







Pilbara Iron Ore and Infrastructure Project
Sediment Contaminants and Acid Sulphate Soils

March 2005





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1. Introduction

1.1 Objectives

The Public Environmental Review (PER) for the Pilbara Iron Ore and Infrastructure Project is currently out for public review. Fortescue Metals Group (FMG) were requested by the EPA to undertake some additional work prior to the EPA's assessment as follows:

- Undertake a sediment quality survey in accordance with guidance provided in the National Ocean Disposal Guidelines (Commonwealth of Australia, 2002) and an assessment of return water quality; and
- Undertake a comprehensive sampling program to determine the extent of Acid Sulphate Soils (ASS) in the FMG port project area, based on expert advice in accordance with the Department of the Environment (DoE) guidelines for "Identification and investigation of acid suphate soils and groundwater" (DEP/WRC, 2003) and the National Strategy for the Management of Coastal Acid Sulfate Soils.

1.2 Proposed works

1.2.1 Port facility

FMG's proposed port facility will be developed at Anderson Point on the south-western side of Port Hedland harbour. The final location and configuration for the facility has been selected on the basis of environmental investigations, engineering constraints and land access issues (Figure 1.1).

Dredging of the harbour will be required to accommodate the additional berths at Anderson Point. An area of approximately 36.8 ha requires dredging, with this area being divided into intertidal mudflats (17.8 ha) and subtidal substrate (19.0 ha). Dredging is proposed to a depth of 14.6 m to 19.5 m CD and this will require the removal of approximately 3,300,000 Mm³ of material.

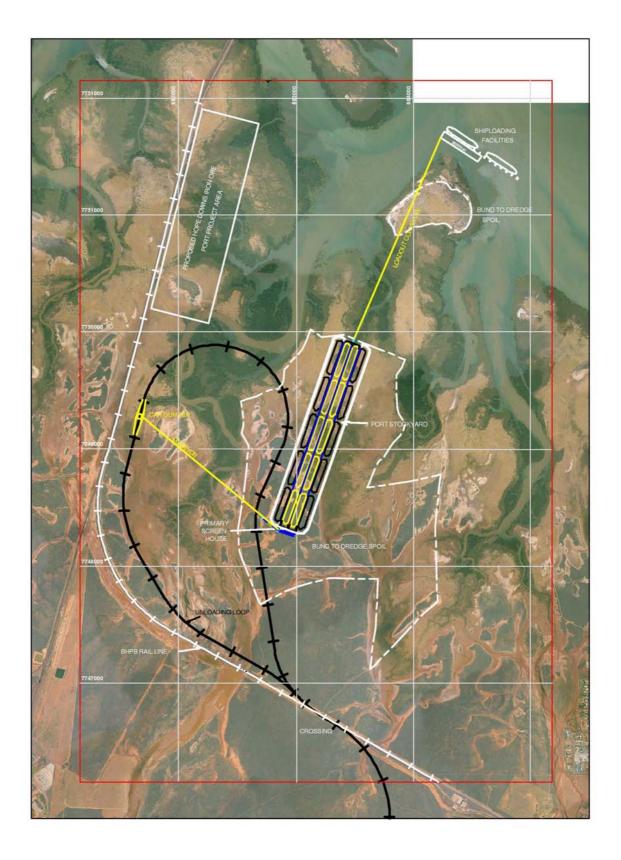


Figure 1.1 Configuration and location of proposed FMG port facility

Dredging works will be undertaken in four separate areas (Figure 1.2) as follows:

- 1. Area A: This area, located immediately north of the proposed berths, has been previously dredged to approximately -9.1 m Chart Datum (CD), and will require further dredging to approximately -14.6 m CD.
- 2. Area B: This area is located immediately east of the proposed berths and will require dredging from its existing level, at approximately +0.5 m CD to -9.1 m CD.
- 3. Area C: This area is located within the berthing pocket and will require dredging from its current level of 0.0 to +2.5 m CD, to a depth of -19.5 m CD.
- 4. Area D: This area is located immediately south of the berthing pocket and will require battered dredging from its current level to tie in with Area C at -19.5 m CD.

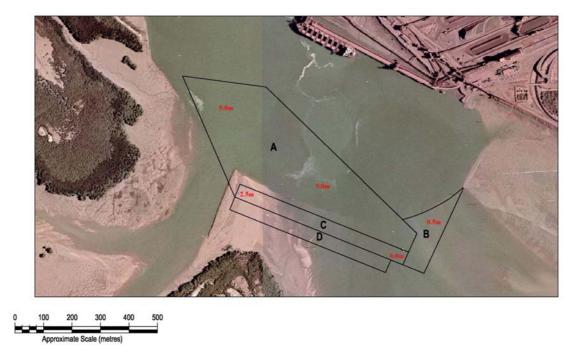


Figure 1.2 Fortescue Metals Group Port Dredging Areas with approximate depths in red (Source: Coffey Geosciences Pty Ltd, 2004a)

1.2.2 Ore Stockpile and conveyor

The stockpile areas are proposed to provide the necessary flexibility to ensure continuity of supply of iron ore to steel mill customers. The construction of a dredge spoil bund is planned, using sandy material from within the proposed stockpile (~40 ha) and sourced externally if required (Figure 1.3). These works will involve the drainage of the area and the disturbance of soils to a depth of approximately 2 m (FMG pers comm). Works at the Train Unloader site are anticipated to reach a depth of 15 m.

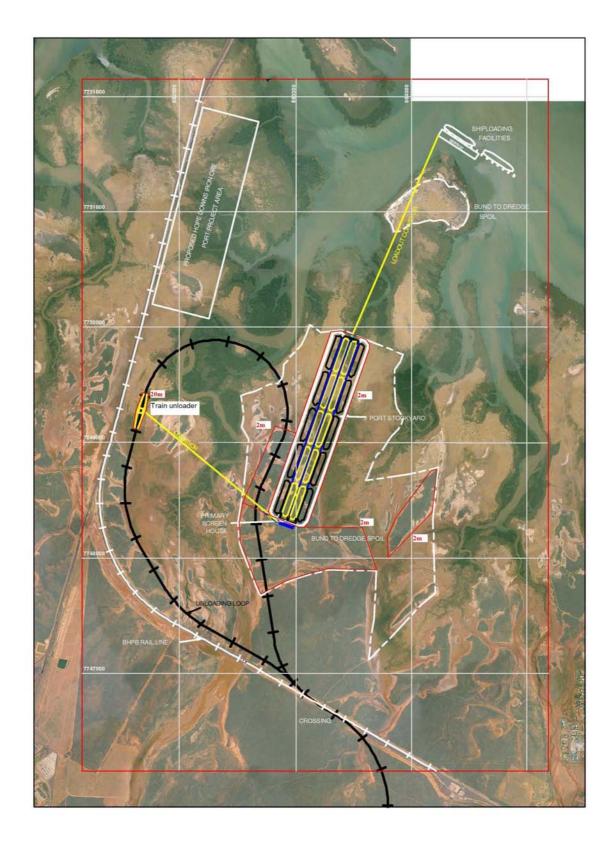


Figure 1.3 Proposed areas and depths of disturbance surrounding stockpile

2. Methods

2.1 Sediment sampling

2.1.1 Contaminants of concern

Recent sediment sampling work carried out within Port Hedland harbour suggested that surface sediments towards the centre of the turning basin (in the north-west of the proposed FMG dredging areas) are currently contaminated by Tributyltin (TBT) and nickel, with chromium, silver and cadmium levels also found to be close to the screening levels (or detection limits close to screening levels) (Oceanica, 2004; DALSE, 2004; URS, 2004). Much of this surface material has since been removed by maintenance dredging (PHPA, 2004), but these contaminants could potentially have become re-established as siltation of the basin recommences. In addition, following discussion with the EPA's Marine Branch, Arsenic, Mercury, Lead, Copper and Zinc were added to the contaminants of concern list having been recorded at concentrations approaching or exceeding the screening levels in sediments within Port Hedland harbour during a recent survey commissioned by BHP Billiton.

In addition to these analyses, selected samples were taken for the identification of acid sulphate soils (ASS) within the proposed dredging areas. Acid sulphate soils contain iron sulphides which, when exposed to air, produce sulphuric acid. They can be classed as either; actual acid sulphate soils which are already exhibiting acidity, or potential acid sulphate soils which have the potential to oxidise and produce acidity following drainage or excavation.

2.1.2 Site selection

The number of sampling sites required was calculated on the basis of the volume of soft sediment being dredged, using Table 2 of the Guidelines (Commonwealth of Australia, 2002), with the availability of recent sediment quality data from the proposed dredging areas allowing the requirements for sampling sites to be halved (Oceanica, 2004).



Figure 2.1 Anderson Point intertidal area

Sampling was undertaken at 10 sampling locations within the mid-basin subtidal area (MB), 10 within the Anderson Point subtidal area (AP) and 7 in the intertidal area (INT) (Figure 2.1; Figure 2.2; Figure 2.3) (total 27). A combination of benthic grabbing and hand coring was used to sample sediments to a maximum depth of 1.2 m.

The positions of these sampling locations were chosen using a grid to cover each area and the selection of squares using a random number table (Zar, 1984).

Sediment contaminant data from these locations adds to the data previously obtained (March 2004) from sites within the intertidal area of Anderson Point (Figure 2.4) (DALSE, 2004).





Figure 2.2: Sediment sampling locations (Hand-cores)



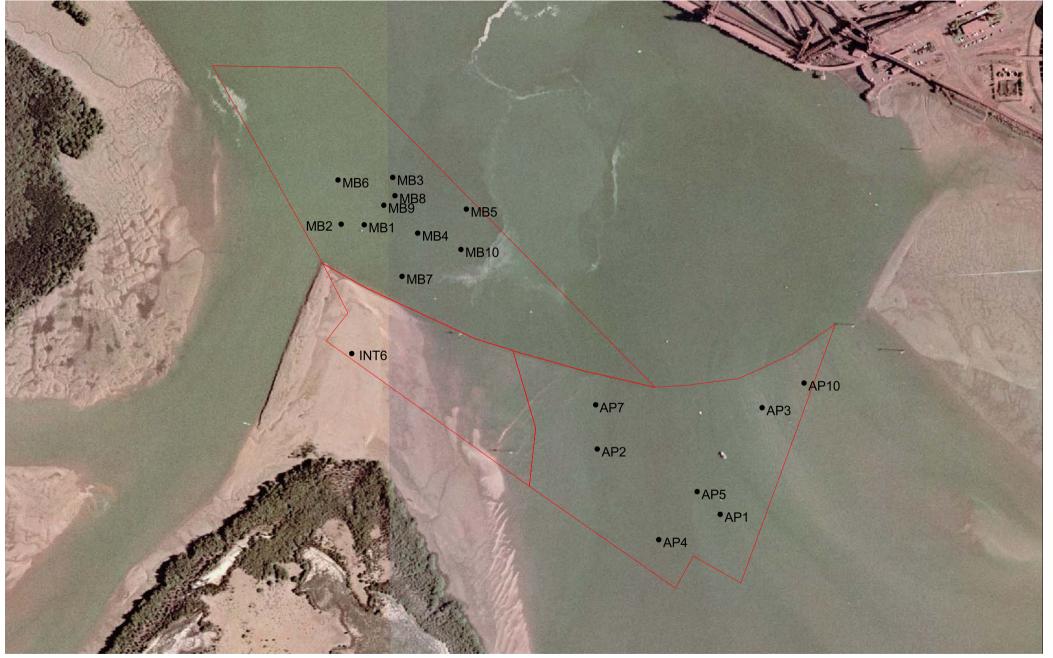




Figure 2.3: Surface sediment sampling locations (grabs/surface scrapes)









Figure 2.4: Recent surface sediment sampling sites at Anderson Point (DALSE,2004)



2.1.3 Sediment sampling methods

The collection of sediment contaminant data from the top 0.1 m of the sediment column was undertaken using grab sampling, with 17 locations within the subtidal areas being sampled (Figure 2.3). Sub-samples were taken from within the grab, taking care to avoid sediment in contact with the grab sides. Samples were obtained from the top 0.1 m of the sediment profile, including the ≤ 0.05 m deep silt layer identified at the top of the sediments. Surface samples were also obtained at site INT2 and INT6 to supplement the ten surface sediment samples previously taken within the intertidal area (DALSE, 2004).

