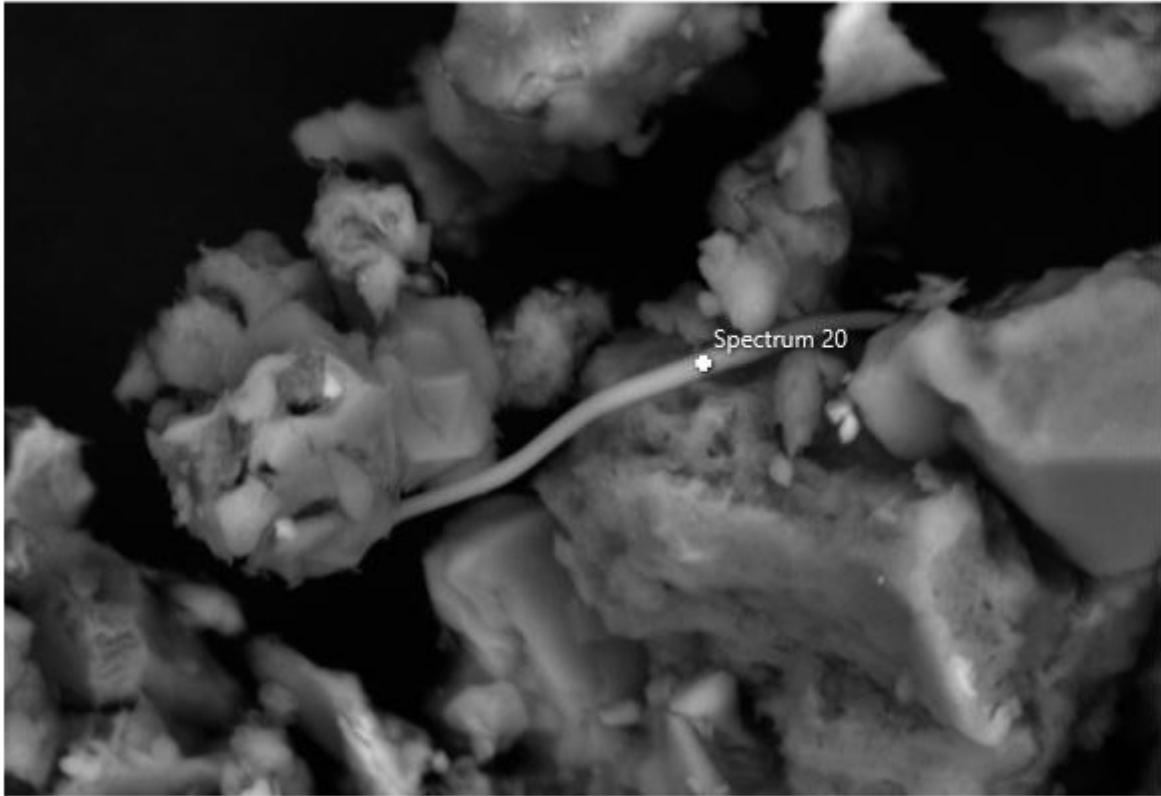
Electron Image 15 (SE)



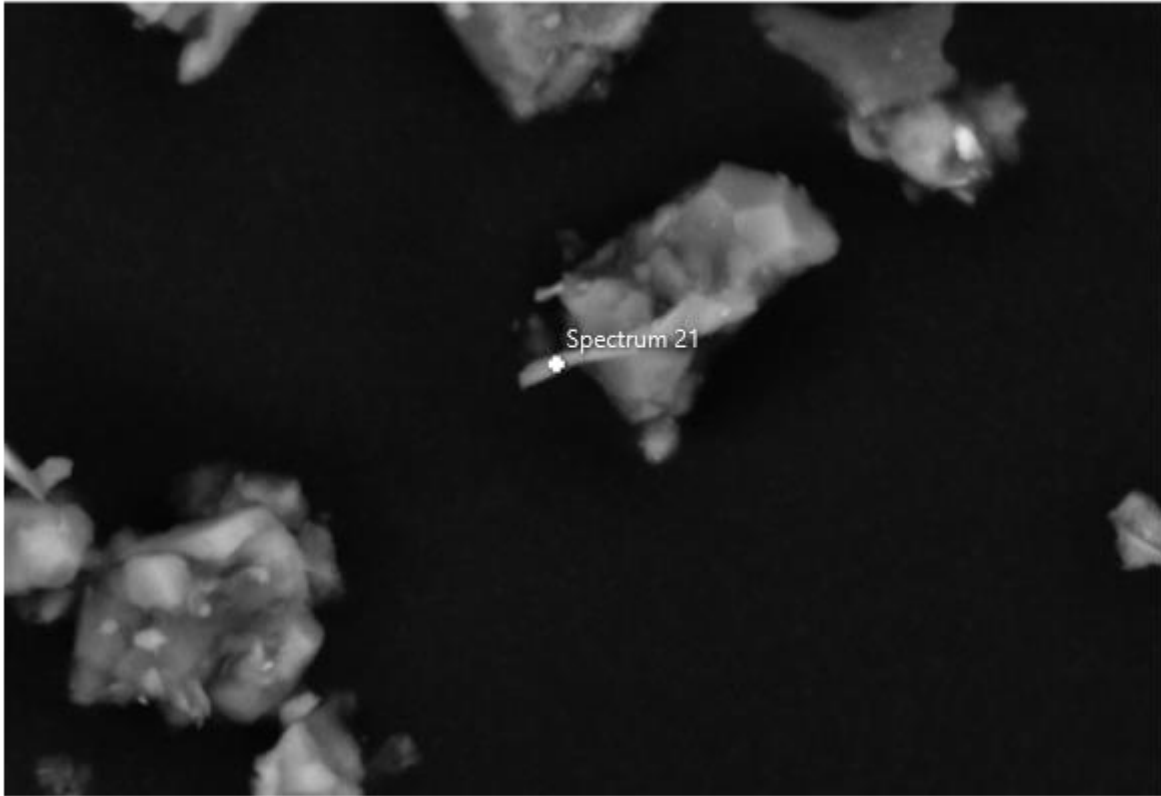
10µm



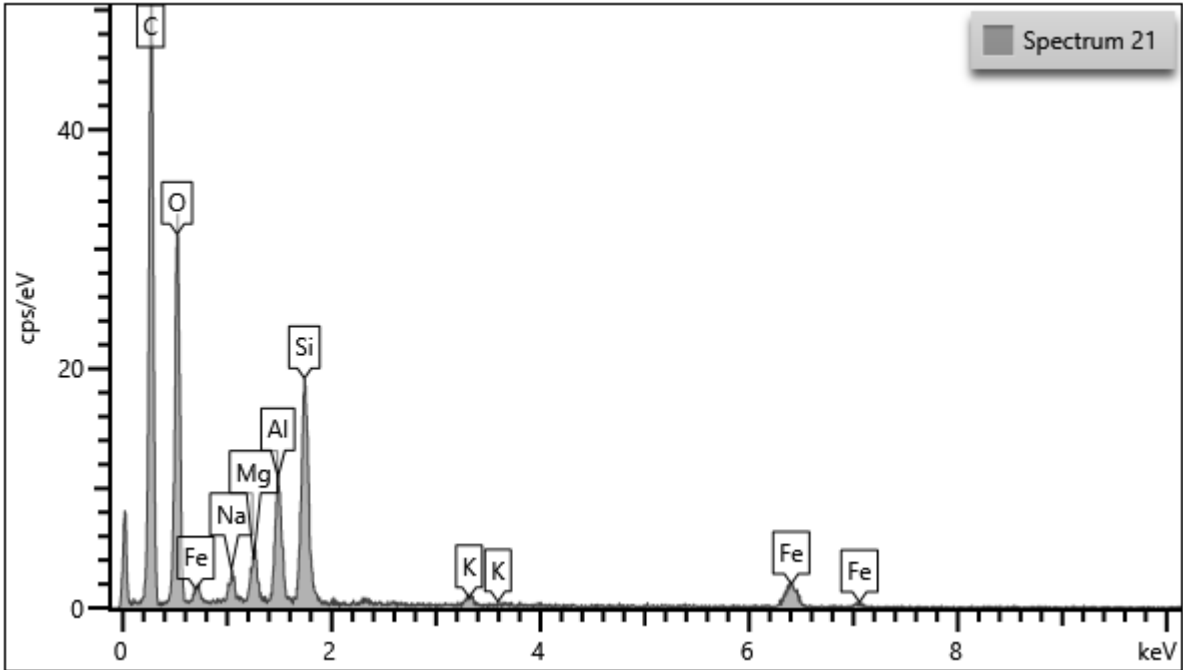
Electron Image 16 (SE)



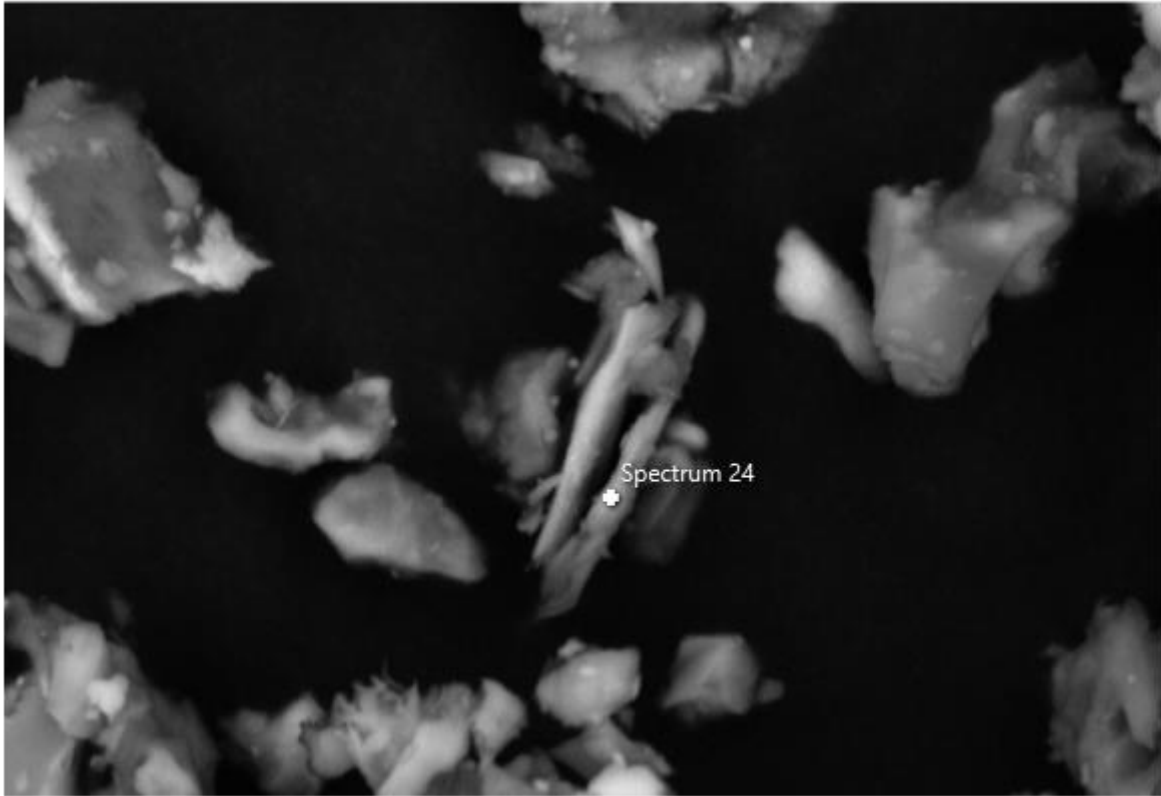
Electron Image 17 (SE)



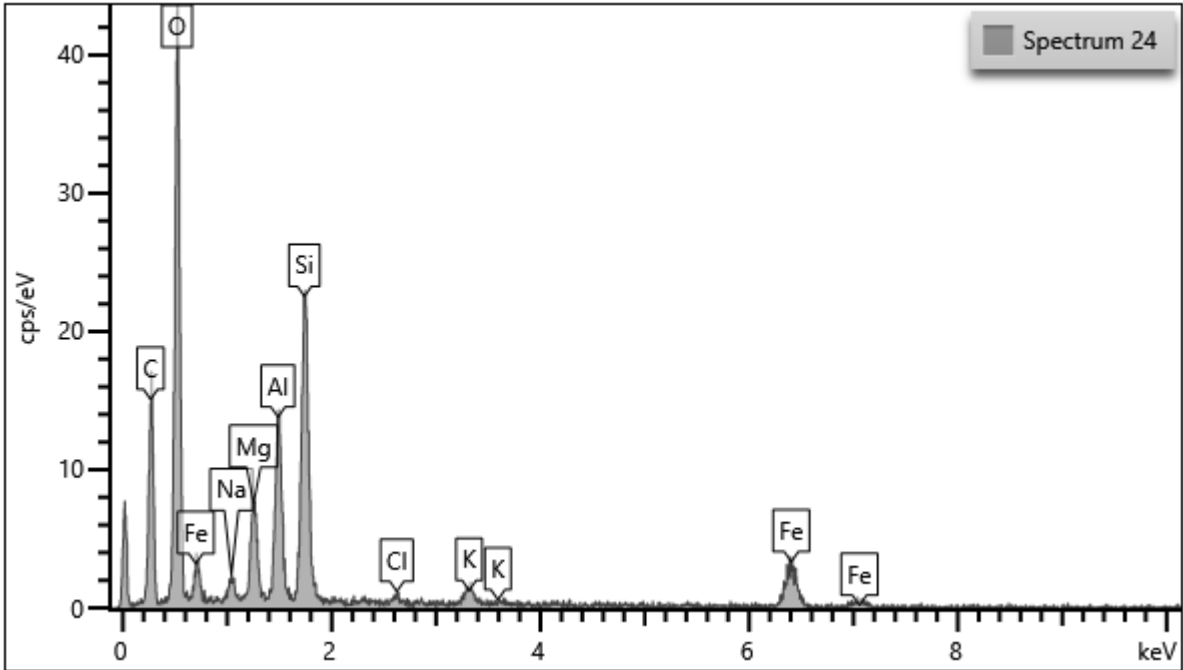
10µm



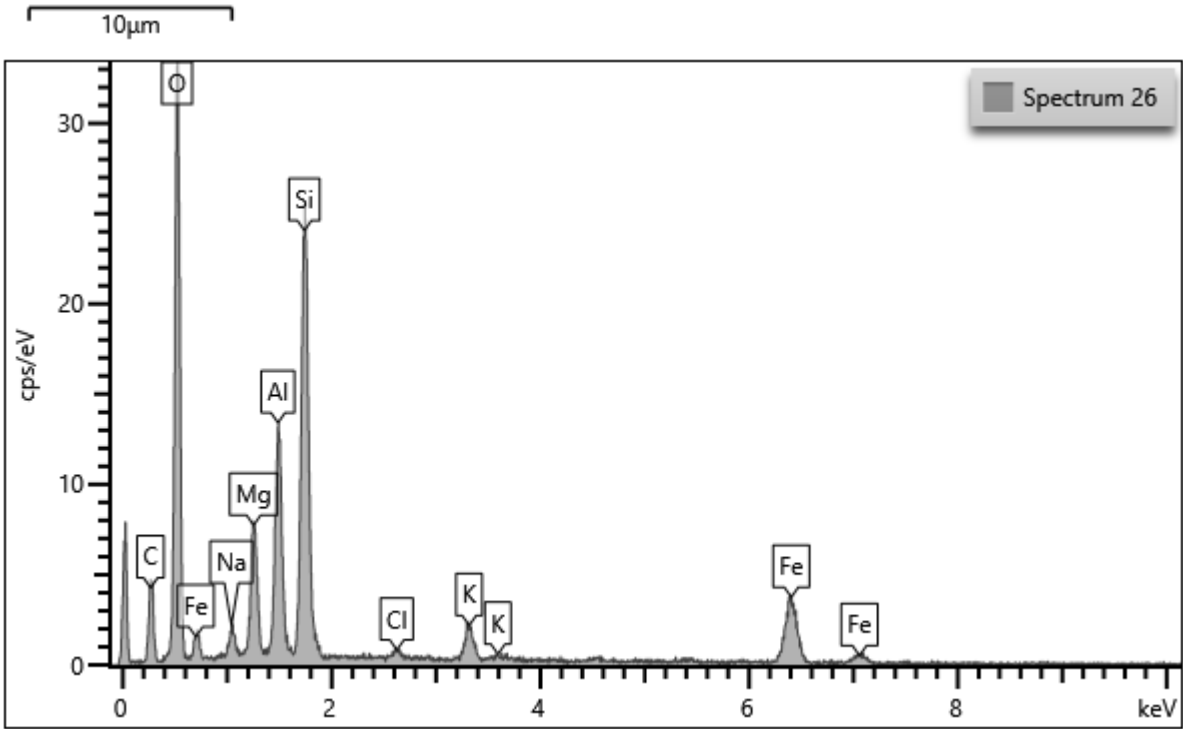
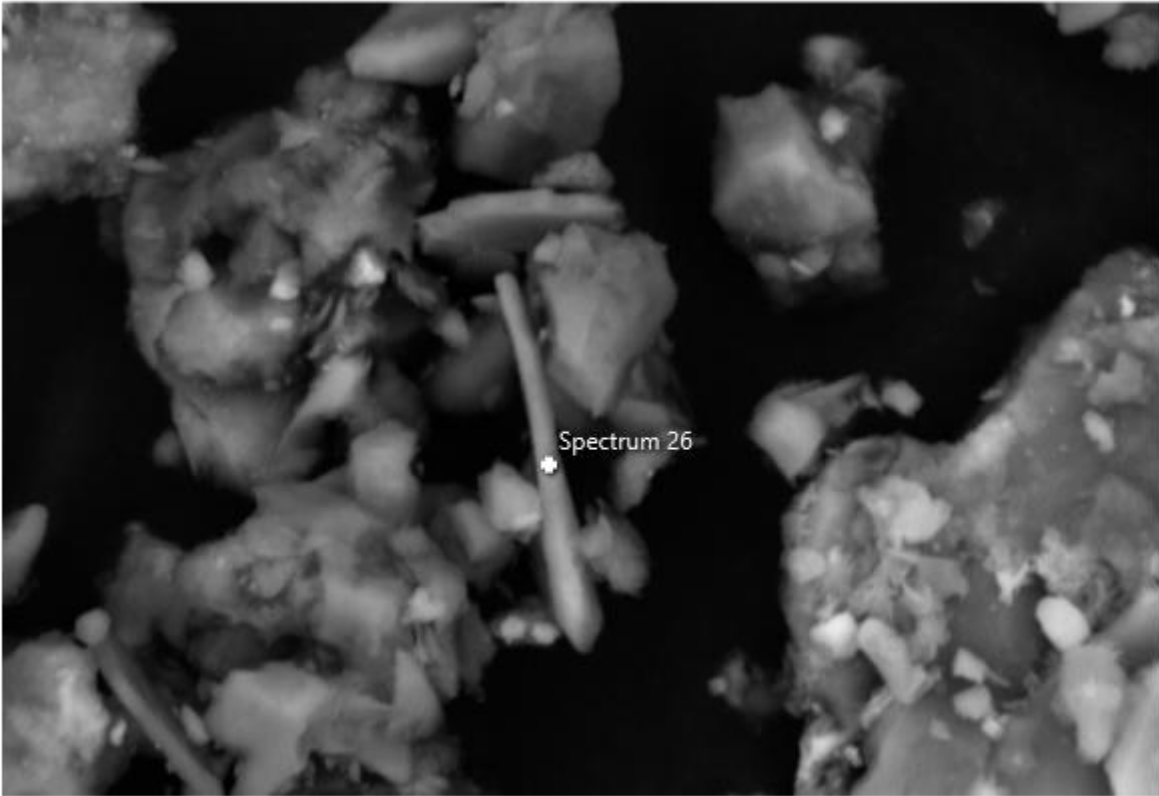
Electron Image 19 (SE)



10µm



Electron Image 21 (SE)



Client: Landloch
Job number: 25_0206
Lab ID: 25_0206_006
Client ID: RWC123
Analysis: Fibre characterisation by scanning electron microscopy (SEM) with elemental analysis by energy dispersive spectroscopy (EDS) following AS 4964-2004 (modified for SEM)
Revision number: 0
Comments: None

Date received: 30/01/2025
Date analysed: 28/02/2025
Date reported: 28/02/2025

Executive summary

The sample was determined to contain < **0.01 wt % asbestos** mineral fibre.

Sample preparation

The sample was supplied to Microanalysis Australia as a bulk sample. The sample was screened at 2 mm, and each size fraction was inspected visually (both with the naked eye and with a low-powered microscope) for fibrous material.

As no fibrous material was visually apparent for immediate pre-selection, a representative sub-sample of the <2 mm fraction was disaggregated and was placed on top of at least one double-sided carbon tab before being carbon coated. Non-conducting samples require coating prior to SEM analysis to prevent charging whilst being analysed by the electron beam.

Analysis

The sample was analysed using a Carl Zeiss EVO50 scanning electron microscope (SEM) fitted with an Oxford INCA energy dispersive spectrometer (EDS). The sample was scanned at low magnification to identify any possible fibre clusters before the magnification was increased to 2000x magnification for closer examination.

EDS is a semi-quantitative technique (at best) on well prepared, optically flat samples. Factors such as sample unevenness may adversely bias elemental concentration interpretation. EDS has a spatial resolution of ~5 µm meaning spectra from particles less than this size may contain elemental concentrations biased by their surroundings.

All images were acquired using backscatter electrons. Image brightness is proportional to average atomic number – the brighter the pixel, the higher the atomic number of the element.

Summary

Following AS 4964, asbestos is defined as:

- Many particles with aspect ratios (i.e. length to width ratios) ranging from 20:1 to 100:1 or higher for particles >5 micrometers in length. Bundles of fibres may have lower aspect ratios;
- Sets of fibre bundles generally less than 0.5, but always less than 1.0 micrometres in width, unless in thick bundles;
- In addition to the mandatory fibrillar crystal growth, one or more, and preferably three of the following aspects:
 - Parallel fibres occurring in bundles;
 - Fibre bundles displaying splayed ends;
 - Matted masses of individual fibres;
 - Fibres showing characteristic curvature.
- Respirable asbestos fibres are defined as:
 - Asbestos fibres less than 3 µm in width, and greater than 5 µm in length, and with a length to width ratio greater than 3 to 1.

Less than 5 observed fibres had an elemental composition and morphology indicative of asbestos mineral fibre, according to the definition in AS 4964.

Depending upon sample condition, composition and fibre type, the detection limit of AS 4964 has been found to lie generally in the range of 1 in 1,000 (0.1 wt %) to 1 in 10,000 (0.01 wt %) parts by weight, equivalent to 1 to 0.1 g/kg.

For this report, a limit of reporting (LoR) of 0.01 wt % has been assumed, as stated in AS 4964.

A selection of images/fields and associated elemental spectra are reported below. The fields are not representative of the area analysed. The images/fields were reported due to their higher fibre count.

Fibre #	Image/Field #	Diameter (µm)	Length (µm)	Aspect ratio	Major Elements	Minor Elements	Morphology	Assigned mineralogy
1	22/27	1.4	12.5	9 :1	O, Si	Mg, Ca, Fe, Al, Na	Non-parallel sides	Hornblende
2	23/28	1.5	24.1	16 :1	O, Si	Al, Mg, Fe	Non-parallel sides	Chlorite group
3	24/29	2.9	16.1	6 :1	O, Si	Al, Na	Non-parallel sides	Albite
4	25/30	1.2	15.4	13 :1	O, Si	Mg, Ca, Fe, Al, Na	Non-parallel sides	Hornblende
5	27/32	1.0	23.9	24 :1	O, Si	Mg, Ca, Fe, Al, Na	Non-parallel sides	Hornblende
6	28/33	0.3	6.7	22 :1	Na, Cl	-	Non-parallel sides	Halite
7	29/34	1.2	20.9	17 :1	O, Si	Al, Mg, Fe	Non-parallel sides	Chlorite group

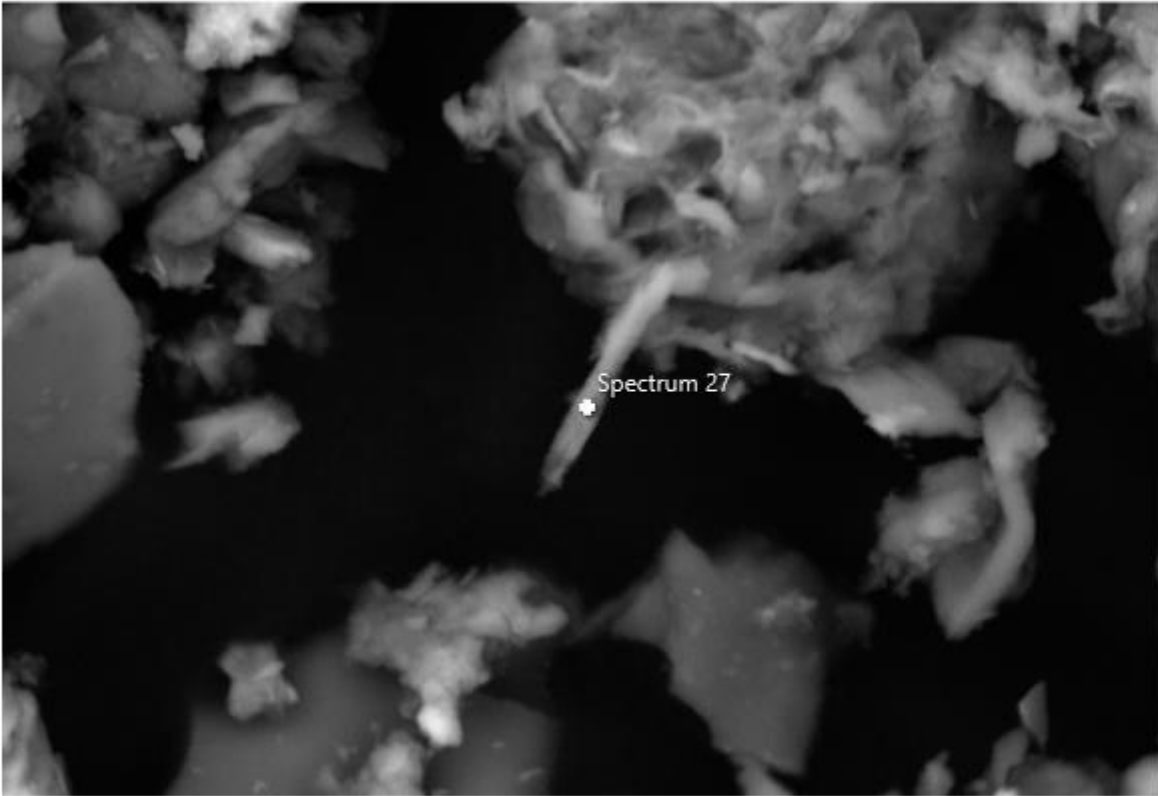
It should be noted that the higher resolution of the SEM may increase the number of fibres observed when compared with optical microscopy. Positive identification of asbestos fibres by SEM EDS uses elemental information that is not available by optical methods.