Hand-coring, using a 50 mm diameter PVC pipe, was employed to obtain sediment samples from deeper than 0.1 m and determine the depth of any potential contamination, samples were collected at 12 sites (in water depths ≤1.0 m) (Figure 2.2). Sub-samples were taken from the bottom of the core tube using a clean plastic scoop. Samples were obtained from a depth of 0.30-0.50 m (core penetration 0.5 m) for sampling the top 0.5 m of the sediment, and from a depth of 0.80-1.0 m (core penetration 1.0 m) to sample the 0.5-1.0 m sediment depth layer. This method enabled the collection of sediment samples covering a range of sediment depths whilst avoiding cross-contamination. Sediment samples were stored on ice until they could be frozen on completion of the day's sampling.

2.1.4 Geotechnical coring

Although the depth of soft sediment, and therefore the sediment likely to exhibit contamination, is generally less than 1 m in the areas to be dredged (Coffey, 2004a), the guidelines and the EPA require the characterisation of the sediment throughout the column to be dredged. Therefore the collection of sediment samples from the geotechnical work has been carried out (Figure 2.5). A selection of the sediment core lengths collected were sub-sampled for sediment contaminants and acid sulphate soils alongside the sub-sampling for geotechnical testing (Table 2.1; Figure 2.6). Sediment sub-samples were taken from a range of sediment horizons covering the total sediment depth to be dredged, although many were composed of rock which underlies the surface sediments. These samples were considered to be additional to the hand grab and core samples detailed above. It was acknowledged that some contamination of the sediment core samples by the cellulose ether-based drilling muds was likely. This was not, however, expected to affect the concentrations of the contaminants of concern (Section 2.1.1) as none of these elements are present within the drilling muds (Australian Mud Company Ltd). Further, the results of any analyses are considered to be conservative as the collection by this method will not result in less contamination.

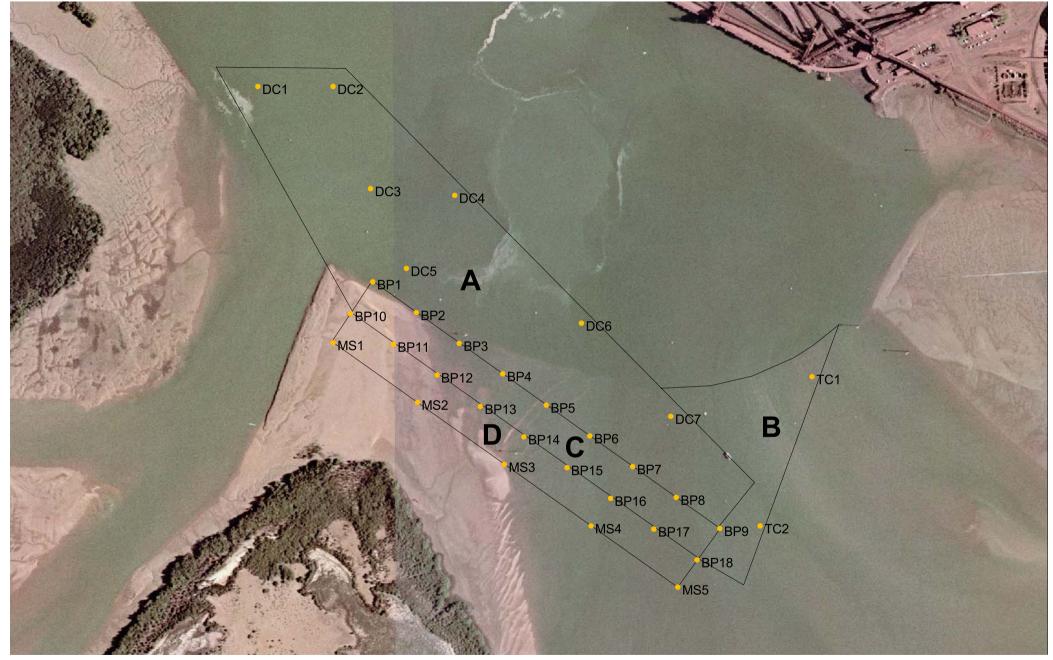




Figure 2.5: Location of proposed geotechnical core sampling sites (Coffey)



Table 2.1 Coffey geotechnical core samples taken for analysis

Hole ID	Depth of sample (m)	Sample taken for analysis						
BP5	1.4	ASS	ASS TBT					
	3.7	ASS	TBT					
	6.2	ASS	TBT					
BP7	0.0-0.5	ASS						
	1.5	ASS	TBT					
	3.8	ASS	TBT					
	5.2	ASS	TBT	METALS				
PB9	0.0-0.5	ASS	.5.	IIIZ I / IZO				
. 20	1.5	ASS	TBT	METALS				
	3.0	ASS	TBT	METALS				
	6.0	ASS	TBT	METALS				
	7.5	ASS	TBT	METALS				
	9.0	ASS	TBT	METALS				
	10.25	ASS	TBT	WETTE				
	14.8	ASS	TBT	METALS				
	16.3	ASS	TBT	METALS				
	17.8	ASS	TBT	METALS				
BP10	1.3	ASS	TBT	METALS				
טר וע	3.0	ASS	TBT	METALS				
	4.2	ASS	TBT	METALS				
	6.0	ASS	TBT	METALS				
	10	ASS	TBT	METALS				
	11.3-11.75	ASS	TBT	METALS				
	12.5	ASS	TBT	METALS				
	14.5	ASS	TBT	METALS				
	16.0	ASS	TBT	METALS				
	17.5	ASS	TBT	METALS				
DD4.4	19.0	ASS	TBT	METALS				
BP14	4.0	ASS	TBT	METALS				
BP18	1.5	ASS	TBT	METALO				
BP16	1.4	ASS	TBT	METALS				
	3.0	ASS	TBT	METALS				
D04	4.5	ASS	TBT	METALS				
DC1	0.0	ASS						
	1.8	ASS						
DO:	3.0	ASS						
DC4	0.3	ASS	TBT	METALS				
	1.0	ASS	TBT	METALS				
	3.0	ASS	TBT	METALS				
D.C.=	4.4	ASS	TBT	METALS				
DC7	0.3-0.75	ASS	TBT	METALS				
	1.5	ASS	TBT	METALS				
	3.0	ASS	TBT	METALS				
	4.5	ASS	TBT	METALS				
	7.5	ASS	TBT	METALS				
	12	ASS	TBT	METALS				
MS1	0.0-0.45	ASS						
	0.8-1.25	ASS						
	2.6-3.05	ASS						
	3.7-4.15	ASS						
	5.1-5.4	ASS						
MS2	1.5	ASS	TBT	METALS				
	3.0	ASS	TBT	METALS				
	4.5	ASS	TBT	METALS				
	6.0	ASS	TBT	METALS				

Hole ID	Depth of sample (m)	S	sample taken for	analysis
MS3	1.5	ASS	TBT	METALS
	3.0	ASS	TBT	METALS
	6.0	ASS	TBT	METALS
MS5	1.5	ASS	TBT	METALS
	3.25	ASS	TBT	METALS
	4.65		TBT	METALS
	5.95	ASS	TBT	METALS
TC1	0.0-0.45	ASS		
	2.0	ASS	TBT	METALS
	3.8	ASS	TBT	METALS
	5.0	ASS	TBT	METALS
	6.5	ASS	TBT	METALS
	8.0	ASS	TBT	METALS



Figure 2.6 Geotechnical core

2.1.5 Sample analysis

A selection of the hand coring and grab samples (>20% of the total) collected towards the beginning of the program were analysed initially in order to generally characterise the sediments and identify any contaminants present. Contaminants not detected could then be eliminated from the analysis list (Section 2.1.2 of Guidelines (Commonwealth of Australia, 2002).

The selection of samples for initial analysis for all potential contaminants included samples from each of the three areas identified (Figure 2.2; Figure 2.3) but was targeted towards sites and surface sediment layers considered to have the greatest chance of exhibiting contamination. In addition to these samples, a number were selected from deeper sediment layers to investigate the potential contamination depth.

Under guidance from the EPA marine branch, further sediment samples in excess of the recommended 20% were analysed for the full metals suite. These included samples obtained from the geotechnical survey work which enabled characterisation of the deeper sediment layers. Following exceedence of the screening levels, a number of samples were also tested to determine the bioavailability of nickel and chromium within the sediments.

2.2 Soil investigation

2.2.1 Sampling methods

Soil sampling was required within the proposed dredge spoil reclamation area, in five areas of excavation of approximately 16 ha (North), 52 ha (South), 18 ha (East), 2 ha (Train Unloader) and 40 ha (Iron Ore Stockpile) (Figure 2.8) for the identification of acid sulphate soils.

The latest guidelines (DoE, 2004a) suggest a minimum of two cores per hectare of disturbance (when extent of site project exceeds 4 ha) to provide sufficient sample coverage for ASS investigations (Table 1; DoE, 2004a). However, due to the remoteness and extent of the site requiring investigation, and following consultation with Dr Steve Appleyard (DoE/WRC), the number of sampling locations was reduced to 39 throughout the areas to be disturbed (Figure 2.8), with the use of hand auger sampling (Figure 2.7).