Analyst: Damon Blakey, *B.Sc. (Forensic Biology and Toxicology)*

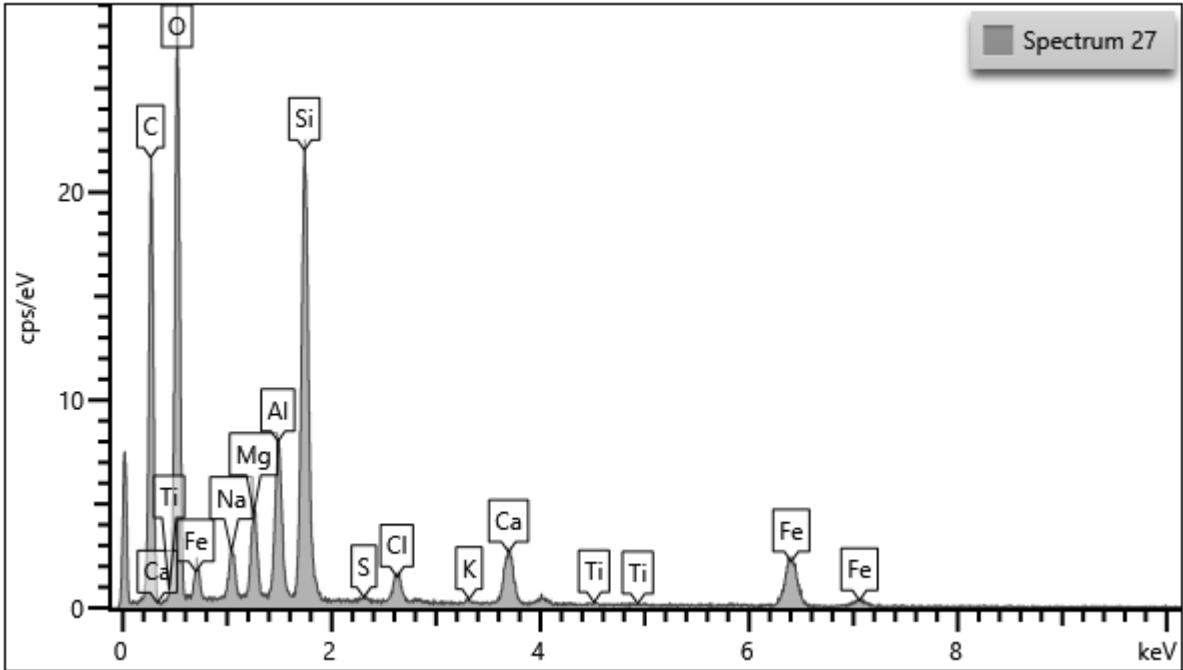
Reported by: Damon Blakey, *B.Sc. (Forensic Biology and Toxicology)*

Approved: Nimue Pendragon, *B.Sc.(Nanotechnology)*

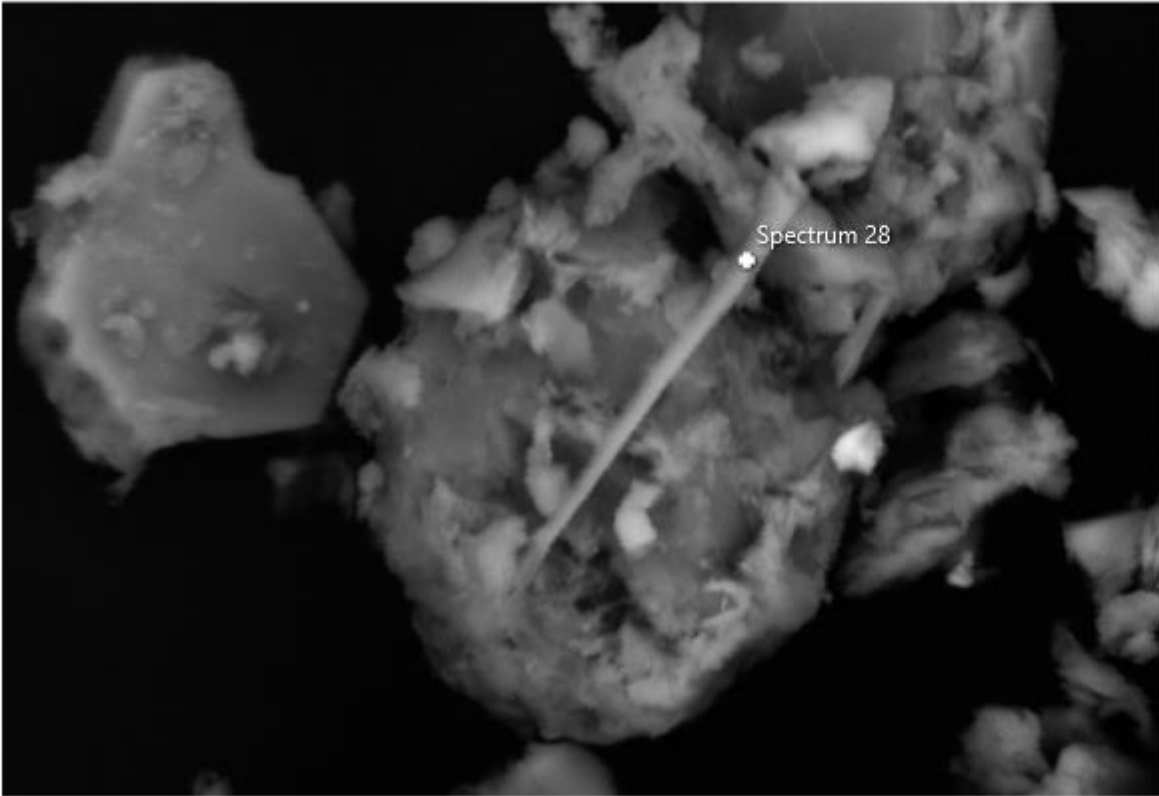
Electron Image 22 (SE)



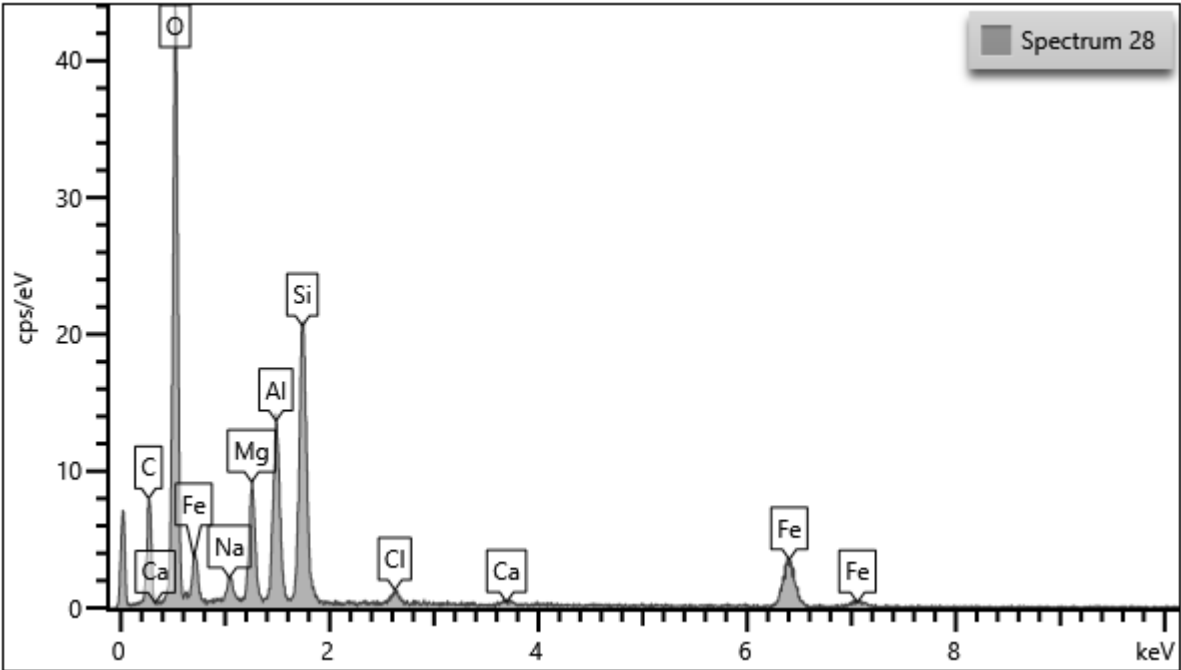
10µm



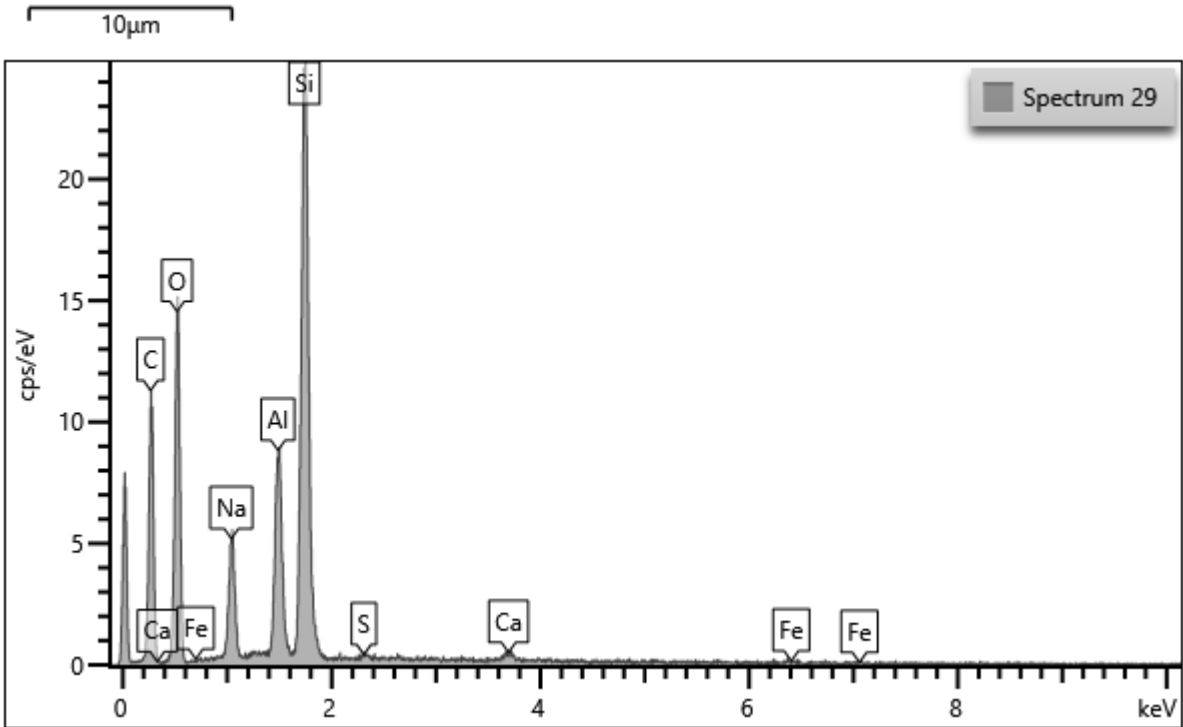
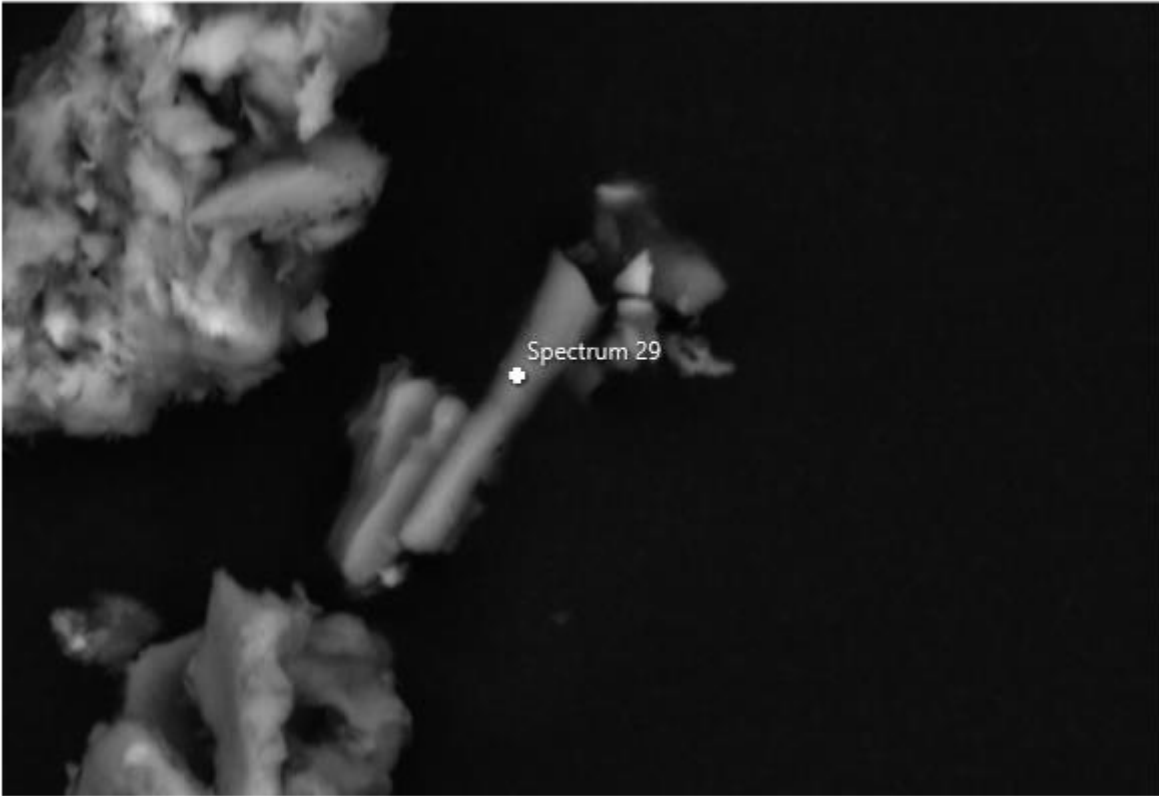
Electron Image 23 (SE)



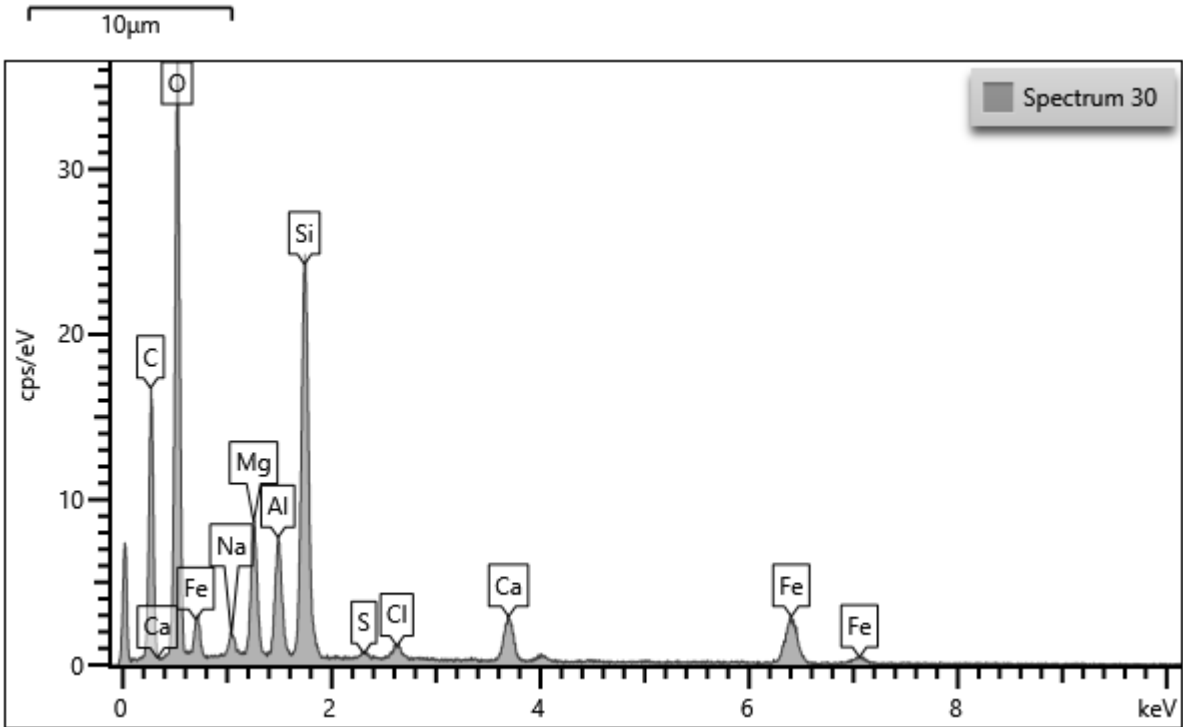
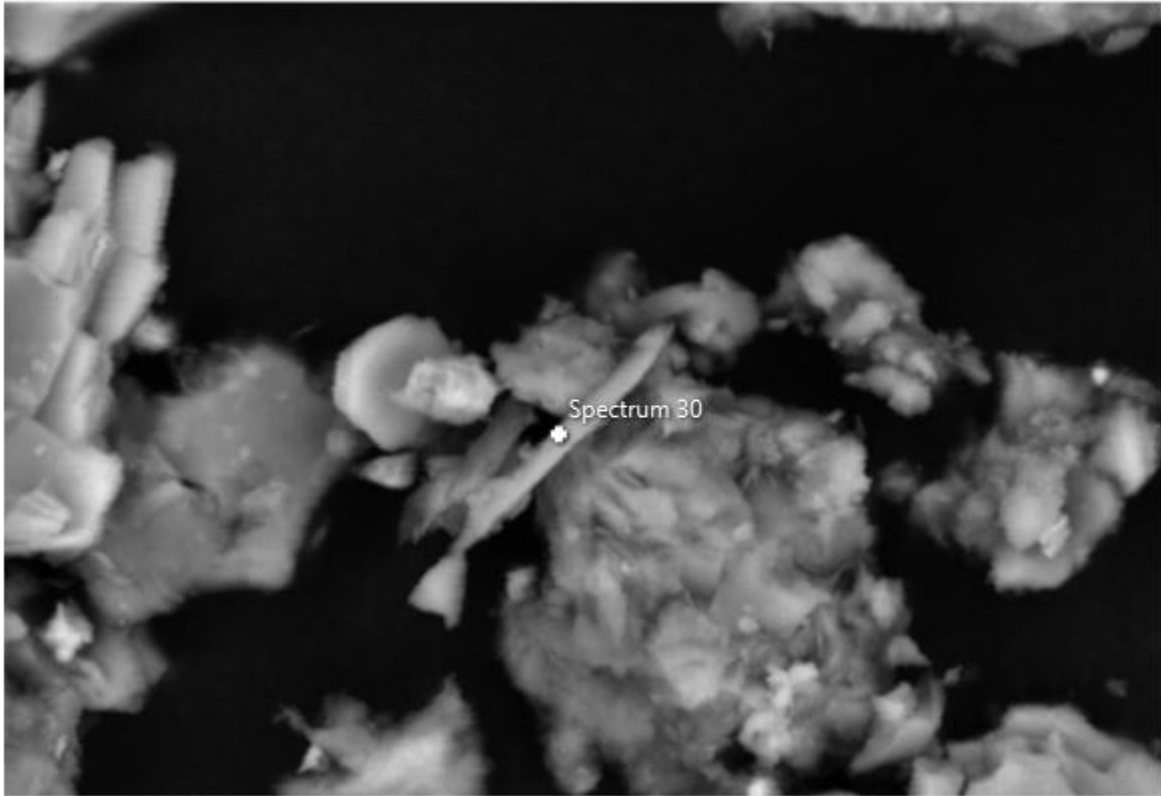
10µm



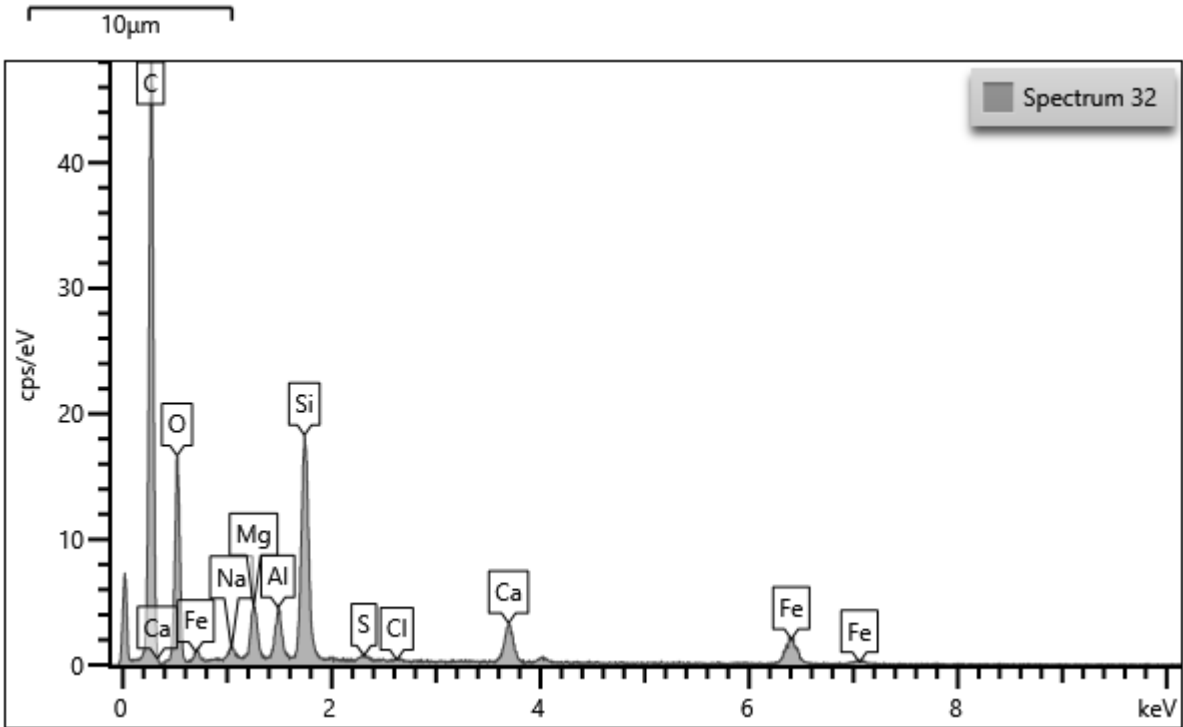
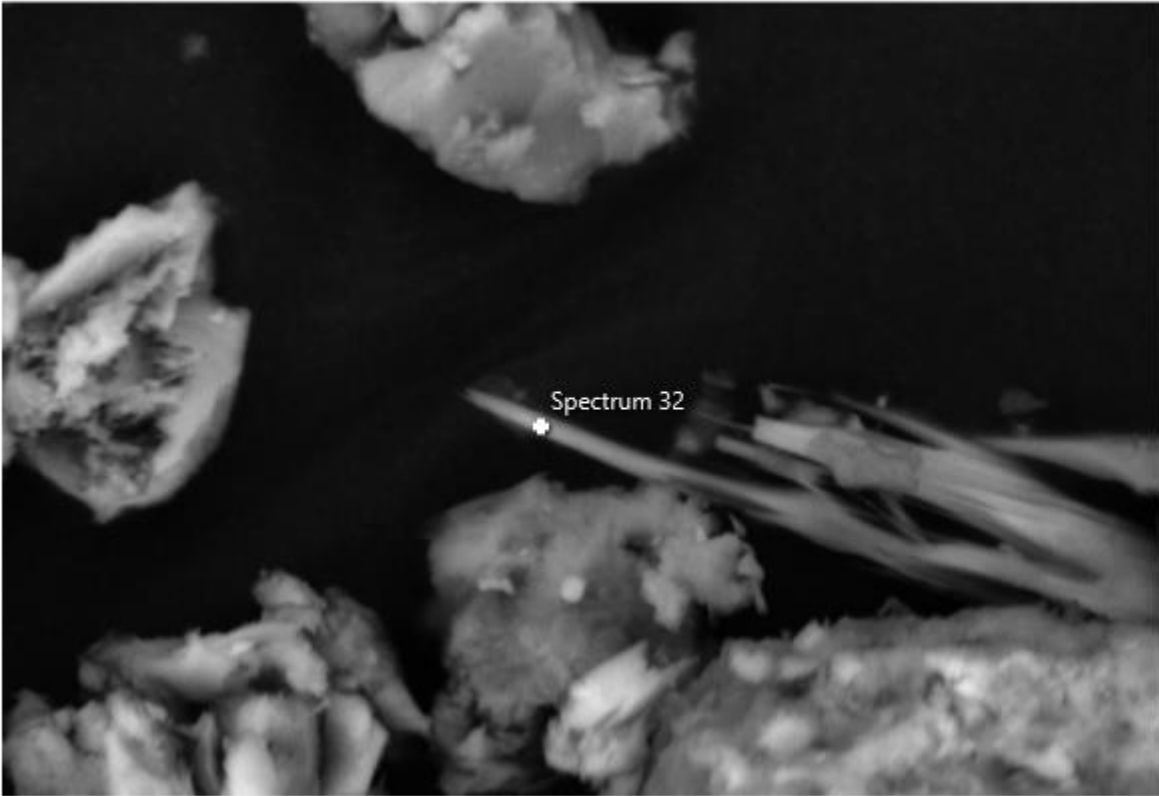
Electron Image 24 (SE)