Figure 2.7 Hand augering to obtain soil samples to a depth of 2.6 m

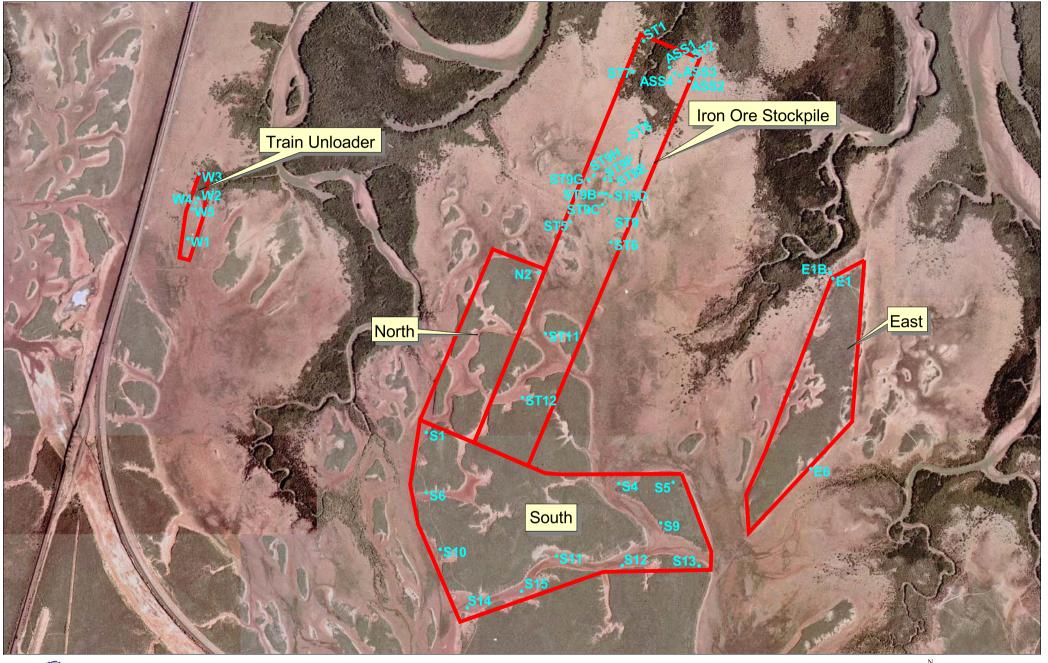




Figure 2.8: Acid Sulphate Soil sampling locations (Oceanica)



2.2.2 Field testing

Selected sub-samples of soil from the different horizons identified during the first two days of the survey were field tested for potential and actual acidity (pH_{Field} and pH_{FOX} tests). The results of this testing allowed the identification of any acid sulphate soils present. Following further consultation with Dr Steve Appleyard, the soil sampling programme was then adjusted to focus more closely on those areas exhibiting or expected to exhibit acid sulphate soils, in order to allow the future development of an acid sulphate soils management plan. This follows the staged approach as outlined in Ahern *et al.* (1998). The more targeted sampling sites were chosen following an examination of all the proposed excavation areas, with areas physically similar to those already sampled and not exhibiting acid sulphate soils given lowest priority for further sampling.

Sampling was carried out to a maximum depth of 2.6 m. The maximum depth of disturbance at all sites is 2 m, except at the Train Unloader site at which excavation to 15 m is proposed.

2.2.3 Laboratory testing

In addition to the sub-samples taken for field testing, a number were taken for submission for laboratory analysis using the chromium suite for the detailed description of ASS properties. The selection process was not carried out randomly, but with the aim of characterising each sediment horizon identified and obtaining the 'worst case' in terms of acidity (Dr Steve Appleyard pers comm.; Ahern *et al*, 1998). Results from the initial laboratory analyses will be discussed with DoE to determine the need for the laboratory analysis of further samples.

The chromium reducible sulphur suite method for the assessment of acid sulphate soils is due to become a published Australian Standard in late-2004/early-2005. The chromium reducible sulphur suite enables an accurate measure of the reduced inorganic sulphur compounds present within the sediment via a series of steps and provides a measure of the potential acid sulphate soil (Appendix A).

The first step in the chromium reducible sulphur method is the determination of the reduced inorganic sulphur content (SCR) which provides an estimate of the potential sulphuric acidity of the sediment. Following this, the soil pH, in a potassium chloride suspension (pHKCL), is determined as a means of estimating the actual acidity of the sediment. Depending upon these results (Appendix A), it may be necessary to analyse for Titratable Actual Acidity (TAA) to determine the actual acidity and/or analyse for the Net Acid Soluble Sulphur (SNAS) to estimate the retained acidity. The acid neutralising capacity (ANC) of the sediment provides an estimate of the ability of the soil to naturally neutralise any acid produced (for example due the presence of carbonate material).

3. Results

3.1 Sediment contaminants

Sediment samples were taken from a number of surface sediment layers (\leq 1.2 m depth) within the proposed dredging areas and the sediment type described (Table 3.1).

Table 3.1 Sediment sample description

Site	GDA94	zone 50	Sampling	Samples	Sadiment description
Site	Easting	Northing	depth (m)	taken	Sediment description
INT1	664538	7751515	0.3-0.5	Metals	Clean medium sand with pebbles
INT2	664368	7751657	0.0-0.1	ТВТ	Muddy coarse quartz sand with shell and stones. Water on surface
			~0.3-0.5	Metals	Muddy coarse quartz sand with shell and stones
			~0.6-0.7	ASS	Muddy coarse quartz sand with shell and stones
INT3	664425	7751600	0.3-0.5	Metals & TBT	Sandy mud
			0.8-1.0	Metals & TBT	Grey muddy medium/coarse sand
			1.2	ASS	Brown compact clay
INT4	664481	7751572	0.3-0.5	Metals & TBT	Compact red/brown clay
INT5	664368	7751685	~0.3-0.5	ASS	Muddy clay
			~0.5-0.7	Metals and TBT	Black medium/coarse sand.
INT6	664311	7751685	0.0-0.1	Metals and TBT	Muddy clay. Filamentous green algae on surface.
			~0.8-1.0	Metals, TBT & ASS	Muddy coarse sand
INT7	664283	7751798	~0.6-0.75	TBT	Clean coarse quartz sand.
			~0.6-0.75	Metals	Clean coarse quartz sand.
			~0.75-0.85	ASS	Clean coarse quartz sand.
AP1	664970	7751397	0.1	Metals & TBT	Muddy medium/coarse sand
AP2	664750	7751514	0.1	Metals, TBT & ASS	Muddy coarse/medium sand
AP3	665045	775188	0.1	Metals & TBT	Clean coarse/medium sand
AP4	664860	7751352	0.5-0.7	Metals & ASS	Not recorded-sample frozen
			0.1	Metals & TBT	Not recorded-sample frozen
AP5	664929	7751438	0.1	Metals & TBT	Dark grey mud with medium sand
AP6	664856	7751332	0.5-0.6	Metals & TBT	Red/brown compact clay
AP7	664747	7751593	0.1	Metals & TBT	Clean medium/coarse sand
AP8	664700	7751475	0.8-1.0	Metals & TBT	Clean grey coarse sand
AP9	664824	7751322	0.8-1.0	Metals, TBT & ASS	Grey muddy coarse sand with pebbles and shell
AP10	665120	7751632	0.8-1.0	Metals & ASS	Grey medium/coarse sand
			0.1	Metals & TBT	Muddy coarse sand

	GDA94	zone 50	Sampling	Samples	
Site	Easting	Northing	depth (m)	taken	Sediment description
MB1	664333	7751915	0.1	Metals & TBT	1-2 cm silt over muddy medium/coarse sand
MB2	664292	7751916	0.1	Metals, TBT & ASS	1-2 cm silt over muddy medium/coarse sand
MB3	664384	7752000	0.1	Metals, TBT & ASS	1 cm silt over muddy sand
MB4	664429	7751900	0.1	Metals & TBT	1 cm silt over 2 cm thick black mud overlying grey muddy clay
MB5	664516	7751943	0.1	Metals, TBT & ASS	1 cm brown silt over grey muddy clay
MB6	664286	7751995	0.1	Metals, TBT & ASS	1 cm silt over muddy medium sand
MB7	664401	7751823	0.1	Metals & TBT	1 cm silt over muddy medium/coarse dark grey sand
MB8	664388	7751967	0.1	Metals & TBT	0.5 cm brown silt over grey mud
MB9	664368	7751950	0.1	Metals & TBT	Grey mud/clay
MB10	664506	7751871	0.1	Metals, TBT & ASS	1 cm silt over dark grey/brown mud

Analysis of the initial samples showed that the concentration of all contaminants, with the exception of nickel, fell below the National Ocean Disposal Guideline Screening Levels (Commonwealth of Australia, 2002) (Table 3.2). Both the intertidal (INT) and Anderson Point subtidal (AP) areas were found to exhibit no contamination over Screening Levels.

Table 3.2 Sediment contaminant results (batch 1)

Site	Sampling Depth (m)		Concentration (mg/kg)									(μgSn/kg)			
		Ag	As	Cd	Cr	Cu	Ni	Pb	Zn	Hg	MBT	DBT	ТВТ		
Reporting Limit		<1	<1	<0.06	<0.2	<0.2	<0.4	<1	<0.5	<0.01	0.5	0.5	0.2		
Screening Level		1.0	20	1.5	80	65	21	50	200	0.15	n/a	n/a	5		
AP8	1.0	<1	5	<0.06	28	3.2	5.4	3	15	<0.01	0.6	1.7	<0.2		
AP10	0.1	<1	10	<0.06	21	3.8	7.5	3	19	<0.01	1.7	1.9	0.3		
AP10	1.0	<1	9	<0.06	24	3.7	8.2	2	8.3	<0.01	ı	-	-		
INT6	0.1	<1	12	<0.06	42	7.6	18	8	25	<0.01	8.0	0.6	0.2		
INT6	1.0	<1	11	<0.06	19	1.9	5.0	2	3.8	<0.01	#	#	#		
MB2	0.1	<1	10	<0.06	54	12	25	6	34	<0.01	1.0	2.7	1.5		
MB3	0.1	<1	12	<0.06	66	18	33	11	47	0.01	1.1	3.0	1.5		
MB10	0.1	<1	13	<0.06	64	15	31	9	45	0.02	<0.5	1.7	1.9		

Sample destroyed in transit
Screening level exceeded

A second selection of samples was subsequently sent for analysis for the full metals suite, with the nickel, chromium and arsenic concentrations recorded from some samples exceeding the screening levels (Table 3.3). It is of note, however, that the nickel and chromium levels within several of the deep samples (≥3 m below the sediment surface), considered to represent deep, consolidated and uncontaminated sediment layers, also exceeded the screening levels (Table 3.3). This supports the view that these elements naturally occur at relatively high levels within the region. The elevated concentrations at depth are also likely to be a result of the higher fines content of the sediments (for example sample BP10b was composed of medium/coarse sand whilst BP10c was composed of sandy clay (Coffey, unpublished)), since particle size is one of the dominant influences on contaminant levels in sediments (Commonwealth of Australia, 2002).