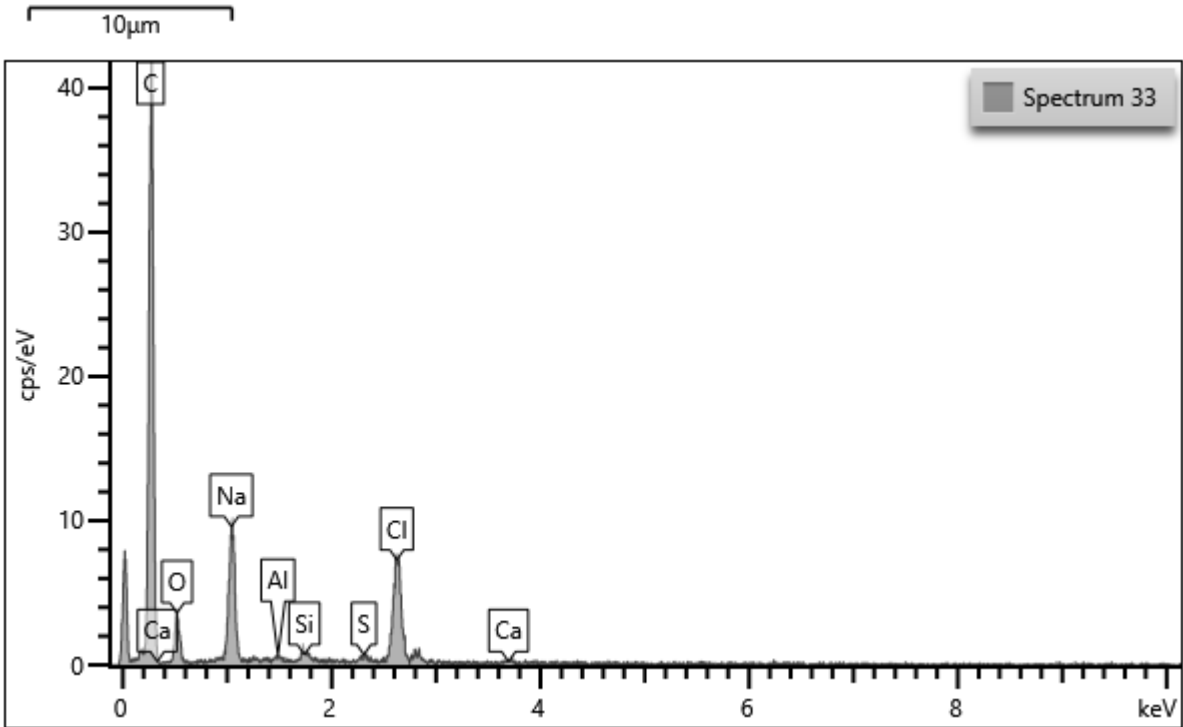
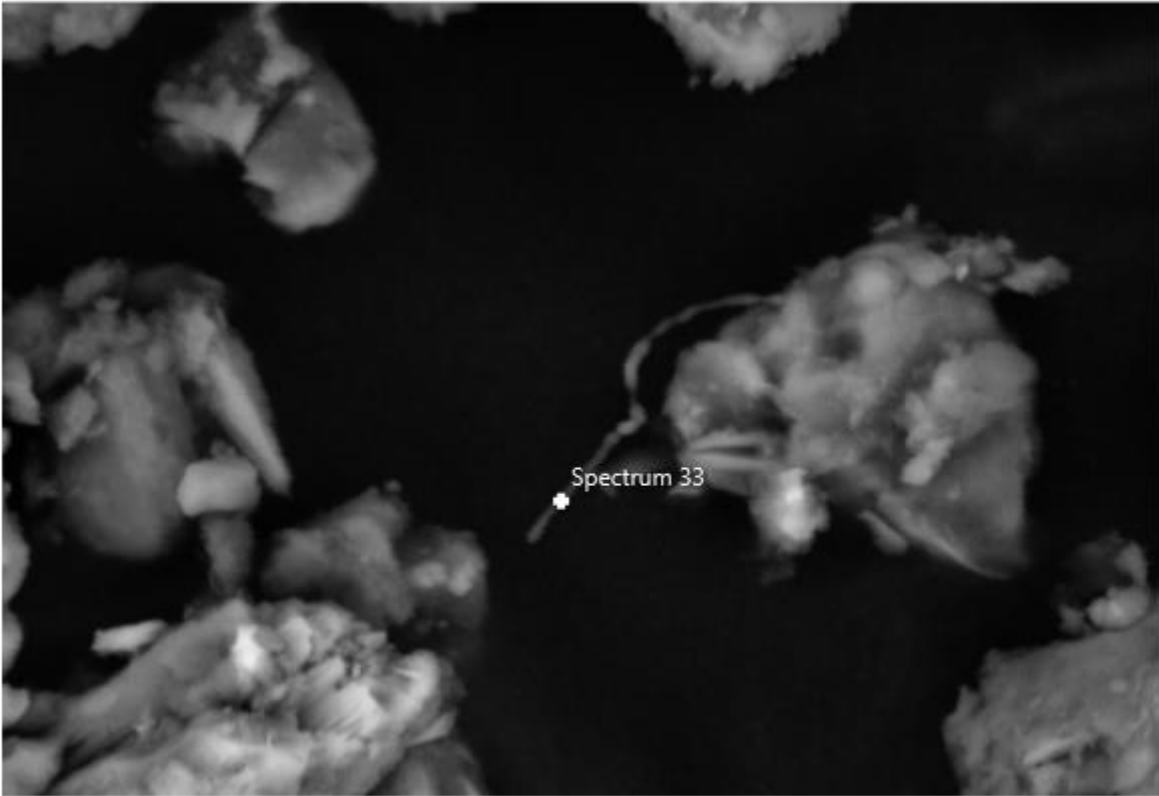
Electron Image 25 (SE)



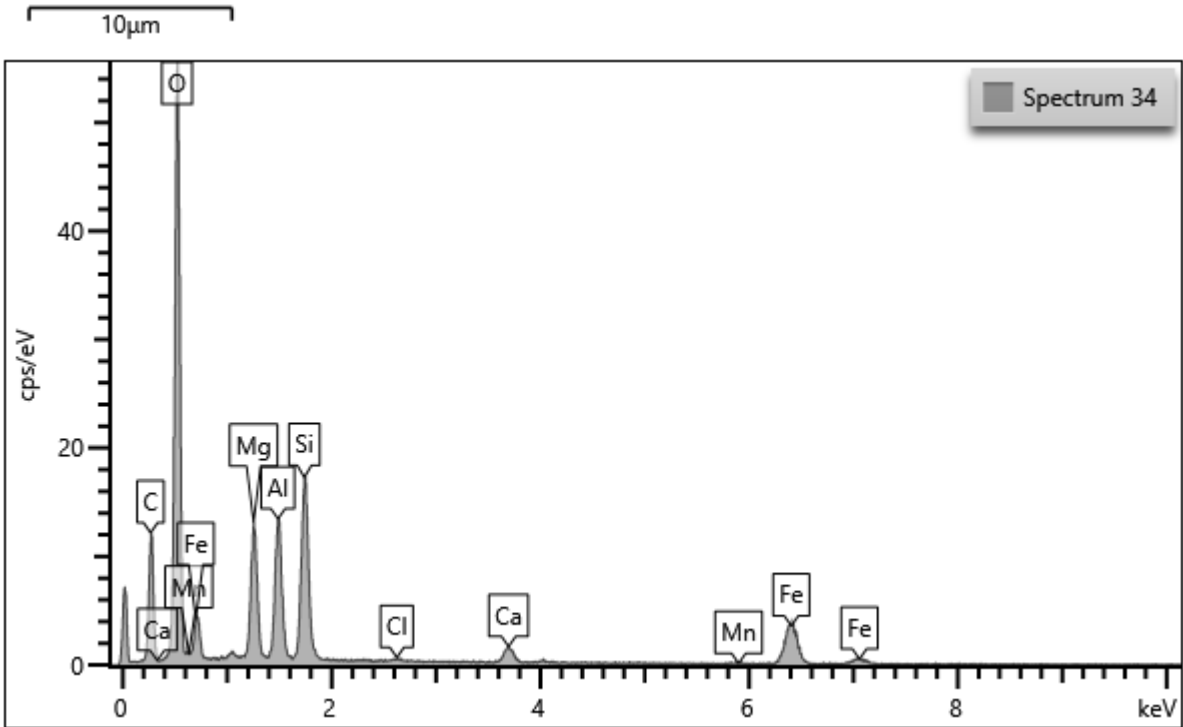
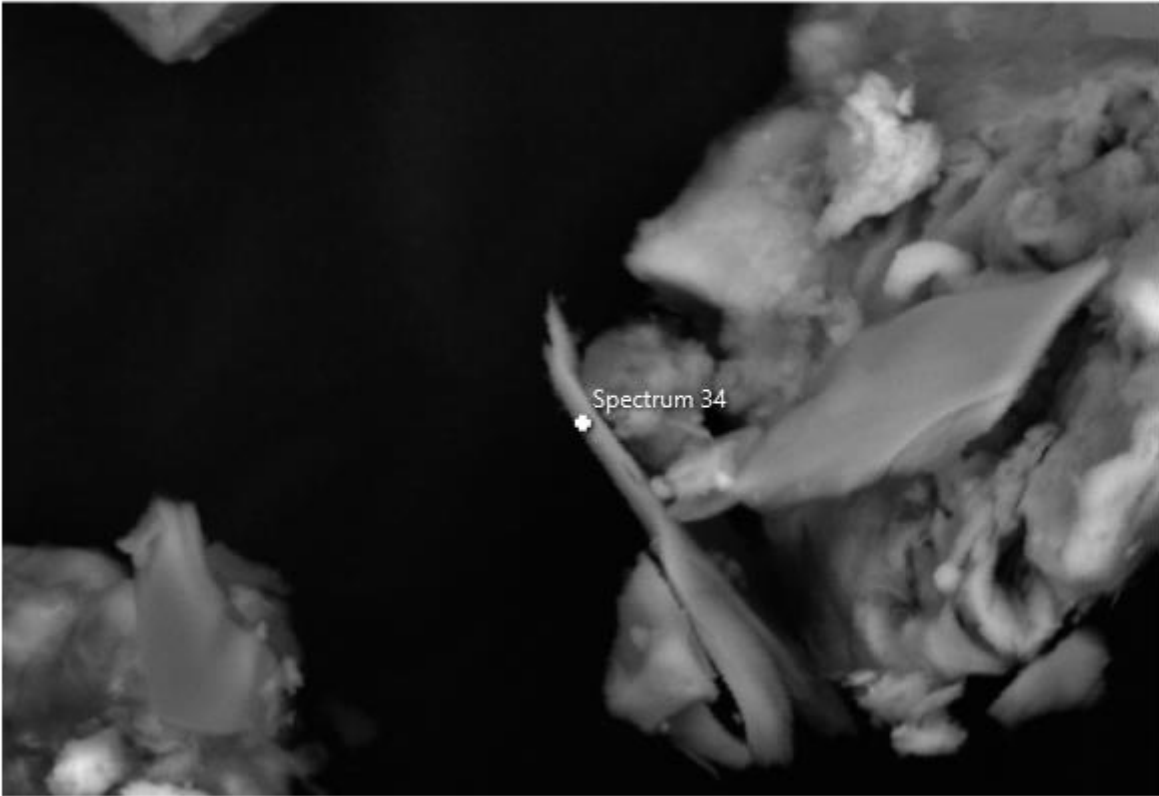
Electron Image 27 (SE)



Electron Image 28 (SE)



Electron Image 29 (SE)



Client: Landloch
Job number: 25_0206
Lab ID: 25_0206_007
Client ID: RWC128
Analysis: Fibre characterisation by scanning electron microscopy (SEM) with elemental analysis by energy dispersive spectroscopy (EDS) following AS 4964-2004 (modified for SEM)
Revision number: 0
Comments: None

Date received: 30/01/2025
Date analysed: 28/02/2025
Date reported: 28/02/2025

Executive summary

The sample was determined to contain < **0.01 wt % asbestos** mineral fibre.

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2	33/38	0.5	6.7	13 :1	O, Fe	-	Non-parallel sides	Iron oxide
3	34/40	1.9	7.8	4 :1	O, Si	Mg, Ca, Fe, Al, Na	Non-parallel sides	Hornblende
4	35/41	0.7	10.3	15 :1	O, Si	Mg, Ca, Fe, Al, Na	Irregular ends	Hornblende
5	36/42	0.7	11.3	16 :1	O, Si	Al, Mg, Fe	Non-parallel sides	Chlorite group
6	38/44	1.2	12.2	10 :1	O, Si	Mg, Ca, Fe, Al, Na	Irregular ends	Hornblende
7	41/48	1.9	15.4	8 :1	O, Si	Al, Mg, Fe	Non-parallel sides	Chlorite group

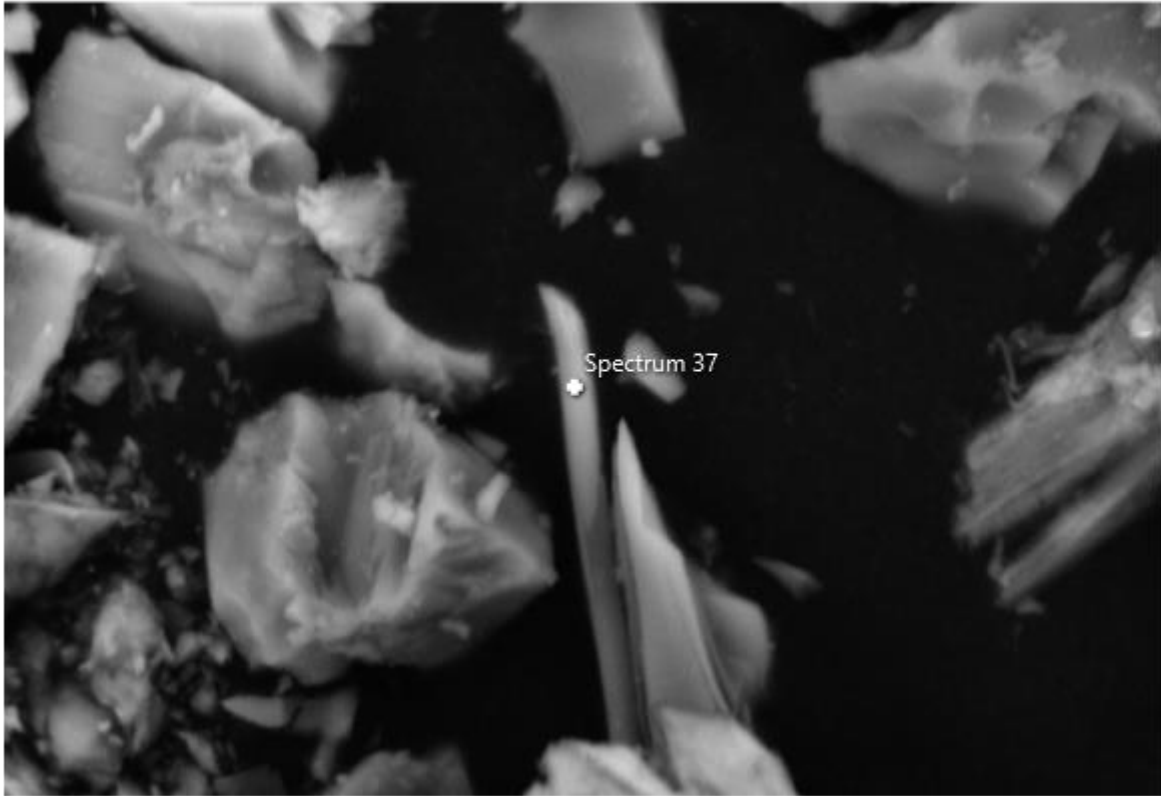
It should be noted that the higher resolution of the SEM may increase the number of fibres observed when compared with optical microscopy. Positive identification of asbestos fibres by SEM EDS uses elemental information that is not available by optical methods.

Analyst: Damon Blakey, *B.Sc. (Forensic Biology and Toxicology)*

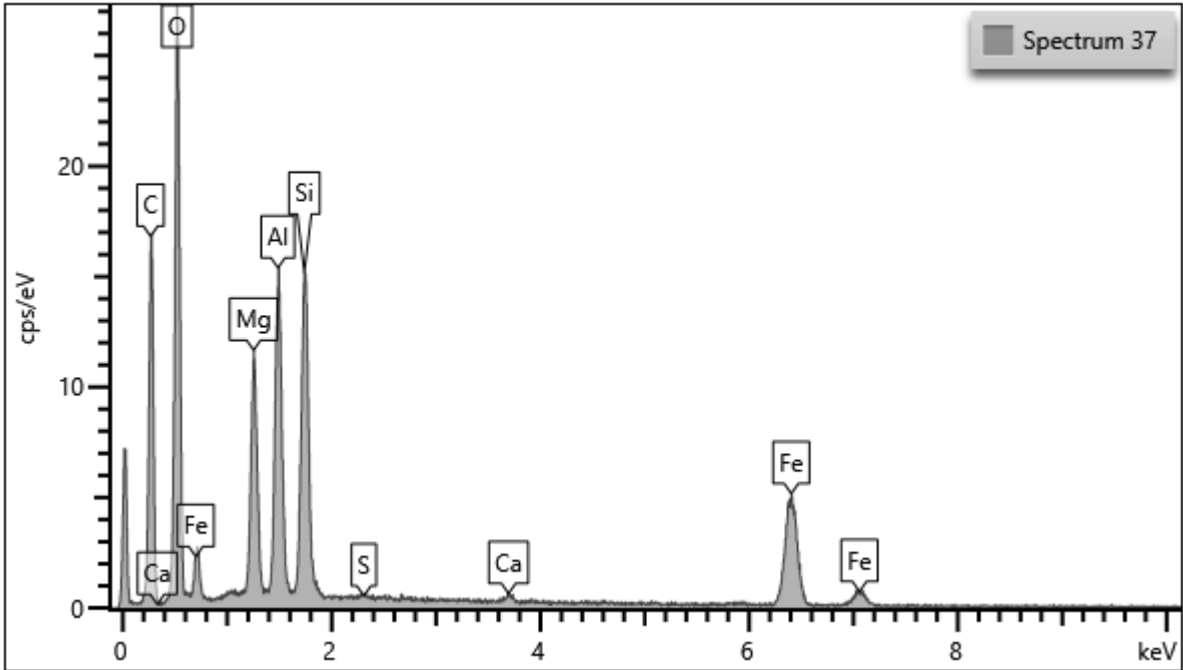
Reported by: Damon Blakey, *B.Sc. (Forensic Biology and Toxicology)*

Approved: Nimue Pendragon, *B.Sc.(Nanotechnology)*

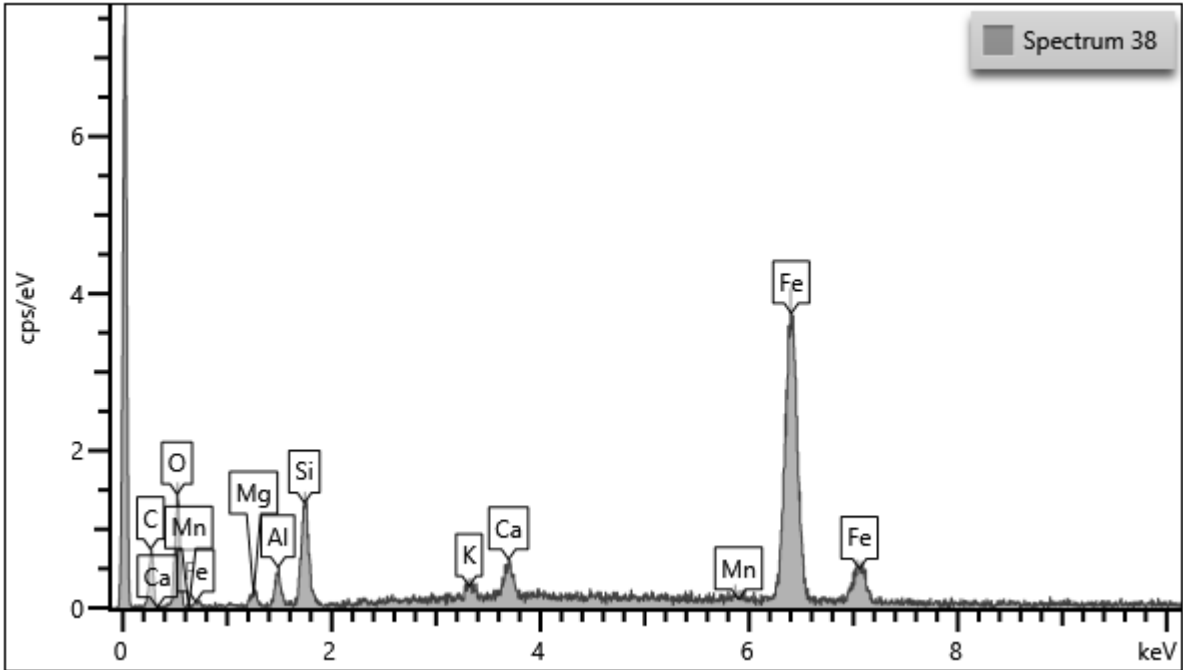
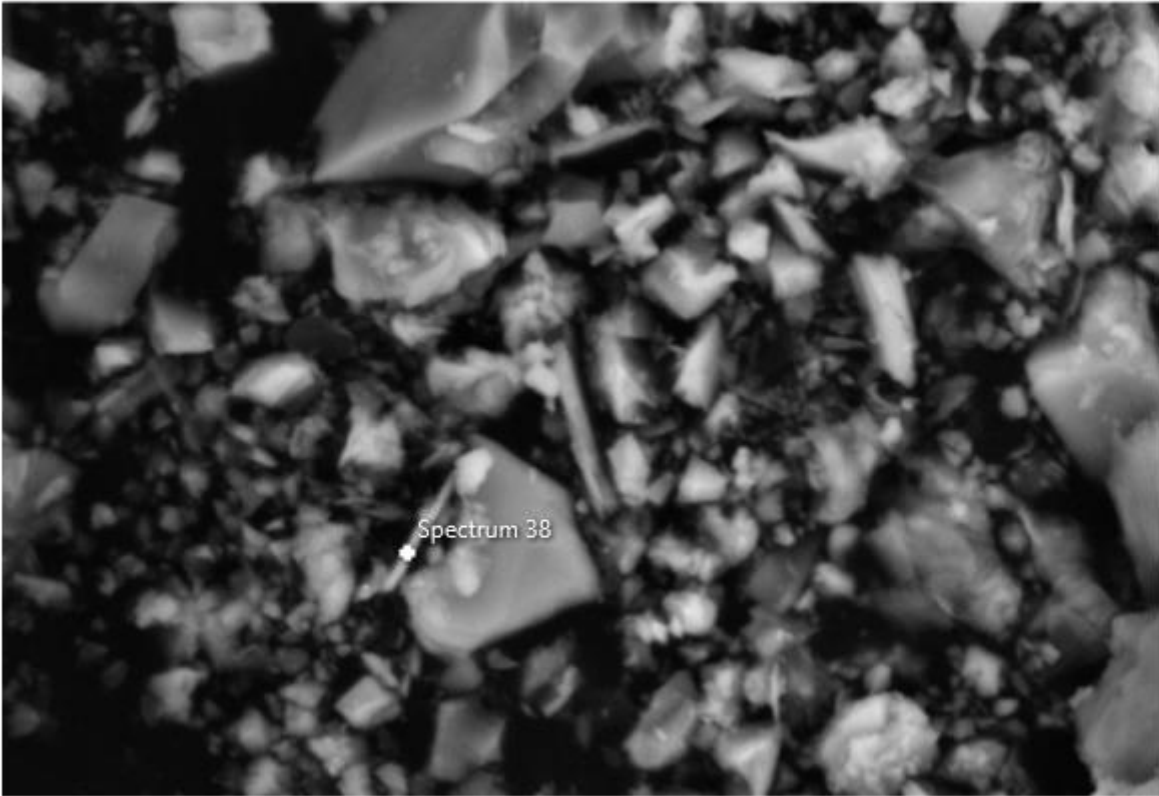
Electron Image 32 (SE)