A dilute acid extraction of nickel and chromium was added to the analyses carried out on some of these samples to investigate the bioavailability of these elements (Section 3.10.4 of guidelines) (Section 3.5.5.2 of ANZECC/ARMCANZ, 2000), and showed that despite high concentrations being present within the Anderson point (AP) and intertidal (INT) sediments, only very low levels are likely to be bioavailable (less than screening levels) (Table 3.3).

As these studies indicate that the bioavailability of nickel and chromium is acceptably low (compared against the National Ocean Disposal Guideline Screening Levels (Commonwealth of Australia, 2002)), the sediments can be classified as nontoxic in terms of the nickel and chromium content (Commonwealth of Australia, 2002). A single exceedence of the arsenic screening level from a sample obtained at BP10b from a sediment depth of 3 m (36 mg/kg), was investigated, and repeat analysis of the sample gave a result of 11 mg/kg, below the screening level. A dilute acid extraction of arsenic from this sample showed this element to also be non-bioavailable (below screening levels) (Table 3.3).

The 95% UCLs of the surface sample results have been calculated (95% confidence limits of the mean calculated within STATISTICA Release 6) and compared to the Screening Levels (as per the guidelines) (Table 3.4). This analysis shows that within the surface (<0.1 m depth) sediments of the Anderson Point (AP) and mid-basin (MB) areas, only nickel levels within the MB area exceeded the screening levels. The 95% UCLs for the surface intertidal (INT) sediments were also determined (Table 3.5) and showed no metals to exceed the screening levels. Calculation of the 95% UCLs was not carried out on the deep (>0.1 m) sediment results because of the low number of samples analysed from comparable locations and depths.

Table 3.3 Sediment contaminant results (batch 2)

Site		Sampling Depth (m)		Concentration (mg/kg)										
			Ag	As	As (DAE*)	Cd	Cr	Cr (DAE*)	Cu	Ni	Ni (DAE*)	Pb	Zn	Hg
Reporting Limit			<1	<1	<1	<0.06	<0.2	<0.2	<0.2	<0.4	<0.4	<1	<0. 5	<0.01
Screening	Level		1.0	20	20	1.5	80	80	65	21	21	50	200	0.15
AP1		0.1	<1	6	-	<0.06	36	-	1.7	5.0	0.4	2	13	<0.01
AP4		0.1	<1	10	-	<0.06	36	-	7.4	14	1.0	5	29	<0.01
AP4		0.5-0.7	<1	7	-	<0.06	84	1.1	13	40	1.8	12	18	<0.01
AP7		0.1	<1	5	-	<0.06	17	-	1.4	2.9	<0.4	2	8.6	<0.01
BP9a		0.0-0.45	<1	4	-	<0.06	78	-	11	42	0.9	10	16	<0.01
BP9b	AP	1.5	<1	5	-	<0.06	82	ı	16	46	1.5	10	20	<0.01
BP9c		3.0	<1	3	-	<0.06	54	-	8.2	31	0.8	6	13	<0.01
BP10a		1.3	<1	14	-	<0.06	41	-	6.2	13	1.1	5	11	<0.01
BP10b	INT	3.0	<1	11	<1	<0.06	34	-	6.2	14	1.9	4	10	<0.01
BP10c		4.2	<1	9	-	<0.06	100	2.0	15	53	2.4	18	23	<0.01
BP10d		6.0	<1	8	-	<0.06	150	3.7	20	78	2.1	15	45	<0.01
MB1		0.1	-	-	-	-	-	-	-	-	1.1	-	-	-
MB2		0.1	-	-	-	-	-	-	1	-	1.2	-	-	-
MB3		0.1	-	-	-	-	-	-	ı	-	1.8	-	-	-
MB8		0.1	-	-	-	-	-	-	-	-	2.1	-	-	-
MB9		0.1	-	-	-	-	-	-	•	-	2.1	-	-	-
MB10		0.1	-	-	-	-	-	-	-	-	1.7	-	-	-

* Dilute acid extraction (DAE)

Screening level exceeded

DAE showing element not to be bioavailable at this site/depth

Table 3.4 Comparison of 95% UCLs of surface samples against the screening levels

Site	Sampling Depth (m)		Concentration (mg/kg)								
		Ag	As	Cd	Cr	Cu	Ni	Pb	Zn	Hg	ТВТ
Reporting Limit		<1	<1	<0.0 6	<0.2	<0.2	<0.4	<1	<0.5	<0.01	0.2
Screening Level		1.0	20	1.5	80	65	21	50	200	0.15	5
AP1	0.1	<1	6	<0.06	36	1.7	5.0	2	13	<0.01	-
AP4	0.1	<1	10	<0.06	36	7.4	14	5	29	<0.01	-
AP7	0.1	<1	5	<0.06	17	1.4	2.9	2	8.6	<0.01	-
AP10	0.1	<1	10	<0.06	21	3.8	7.5	3	19	<0.01	-
95% UCL		n/a	12	n/a	43	8.0	15.0	5	31.5	n/a	-
MB2	0.1	<1	10	<0.06	54	12	25	6	34	<0.01	1.5
MB3	0.1	<1	12	<0.06	66	18	33	11	47	0.01	1.5
MB10	0.1	<1	13	<0.06	64	15	31	9	45	0.02	1.9
95% UCL		n/a	15	n/a	77	22	40	15	59	0.03	2.2

Screening level exceeded

The surface sediments of the intertidal area can be considered 'clean' as none of the metals concentrations exceeded the Screening Levels (Table 3.2; Table 3.5).

Table 3.5 Sediment contaminant concentrations recorded from Anderson Point in March 2004 (DALSE, 2004)

	Contaminant Levels recorded							
Site	As	Cd	Со	Cr	Cu	Ni	Pb	Zn
		(mg/kg)						
Screening Level	20	1.5	NA	80	65	21	50	200
1	5	<0.06	0.7	14	0.3	1.2	<1	4.9
2	8	<0.06	2.1	23	2.4	5.9	2	12
3	7	<0.06	1.5	18	1.7	4.2	1	11
4	6	<0.06	0.7	13	0.5	1.4	<1	5.8
11	9	<0.06	1.6	19	1.6	4.5	1	8.2
12	7	<0.06	1.8	21	2.2	4.9	2	9.8
95% UCL	8	n/a	2.0	22	2.4	5.7	2	12

The mid-basin area has recently been dredged (PHPA, 2004). It is likely that any contamination previously present within the surface sediments of the mid-basin area (MB) was removed during the recent maintenance dredging operation (PHPA, 2004). The persistence of high nickel levels suggests an active source within the area.

3.2 Acid Sulphate Soils

3.2.1 Sediments – Port Facility

A small selection of hand core samples taken from within the Anderson Point (AP) and Intertidal (INT) proposed dredging areas (Figure 2.2) were analysed for acid sulphate soils (Table 3.6). Although the sediments did not exhibit the characteristics of acid sulphate soils, the samples selected for analysis were those thought to be most likely to contain ASS following an examination of their physical properties.

As noted above (Section 2.2.3), the actual *in situ* acidity may be determined from the pH_{KCL} values. For all samples analysed, the pH_{KCL} values were greater than 9 and indicate none of the samples taken were actually acidic (Table 3.6).

One of the chromium reducible sulphur values (%S_{CR}) exceeded the values of the Action Criteria (0.03%)(DoE, 2004a), indicating the presence of slight potential acidity at this site at a depth of 1.2 m. Due to limited hand core penetration, sampling to this depth was not carried out at any other sites by this method. The geotechnical work has yielded samples from approximately this depth (Table 2.1), and comprehensive core logs from this work describe the nature of sediments at each core location.

Table 3.6 Acid sulphate soils laboratory results for dredging area samples

Site	Sediment depth (m)	pH _{KCL}	Action Criteria for soils* (%S)	%S (S _{CR})	ANC _{BT} (% CaCO ₃)
AP4	0.5-0.7	9.5	0.03	<0.01	0.9
INT3	1.2	9.5	0.03	0.08	8.2
INT7	0.75-0.84	9.9	0.03	<0.01	3.7

^{*} Action Criteria for disturbance of >1000 tonnes of soils (DoE, 2004a)

Action criteria exceeded

To determine whether acid sulphate soils were likely to be present within the deeper sediment layers, field testing was carried out on a small selection of samples taken during the Coffey geotechnical study (Table 3.7).

Table 3.7 Acid sulphate soils field testing results for dredging areas

Hole ID	Depth of sample (m)	pH Field	Reaction	pH FOX
AP10	1.0	8.71	Low	6.63
BP9	0.0-0.45	9.19	Low	7.21
BP9	1.5	8.73	None	6.79
BP9	3.0	8.23	None	6.67
BP9	6.0	8.35	None	6.47
BP10	1.3	8.17	None	6.67
BP10	3.0	7.94	None	6.41
BP10	4.2	8.62	None	7.06
BP10	6.0	8.63	Low	6.64
MS2	1.5	9.00	Low	6.70

Results interpretation:

- $pH_F < 4 AASS$ present
- $pH_{FOX} < 3 strongly indicates PASS$
- Change in pH (pH_F \rightarrow pH_{FOX}) \geq 2.5 suggests PASS
- Volcanic or high reaction indicates likely ASS but other constituents e.g. organic matter can cause reaction.

None of the deeper samples field tested were found to exhibit actual or potential acidity. Examination of the core log from MS2, approximately 15 m WNW of INT3, revealed a sediment layer composed of a 'clayey sand, dense/very stiff, grey brown, medium grained quartz, low plasticity material' (Coffey, unpublished) to lie between 1.14 m and 1.50 m below the sediment surface. A sample from 1.5 m at this site, but composed of brown clayey sand and not the grey material more likely to be ASS, was field tested in the laboratory and did not produce the drop a pH indicative of ASS (Table 3.7). A similar sediment layer was not recorded from any adjacent core locations (MS1, MS3, BP1, BP3, BP5, BP10, BP12, BP14) (Coffey, unpublished) so the ASS material is thought to be extremely limited in its horizontal extent, likely to extend inshore from MS2 but not far within the proposed dredging areas. The very limited vertical extent of the ASS layer also suggests that the amount of this material within the areas to be dredged is relatively small. This material contains a high proportion of calcium carbonate (Table 3.6), as does the surrounding material

(Coffey, unpublished), so it is likely that any acid produced following excavation and subsequent disposal would be rapidly neutralised.

To investigate the availability and reactivity of the carbonate within the material to be dredged, two bulked samples from BP5 (samples from 1.4 m, 3.7 m and 6.2 m combined) and BP16 (samples from 1.4 m, 3.0 m and 4.5 m combined) were tested for ANC and for their pH-buffering properties (acidimetric auto-titration). It was found that sample BP16 contained a high proportion of reactive carbonate minerals and would be effective in the near-neutral buffering of acidity produced from any of the PASS layers. Sample BP5 was found to contain little carbonate and to be ineffective as a buffer (Appendix F).

3.2.2 Soils – Iron Ore Stockpile and Conveyor

Soil layers identified from the 39 hand auger sites (Figure 2.8) were described and selectively sampled for acid sulphate soils (Appendix B). The high relief vegetated areas within the proposed stockpile, and within the areas to be used for sourcing bund material ('North', 'South' and 'East') (Figure 2.8), were found to consist of dry sandy soil overlying red clay. A limestone layer was also identified at some locations (Appendix C). The creek beds were found to be composed of red sand and clay in varying proportions (Appendix B). Soils within low-lying areas within the Train Unloader and northern stockpile areas were found to consist of brown muds and clays overlying red clay, with an intermediate grey clay layer identified at some sites. The presence of a grey clay layer, identified as likely to contain potential acid sulphate soils (Table 3.8), seemed to be related to the proximity of the site to mangroves, although some areas separated from mangroves but distinguished by numerous crab burrows also exhibited this soil layer.

Acid sulphate soils field testing

Selected sub-samples of soil from the different horizons identified during the first two days of the survey were field tested for potential acid sulphate soils (PASS) and actual acid sulphate soils (AASS) (pH $_{\rm Field}$ and pH $_{\rm FOX}$ tests) (Table 3.8). The results of this testing allowed the identification of any acid sulphate soils present and the focusing of the remaining investigation on those areas anticipated to contain acid sulphate soils.

Table 3.8 Acid sulphate soils field testing results for stockpile and conveyor sites

Sample	Soil depth (m)	pH Field	Reaction	pH FOX
S4	0.9	8.04	High/volcanic	7.8
S11	0.5	8.19	Low	6.88
S11	1.0	8.26	None	6.93
S11	1.8	8.28	Low	6.81
S12	0.4	8.12	Low	5.79
W2	0.3	8.08	Medium (delayed)	6.95
W2	1.0	7.69	Medium	4.96
W2	1.2	7.84	Volcanic	1.49
W2	1.4	7.93	Volcanic	2.16
W2	1.8	8.16	None	6.37
ST1	0.75	7.46	Volcanic	1.48
ST1	1.0	7.55	Volcanic	1.65
ST1	1.1	7.65	Low	6.29
ST8	0.4	7.67	None	6.35

Results interpretation:

- $pH_F < 4 AASS present$
- $pH_{FOX} < 3 strongly indicates PASS$
- Change in pH (pH_F \rightarrow pH_{FOX}) \geq 3 suggests PASS
- Volcanic or high reaction indicates likely ASS but other constituents e.g. organic matter can cause reaction.

Likely potential ASS

Acid sulphate soils laboratory testing

For all samples analysed, the pH_{KCL} values were greater than 7 and indicate none of the samples taken were actually acidic (Table 3.9).

Six of the chromium reducible sulphur values ($\%S_{CR}$) exceeded the values of the Action Criteria (0.03%)(DoE, 2004a). Hence, the chromium method confirms that samples ST1 (0.75), ST2 (1.0), W2 (1.0), W2 (1.2), W2 (1.5) and W5 (2.0) were potential acid sulphate soils (PASS) (Table 3.9).

Table 3.9 Acid sulphate soils laboratory results for stockpile and conveyor sites

Site	Soil depth (m)	pH _{KCL}	Action Criteria for soils* (%S)	%S (S _{CR})	ANC _{BT} (% CaCO ₃)
N2	0.5	8.5	0.03	<0.01	0.2
N2	2.0	8.2	0.03	<0.01	0.1
S6	1.0	7.9	0.03	<0.01	0.0
S6	2.5	7.8	0.03	<0.01	0.1
S15	0.5	8.5	0.03	<0.01	0.3
S15	1.0	8.0	0.03	<0.01	0.1
S15	1.8	8.2	0.03	<0.01	0.1
ST1	0.75	7.9	0.03	0.76	0.4
ST1	1.1	8.4	0.03	0.03	0.2
ST2	1.0	8.5	0.03	0.34	0.5
ST9	0.3	9.4	0.03	<0.01	8.4
ST12	0.5	8.2	0.03	<0.01	0.4
ST12	1.0	8.1	0.03	<0.01	0.4
W2	1.0	8.9	0.03	0.36	1.4
W2	1.2	8.6	0.03	3.90	7.0
W2	1.5	9.4	0.03	0.07	2.7
W5	1.5	9.1	0.03	1.10	9.5

^{*} Action Criteria for disturbance of >1000 tonnes of soils (DoE, 2004a)

Action criteria exceeded

The acid neutralising capacity (ANC) of the soil layers indicates their ability to neutralise any acid produced. Several of the sediment layers tested exhibited relatively high ANCs (>7% CaCO₃) (Table 3.9). This neutralising capacity can be compared against the potential acidity (mol H⁺/tonne) to determine whether effective neutralisation is likely to occur following the oxidation of the soil (Table 3.10).

Table 3.10 Acid neutralising capacity compared against potential acidity

Site	Soil depth (m)	%S (S _{CR})	Equivalent acidity (mol H ⁺ /tonne)	ANC _{BT} (% CaCO ₃)	ANC (mol H ⁺ /tonne)	Net acidity (acidity-ANC) (mol H ⁺ /tonne)*
ST1	0.75	0.76	474.0	0.4	79.9	+394.1
ST2	1.0	0.34	212.1	0.5	99.9	+112.2
W2	1.0	0.36	224.5	1.4	279.7	-55.2
W2	1.2	3.90	2432.4	7.0	1398.6	+1033.8
W2	1.5	0.07	43.7	2.7	539.5	-495.8
W5	1.5	1.10	686.1	9.5	1898.1	-1212.0

^{*} Positive value = excess acid, negative value = excess neutralising capacity

The comparison of relative potential acidity and acid neutralising capacity for each soil layer (Table 3.10) shows three of the sediment layers exhibiting potential acidity (W2 1.0 and W5 1.5) to be effectively buffered by the other components of the soil. The excess neutralising capacity within the soils at W2 (1.5) and W5 (1.5) also exceeds the recommended safety factor (addition of 1.5 times the required neutralising agent) when calculating the neutralising requirements (DoE, 2004a).

4. Conclusions

4.1 Sediment contamination

The surface (≤ 1 m deep) sediment layers within the areas to be dredged have been found to be uncontaminated by metals or TBT with the exception of nickel and chromium.

High concentrations of nickel and chromium have previously been recorded within marine sediments in the Port Hedland region, often exceeding the screening levels (ENV, 2001; URS, 2003), even at sites remote from the harbour (URS, 2004). This suggests that high background levels, possibly a result of the weathering of terrestrial rock, exist in this region. The EPA, however, requested that further testing be carried out on nickel within the sediments to be dredged, as per the guidelines.

Having recorded nickel levels within the mid-basin surface sediments that exceed the screening levels, bioavailability testing was carried out. These studies indicated that the bioavailability of nickel was acceptably low (compared against the National Ocean Disposal Guideline Screening Levels (Commonwealth of Australia, 2002)), and therefore the spoil can be classified as non-toxic (Commonwealth of Australia, 2002). Similarly, bioavailability testing of chromium was carried out on a small number of the INT and AP sediment samples following exceedence of the screening levels during analysis of the second batch of samples, with chromium found to be non-bioavailable.

4.2 Acid sulphate soils

Acid sulphate soils have been identified within the proposed Stockpile and Train Unloader areas. The material consists of a readily visually identifiable grey clay layer which is generally in close association with mangroves. The layer was found to have an upper depth limit of 0.5 to 1.0 m.

For projects that disturb ≥ 1000 tonnes of ASS with ≥ 0.03 %S, a detailed management plan (ASSMP) is required (DoE, 2004a). The results suggest that the grey clay material identified within the northern Stockpile area, and within the northern Train Unloader area, exceeds both the acidity and weight criteria for the requirement of a detailed ASSMP covering both these areas. FMG will produce an ASSMP prior to the commencement of excavation. Estimation of the volume of the acid sulphate soils in each area, together with consideration of the natural acid neutralising capacity of the different sediment/soil layers, will be fundamental to the management options.

Acid sulphate soils were recorded from one site (INT3) and suggested at another site (MS2) within the proposed dredging areas during the marine component of the acid sulphate soils investigation. However, following an examination of the geotechnical core logs, it was found that this material, composed of a clayey grey brown sand (Coffey, unpublished), was not present within any adjacent cores. In addition, this material was found to lie between 1.14 m and 1.50 m below the sediment surface. Thus the ASS material is thought to be extremely limited in its horizontal and vertical extents, suggesting that the amount of this material within the areas to be dredged is relatively small. This material is not highly acidic (upon oxidation) and contains, as do the surrounding sediment layers, a high proportion of calcium carbonate, so it is likely that any acid produced following excavation would be rapidly neutralised. Tests on the availability and reactivity of the calcium carbonate within the sediments have shown this material to be reactive and available for the neutralisation of any acid produced.

5. References

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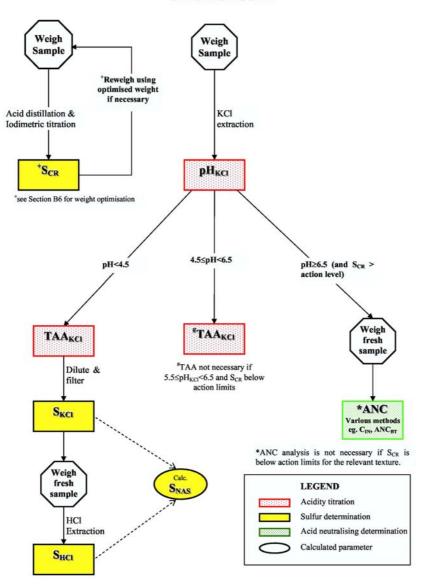
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Appendix A Chromium Suite – ASS Analysis

Appendix A Chromium suite - ASS analysis

CHROMIUM SUITE



Appendix B Acid sulphate soil investigation notes

Appendix B Acid sulphate soil investigation notes

Site	Easting	Northing	Sediment depth (m)	Samples taken	Sediment description	Sampling details
						1m penetration. Full core
ASS4	663537	7749814	0.0-1.0	Core		taken.
			0.0-0.3		Light brown muddy medium sand	
			0.40		As above + grey mud marbling	
			0.50		As above plus grey clay	
			0.60		Grey clay	
			1.0-1.1		End of grey clay, beginning red clay	
ASS3	663556	7749806	0.0-0.3		Muddy light brown medium sand	
			0.40		As above with grey marbling	
			0.50		As above. Water table.	
			0.60		Red muddy sand	
			0.70		Red clay with sand	
			0.80		Red clay with some medium sand	
ASS2	663593	7749783	0.00	1	Compact brown sandy clay	
					Slight grey clay	
			0.60		marbling	
			0.70		As above Red/brown mud with	
			0.75		medium sand	
			0.05		Red/brown clay with	
			0.85 1.10		medium sand Red clay. Water table.	
			1.10		Neu ciay. Water table.	
	000540					1m penetration. Compressed to 0.7m. Recovered bottom
ASS1	663516	7749837	0.0-1.0	Core	Brown/red wet clay with	0.5m.
			0.0-0.5		grey clay marbling	
			0.55	1	Black organic material with grey clay	
			0.70		Grey clay	
			1.50		Dark grey muddy coarse sand.	
E1	664440	7749021	0.0055		Light brown muddy sand	
EI	664149	1173021	0.0-0.55		As above with slight	
			0.55		grey clay marbling	
			0.60		Red clay	
E1B	664138	7749042	0.0-0.4		Light brown muddy sand	
			0.50		Red/brown clay	
E 6	664063	7748284	0.0-0.3		Brown clay with medium sand	
			0.3-0.5		As above plus grey clay marbling	