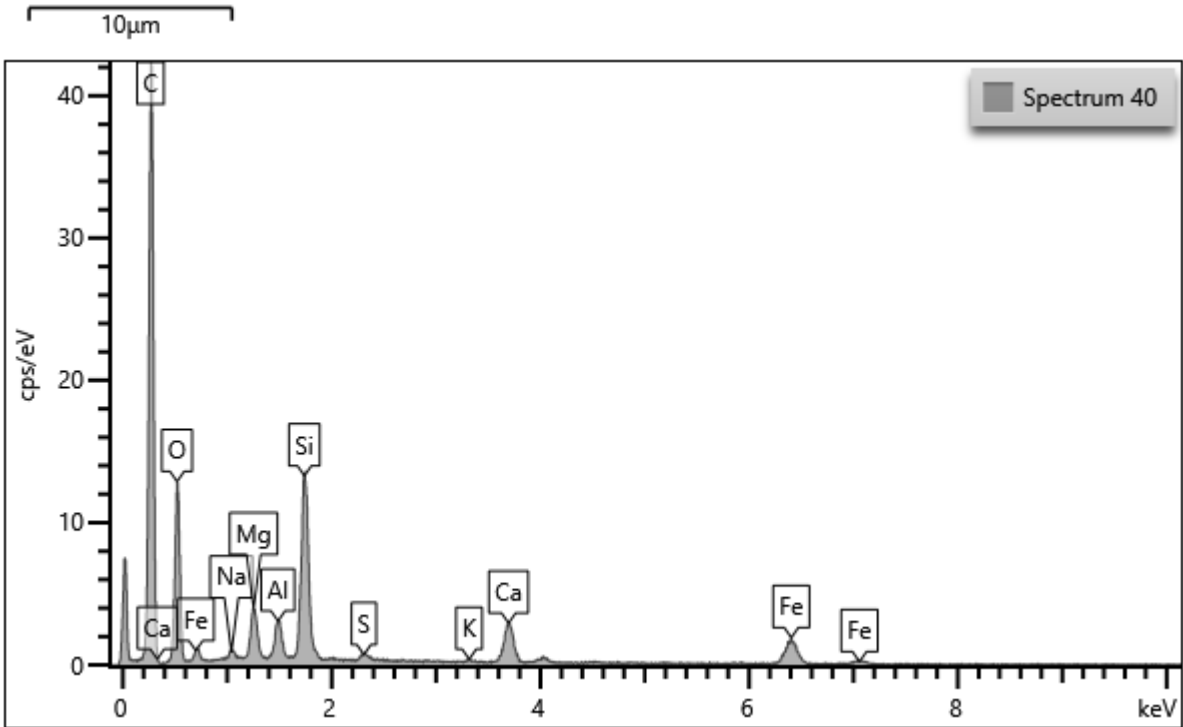
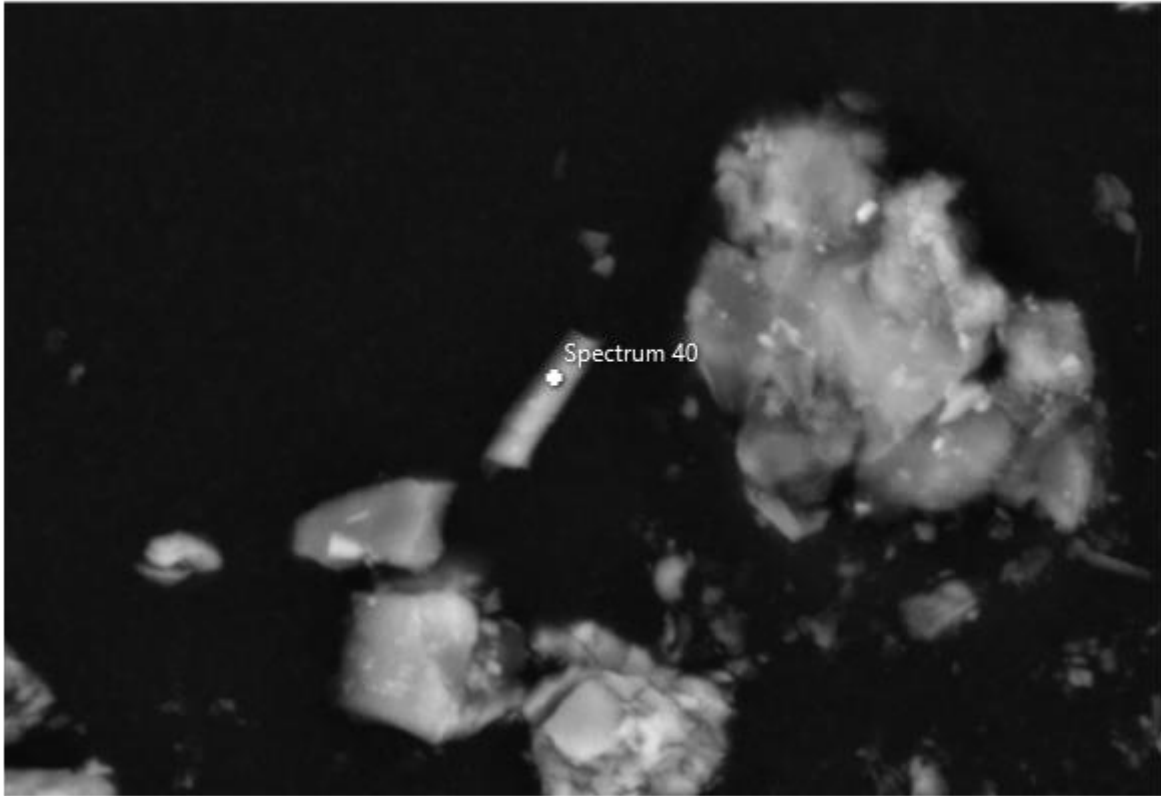
10µm



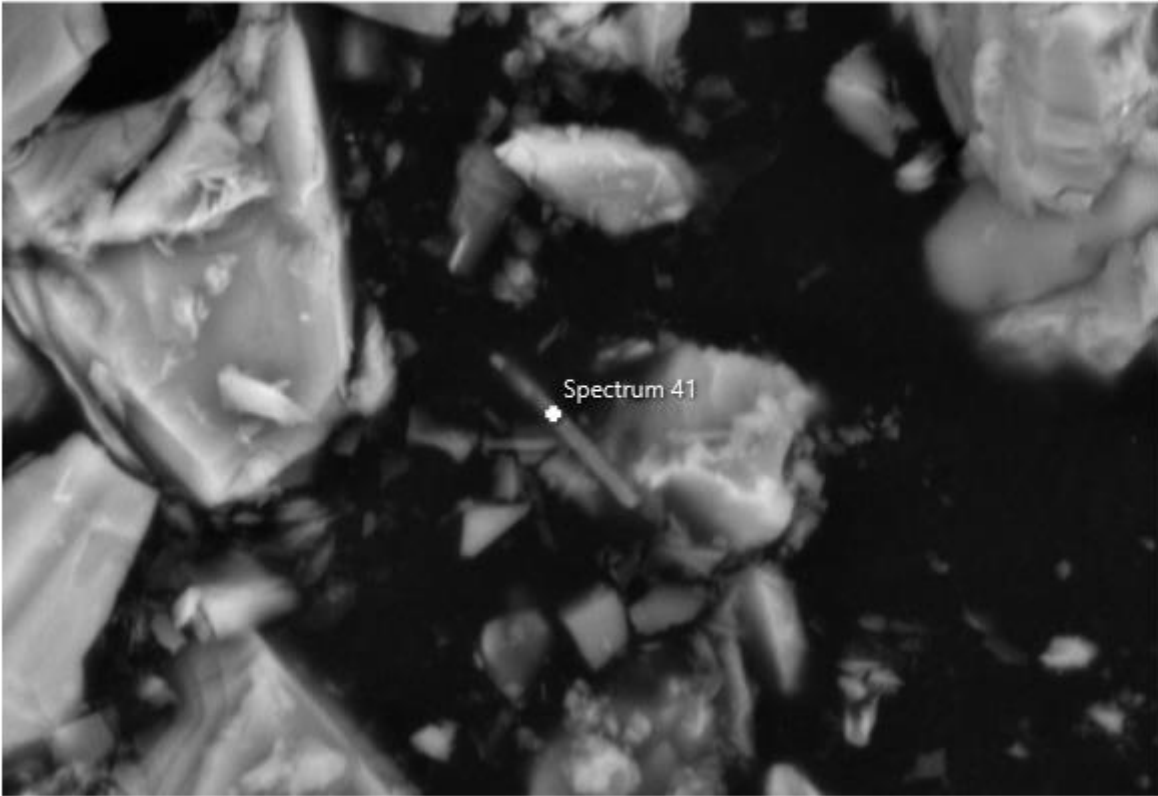
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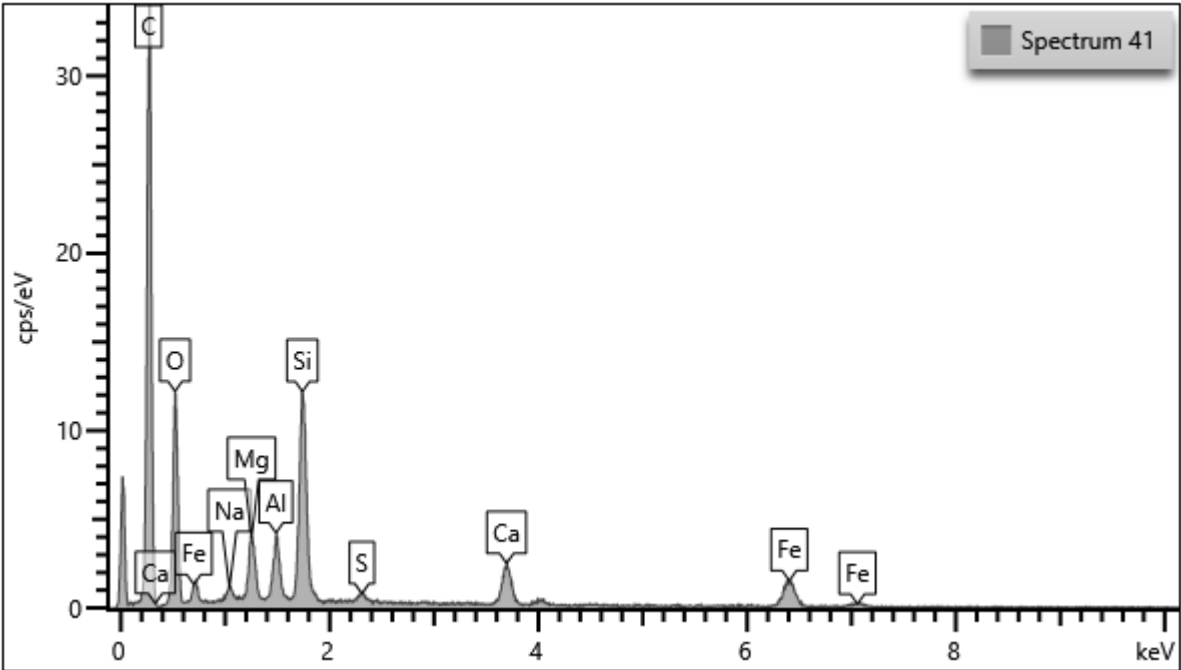
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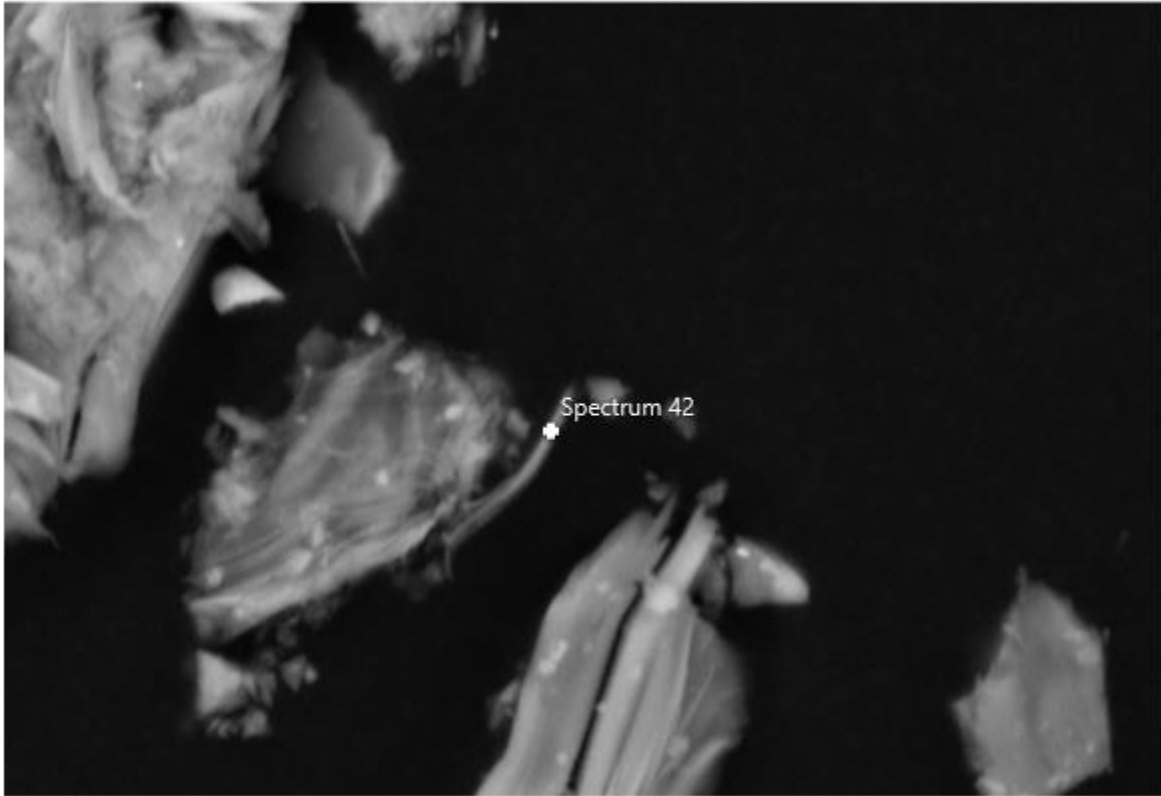
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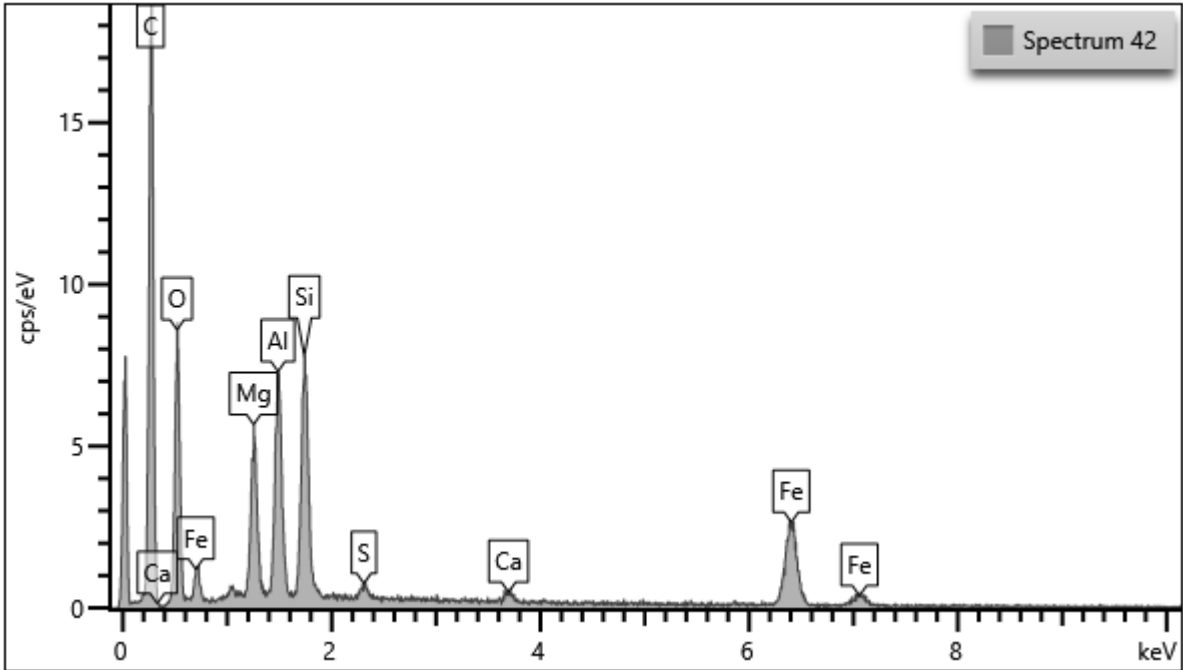
10µm