Site	Easting	Northing	Sediment depth (m)	Samples taken	Sediment description	Sampling details
			0.5-1.0		Sandy red/brown clay	
			1.10		Dry sandy brown clay with pebbles	
			1.1-1.3		Brown/red clay with some coarse sand	
S1	662575	7748424	0.0-0.5	1	Clay with medium sand	
S4	663320	7748227	0.20	1	Clay with sand	
			0.50	1	Wet coarse sand and clay	
			0.90	Field	As above	
S5	663530	7748232	0.0-0.8		Dry sandy soil	
			0.8-1.0	1	Sandstone fragments, clay and medium sand	
			1.00	'	Dry soil with sand	
S6	662575	7748192	0.0-0.2		Medium sand Thick clay with medium	
			0.2-2.3	3	sand	
			2.30	1	Wet mud with sand	
S9	663482	7748074	0.0-0.8	1	Dry sandy soil	
			0.8-1.0	1	Moist sandy soil	
S10	662628	7747973	0.0-1.0		Sandy dry soil	
				2 &	Brown medium sand	
S11	663079	7747947	0.0-1.5	2xField	with clay Grey clay with coarse	
			1.5-1.9	2 & Field	sand	
			1.90	1	Red clay	
S12	663333	7747909	0.0-0.08	1	Muddy coarse sand	
			0.08-0.4	Field	Red clay with coarse sand	
			0.4-1.0	2	Red clay with coarse sand and pebbles	
S13	663631	7747908	0.0-0.6		Dry sandy soil	
0.0	000001	77 11 000	0.6-1.0	2	Sandy moist soil	
			1.0-1.9	1	Red clay with medium sand	
			1.90	1	Coarse sand with pebbles and clay	
S14	662733	7747746	0.0-0.5	2	Coarse red muddy sand	
314	002100	1141140	0.5-0.6	1	Coarse red sand with clay	
			0.6-1.1	·	Wet red clay with sand	
					01: "	
S15	662943	7747811	0.0-0.5		Clay with some medium sand	
			0.5-1.0	1	As above	
			1.0-1.6 1.6-1.8	1	Wet clay with sand Coarse sand with mud	
				·		
ST1	663416	7749939	0.0-0.65		Brown mud with medium sand	
			0.65-1.1	2 &	Grey clay/mud	

Site	Easting	Northing	Sediment depth (m)	Samples taken	Sediment description	Sampling details
			4.4.0.0	2xField	5 1 11 1	
			1.1-2.0	1 & Field	Brown clay with sand	
ST2	663604	7749865	0.0-0.9		Light brown sandy clay	
012	000004	1143000	0.9-1.55	1	Grey clay	
			1.55	•	Red clay	
					•	
ST5	663131	7749240	0.00		Brown sandy clay	
			0.05		As above with grey clay	
			0.25 0.30		marbling Red clay	
			0.30		Neu clay	
ST6	663290	7749161	0.0-0.25		Light brown muddy medium sand	
			0.25		Red clay	
ST7	663375	7749821	0.0-0.7		Sandy mud	
017	000070	77 10021	0.7-0.75		Grey clay	
					Brown mud with	
			0.75-1.0		medium/coarse sand	
6=6	0000=0	77.405-0	000:		NA/1-1	
ST8	663359	7749558	0.0-0.4 0.4-0.55	Field	Wet brown mud	
			0.4-0.55	rieiū	Grey sandy mud Red/brown muddy sand	
			0.55		rearbiowii iliaaay sana	
ST9	663271	7749348	0.00		Wet light brown clay	
					Brown clay with slight	
			0.10		grey marbling	
			0.20		Sand brown clay	
			0.30	1	Grey clay with black organic material	
			1.90		Grey muddy coarse sand	
ST9B	663246	7749348	0.00		Light brown clay with sand As above with grey clay	
			0.3-0.5		marbling	
			0.50	1	As above	
ST9C	663256	7749306	0.0-0.5		Red/brown clay with sand	
ST9D	663289	7749337	0.0-0.15		Brown medium/coarse sand	
			0.30		Brown mud/clay	
			0.60		Grey mud	
ST9E	663288	7749399	0.30		Brown clay with grey clay marbling	
			0.60		Grey clay	
0===	00000=	77.40.40-	0000			
ST9F	663265	7749405	0.0-0.6		Brown clay	
			0.60		Grey clay	
ST9G	663203	7749403	0.0-0.1		Brown clay/mud/sand	
					As above with grey	
			0.10		marbling	
			0.90		Red clay	
					Brown sandy clay with	
ST9H	663223	7749419	0.40		some grey marbling	
			0.60	1	More grey marbling	

			Sediment	Samples		Sampling
Site	Easting	Northing	depth (m)	taken	Sediment description	details
			0.70		Red clay	
ST11	663036	7748808	0.0-1.5	2	Medium sand with clay	
			1		Grey clay with coarse	
			1.5-1.8	1	sand	
			1.90	1	Red clay	
ST12	662947	7748560	0.0-0.08	1	Wet muddy coarse sand	
3112	002941	7740300	0.0-0.08	ı	Red clay with coarse	
			0.08-0.4		sand	
					Red clay with pebbles	
			0.4-1.0	2	and coarse quartz sand	
					Red mud with coarse	
W1	661651	7749173	0.0-0.3		sand	
			0007	4	Brown clay with	
			0.3-0.7	1	fine/medium sand	
W2	661691	7740221	0.0.0.3	1 9 Field	Light brown alov	
VVZ	001091	7749331	0.0-0.3 1.00	1 & Field Field	Light brown clay Light grey clay	
			1.00	1 leiu	Dark grey clay with	
			1.20	1 & Field	organic matter	
			1.40	1 & Field	Light grey clay	
			1.50	1	Brown mud with sand	
			1.5-1.7	Field	As above	
					Brown mud with some	
W3	661695	7749425	0.0-0.7		fine sand	
			0.7-1.0		Grey clay	
			1011	1	Grey clay with brown	
			1.0-1.1	l	clay marbling Brown mud with fine	
			1.10		sand	
			0		55.10	
					Brown mud with fine	
W4	661683	7749313	0.0-0.6		sand	
			0.60	1	Grey clay	
			0.65-1.1		Grey sandy clay	
			1.10		Red clay	
	00105					
W5	661667	7749290	1.00	,	Grey clay	
			1.90	1	Compact grey clay	
			2.00	1	Compact grey clay	
NO	662040	7740040	0004		Musalahu a a :l	
N2	663010	7749043	0.0-0.1		Muddy sand	
			0.1-1.0	1	Clay with medium/coarse sand	
			2.00	1	Wet clay	
			2.0-2.6	1	Coarse muddy sand	
Notes:			ils (from field tes		Dodise maday sand	

Notes: ____ - Suspected acid sulphate soils (from field testing) Field – sample taken for field testing $(pH_F \& pH_{FOX} (Section 3.2.2))$.

Appendix C Acid sulphate soil investigation photos

Appendix C Acid sulphate soil investigation photos



Site S5 – Vegetated dry sandy soil



Site ST3 – Side of vegetated area showing limestone layer

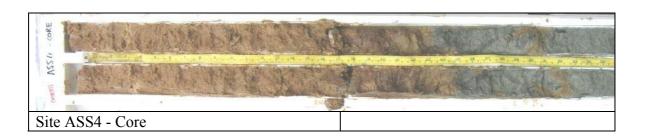


Site S12 – Creek bed



Site S15 – Creek bed





Appendix D

Interpretation of acid sulphate soil field testing

Appendix D Interpretation of acid sulphate soil field testing (Reproduced from Appendix 2 (DoE, 2004a)

It is important to note that whilst a useful exploratory tool, soil field pH tests are indicative only and cannot be used as a substitute for laboratory analysis to determine the presence of ASS. Laboratory analysis is needed to quantify the amount of existing plus potential acidity. This appendix provides information on how to interpret the results from soil field pH tests. For further information on how to conduct and interpret these tests, consult the latest version of the *Guidelines for Sampling and Analysis of Lowland Acid Sulfate Soils (ASS) in Queensland 1998*.

1. Field pH test (pH_F) i.e. pH of soil and water paste

The pH_F test measures the existing acidity of a 'soil:water' paste, and is therefore used to help identify if ASS are present. If the measured pH of the soil paste is pH_F < 4, oxidation of sulfides has probably occurred in the past, indicating the presence of AASS. Highly organic soils or heavily fertilised soils may also return a pH_F close to 4. A pH_F >4 but <5 indicates an acid soil, but the cause of the acidity will need to be further investigated by laboratory analysis. The pH_F test does not detect any unoxidised sulfides (i.e. PASS). For this reason, this test must be used in conjunction with the pH_Fox test.

2. Field pH peroxide test (pHFox) i.e. pH of soil and peroxide mix and reaction with peroxide

The pH_{FOX} test is used to indicate the presence of iron sulfides or PASS. This test involves adding 30%

hydrogen peroxide (pH adjusted to 4.5–5.5) to a sample of soil. If sulfides are present a reaction will occur.

The reaction can be influenced by the amount of sulfides present in the sample, the presence of organic matter or the presence of manganese. Once the reaction has occurred, the pH is measured.

A combination of three factors is considered in arriving at a 'positive field sulfide identification':

A reaction with hydrogen peroxide. The strength of the reaction with peroxide is a useful indicator but cannot be used alone. Organic matter, coffee rock and other soil constituents such as manganese oxides can also cause a reaction. Care should be exercised in interpreting a reaction on surface soils and high organic matter soils such as peats and coffee rock, and some mangrove/estuarine muds and marine clays.

The actual value of pH_Fox. If the pH_Fox <3, and a significant reaction occurred, then it strongly indicates a PASS. The more the pH_Fox drops below 3 the more positive the presence of inorganic sulfides.

A much lower pHFox than field pHF. The lower the final pHFox value and the greater the difference between the pHFox compared to the pHF, the more indicative the presence of PASS. This difference may not be as great if starting with an already very acid pHF (close to 4), but if the starting pH is neutral or alkaline then a larger change in pH should be expected. Where fine shell, coral or carbonate, is present the change in pH may not be as large due to buffering.