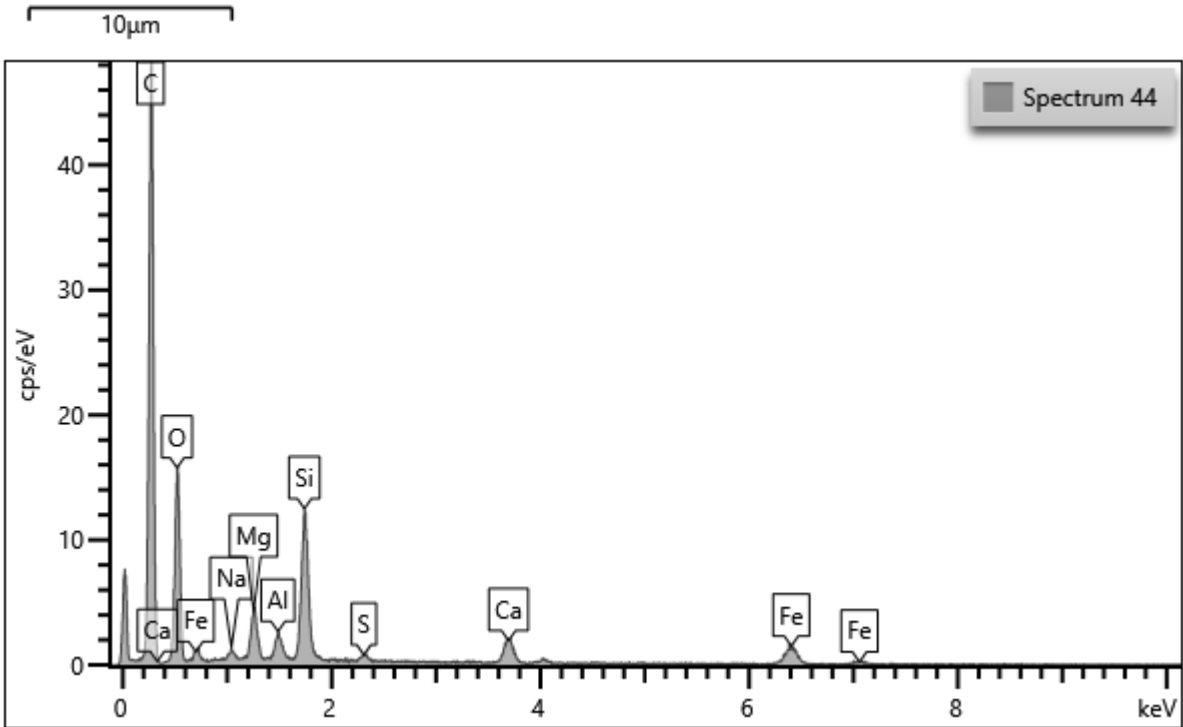
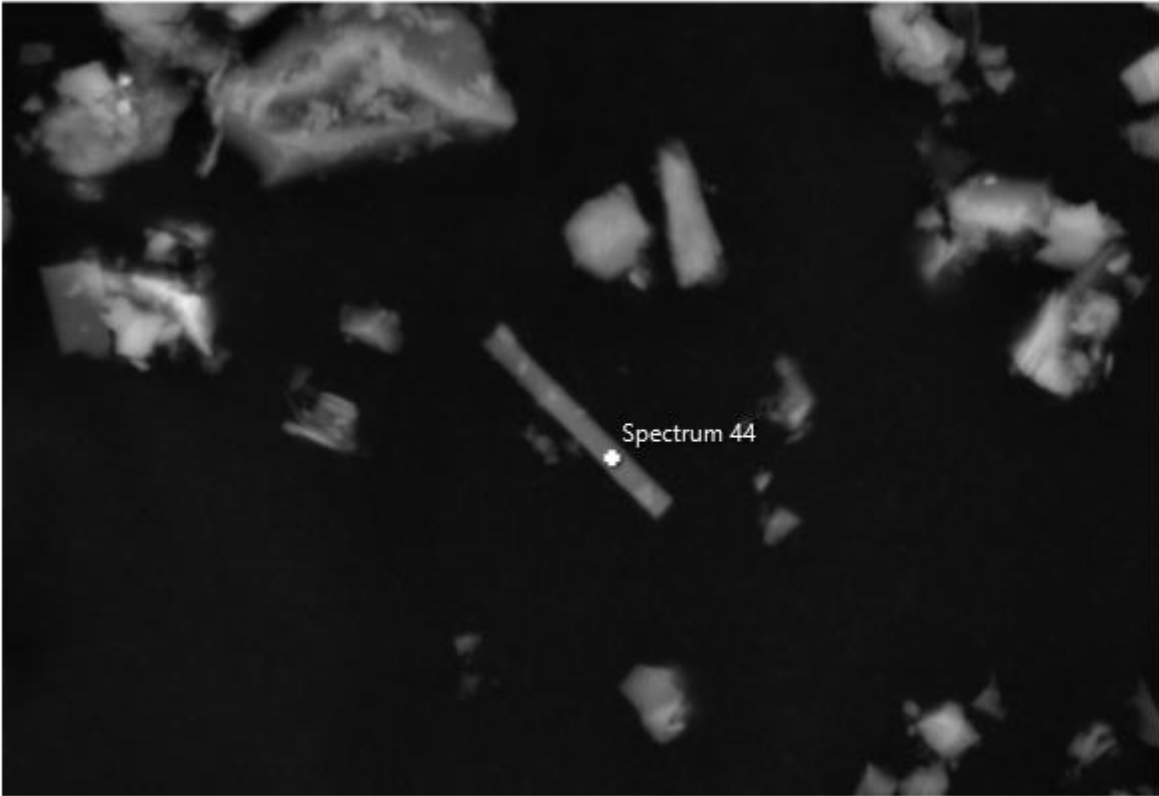
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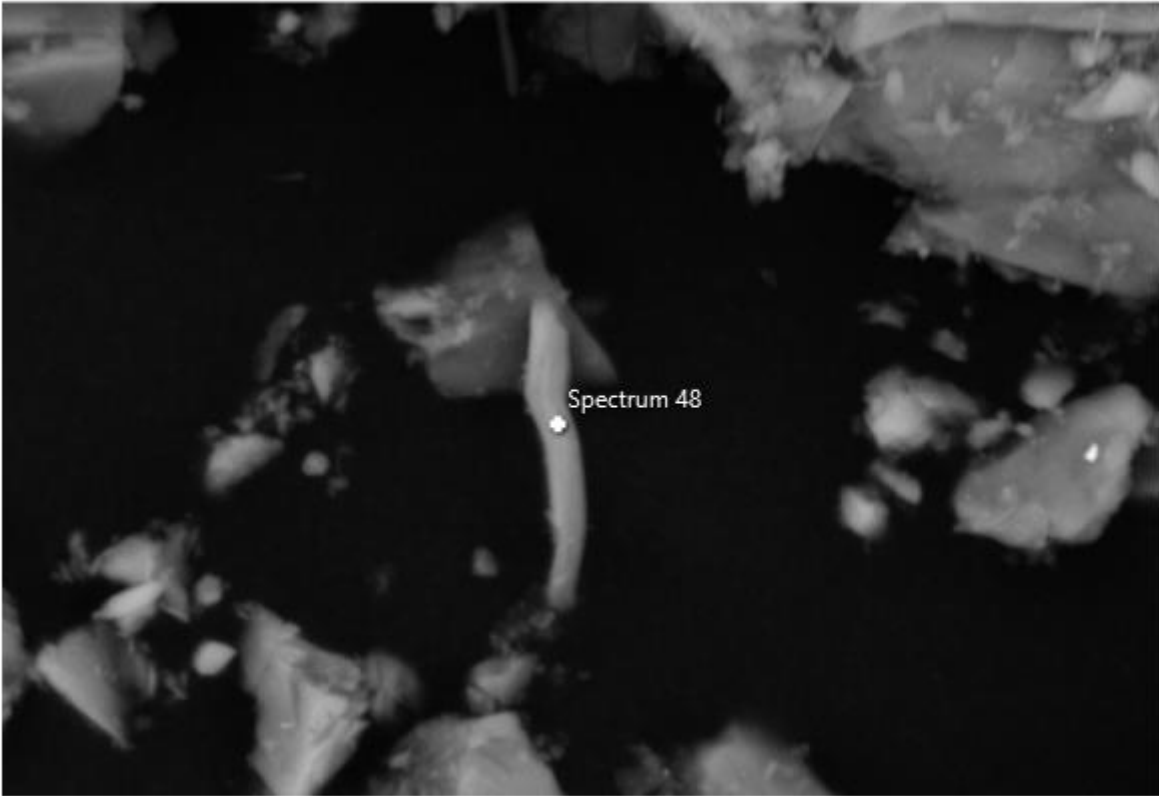
10µm



Electron Image 38 (SE)



Electron Image 41 (SE)



10µm