Appendix E Texture-based ASS 'Action Criteria'

Appendix E Texture-based ASS 'Action Criteria' (Reproduced from Appendix 3 (DoE, 2004a))

The *Action Criteria* are based on the sum of existing plus potential acidity, calculated as equivalent sulfur (e.g. s-TAA + SCR in %S units) or equivalent acidity (e.g. TAA + a- SCR in mol H+/tonne). The highest laboratory result(s) is always used to assess against the action criteria. For further information refer to *Guidelines for Sampling and Analysis of Lowland Acid Sulfate Soils (ASS) in Queensland 1998.*

As clay content tends to influence a soil's natural pH buffering capacity, the action criteria are grouped by three broad texture categories – coarse, medium and fine. The criteria are used to define when ASS disturbed at a site will need to be treated and managed. For projects that disturb >1000 tonnes of ASS with >0.03 %S or >18 mol H+/tonne equivalent acidity, a detailed management plan and development consent will be required.

Texture-based acid sulfate soils 'action criteria'

Type of materia	1	Action Criteria of materials is d Existing + Poten		Action Criteria of materials is d Existing + Poter	isturbed
Texture range McDonald et al. (1990)	Approx. clay content (%)	Equivalent sulfur (%S) (oven-dry basis)	Equivalent Acidity (mol H+/tonne) (oven-dry basis)	Equivalent sulfur (%S) (oven-dry basis)	Equivalent Acidity (mol H+/ tonne) (oven-dry basis)
Coarse Texture Sands to Loamy sands	≤5	0.03	18	0.03	18
Medium Texture Sandy loams to light clays	5 – 40	0.06	36	0.03	18
Fine Texture Medium to Heavy clays and silty clays	≥ 40	0.1	62	0.03	18

The action criteria refer to existing and potential acidity for given volume of ASS. The highest result(s) should always be used to assess if the relevant action criteria level has been made or exceeded; using the average or mean of a range of results is no longer considered appropriate.

Total actual acidity (TAA) is determined by titration of a 1M KCl salt solution to pH 5.5 using NaOH. This is a measurement of the soil's existing acidity prior to oxidation of sulfidic material.

Total potential acidity (TPA) is determined by peroxide double oxidation. This is estimated by titration to pH 5.5 of total acidity after oxidation of the soil with 30% hydrogen peroxide. When determining lime requirements, subtracting TAA from Total Potential Acidity (TPA) to get Total Sulfidic Acidity (TSA) is acceptable.

Appendix F Laboratory analysis reports

INTERIM REPORT

MARINE AND FRESHWATER RESEARCH LABORATORY

SEDIMENT DATA

Client: Spencer Shute, Oceanica Address: PO Box 3172, Broadway, Nedlands, WA 6009

Method Sample Code	Date	ICP002 Ag	ICP002	ICP002 Cd	ICP002 Cr	ICP002 Cu	ICP002 Ni	ICP002	ICP002 Zn	
Reporting Limit		^1 	^1 	<0.06	<0.2	<0.2	<0.4	4	<0.5	
File:		04111001	04111001	04111001	04111001	04111001	04111001	04111001	04111001	
AP8 1.0	15/10/2004	<u>^</u>	Ŋ	<0.06	28	3.2	5.4	ω	15	
AP10 1	15/10/2004	<u>^</u>	9	<0.06	24	3.7	8.2	2	8.3	
AP10 2	15/10/2004	<u>^</u>	10	<0.06	21	3.8	7.5	ω	19	
INT6	15/10/2004	<u>^</u>	12	<0.06	42	7.6	18	8	25	
INT6 (1.0)	15/10/2004	<u>^</u>	1	<0.06	19	1.9	5.0	2	3.8	
MB2	15/10/2004	<u>^</u>	10	<0.06	54	12	25	0	34	
MB3	15/10/2004	<u>^</u>	12	<0.06	66	18	33	11	47	
M10	15/10/2004	<u>^</u>	13	<0.06	64	15	31	9	45	

Date of Issue: Reference: OCA04-17

Client: Batch: Sub Batch: Date of Issue: PEB1576

11/11/2004

CERTIFICATE OF ANALYSIS



Client Reference:

			-					SAMPLE IDI	ENTIFICATIO	Ź	
		Laboratory I.D	yy I.D.	1	2	ω	4	5ī	5		
		Date Sampled	mpled	15/10/2004	15/10/2004	15/10/2004	15/10/2004	2	15/10/2004		
				AP8	AP10	INT 6 ¥	MB2	MB3	MB10		
METHOD	ANALYSIS DESCRIPTION	TINU	LOR	1.0	- o						
EA-055	Moisture Content (dried @ 103'C)	%	0.1	19.0	17.8	30.5	37.6	44.5	44.0		
EP-090-SS	ORGANOTIN COMPOUNDS								-		
EP-090-SS	Monobutyltin	ugSn/kg	0.5	0.6	1.7	0.8	1.0	1.1	<0.5		
EP-090-SS	Dibutyltin	ugSn/kg	0.5	1.7	1.9	0.6	2.7	3.0	1.7		
EP-090-SS	Tributyltin	ugSn/kg	0.2	<0.2	0.3	0.2	1.5	1.5	1.9		
EP-090S-SS	ORGANOTIN COMPOUND SURROGATE							-			
EP-090S-SS	Tripropyltin	%	_	64	63	57	57	59	40		

* INT 6 (10m) DESTROYED IN TRANSIT (PERTH ALS - QLO ALS)

BATCH QUALITY CONTROL

ALS EP-090: ORGANOTINS

QC LOT No. :

TBTS603

ANALYST: Craig Beinke

MATRIX:

Sediment

	Blank	Spike		SPIKE QC F	RESULTS		Co	ntrol Lin	nits
	Conc	Level	SCS	DCS	Average	RPD	Re	ec.	RPD
COMPOUND			Conc	Conc	Rec.				
	ug Sn/kg	ug Sn/kg	ug Sn/kg	ug Sn/kg	%	%	Low	High	%
Surrogate Recovery									
Tripropyltin	18.9	25	19.3	19.6	77	0	0	121	20
Target Compounds									
Monobutyltin	<0.5	25	16.7	17.8	69	6	28	99	50
Dibutyltin	<0.5	25	17.1	17.0	68	1	15	106	35
Tributyltin	<0.2	25	7.5	8.9	33	17	12	111	20

COMMENTS:

1) * : Recovery or RPD falls outside of the recommended control limits.

GRAEME CAMPBELL & ASSOCIATES PTY LTD Specialists in Mine-Waste Geochemistry, & Soil-Moisture-Retention Testing

P.O. Box 247, Bridgetown, Western Australia 6255

Phone: (61 8) 9761 2829 Fax: (61 8) 9761 2830

E-mail: gca@wn.com.au

0417/5

COMPANY: Fortescue Metals Group Ltd

ATTENTION: Nicky Hogarth

FROM: Graeme Campbell

SUBJECT: Assessment of Acid-Consuming Properties of Marine-

Sediment Samples

NO. PAGES (including this page): 9 DATE: 27th January 2005

Nicky,

Please find herein the findings of the testing carried out on the marine-sediment samples earlier provided by Mr Spencer Shute (Oceanica, Nedlands). It is understood that these samples are derived from recent drilling investigations for geotechnical assessment.

1.0 SAMPLES AND TESTING

The samples provided to GCA for testing were:

- BP16a 1.4 m, BP16b 3.0 m, and BP16c 4.5 m
- BP5a 1.4 m, BP5b 3.7 m, and BP5c 6.2 m

Each sample (c. 0.5 kg dry-solids basis) was in the form of moist sediments, and were oven-dried (80 0 C for c. 24 hrs), and crushed in a jaw-crusher with clearance set to 2-3 mm (nominal). Equal-weight-based composites (viz. **BP16** and **BP5**) were prepared from the respective individual samples.

The Acid-Neutralisation-Capacity (ANC) values of the samples were determined by SGS Environmental Services (Welshpool). The acidimetric auto-titrations were performed in the GCA Testing-Laboratory (Bridgetown). Copies of the laboratory reports are attached.

2.0 TESTWORK RESULTS

2.1 ANC Values

Sample BP16 had an ANC value of 83-92 kg H_2SO_4 /tonne, and was calcareous as inferred from the vigorous effervescence (viz. "fizzing") produced upon addition ('inthe-cold') of HCl in the ANC-testwork.

Sample BP5 had an ANC value of 5.3 kg H₂SO₄/tonne, and a paucity of carbonate-minerals (e.g. calcite), as inferred from the lack of "fizzing" upon reaction with HCl.

2.2 pH-Buffering Properties

The pH-buffering curves are shown on Figures 1 and 2.

The shape of the pH-buffering curve for sample BP16 (Figure 1) is consistent with that for the consumption of acid by reactive carbonate-minerals (e.g. calcite, aragonite, etc.). The consumption of c. 80-90 kg H_2SO_4 /tonne when the pH=3 end-point was reached is also consistent with the ANC value for this sample.

The shape of the pH-buffering curve for sample BP5 (Figure 2) is consistent with that for a "gutless-groundmass" deficient in carbonate-minerals.

3.0 CONCLUSIONS

The testwork results indicate that the location corresponding to sample BP16 comprise calcareous marine-sediments in which reactive carbonate-minerals are an accessory component, and for which alkalinity-forms are readily-available for circum-neutral buffering.

The location corresponding to sample BP5 is deficient in carbonate-minerals.

.....

I trust the above is satisfactory to you.

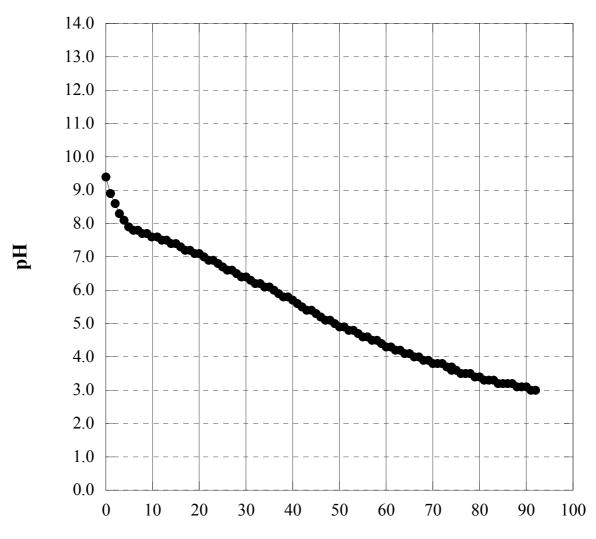
Regards,

Dr GD Campbell Director

Encl. Figures (2 pages), and laboratory reports (5 pages).

Figure 1

pH-Buffering Curve for Sample B16

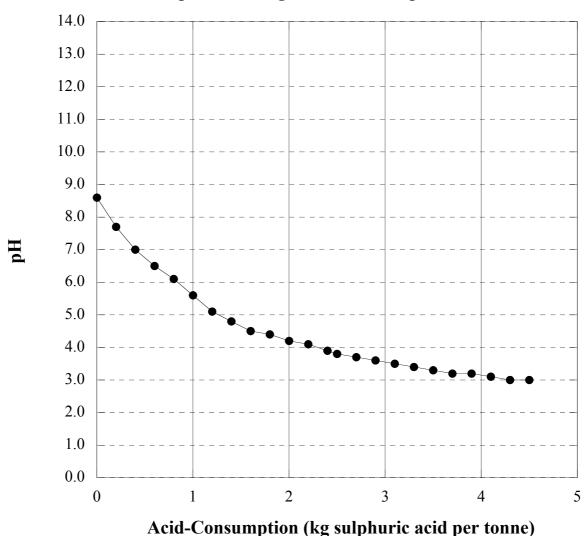


Acid-Consumption (kg sulphuric acid per tonne)

Note: The H_2SO_4 -addition rate employed in the auto-titration corresponds to a sulphide-oxidation rate (SOR) of c. 6-7 x 10^5 mg SO_4 /kg/week (= c. 3-4 x 10^4 kg H_2SO_4 /tonne/year).

This SOR is therefore up to 10^4 - 10^5 **faster** than that typical for the weathering (at circum-neutral-pH) of soils/sediments, regoliths and bedrocks that contain "minute/trace-sulphides".

Figure 2
pH-Buffering Curve for Sample BP5



Note: The H_2SO_4 -addition rate employed in the auto-titration corresponds to a sulphide-oxidation rate (SOR) of c. 1-2 x 10^5 mg SO_4 /kg/week (= c. 5-6 x 10^3 kg H_2SO_4 /tonne/year).

This SOR is therefore up to 10^3 - 10^4 faster than that typical for the weathering (at circum-neutral-pH) of soils/sediments, regoliths and bedrocks that contain "minute/trace-sulphides".



13 January, 2005

Graeme Campbell & Associates Pty Ltd

Attn: Dr G Campbell

PO Box 247

BRIDGETOWN WA 6255

Our Reference: 86294

Your Reference: GCA0417/5 NATA Accreditation: 2562(1705)

Dear Sir

On the 10th January 2005 you forwarded test work instructions for two (2) waste rock samples that were received on the 11th January 2005 at our laboratory.

Results of all test work performed follow:

Acid Neutralisation Capacity (ANC):

Sample	Fizz	Sampl	Titre	Normalit	Initial	Effervescen	ANC	ANC
Number	Ratin	e	NaOH	y	Effervescen	ce on	Solution	(kg
	g	Weigh	(mL)	HCl/NaO	ce	Warming	pН	H ₂ SO ₄ /tonn
		t (g)		H (N)				e)
BP 5	0	4.9869	19.60	0.1	Nil	Nil	1.4	5.3
BP 16	2-3	1.9570	17.80	0.5	Slight	Nil	0.7	92
RPT BP 16	2-3	2.0208	18.30	0.5	Slight	Nil	0.7	83
ANC Std 20	i	1.9935	17.20	0.1	-	-	1.4	19
ANC Std 200	-	1.4738	13.60	0.5	-	-	0.7	191

NOTES:

- 1. Acid neutralisation capacity was determined on the as received crushed sample. Unless otherwise stated, 25mL of HCl is used. Reagent blank titre of 0.5N NaOH was 25.20mL and 0.1N NaOH was 25.00mL.
- 2. ANC Std20 and Std200 are internally produced standards of CaCO₃ and quartz pulped to a nominal 75 μ m particle size which has nominal ANC of 20kg and 200kg of H₂SO₄/tonne, respectively.
- 3. This procedure is based on Sobek et al, 1978.



CLIENT: Graeme Campbell & Associates Pty Ltd OUR REFERENCE: 86294

PROJECT NO: GCA0417/5

Yours faithfully,

Steven Ednett.

8. Venning

STEVEN EDMETT Client Liaison Manager JANICE VENNING Manager, Perth

This report supersedes our preliminary results sent by facsimile on the 12th January 2005.



NATA Endorsed Test Report

This document may not be reproduced except in full.

NATA Accredited Laboratory No. 2562

Graeme Campbell & Associates Pty Ltd

Laboratory Report

pH-BUFFERING TESTWORK (B16)

Cumulative Volume of Acid	Cumulative Acid Consumption	pН	Cumulative Volume of Acid	Cumulative Acid Consumption	pН
Added (mL)	(kg H ₂ SO ₄ /tonne)		Added (mL)	(kg H ₂ SO ₄ /tonne)	•
0.00	0.0	9.4	14.00	34	6.1
0.40	1.0	8.9	14.40	35	6.1
0.80	2.0	8.6	14.80	36	6.0
1.20	2.9	8.3	15.20	37	5.9
1.60	3.9	8.1	15.60	38	5.8
2.00	4.9	7.9	16.00	39	5.8
2.40	5.9	7.8	16.40	40	5.7
2.80	6.9	7.8	16.80	41	5.6
3.20	7.8	7.7	17.20	42	5.5
3.60	8.8	7.7	17.60	43	5.4
4.00	9.8	7.6	18.00	44	5.4
4.40	11	7.6	18.40	45	5.3
4.80	12	7.5	18.80	46	5.2
5.20	13	7.5	19.20	47	5.1
5.60	14	7.4	19.60	48	5.1
6.00	15	7.4	20.00	49	5.0
6.40	16	7.3	20.40	50	4.9
6.80	17	7.2	20.80	51	4.9
7.20	18	7.2	21.20	52	4.8
7.60	19	7.1	21.60	53	4.8
8.00	20	7.1	22.00	54	4.7
8.40	21	7.0	22.40	55	4.6
8.80	22	6.9	22.80	56	4.6
9.20	23	6.9	23.20	57	4.5
9.60	24	6.8	23.60	58	4.5
10.00	25	6.7	24.00	59	4.4
10.40	25	6.7	24.40	60	4.3
10.80	26	6.6	24.80	61	4.3
11.20	27	6.6	25.20	62	4.2
11.60	28	6.5	25.60	63	4.2
12.00	29	6.4	26.00	64	4.1
12.40	30	6.4	26.40	65	4.1
12.80	31	6.3	26.80	66	4.0
13.20	32	6.2	27.20	67	4.0
13.60	33	6.2	27.60	68	3.9

Cumulative Volume of Acid Added (mL)	Cumulative Acid Consumption (kg H ₂ SO ₄ /tonne)	pН
28.00	69	3.9
28.40	70	3.8
28.80	71	3.8
29.20	72	3.8
29.60	73	3.7
30.00	74	3.7
30.40	74	3.6
30.80	75	3.6
31.20	76	3.5
31.60	77	3.5
32.00	78	3.5
32.40	79	3.4
32.80	80	3.4
33.20	81	3.3
33.60	82	3.3
34.00	83	3.3
34.40	84	3.2
34.80	85	3.2
35.20	86	3.2
35.60	87	3.2
36.00	88	3.1
36.40	89	3.1
36.80	90	3.1
37.20	91	3.0
37.60	92	3.0

Note: Titration performed using a Metrohm[®] 736 Titrino auto-titrator, and 0.05 M-H₂SO₄. Equilibration time between titrant additions was 15 minutes. 2.0 g of crushed (c. 2-3 mm), oven-dried (80 °C) sample initially dispersed in 150 mL of deionised-water.

Test mixture in contact with air, at ambient temperature, and continuously stirred. <u>Calibration of pH-Glass Electrode</u>:

Immediately prior to titration: asymmetry potential = -12 mV (pH=7.00); slope-point = 163 mV (pH=4.00); 99.0 % of Nernstian response for 25 °C.

Immediately following titration: pH=7.00 buffer read pH=7.02 and pH=4.00 buffer read pH=4.03. These discrepancies represent drift in pH-Glass electrode response during course of auto-titration.

Dr GD Campbell 10th January 2005

Graeme Campbell & Associates Pty Ltd

Laboratory Report

pH-BUFFERING TESTWORK (B5)

G 1.1	C 1.:	
Cumulative	Cumulative	
Volume of Acid	Acid Consumption	pН
Added (mL)	(kg H ₂ SO ₄ /tonne)	
0.00	0.0	8.6
0.40	0.2	7.7
0.80	0.4	7.0
1.20	0.6	6.5
1.60	0.8	6.1
2.00	1.0	5.6
2.40	1.2	5.1
2.80	1.4	4.8
3.20	1.6	4.5
3.60	1.8	4.4
4.00	2.0	4.2
4.40	2.2	4.1
4.80	2.4	3.9
5.20	2.5	3.8
5.60	2.7	3.7
6.00	2.9	3.6
6.40	3.1	3.5
6.80	3.3	3.4
7.20	3.5	3.3
7.60	3.7	3.2
8.00	3.9	3.2
8.40	4.1	3.1
8.80	4.3	3.0
9.20	4.5	3.0
N. d. Trid die	1 · M · 1 ® 72	(T')

Note: Titration performed using a Metrohm[®] 736 Titrino auto-titrator, and 0.05 M-H₂SO₄. Equilibration time between titrant additions was 15 minutes. 10.0 g of crushed (*c*. 2-3 mm), oven-dried (80 °C) sample initially dispersed in 150 mL of deionised-water.

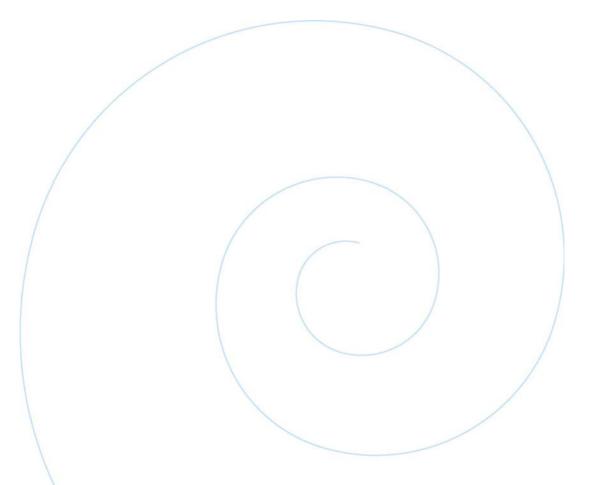
Test mixture in contact with air, at ambient temperature, and continuously stirred.

Calibration of pH-Glass Electrode:

Immediately prior to titration: asymmetry potential = -21 mV (pH=7.00); slope-point = 155 mV (pH=4.00); 99.6 % of Nernstian response for 25 °C.

Immediately following titration: pH=7.00 buffer read pH=7.02 and pH=4.00 buffer read pH=4.03. These discrepancies represent drift in pH-Glass electrode response during course of auto-titration.

Dr GD Campbell 15th January 2005



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